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Palladium-Catalyzed Construction of Phthalides Bearing Two Adjacent Stereocenters through Retro-oxa-Michael Addition

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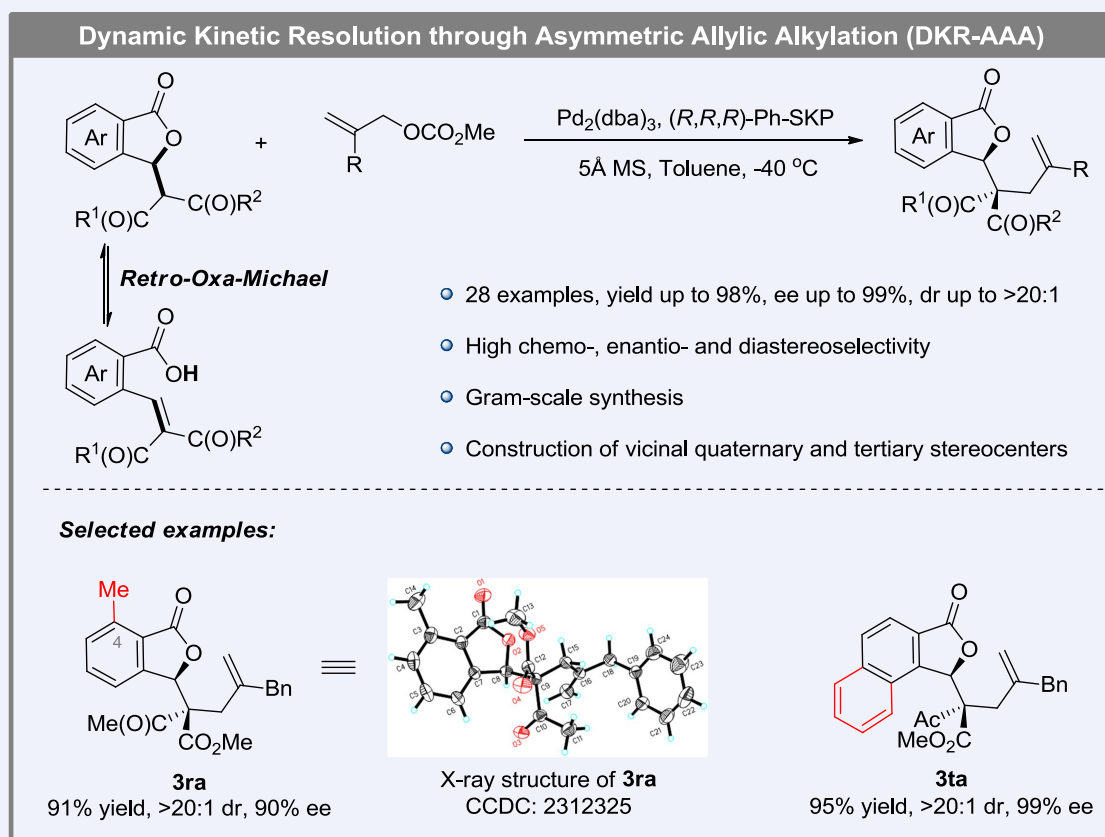
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Keywords

Retro-oxa-Michael addition | Dynamic kinetic resolution | Palladium catalysis | Asymmetric allylic alkylation | Chemoselectivity | Stereochemistry | Lactones

Comprehensive Summary



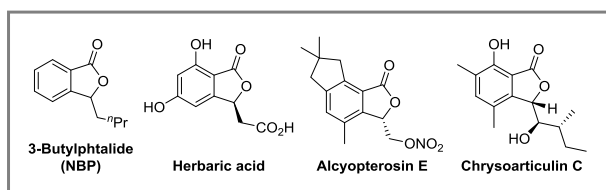
Optically active phthalides are prevalent in many natural and bioactive products. Herein, a novel dynamic kinetic resolution of isobenzofuranone derivatives through palladium-catalyzed asymmetric allylic alkylation has been developed to synthesize phthalide derivatives bearing vicinal quaternary and tertiary stereocenters with high yields, showing excellent chemo-, enantio- and diastereoselectivity. Furthermore, gram-scale experiment underwent smoothly and the transformation of product could build a bridged bicyclic skeleton.

