Organic & Biomolecular Chemistry



View Article Online

PAPER

Check for updates

Cite this: Org. Biomol. Chem., 2023, 21, 8516

Palladium-catalyzed stereoselective construction of chiral allenes bearing nonadjacent axial and two central chirality[†]

Li-Xia Liu, ம a,b Wen-Jun Huang, 🕩 a Chang-Bin Yu 🕩 *a and Yong-Gui Zhou 🗐 *a

It is challenging to enantioselectively construct molecules bearing multiple nonadjacent stereocenters, in contrast to those bearing a single stereocenter or adjacent stereocenters. Herein, we report an enantioand diastereoselective synthesis of substituted chiral allenes with nonadjacent axial and two central chiral centers through a combination of retro-oxa-Michael addition and palladium-catalyzed asymmetric allenylic alkylation. This methodology exhibits good functional-group compatibility, and the corresponding allenylic alkylated compounds, including flavonoid frameworks, are obtained with good yields and diastereoselectivities and excellent enantioselectivities (all >95% ee). Furthermore, the scalability of the current synthetic protocol was proven by performing a gram-scale reaction.

Received 17th August 2023, Accepted 12th October 2023 DOI: 10.1039/d3ob01315d

rsc.li/obc

Introduction

Bearing a peculiar three-carbon axially chiral skeleton, chiral allenes are featured in many naturally occurring products, materials,² pharmaceuticals,¹ and chiral ligands.³ Additionally, as important building blocks in organic synthesis,⁴ allenes have attracted increasing interest. Recently, some novel synthetic methods to access functionalized chiral allenes have been designed and developed, and mainly control either the axial or the central chirality.⁵ However, the efficient simultaneous control of the axial and central chirality,⁶ especially multiple central chiral centers, remains a challenge, despite their presence in natural products and pharmaceuticals. For instance, panacene is a metabolite of the sea hare Aplysia brasiliana, acting as a fish antifeedent.^{1,7} (Scheme 1) Besides this, enprostil is a prostaglandin E2 (PGE₂) analogue that strongly inhibits gastric-acid secretion and carbacyclin is a promising antithrombotic agent.^{1,8} Thus, the development of efficient methods to prepare allenes bearing multiple chiral elements is highly desirable. A few asymmetric catalytic reactions have been developed to construct chiral allene derivatives bearing multiple adjacent stereocenters.9 As for the more difficult stereoselective construction of allenes with multiple

^bUniversity of Chinese Academy of Sciences, Beijing 100049, China

nonadjacent stereocenters, to the best of our knowledge, limited methodologies have been devised. In 2018, Trost and co-workers synthesized chiral exocyclic allenic compounds bearing nonadjacent axial and two central chiral centers through a palladium-catalyzed asymmetric [3 + 2] cycloaddition reaction.¹⁰ Recently, our group realized the synthesis of enantioenriched substituted allenes with multiple nonadjacent stereocenters through the palladium-catalyzed asymmetric allenylic alkylation of thiochromanone derivatives.¹¹ Despite these advances in synthesis of chiral allenes, the preparation of diverse chiral allenes bearing multiple nonadjacent stereo-centers is still a conundrum.

The flavonoid structural motif is widely distributed in numerous naturally occurring compounds.¹² Previously, we synthesized substituted flavonoids by utilizing the strategy of retro-oxa-Michael addition, constructing two contiguous stereogenic centers in nucleophilic flavonoids.¹³ As a nontrivial extension to this work, we devoted our efforts to exploring the combination of the retro-oxa-Michael addition process and palladium-catalyzed asymmetric allenylic alkylation, which



Scheme 1 Representative examples of axially chiral allenes in natural products and pharmaceuticals.

^aState Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, China. E-mail: cbyu@dicp.ac.cn, ygzhou@dicp.ac.cn

[†]Electronic supplementary information (ESI) available: Experimental procedures, spectroscopic data of compounds and crystallographic data. CCDC 2226434 for (-)-**3bl**. For ESI and crystallographic data in CIF or other electronic format see DOI: https://doi.org/10.1039/d3ob01315d

could construct chiral allene products with multiple nonadjacent stereocenters (Scheme 2).

Results and discussion

Our initial investigations focused on condition optimization of the asymmetric allenylic alkylation by selecting a racemic 2,3-



Scheme 2 Palladium-catalyzed asymmetric allenylic alkylation.

