Stereoselective Construction of Trisubstituted 1,3-Enynes *via* Aryl to Vinyl 1,4-Palladium Migration

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Abstract: Palladium-catalyzed stereoselective olefinic C–H alkynylation of *gem*-diarylsubstituted ethylenes with propargylic alcohols was achieved to access diverse unsymmetrical 1,3-enynes. The regio- and stereoselectivities were established through a 1,4-palladium migration from aryl to vinyl in the presence of 2- FC_6H_4OH as additive. Mechanistic investigations suggest that cleavage of the olefinic C–H bond might not be involved in the rate-determining step of the catalytic process.

Keywords: C-H activation; Propargylic alcohols; 1,3-Enynes; 1,4-Palladium migration

Introduction

Transition metal-catalyzed C-H activation has emerged as an increasingly powerful tool for C-C bond formation in organic synthesis.^[1] Efficient 1,nmigration of a transition metal has been used as a prospective C-H activation approach for converting C-H bonds in intricate substrates into beneficial functional moieties.^[2] In this regard, 1,4-palladium migration has provided a unique reaction mode adopted to selectively achieve remote C-H bond modification.^[3] Various processes involving 1,4-palladium migration such as aryl to aryl,^[4] vinyl to aryl,^[5] alkyl to acyl,^[6] and aryl to alkyl,^[7] have been explored. Following the work by Lin and Feng, et al. on palladium-catalyzed vinylic C-H borylation of orthovinyl bromobenzenes,^[8a] efforts have been focused on the functionalization of alkenes through aryl to vinyl 1,4-palladium migration,^[8b-10] especially for the construction of C-C bonds. Olefinic C-H alkenylation^[8b,c] (Scheme 1a) and arylation^[8d] of gem-disubstituted ethylenes were established through a 1,4-palladium migration/Heck or Suzuki-Mivaura coupling sequence. respectively. Strategies for intercepting the vinylpalladium species generated via 1,4-palladium migration to enable vinylic C-H carbonylation have been separately

(a) vinylic C-H alkenylation^[8b,8c]



Scheme 1. C–C Bond formation *via* aryl to vinyl 1,4-palladium migration.

described (Scheme 1b).^[11] Recently, we documented highly regioselective C–H alkylation of *o*-vinyl bromobenzenes with cyclobutanols through a 1,4-palladium migration/C–C cleavage protocol



(Scheme 1c),^[12a] C–H allenylation/alkenylation of styrenes with N-tosylhydrazones via such a palladium migration/carbene migratory insertion/ β -hydride elimination sequence,^[12b] and C-H polyfluoroarylation of gem-diaryl alkenes employing polyfluoroarenes.^[12c]

Propargylic alcohols are widely used as reactive handles in organic synthesis.^[13] For example, alkynefunctionalized dihydrobenzofurans and indolines were accessed with propargylic alcohols as coupling components through a vinyl to acyl 1,5-palladium migration.^[14a] A Pd/Rh cooperatively catalyzed cascade reaction was developed to merge an alkynylidene moiety with benzosilacycle.^[14b] Regarding C-C alkenylation of propargylic alcohols, the method for the regio- and stereoselective synthesis of 1,3-envnes has not yet been established. In line with the known work,^[8-11] and considering our ongoing research on olefinic C–H functionalization,^[12,15] it was reasonably envisioned that 1,4-palladium migration of o-vinyl bromobenzenes may be potent for the alkenvlation of

propargylic alcohols to execute highly regio- and stereoselective formation of 1,3-envnes, which are important building blocks in organic synthesis,[16,17] and also occur widely in natural products.^[18] Herein, we report a concise protocol to access synthetically useful unsymmetrical 1,3-envnes via aryl to vinyl 1,4palladium migration under mild conditions (Scheme 1d). During our submission process, Feng's group^[19] disclosed such a metal migration using the terminal alkyne ethynyltriisopropylsilane (H-C=C-TIPS) via a 1,4-palladium migration/Sonogashira sequence to access TIPS-substituted envnes.