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Background and Originality Content

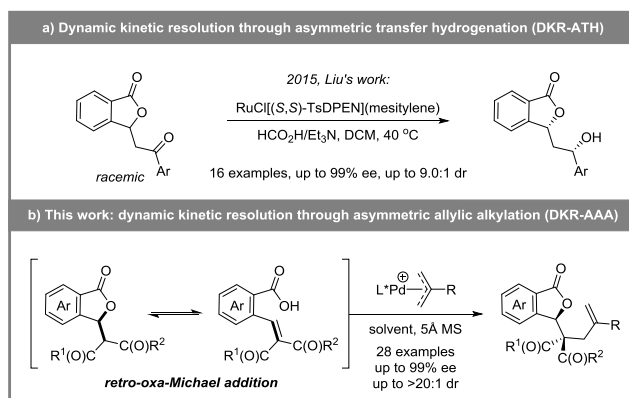
Isobenzofuranones, commonly termed as phthalides, are prevalently found in naturally occurring products and exhibit a myriad of biological activities,^[1] which also act as pragmatic building blocks in organic synthesis.^[2] For instance, 3-butylphthalide (NBP)^[3] is a cardiovascular drug for the treatment of cerebral ischemia. It is also a natural antioxidant and has anticonvulsant activity. Herbaric acid^[4] has antibacterial property and alcyopterosin E^[5] showed mild cytotoxicity toward human tumor cell lines. Chrysoarticulin C^[6] exhibited weak cytotoxicity against the K562 and A549 cancer cell lines and was also moderately active against sortase A (Scheme 1).

Scheme 1 Representative examples of natural phthalides



In recent years, various elegant methods have been developed for the asymmetric catalytic synthesis of phthalides, involving metal,^[7] organic^[8] and biological catalysis.^[9] These reported literatures mainly focused on the enantioselective construction of isobenzofuranone framework. In contrast, it is rare to achieve enantioselective synthesis of phthalide derivatives based on this framework. Indubitably, the dynamic kinetic resolution (DKR)^[10] has been demonstrated as a powerful tool. Hence, for the construction of enantiopure phthalides, dynamic kinetic resolution of compounds bearing isobenzofuranone framework is effective. In 2015, Liu and coworkers^[7f] developed DKR of phthalides through asymmetric transfer hydrogenation^[11] to construct 3-(2-hydroxy-2-arylethyl)isobenzofuran-1(3*H*)-one bearing 1,3-distereocenters with excellent enantioselectivities and acceptable diastereomeric ratios, which utilized the racemization through retro-oxa-Michael addition (Scheme 2a). Previously, our group made efforts in the dynamic kinetic resolution of substrates based on retro-oxa-Michael addition process.^[12] Among them, we have realized dynamic kinetic resolution of 2,3-disubstituted flavonoids *via* palladium-catalyzed asymmetric allylic alkylation,^[13] and two contiguous stereogenic centers were constructed on the endocyclic structure.^[12a] As is well known, it is more difficult to stereoselectively build multiple stereocenters on the acyclic structure. Herein, utilizing retro-oxa-Michael addition process, we reported an efficient dynamic kinetic resolution through asymmetric allylic alkylation to provide a wide range of phthalide derivatives bearing two