Table 1 Optimization of reaction conditions⁴

disubstituted flavanone (1a) as the model substrate, a [Pd]/L1 complex as the catalyst and Cs₂CO₃ as the base. First, the solvent effect was examined (Table 1, entries 1-3). Fortunately, the desired product 3aa was obtained in 93% yield with 11:1 dr and 97% ee for the major diastereoisomer when tetrahydrofuran was used. After screening a series of commercially available bisphosphine ligands with large steric hindrance (entries 4-7), L1 was found to be the best choice for the allenvlic alkylation. Subsequently, allene partners with various protecting groups were tested (entries 8-11), amongst which the one with an ethoxycarbonyl group performed better in the control of diastereoselectivity (entry 8). Then, the palladium precursor was investigated and it was found that Pd(dba)₂ was still optimal (entry 8), while other catalyst precursors, such as Pd₂(dba)₃, [Pd(C₃H₅)Cl]₂ and Pd(OAc)₂, showed lower stereoselectivity (entries 12-14). Both organic and inorganic bases were also screened (entries 15 and 16). Considering the low solubility of cesium carbonate at decreased temperature, 1,8diazabicyclo-[5.4.0]undec-7-ene (DBU) was selected as the base for the next investigation. To further improve the diastereoselectivity and enantioselectivity, the reaction temperature was decreased to -20 °C, affording 3aa in 92% yield with 16:1 dr and 99% ee for the major diastereoisomer.



Entry	Solvent	L	PG	[Pd]	Base	Yield ^b (%)	dr ^b	ee ^c (%) Major/minor
1	Toluene	L1	CO ₂ Me	Pd(dba) ₂	Cs_2CO_3	>95	6:1	93/61
2	DMF	L1	CO ₂ Me	Pd(dba) ₂	Cs_2CO_3	63	10:1	98/42
3	THF	L1	CO_2Me	$Pd(dba)_2$	Cs_2CO_3	93	11:1	97/78
4	THF	L2	CO ₂ Me	Pd(dba) ₂	Cs_2CO_3	92	9:1	96/51
5	THF	L3	CO_2Me	Pd(dba) ₂	Cs_2CO_3	91	9:1	89/55
6	THF	L4	CO ₂ Me	Pd(dba) ₂	Cs_2CO_3	93	6:1	39/42
7	THF	L5	CO ₂ Me	Pd(dba) ₂	Cs_2CO_3	>95	9:1	50/10
8	THF	L1	CO_2Et	Pd(dba) ₂	Cs_2CO_3	>95	11:1	98/72
9	THF	L1	CO ₂ ⁱ Pr	Pd(dba) ₂	Cs_2CO_3	>95	10:1	97/64
10	THF	L1	Ac	Pd(dba) ₂	Cs_2CO_3	>95	9:1	94/41
11	THF	L1	$P(O)(OEt)_2$	Pd(dba) ₂	Cs_2CO_3	93	11:1	96/65
12	THF	L1	CO ₂ Et	$Pd_2(dba)_3$	Cs_2CO_3	66	7:1	94/08
13	THF	L1	CO_2Et	$[Pd(C_3H_5)Cl]_2$	Cs_2CO_3	96	1:1	51/27
14	THF	L1	CO_2Et	$Pd(OAc)_2$	Cs_2CO_3	94	11:1	97/57
15	THF	L1	CO_2Et	Pd(dba) ₂	KO ^t Bu	71	8:1	96/55
16	THF	L1	CO_2Et	$Pd(dba)_2$	DBU	87	11:1	96/52
17^d	THF	L1	CO ₂ Et	$Pd(dba)_2$	DBU	92^e	16:1	99/06

^{*a*} Reactions were carried out with **1a** (0.1 mmol), **2** (1.5 eq.), [Pd] (10 mol%), **L** (11 mol%), base (1.2 eq.), solvent (1.0 mL), 5 Å molecular sieves (MS) (50 mg), 30 °C, 24 h. ^{*b*} Yield and diastereomeric ratio were measured by analysis of ¹H NMR spectra using 1,3,5-trimethoxybenzene as the internal standard. ^{*c*} Determined by chiral HPLC. ^{*d*} -20 °C instead of 30 °C. ^{*e*} Isolated yield for the reaction with 0.2 mmol scale after 72 h.