Results and Discussion

Initially, 1-bromo-2-(1the reaction of phenylvinyl)benzene (1a) with 2-methyl-4-phenylbut-3-yn-2-ol (2a) was performed to optimize the reaction conditions (Table 1). Referring to the reaction of olefinic C-H alkylation previously reported by our

Table 1. Optimization of the reaction conditions.^[a]

| | | Ph Br + | $Ph \qquad \qquad$ | | | | |
|-------------------|------------------|--|---|-------------------------------|---------------------------------|-----------------------|--|
| | | 1a | 2a | 3a | 3a' | | |
| Entry | Ligand | 2-FC ₆ H ₄ OH (mol%) | | Solvent | Yield of 3 $a^{[b]}$ (%) | $rr^{[c]}(3 a: 3 a')$ | |
| 1 | PPh ₃ | 10 | | toluene | 14 | 1:1.8 | |
| 2 | Sphos | 10 | | toluene | 72 | 8.4:1 | |
| 3 | ĹĨ | 10 | | toluene | 19 | 1.3:1 | |
| 4 | L2 | 10 | | toluene | 49 | 3.8:1 | |
| 5 | L3 | 10 | | toluene | 39 | 1.8:1 | |
| 6 | L4 | 10 | | toluene | 47 | 5.2:1 | |
| 7 | L5 | 10 | | toluene | 68 | 6.8:1 | |
| 8 | L6 | 10 | | toluene | 36 | 16:1 | |
| 9 | Sphos | 10 | | THF | 82 | >20:1 | |
| 10 | Sphos | 10 | | 2-MeTHF | 73 | >20:1 | |
| 11 | Sphos | 20 | | THF | 87 | >20:1 | |
| 12 | Sphos | | | THF | 5 | 1:9 | |
| 13 ^[d] | Sphos | 20 | | THF | 87 (85) ^[e] | >20:1 | |
| | | MeO OMe | PCy ₂ | Me PCy ₂ | Me ₂ N | | |
| | | Sphos | L1, CyJohnphos | L2, Mephos | L3, Davephos | | |
| | | MeO L4, Ph-Sphos | ⁱ PrO L 5 , Ruphos | PCy2 Pr Pr L6, Xphos | | | |

^[a] Conditions: **1a** (0.2 mmol), **2a** (0.3 mmol), Pd(OAc)₂ (5 mol%), ligand (10 mol%), Cs₂CO₃ (2 equiv.), 2-FC₆H₄OH, solvent (2 mL), argon, 80 °C, 0.5 h.

^[b] Determined by ¹H NMR analysis using 1,3,5-trimethoxylbenzene as the internal standard.

^[c] The rr represents the ratio of the target product **3a** to its regioisomer **3a**', determined by ¹H NMR analysis.

^[d] **1 a** (0.3 mmol), **2 a** (0.45 mmol), THF (3 mL).

^[e] Isolated yield given in parentheses.

lab,^[12a] with PPh₃ as the ligand and Cs₂CO₃ as the base the reaction only gave the target migratory product 3 a in 14% yield along with major generation of the direct coupling product 3a' (3a/3a' = 1:1.8) (Table 1, entry 1). Using Sphos as the ligand obviously enhanced both the yield and regioselectivity (72%) (8.4:1 rr) of the target product (Table 1, entry 2). Screening of phosphine ligands has shown that the desired 1,4palladium migration process could be accelerated by the biphenyl-type phosphines, especially those (Sphos and L5) bearing both alkoxyl and dicyclohexyl groups (Table 1, entries 2 and 7). With THF (82%) or 2-MeTHF (73%) as the reaction medium, 3 a was formed in a more efficient manner with high regioselectivity (>20:1 rr), demonstrating that THF solvent was crucial to avoid formation of the side-product 3a'(Table 1, entries 9 and 10). Addition of 2-FC₆H₄OH significantly expedited the intended aryl-to-vinyl 1,4palladium migration process and suppressed generation of the regioisomer 3a', presumably due to a thermodynamically favourable process assisted by $2\text{-FC}_6\text{H}_4\text{OH}$ according to our previous DFT calculations.^[12a] Increasing the loading of 2-FC₆H₄OH additive from 10 mol% to 20 mol% further optimized the reaction efficiency (Table 1, entries 2 and 11). However, in the absence of 2-FC₆H₄OH compound 3a (5%) could hardly be formed, and direct cross-coupling of **1** a with 2a predominantly occurred to give the direct coupling product **3**a' in 45% yield (Table 1, entry 12). The choice of bases exerted a decisive influence on this transformation, and Cs₂CO₃ was identified as the most suitable one (see the details in SI). Eventually, the reaction was conducted on a 0.3 mmol scale of 1a, yielding the migratory product 3a in 85% isolated yield with a > 20:1 rr (Table 1, entry 13).