Scheme 2 Dynamic kinetic resolution of phthalide derivatives



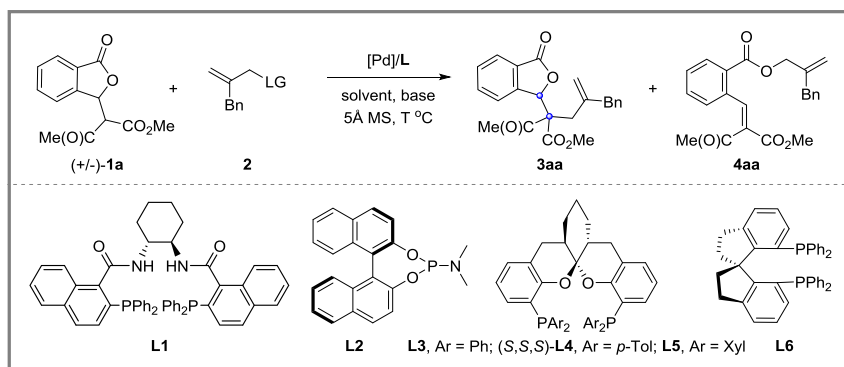
contiguous stereocenters (including an all-carbon quaternary stereocenter) with high yields, excellent enantio- and diastereoselectivities (Scheme 2b).

Results and Discussion

Encouraged by our antecedent work on retro-oxa-Michael addition,^[12a] we chose Pd₂(dba)₃/(*R,R*)-DACH-Naphthyl Trost Ligand (**L1**) as catalyst to investigate the asymmetric allylic alkylation between 2-substituted allyl carbonate (**2a**) and isobenzofuranone derivative (**1a**). Under this condition, the target product **3aa** was obtained with only 24% yield and 5.0 : 1 dr. At the same time, there was also the generation of chemoselective sideproduct **4aa** with 10% yield (Table 1, entry 1). We deem that chiral ligands may play a pivotal role in this alkylation reaction. Thus, the chiral ligand effect was initially examined (entries 2–6). Fortunately, the chemo- and diastereoselectivity of asymmetric allylic alkylation were excellent when the chiral spiroketal-based diphosphine SKPs^[14] were used, in which (*R,R,R*)-Ph-SKP **L3** was optimal. Then, the different leaving groups of allyl reagents were investigated, among which allyl carbonate was superior to allyl acetate and allyl chloride (entries 7–8). Subsequently, the different solvents were tested (entries 9–12). It was found that toluene as the solvent afforded the desired product in 93% yield with > 20 : 1 dr and 79% ee. Next, triethylamine was used instead of 1,8-diazabicyclo-[5.4.0]undec-7-ene (DBU) as the base and found that the results of alkylation reaction were similar (entries 9 and 13). Inferring the base might not be necessary, asymmetric allylic alkylation without base was attempted, which furnished the desirable product in the identical yield with 80% ee (entry 14). Therefore, we chose not to add the base. Then, the palladium precursors were investigated, showing that Pd₂(dba)₃ was still better for the enantioselectivity compared with Pd(dba)₂, [Pd(C₃H₅)Cl]₂ and Pd(OAc)₂ (entries 15–17). Naturally, the temperature was decreased to further improve the stereoselectivity (entries 18–19). Reaction temperature at –40 °C was determined to be the optimal, delivering the desired product with > 20 : 1 dr and 93% ee. Ultimately, the amount of ligand was decreased to 6.0 mol% and the optimal conditions were established in entry 20. In addition, we found that when **4aa** reacted at 50 °C under above conditions, it can be completely converted to **3aa** with 78% ee.

After identifying the optimal conditions, the substrate scope of this methodology was evaluated. First of all, a range of allyl substrates were investigated as shown in Scheme 3. It was found that substituents on the allyl carbonates have a vast effect on the stereoselectivity. With simple allyl methyl carbonate (**2b**) as the electrophile, the alkylation reaction displayed a lower enantioselectivity (42% ee). Cinnamyl methyl carbonate (**2c**) performed very poorly, providing the corresponding product **3ac** in 30% yield with only 4.0 : 1 dr and 24% ee for the major diastereoisomer. When the substituents on the 2-position of allyl methyl carbonate were tested, such as phenyl (**2d**) and methyl (**2e**), the reaction exhibited moderate enantioselectivity. 2-Benzyl substituted allyl substrates played an important role in the stereoselectivity. Accordingly, the different substituents on the aryl group of allyl substrate (**2f–j**) were explored. When electron-withdrawing substituent (F) was introduced, the reactivity and diastereoselectivity was not diminished, but enantioselectivity slightly decreased to 85%. When other substituents, such as methoxy (**2g**) and methyl on different positions (**2h–j**), were introduced, the reactivity would be affected and the reaction needed to be carried out at –30 °C.

