Literature Report II

Catalytic Regio- and Enantioselective Boracarboxylation of Arylalkenes with CO₂ and Diboron

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Date: 2024-03-18

CV of Prof. Tang Yong



Background:

- □ 1982-1986 B.S., Sichuan Normal University
- **□ 1990-1993** M.S., SIOC (Prof. Huang Yao-Tseng)
- **□ 1993-1996** Ph.D., SIOC (Prof. Dai Li-Xin)
- □ 1996-1996 Associate Professor, SIOC
- **□ 1996-1997** Postdoc., Colorado State University
- **□ 1997-1999** Postdoc., Georgetown University
- □ 1999-now Professor, SIOC

Research:

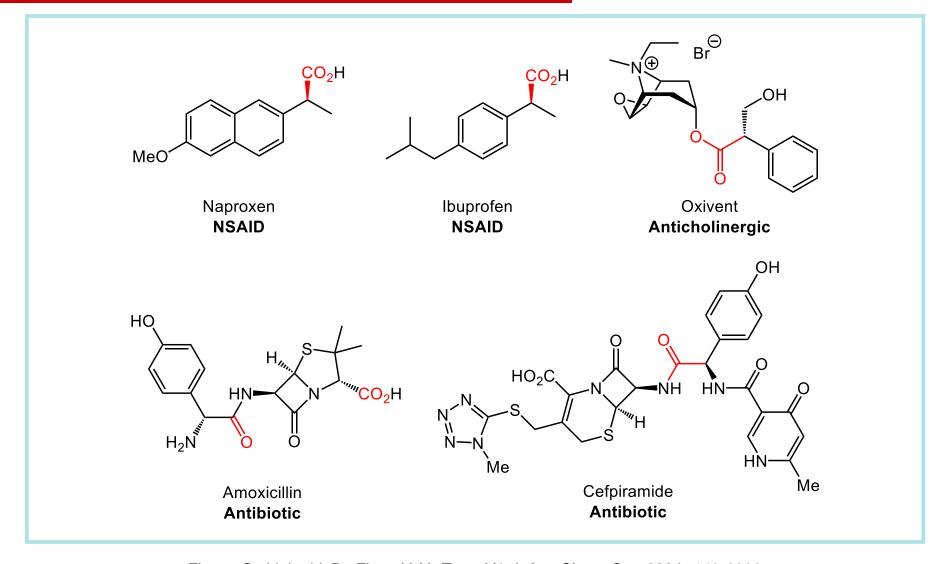
- > Asymmetric Catalysis
- > Design, Synthesis and Application of Olefin Polymerization Catalysts
- Total Synthesis of Natural Products