Paper

With the optimal conditions established in entry 17, we further set out to explore the generality of the methodology. First, flavanone moieties bearing various alkyl substituents on the ester group, including ethyl (1a), methyl (1b), iso-propyl (1c), *tert*-butyl (1d), cyclohexyl (1e) and benzyl (1f), were investigated (Scheme 3). The diastereoselectivity of this reaction was slightly sensitive to the steric bulk in flavanones 1. When *tert*-butoxycarbonyl (1d) was introduced, a moderate 12:1 dr was obtained. By comparison, a 19:1 dr was observed when methoxycarbonyl (1b) was introduced. When the nitro group (1g) was used as the electron-withdrawing group (EWG), lower diastereoselectivity was observed. Thus, the methoxycarbonyl group was identified as the optimal EWG for the next substrate scope investigation.

The effect of the substituents of allenylic carbonates 2 was further probed, and it was found that they have a strong impact on the reactivity and stereoselectivity in the allenylic alkylation (Scheme 4). It seemed that allenylic carbonates containing any groups were better than those with alkyl groups in terms of reactivity and diastereoselectivity. The methyl-substituted allenvlic carbonate (2b) reacted with 1b and generated the product 3bb with 95% ee for the major diastereoisomer, albeit in moderate 47% yield and 5:1 dr. When the cyclohexyl group (2c) was introduced, the result for the reactivity and diastereoselectivity was even worse. We further tested the effect of the substituent at ortho (2d), meta (2e) or para-positions (2f) of the aryl group, and found that better diastereoselectivity (24:1 dr) was obtained with the meta-position substituent. Subsequently, when a fluoro, chloro, bromo or methoxy group was introduced at the *meta*-position, the alkylation reaction proceeded smoothly to deliver 3bg, 3bh, 3bi or 3bj with excellent ee and dr, but moderate yields were observed with the halogen atoms. Furthermore, a thienyl substituent (2k) was tolerable, affording 3bk with 19:1 dr and 98% ee for the major diastereoisomer, although a low yield was observed. The allenylic carbonate with a naphthyl substituent (21) was also suitable. Regretfully, trisubstituted allene partners were not suitable for this protocol, giving low reactivity even under



Scheme 3 Substrate scope for the ester group of flavanones 1.

Organic & Biomolecular Chemistry



Scheme 4 Substrate scope for allenylic carbonates 2.

more stringent reaction conditions. The absolute configuration of product **3bl** was assigned as $(2R,3S,R_a)$ by X-ray diffraction analysis (for the details, please see the ESI†), and other products were assigned by analogy.

Subsequently, the substrate scope was further investigated with a series of 2,3-disubstituted flavanones or chromanones by using ethyl(4-(m-tolyl)buta-2,3-dien-1-yl) carbonate (2e) as the allenylic carbonate partner (Scheme 5). As expected, a variety of aryl-substituted flavanones were smoothly converted into the corresponding products in high isolated yields, with excellent dr and ee values (3ke-3se), while moderate diastereoselectivity and lower reactivity were observed with the alkyl-substituted chromanones (1h-1j). Additionally, the position of the substituents on the phenyl group was probed, showing that a flavanone bearing an ortho-position substituent (1k) reacted with a slightly lower dr than that bearing a meta- (11) or para-position (1m) substituent. To our delight, when R² was a 4-fluoro-, 4-chloro- or 4-bromo-phenyl group (1n-1p), those substrates were all compatible with this reaction and led to the corresponding products (3ne-3pe) in excellent yields, with high dr and ee values. By contrast, when a methoxy group was introduced at the para-position, the alkylation reaction proceeded smoothly but delivered 3ge with slightly lower diastereoselectivity. Finally, substituents on the carbocyclic ring of the flavanones 1, such as methyl (1r) and chloro (1s), had negligible effect on the reactivity and enantioselectivity but a small negative influence on the diastereoselectivity.



Scheme 5 Substrate scope for the flavanones/chromanones 1.

a) Gram-scale experiment



Scheme 6 Gram-scale experiment and product transformations.