Afterwards, the substrate scope was further investigated with a series of isobenzofuranone derivatives, using 2-benzylallyl methyl carbonate (**2a**) as the electrophile (Scheme 3). The effect of electron-withdrawing substituents of isobenzofuranone derivatives was first probed. It was found that isobenzofuranone

Table 1 Optimization of reaction conditions^a

entry	L	LG	solvent	base	[Pd]	yield (%) of 4aa ^b	yield (%) of 3aa ^b	dr of 3aa ^b	ee (%) of 3aa (major isomer) ^c
1	L1	OCO ₂ Me	PhCF ₃	DBU	Pd ₂ (dba) ₃	10	24	5.0 : 1	77
2	L2	OCO ₂ Me	PhCF ₃	DBU	Pd ₂ (dba) ₃	9	21	4.3 : 1	17
3	L3	OCO ₂ Me	PhCF ₃	DBU	Pd ₂ (dba) ₃	< 5	89	> 20 : 1	79
4	L4	OCO ₂ Me	PhCF ₃	DBU	Pd ₂ (dba) ₃	< 5	88	~20 : 1	78
5	L5	OCO ₂ Me	PhCF ₃	DBU	Pd ₂ (dba) ₃	< 5	69	7.6 : 1	14
6	L6	OCO ₂ Me	PhCF ₃	DBU	Pd ₂ (dba) ₃	< 5	36	5.0 : 1	16
7	L6	OAc	PhCF ₃	DBU	Pd ₂ (dba) ₃	< 5	47	~20 : 1	78
8	L6	Cl	PhCF ₃	DBU	Pd ₂ (dba) ₃	< 5	73	~20 : 1	78
9	L3	OCO ₂ Me	toluene	DBU	Pd ₂ (dba) ₃	< 5	93	> 20 : 1	79
10	L3	OCO ₂ Me	THF	DBU	Pd ₂ (dba) ₃	< 5	> 95	~20 : 1	76
11	L3	OCO ₂ Me	DCM	DBU	Pd ₂ (dba) ₃	< 5	79	18.8 : 1	72
12	L3	OCO ₂ Me	CH ₃ CN	DBU	Pd ₂ (dba) ₃	< 5	25	7.3 : 1	52
13	L3	OCO ₂ Me	toluene	Et ₃ N	Pd ₂ (dba) ₃	< 5	> 95	> 20 : 1	78
14	L3	OCO ₂ Me	toluene	—	Pd ₂ (dba) ₃	< 5	> 95	> 20 : 1	80
15	L3	OCO ₂ Me	toluene	—	Pd(dba) ₂	< 5	> 95	> 20 : 1	79
16	L3	OCO ₂ Me	toluene	—	[Pd(C ₃ H ₅ Cl) ₂]	< 5	> 95	> 20 : 1	79
17	L3	OCO ₂ Me	toluene	—	Pd(OAc) ₂	< 5	> 95	> 20 : 1	79
18 ^d	L3	OCO ₂ Me	toluene	—	Pd ₂ (dba) ₃	< 5	> 95	> 20 : 1	89
19 ^e	L3	OCO ₂ Me	toluene	—	Pd ₂ (dba) ₃	< 5	> 95	> 20 : 1	93
20 ^{e,f}	L3	OCO ₂ Me	toluene	—	Pd ₂ (dba) ₃	< 5	> 95 (96 ^g)	> 20 : 1	93