Contents

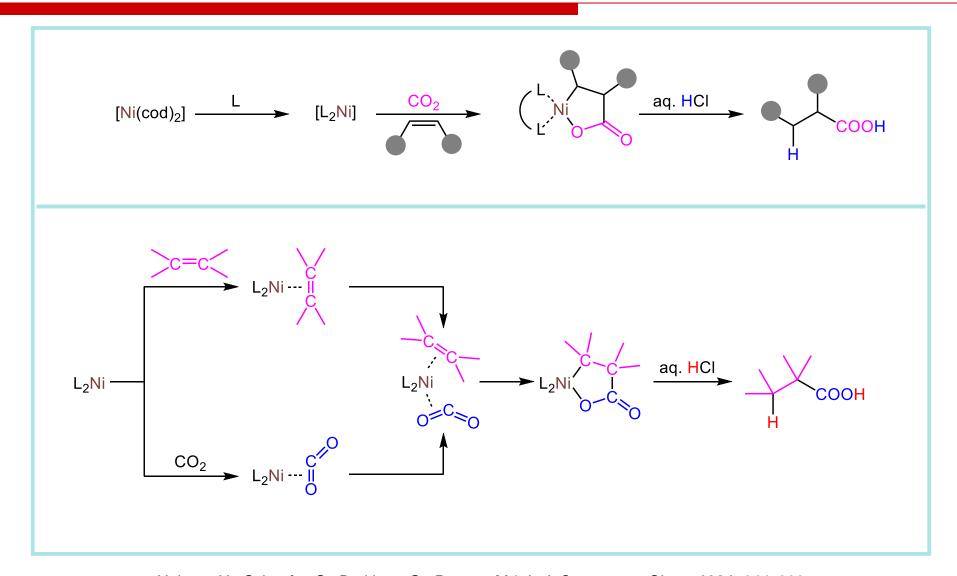
1 Introduction

2 Enantioselective Boracarboxylation of Arylalkenes with CO₂ and Diboron

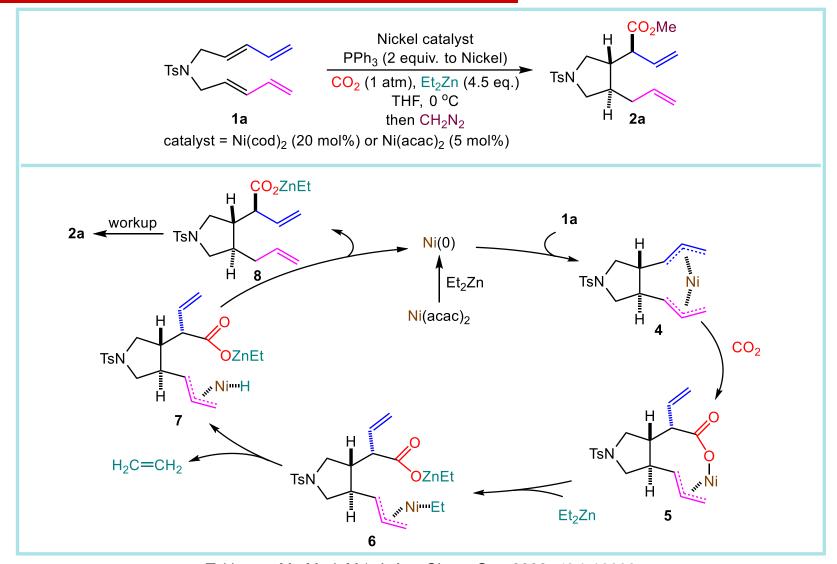
3 Summary



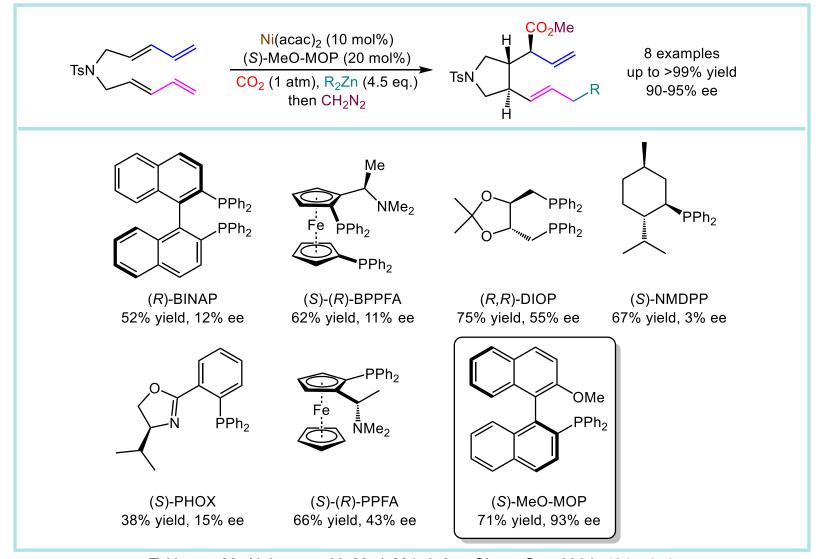
Zhang, S.; Li, L.; Li, D.; Zhou, Y.-Y.; Tang, Y.* J. Am. Chem. Soc. 2024, 146, 2888