To further showcase the synthetic utility of the current strategy, a gram-scale reaction was carried out under the standard conditions, delivering the product **3be** and maintaining the yield, with slightly diminished stereoselectivity (Scheme 6a). In addition, the transformations of **3be** were concentrated on the reduction. The carbonyl group of **3be** was selectively reduced with sodium borohydride at room temperature, affording reduction product **4** in 94% yield with 13:1 dr (the relative configuration of the hydroxyl and phenyl groups in compound **4** was assigned as *cis*-**4** by analogy¹¹). Besides this, the single reduction isomer **5** was obtained in 92% yield and 87% ee by hydrogenating the allene moiety of **3be** with 10% Pd/C catalyst, showing that the diastereoisomers of **3be** are ascribed to the axial chirality.

Conclusions

In summary, we have successfully developed an enantio- and diastereoselective construction of substituted chiral allenes bearing nonadjacent axial and two central chiral centers, through a combination of retro-oxa-Michael addition and palladium-catalyzed asymmetric allenylic alkylation under basic conditions. This protocol features high yields, good diastereoselectivities, excellent enantioselectivities and broad substrate scope, achieving straightforward and efficient access to functionalized chiral allenes.

Conflicts of interest

The authors declare no competing financial interests.

Acknowledgements

Financial support from the National Natural Science Foundation of China (22171260), the K. C. Wong Education Foundation (GJTD-2020-08) of the Chinese Academy of Sciences and the Bureau of Science and Technology (XLYC2002080) of Liaoning Province is acknowledged.

References

- 1 A. Hoffmann-Röder and N. Krause, *Angew. Chem., Int. Ed.*, 2004, **43**, 1196–1216.
- 2 P. Rivera-Fuentes and F. Diederich, *Angew. Chem., Int. Ed.*, 2012, **51**, 2818–2828.
- 3 (a) X. Pu, X. Qi and J. M. Ready, *J. Am. Chem. Soc.*, 2009, 131, 10364–10365; (b) F. Cai, X. Pu, X. Qi, V. Lynch, A. Radha and J. M. Ready, *J. Am. Chem. Soc.*, 2011, 133, 18066–18069.
- 4 (a) R. Zimmer, C. U. Dinesh, E. Nandanan and F. A. Khan, Chem. Rev., 2000, 100, 3067-3126; (b) R. W. Bates and V. Satcharoen, Chem. Soc. Rev., 2002, 31, 12-21; (c) S. Ma, Acc. Chem. Res., 2003, 36, 701-712; (d) L.-L. Wei, H. Xiong and R. P. Hsung, Acc. Chem. Res., 2003, 36, 773-782; (e) L. K. Sydnes, Chem. Rev., 2003, 103, 1133-1150; (f) S. Ma, Chem. Rev., 2005, 105, 2829-2872; (g) M. Brasholz, H.-U. Reissig and R. Zimmer, Acc. Chem. Res., 2009, 42, 45-56; (h) S. Ma, Acc. Chem. Res., 2009, 42, 1679-1688; (i) S. Yu and S. Ma, Angew. Chem., Int. Ed., 2012, 51, 3074-3112.
- 5 (a) K. M. Brummond and J. E. DeForrest, Synthesis, 2007, 795–818; (b) M. Ogasawara, Tetrahedron: Asymmetry, 2009, 20, 259–271; (c) S. Yu and S. Ma, Chem. Commun., 2011, 47, 5384–5418; (d) J. Ye and S. Ma, Org. Chem. Front., 2014, 1, 1210–1224; (e) R. K. Neff and D. E. Frantz, ACS Catal., 2014,

4, 519–528; (f) W.-D. Chu, Y. Zhang and J. Wang, *Catal. Sci. Technol.*, 2017, 7, 4570–4579; (g) X. Huang and S. Ma, *Acc. Chem. Res.*, 2019, **52**, 1301–1312; (h) L. Fu, S. Greßies, P. Chen and G. Liu, *Chin. J. Chem.*, 2020, **38**, 91–100; (i) W. Xiao and J. Wu, *Org. Chem. Front.*, 2022, **9**, 5053– 5073; (j) X. Wang, X. Chen, W. Lin, P. Li and W. Li, *Adv. Synth. Catal.*, 2022, **364**, 1212–1222; (k) T. T. Nguyen, *Org. Biomol. Chem.*, 2023, **21**, 252–272.