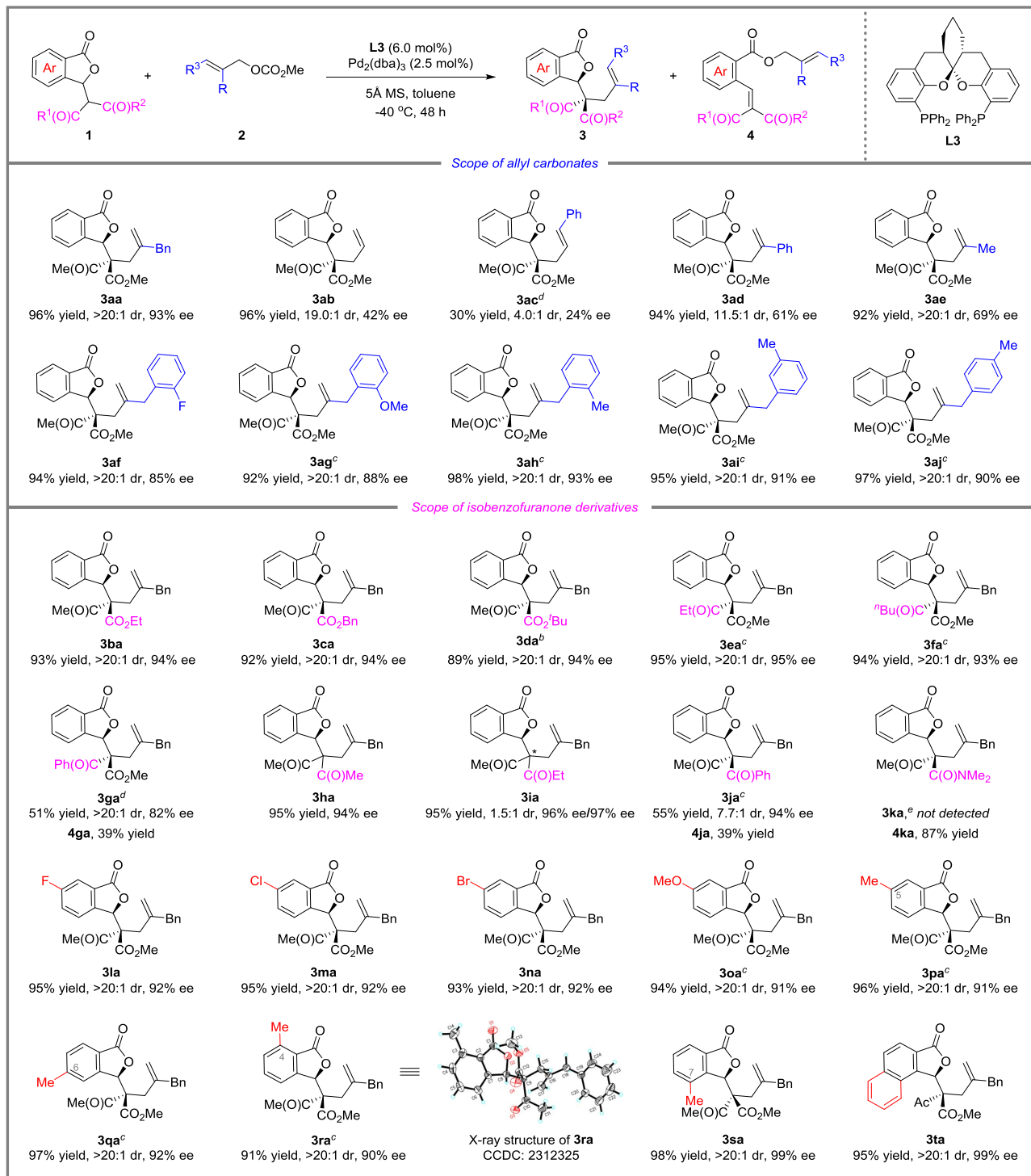
^a Reactions were carried out with **1a** (0.10 mmol), **2** (1.5 eq.), [Pd] (5.0 mol%), L (7.5 mol%), base (1.0 eq.), solvent (1.0 mL), 5 Å MS (50 mg), 30 °C, 1–48 h. ^b Yield and diastereomeric ratio were measured by analysis of ¹H NMR spectra using 1,3,5-trimethoxybenzene as the internal standard. ^c Determined by chiral HPLC. ^d –20 °C instead of 30 °C. ^e –40 °C instead of 30 °C. ^f L3 (6.0 mol%). ^g Isolated yield for the reaction on 0.20 mmol scale and toluene (1.0 mL).