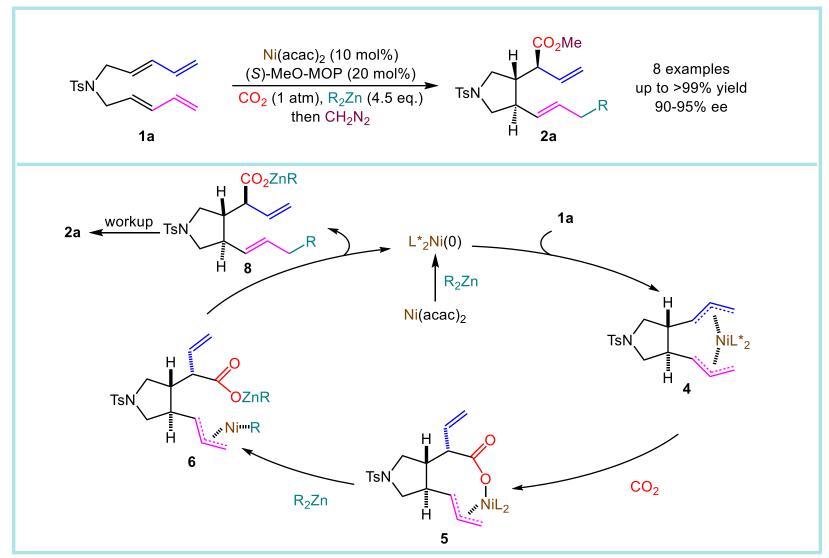
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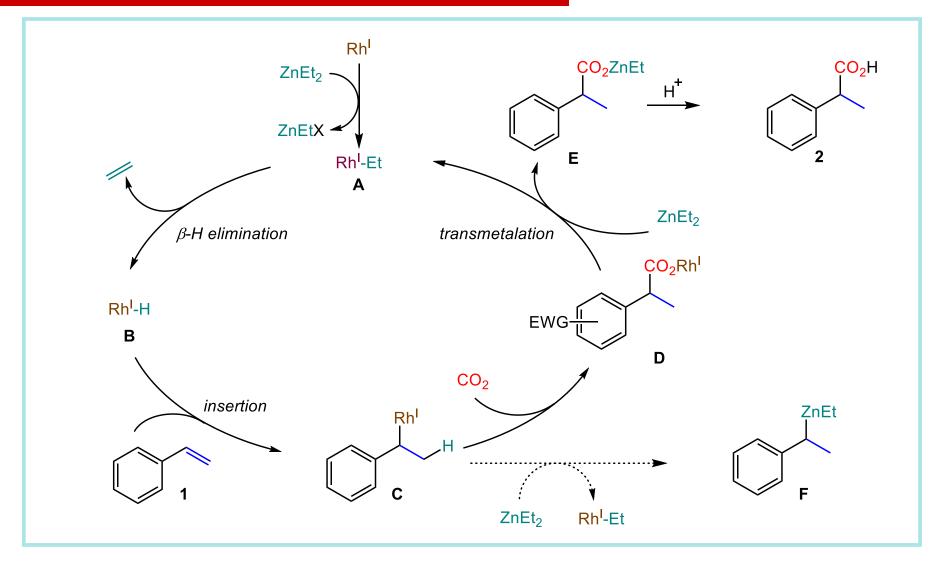
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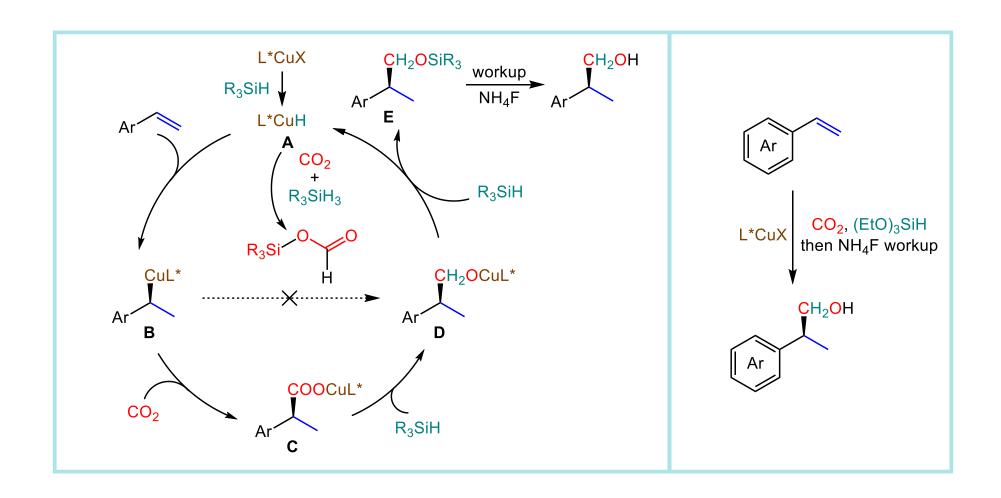