Next, the scope of 1-bromo-2-vinylbenzenes (1) was explored under the optimal conditions (Table 2). In a manner akin to the preparation of **3a** (Table 1, entry 13), 1-bromo-2-(1-(substituted)phenylvinyl)-benzenes (1b-11) coupled with 2-methyl-4-phenylbut-3yn-2-ol (2a) gave the target unsymmetrical 1,1,4triphenyl-substituted 1,3-envnes 3b-31 in 60-86% yields with tolerance of electron-donating and withdrawing substituents such as methyl, tert-butyl, methoxy, phenyl, fluoro, chloro, trifluoromethyl, and trifluoromethoxy at the para-, meta-, or ortho-position of one phenyl of the 1,1-diphenyl moiety in 1. In the cases of 4-^tBu group an obvious negative substituent effect deteriorated the formation of **3 c** (60%), 4-chloro and 4-phenyl showed a moderate substituent effect to diminish the yields of 3e (69%) and 3g (67%), respectively, while both methyl and fluoro facilitated the reaction to afford the target products (3b, 3f, 3j, **3k**, and **3l**) in 78–86% yields. Notably 2-methyl in 1bromo-2-vinylbenzene 11 demonstrated no obvious steric effect on the formation of **3 l** (78%). Gram-scale synthesis of **3 f** was conducted to reach a 74% isolated

 Table 2. Scope of 1-bromo-2-vinylbenzenes.^[a]



^[a] Conditions: 1 (0.3 mmol), 2 a (0.45 mmol), Pd(OAc)₂ (5 mol%), Sphos (10 mol%), 2-FC₆H₄OH (20 mol%), Cs₂CO₃ (2 equiv.), THF (3 mL), 80 °C, argon, 0.5 h.

^[b] 1f (5 mmol), 2a (7.5 mmol), Pd(OAc)₂ (5 mol%), Sphos (10 mol%), 2-FC₆H₄OH (20 mol%), Cs₂CO₃ (2 equiv.), THF (50 mL), 80 °C, argon, 0.5 h.

^[c] Pd(OAc)₂ (10 mol%), Sphos (20 mol%).

yield. 1-(2-Naphthyl) and 1-(2-thienyl)-functionalized 1-bromo-2-vinylbenzenes 1 m and 1 n underwent the reaction with 2 a to produce the corresponding target products 3 m (70%) and 3 n (53%), respectively, but no desired product was detected when one of the phenyl group in the 1,1-diphenyl moiety of 1 was replaced by CO₂Me (30) or methyl (3p). Notably, neither 2bromostyrene nor 1-bromo-2-(2,2-dimethyl-1-meth-



ylenepropyl)benzene could react with 2 a to afford the desired products. These outcomes have suggested that the reactivity of *gem*-diaryl alkenes is particularly sensitive to the electronic surrounding of their vinyl functionality. When the 1-bromophenyl was substituted by electron-donating 4-methyl or methoxy, the target products 3q (73%) and 3r (85%) were efficiently formed, whereas electron-withdrawing 4-fluoro, chloro and trifluoromethyl groups exhibited a remarkable negative electronic impact on the yields of 3 s-3 u (17-30%) which could be enhanced to 50-76% by increasing the catalyst loading to 10 mol%. 3-Methoxy in the 1-bromophenyl moiety facilitated the reaction to form 3v in 84% yield, and 5-fluoro exhibited an obvious steric/electronic effect on the formation of 3 w. It is noteworthy that all the products were regio- and stereoselectively obtained as the (E) or (Z)-isomer with > 20:1 rr, and the molecular structures of compounds 3 were further confirmed by the X-ray single crystal structural determination of compounds 3f and 3s, respectively, which was also confirmed by the ¹H NMR analysis of the reaction mixture of 1 f with 2 a in which only the migratory product 3f, instead of 3swas detected (see the SI).