compounds containing β-keto ester structure were better. Various alkoxy carbonyl groups were well tolerated in the reaction, such as ethoxycarbonyl (**1b**), benzyloxycarbonyl (**1c**) and *tert*-butoxycarbonyl (**1d**). When R¹ was changed from methyl to ethyl (**1e**) or *n*-butyl (**1f**), the enantioselectivity was slightly improved, albeit a little lower reactivity resulting in the reaction conducted at –30 °C. When R¹ was phenyl (**1g**), the chemoselectivity of this reaction was profoundly affected, delivering the sideproduct **4ga** in 39% yield and the desired product **3ga** in 51% yield with 82% ee. For isobenzofuranone compounds containing 1,3-dione structure, there was no change in enantioselectivity. However, for nucleophile (**1i**), asymmetric allylic alkylation could only obtain 1.5 : 1 diastereomeric ratio. Additionally, chemo- and diastereoselectivity of the alkylation reduced when R² was phenyl (**1j**). When substrate **1k** bearing dimethylamino as R² was employed, it was a pity that chemoselective sideproduct **4ka** was generated in 87% yield and the desired **3ka** was not detected. Due to the fact that **4aa** could be converted to **3aa** at 50 °C under standard conditions, we attempted to obtain the desired product **3ka** at higher reaction temperature. However, no **3ka** was also obtained at 50 °C and even at 80 °C.

Furthermore, it was executed to screen substituents on aro-

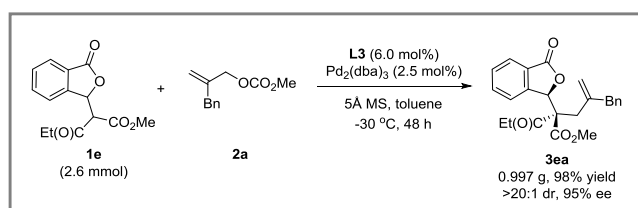
matic ring of the isobenzofuranone derivatives **1**. Halogen atoms such as F, Cl and Br on the C5 position of isobenzofuranone derivatives (**1l–n**) were compatible, and the reactivity and stereoselectivity of asymmetric allylic alkylation could be maintained. When electron-donating groups like methoxy and methyl (**1o,p**) were introduced, the alkylation reaction proceeded at –30 °C, which was the same as the substrates bearing methyl group on the C4 or C6 position (**1q,r**). Gladly, methyl group on the C7 position of isobenzofuranone derivative was advantageous, and the corresponding product **3sa** was obtained with 99% ee. In addition, using substrate bearing naphthalene for asymmetric allylic alkylation, the product **3ta** could also be generated in 95% yield with 99% ee. Besides, to determine the absolute configuration of products, the X-ray crystallographic analysis evidenced that (+)-**3ra** was confirmed as (+)-(R,R)-**3ra** and the absolute configurations of all other products were assigned by analogy.

To demonstrate the practicability of the above methodology, gram-scale reaction on **1e** was executed under the corresponding condition, providing the product **3ea** in 98% yield without loss of stereoselectivity (Scheme 4). Meanwhile, further derivatizations of compound **3ea** were performed (Scheme 5). A tandem epoxidation and acid-promoted intramolecular ring-opening reaction of

Scheme 3 Substrate scopes^a

^a Reactions were carried out with **1** (0.20 mmol), **2** (0.30 mmol), Pd₂(dba)₃ (2.5 mol%), **L3** (6.0 mol%), toluene (1.0 mL), 5 Å MS (50 mg), -40 °C, 48 h. ^b -40 °C, 55 h. ^c -30 °C, 48 h. ^d -30 °C, 72 h. ^e -40 °C, 55 h then -20 °C, 40 h.