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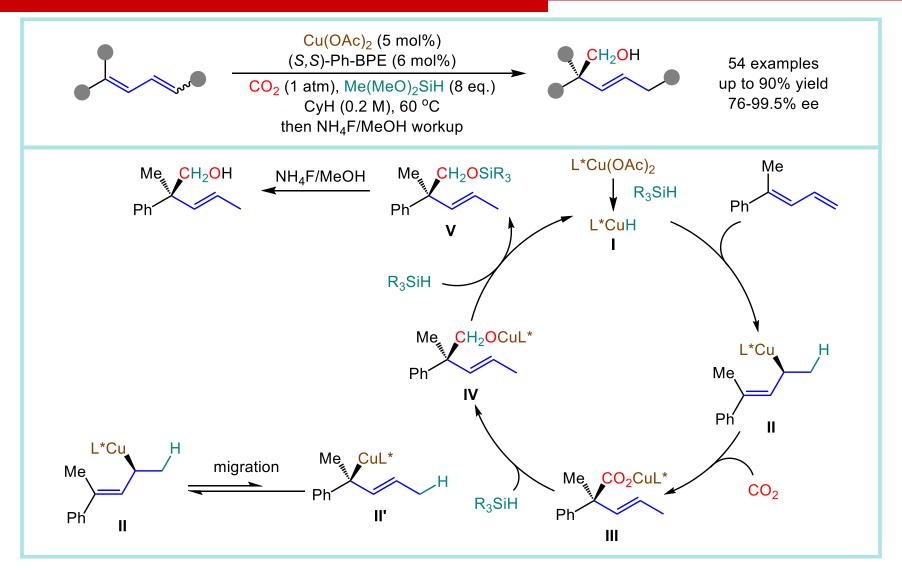


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Gui, Y.-Y.; Hu, N.; Chen, X.-W.; Yu, D.-G.* J. Am. Chem. Soc. 2017, 139, 17011



Gui, Y.-Y.; Hu, N.; Chen, X.-W.; Yu, D.-G.* J. Am. Chem. Soc. 2017, 139, 17011



Chen, X.-W.; Zhu, L.; Gui, Y.-Y.; Yu, D.-G.* J. Am. Chem. Soc. 2019, 141, 18825

Project Synopsis

Cu-catalyzed Asymmetric Boracarboxylation of Olefins with
$$CO_2$$

$$+ CO_2 + B_2(pin)_2$$

Optimization of the Reaction Conditions

Ph +
$$CO_2$$
 + $B_2(pin)_2$ $\xrightarrow{Solvent, T, 72 h}$ $\xrightarrow{EO_2Me}$ $\xrightarrow{B(pin)}$ + $E(pin)$ $\xrightarrow{B(pin)}$ $\xrightarrow{B(pin)}$ $\xrightarrow{B(pin)}$ $\xrightarrow{B(pin)}$ $\xrightarrow{B(pin)}$ $\xrightarrow{B(pin)}$ $\xrightarrow{B(pin)}$