Under the optimal conditions, other alkyne sources, phenylacetylene. 1-phenyl-2-trimethincluding ylsilylacetylene, 2,4-diphenylbut-3-yn-2-ol (2a1),1,1,3-triphenylprop-2-yn-1-ol (2a2),1-(phenylethynyl)cyclobutan-1-ol (2a3),1-(phenylethynyl)cyclopentan-1-ol 1-(2a4),(phenylethynyl)cyclohexan-1-ol (2a5) and 3-isopropyl-4-methyl-1-phenylpent-1-yn-3-ol (2a6), were tested. It was found that phenylacetylene and 1-phenyl-2trimethylsilylacetylene as the coupling reagents led to the migratory product 3a in 30% (1:2.2 rr) and 9% (1:3.7 rr) yields, respectively, along with more efficient formation of regioisomer 3a'. With propargylic alcohols 2a1-2a6 as the masked alkyne source, the migratory reaction efficiently proceeded with a high regioselectivity ($\geq 15:1$ rr). In the cases of using 2a1, 2a2, and 2a3, the target product 3a was formed in > 81% yields with a > 20:1 rr (see Table S2 in SI).

Then, the protocol generality was investigated by using various propargylic alcohols of type 2 as the masked alkyne source (Table 3). Substituted aryl propargylic alcohols 2b-2n proved to be compatible in their reaction with **1**a, giving the target 1,3-envnes (4 a-4 m) in 62–92% yields with >20:1 rr. Electrondonating and withdrawing substituents such as methyl, phenyl, *tert*-butyl. methoxy, fluoro. chloro. trifluoromethyl, and trifluoromethoxy were tolerated. Only in the cases of 4-Cl group, product 4f was obtained in 62% yield, exhibiting an obvious negative substituent effect. In other cases, the target products were accessed in 71-92% yields. In all the cases, the electronic effect seems to be much greater than the effect. For example, 2-methyl-4-(2-methsteric

 Table 3. Scope of propargylic alcohols.^[a]



^[a] Conditions: 1a (0.3 mmol), 2 (0.45 mmol), Pd(OAc)₂ (5 mol%), Sphos (10 mol%), 2-FC₆H₄OH (20 mol%), Cs₂CO₃ (2 equiv.), THF (3 mL), 80 °C, argon, 0.5 h.
 ^[b] Pd(OAc)₂ (10 mol%), Sphos (20 mol%).

ylphenyl)but-3-yn-2-ol (21) led to 4k in 92% yield, while 2-methyl-4-(2-chlorophenyl)but-3-yn-2-ol (2m) resulted in 41 in 77% yield under the same conditions. Bulky 2-methyl-4-(2-naphthyl)but-3-yn-2-ol (2n) also underwent the reaction with 1a, efficiently giving 4m (80%). Heteroaryl-functionalized propargylic alcohols, that is, 2-methyl-4-(2-thienyl)but-3-yn-2-ol (20) and 2methyl-4-(3-pyridyl)but-3-yn-2-ol (2p), also reacted well with 1a to offer 4n (88%) and 4o (73%), respectively. However, 4-alkenyl or alkyl-based propargylic alcohols 2q and 2r only demonstrated a poor to moderate reactivity to 1a, forming 4p (79%) and 4q (44%) by using 10 mol% catalyst. It should be noted that electron-deficient 4-methoxycarbonyl propargylic alcohol (2s) showed no reactivity to 1a. These results have revealed that the reactivity of propargylic alcohols is susceptible to the electronic surrounding around the alkynyl moiety.