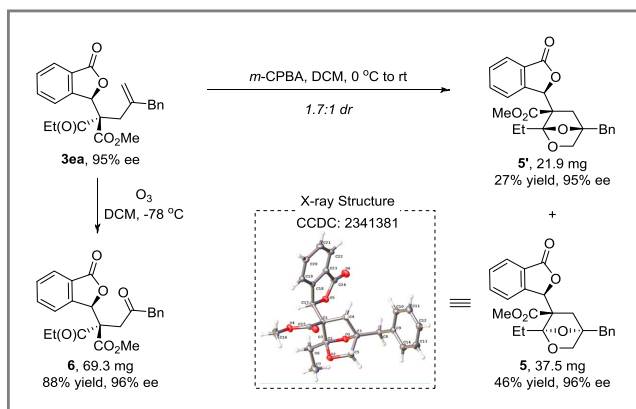
Scheme 4 Gram-scale experiment



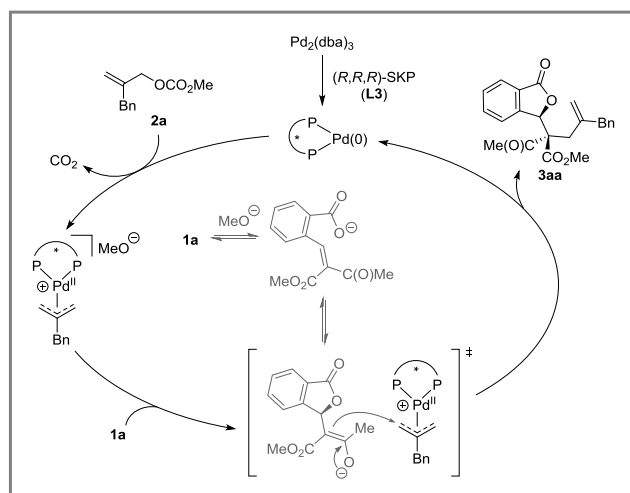
3ea happened in the presence of *m*-chloroperbenzoic acid, delivering the bridged skeleton **5/5'** with 1.7 : 1 dr and maintained optical purity. Finally, oxidative cleavage of the terminal alkene with ozone produced the chiral ketone **6** in 88% yield with 96% ee.

On the basis of the above experimental results and mechanism on palladium-catalyzed allylic alkylation using allyl carbonates,^[15] a plausible mechanism was proposed in Scheme 6. For the stereoselective control, the enolate of the nucleophile approaches the (π-allyl)palladium-L3 complex by its *Si* face.

Scheme 5 Product derivatization



Scheme 6 Plausible mechanism



Conclusions

In summary, we have successfully developed dynamic kinetic resolution of isobenzofuranone derivatives through palladium-catalyzed asymmetric allylic alkylation, based on fast retro-oxa-Michael addition as the racemization step. It is highly efficient to construct a series of enantioenriched phthalide derivatives bearing vicinal tertiary and all-carbon quaternary stereocenters. This strategy has high chemo-, enantio- and diastereoselectivity, showing good functional group tolerance and scale-up reactivity. In addition, the bridged bicyclic skeleton could be built by the transformation of product.

Experimental

General procedure for the palladium-catalyzed asymmetric allylic alkylation

The metal precursor $\text{Pd}_2(\text{dba})_3$ (0.005 mmol, 4.6 mg, 2.5 mol%), (R,R,R) -Ph-SKP ligand (**L3**) (0.012 mmol, 7.9 mg, 6.0 mol%) and toluene (0.5 mL) were placed in a dried Schlenk tube under nitrogen atmosphere. The mixture was stirred at 30 °C for 30 min. Then the mixture was cooled to -40 or -30 °C. Thereafter, the isobenzofuranone derivatives **1** (0.20 mmol) and 5 Å MS (50 mg) were added. Sequentially, allyl carbonates **2** (0.30 mmol) and toluene (0.5 mL) were added slowly. The mixture was stirred at -40 or -30 °C for 48–72 h. After the completion of the reaction, without any treatment, the reaction mixture was directly purified by column chromatography on silica gel using hexanes/ethyl acetate (50/1–5/1) as eluent to give the desirable chiral products **3**.

Supporting Information

The supporting information for this article is available on the WWW under <https://doi.org/10.1002/cjoc.202400612>.

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