6 For selected works on the synthesis of chiral allenes bearing axial and central chirality, see: (a) G. Wang, X. Liu, Y. Chen, J. Yang, J. Li, L. Lin and X. Feng, ACS Catal., 2016, 6, 2482-2486; (b) D. Qian, L. Wu, Z. Lin and J. Sun, Nat. Commun., 2017, 8, 567; (c) Y. Tang, J. Xu, J. Yang, L. Lin, X. Feng and X. Liu, Chem, 2018, 4, 1658-1672; (d) P. Zhang, Q. Huang, Y. Cheng, R. Li, P. Li and W. Li, Org. Lett., 2019, 21, 503-507; (e) L. Zhang, Y. Han, A. Huang, P. Zhang, P. Li and W. Li, Org. Lett., 2019, 21, 7415-7419; (f) F. Zhong, Q.-Y. Xue and L. Yin, Angew. Chem., Int. Ed., 2020, 59, 1562-1566; (g) S.-Q. Yang, Y.-F. Wang, W.-C. Zhao, G.-Q. Lin and Z.-T. He, J. Am. Chem. Soc., 2021, 143, 7285-7291; (h) J. Zhang, X. Huo, J. Xiao, L. Zhao, S. Ma and W. Zhang, J. Am. Chem. Soc., 2021, 143, 12622-12632; (i) Q. Li, Z.-L. Wang, H.-X. Lu and Y.-H. Xu, Org. Lett., 2022, 24, 2832-2836; (j) F.-H. Zhang, X. Guo, X. Zeng and Z. Wang, Angew. Chem., Int. Ed., 2022, 61, e202117114; (k) B. Shi, J.-B. Liu, Z.-T. Wang, L. Wang, Y. Lan, L.-Q. Lu and W.-J. Xiao, Angew. Chem., Int. Ed., 2022, 61, e202117215; (1) J. Dai, L. Li, R. Ye, S. Wang, Y. Wang, F. Peng and Z. Shao, Angew. Chem., Int. Ed., 2023, 62, e202300756.

- 7 R. Kinnel, A. J. Duggan, T. Eisner, J. Meinwald and I. Miura, *Tetrahedron Lett.*, 1977, **18**, 3913–3916.
- 8 P. W. Collins and S. W. Djuric, *Chem. Rev.*, 1993, 93, 1533– 1564.
- 9 (a) P. H. Poulsen, Y. Li, V. H. Lauridsen, D. K. B. Jørgensen, T. A. Palazzo, M. Meazza and K. A. Jørgensen, Angew. Chem., Int. Ed., 2018, 57, 10661–10665; (b) J. Wang, S. Zheng, S. Rajkumar, J. Xie, N. Yu, Q. Peng and X. Yang, Nat. Commun., 2020, 11, 5527; (c) C.-Y. He, Y.-X. Tan, X. Wang, R. Ding, Y.-F. Wang, F. Wang, D. Gao, P. Tian and G.-Q. Lin, Nat. Commun., 2020, 11, 4293; (d) J.-Y. Wang, S. Zhang, X.-Y. Yu, Y.-H. Wang, H.-L. Wan, S. Zhang, W. Tan and F. Shi, Tetrahedron Chem., 2022, 1, 100007; (e) Z. Yue, B. Shen, J. Cao, X. Chen, F. Fang, P. Li, P. Yu and W. Li, Org. Chem. Front., 2023, 10, 3662–3668.
- 10 B. M. Trost, D. Zell, C. Hohn, G. Mata and A. Maruniak, *Angew. Chem., Int. Ed.*, 2018, 57, 12916–12920.
- 11 L.-X. Liu, Y.-Q. Bai, X. Li, C.-B. Yu and Y.-G. Zhou, *Chem. Sci.*, 2023, **14**, 5477–5482.
- 12 (a) R. B. Keller, Flavonoids: Biosynthesis, Biological Effects and Dietary Sources, Nova Science, Hauppauge, 2009, p. 53;
 (b) K. A. Reddy, B. B. Lohray, V. Bhushan, A. S. Reddy, N. V. S. R. Mamidi, P. P. Reddy, V. Saibaba, N. J. Reddy, A. Suryapra-kash, P. Misra, R. K. Vikramadithyan and R. Rajagopalan, J. Med. Chem., 1999, 42, 3265–3278;
 (c) Z. Zhao, J. Jin, J. Ruan, C. Zhu, C. Lin, W. Fang and Y. Cai, J. Nat. Prod., 2007, 70, 1683–1686.
- 13 L.-X. Liu, W.-J. Huang, Q.-X. Xie, B. Wu, C.-B. Yu and Y.-G. Zhou, *ACS Catal.*, 2021, **11**, 12859–12863.