1,1,2-Trisubstituted olefin such as (1-phenyl-1propen-1-yl)benzene (5) was also employed in the reaction. Alkene 5 (E/Z=2.1:1) reacted with 2a under the standard conditions, only producing the direct coupling product 6 (E/Z=3.3:1) in 92% yield, and no migratory product was detected (Scheme 2). This result suggests that compound 5 and product 6 may undergo E/Z isomerization under the standard conditions.

To probe into the reaction mechanism, control experiments were performed (Scheme 3). When alkene 1 r reacted with 2 a in the presence of 4 equiv. of D_2O under the standard conditions, 34% deuterium incorporation was observed at the *ortho*-position (Scheme 3a). Deuterated alkene 1r[D] reacted with 2a to result in



Scheme 2. Reaction with 1,1,2-trisubstituted olefin 5.



Scheme 3. Control experiments.



Scheme 4. Proposed mechanism.

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3r[D] without deuterium incorporation onto the orthoposition (Scheme 3b). These results suggest that the hydrogen atom at the *ortho*-position of the phenyl ring in 3r[D] partially came from the reaction medium, rather than from alkene 1r. In these cases, olefinic hydrogens in the 1,3-envne products were partially deuterated. The kinetic isotope effect (KIE) was measured, and a secondary KIE of 1.01 from the parallel reactions of alkene 1r and its deuterated form 1r[D] with 2a (Scheme 3c) suggests that the olefinic C-H bond cleavage of 1 might not contribute to the rate-determining step throughout the catalytic process^[20] (see the SI for details).

A plausible reaction mechanism is presented in Scheme 4. Initially, oxidative addition of 1a to the in situ generated Pd(0) species forms Pd(II) intermediate A, which undergoes cyclopalladation in the presence of Cs₂CO₃ base, producing palladacycle **B** via a CMD process. Protonation of species B with 2- FC_6H_4OH additive leads to Pd(II) intermediate C via a net aryl to vinyl 1,4-palladium migration, owing to a thermodynamically favourable process assisted by 2-FC₆H₄OH according to our previous DFT calculations.^[12a] Subsequently, intermediate C reacts with the alkyne anion generated in situ from 2a forming Pd(II) intermediate **D** through a *retro*-alkynylation process *via* a β -carbon elimination in the presence of Cs₂CO₃.^[13b] Eventually, species **D** undergoes reductive elimination to afford the target product 1,3-envne **3a**, and while regenerating the catalytically active Pd(0) species, closing the catalytic cycle. Without the assistance of 2-FC₆H₄OH, intermediate A directly reacts with 2a to generate Pd(II) intermediate C', followed by reductive elimination to yield the regioisomer 3 a'.

Conclusions

In summary, an 1,4-palladium migration cascade involving bromo-functionalized gem-diaryl ethylenes and propargylic alcohols has been established, offering a straightforward and rapid method to access diverse synthetically valuable 1,3-envnes. The utilization of 2-FC₆H₄OH additive improves the reaction efficiency by expediting the crucial 1,4-palladium migration process. Mechanistic studies have suggested that the olefinic C–H bond cleavage might not be involved in the ratedetermining step.

Experimental Section

Typical Procedure for the Synthesis of 3 a

Under an argon atmosphere a mixture of alkene 1a (78 mg, 0.3 mmol), 2-methyl-4-phenyl-3-butyn-2-ol 2a (72 mg. 0.45 mmol), Pd(OAc)₂ (3.3 mg, 0.015 mmol), Sphos (12.3 mg, 0.03 mmol), 2-FC₆H₄OH (6.7 mg, 0.06 mmol), and Cs_2CO_3

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(195 mg, 0.6 mmol) in 3 mL of THF was stirred at 80 °C for 30 min. After cooled to ambient temperature, the reaction mixture was diluted with CH_2Cl_2 (10 mL), filtered through a short pad of celite, and rinsed with 5 mL of CH_2Cl_2 . The combined filtrate was concentrated under reduced pressure. The resultant residue was subjected to purification by column chromatography on silica gel (eluent: petroleum ether (60–90 °C) to afford **3a** as a white solid (71 mg, 85%).

CCDC-2327561 (**3 f**), and 2327562 (**3 s**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* www.ccdc.cam.ac.uk/structures.

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