| Entry ^a | L | Base | Solvent | Yield (%) ^b (1a/1b/1c) | Ee (%) (1a) ^c |
|--------------------|----------|---------------------|---------|-----------------------------------|-----------------------------------|
| 1 | L1 | NaO ^r Bu | THF | 96/0/0 | 18 |
| 2 | L2 | NaO ^r Bu | THF | 99/0/0 | 13 |
| 3 | L3 | NaO ^r Bu | THF | 23/45/0 | -40 |
| 4 | L4 | NaO ^r Bu | THF | 43/5/0 | 35 |
| 5 | L5 | NaO ^r Bu | THF | 93/7/0 | -40 |
| 6 | L6-(S,S) | NaO ⁴ Bu | THF | 93/<5/0 | 79 |

^a1 (0.25 mmol), B₂(pin)₂ (0.375 mmol), base (0.50 mmol), CuCl (10 mol%), ligand (11 mol%), solvent (4 mL), 1 atm of CO₂, ambient temperature, 72 h. ^bYields of **1a**, **1b**, and **1c** determined by 1H NMR. ^cee of **1a** determined by HPLC. ^dReaction temperature = −10 °C; reaction time = 96 h.

Optimization of the Reaction Conditions

| Entry ^a | L | Base | Solvent | Yield (%) ^b (1a/1b/1c) | Ee (%) (1a) ^c |
|--------------------|-------------------|------------------------------------|-------------------|-----------------------------------|-----------------------------------|
| 7 | L6-(S,S) | NaO ^t Bu | Et ₂ O | 18/14/<5 | 85 |
| 8 | L6-(S,S) | NaO ^t Bu | ′Pr₂O | 24/14/0 | 84 |
| 9 | L6-(S,S) | LiO ^t Bu | THF | <5/0/0 | / |
| 10 | L6-(S,S) | KO ^t Bu | THF | 53/<5/0 | 39 |
| 11 | L6-(S,S) | NaO ^t Am | THF | 88/<5/0 | 79 |
| 12 | L6-(S,S) | NaOEt | THF | 43/56/0 | 85 |
| 13 | L6-(S,S) | NaOCH ₂ ^t Bu | THF | 75(74)/25/0 | 90 |
| 14 ^d | L6-(<i>S,S</i>) | NaOCH ₂ [#] Bu | THF | 70(70)/30/0 | 92 |
| 15 ^d | L6-(S,S) | NaOCH ₂ ^t Bu | THF | 72(70)/28/0 | -91 |

^a1 (0.25 mmol), B₂(pin)₂ (0.375 mmol), base (0.50 mmol), CuCl (10 mol%), ligand (11 mol%), solvent (4 mL), 1 atm of CO₂, ambient temperature, 72 h. ^bYields of **1a**, **1b**, and **1c** determined by 1H NMR. ^cee of **1a** determined by HPLC. ^dReaction temperature = −10 °C; reaction time = 96 h.

Cu-Catalyzed Boracarboxylation of Styrenes

$$Ar + CO2 + B2(pin)2 \xrightarrow{\begin{array}{c} \text{CuCl (10 mol\%)} \\ \text{L6-(S,S) (11 mol\%)} \\ \hline \\ \text{NaOCH}_2^t\text{Bu} \\ \text{THF, -10 °C, 96 h} \end{array}} \xrightarrow{\begin{array}{c} \text{TMSCHN}_2 \\ \text{MeOH, DCM} \\ \text{2 h} \end{array}} \xrightarrow{\text{Repin}} B(pin)$$

1a, R = H, 70%, 92% ee

2a, R = Me, 87%, -92% ee

3a, R = t Bu, 71%, 93% ee

4a, R = OMe, 86%, 95% ee

5a, R = OPh, 52%, -92% ee

6a, R = F, 63%, 92% ee

7a, R = OAc, 53%, 90% ee

8a, R = SMe, 63%, -88% ee

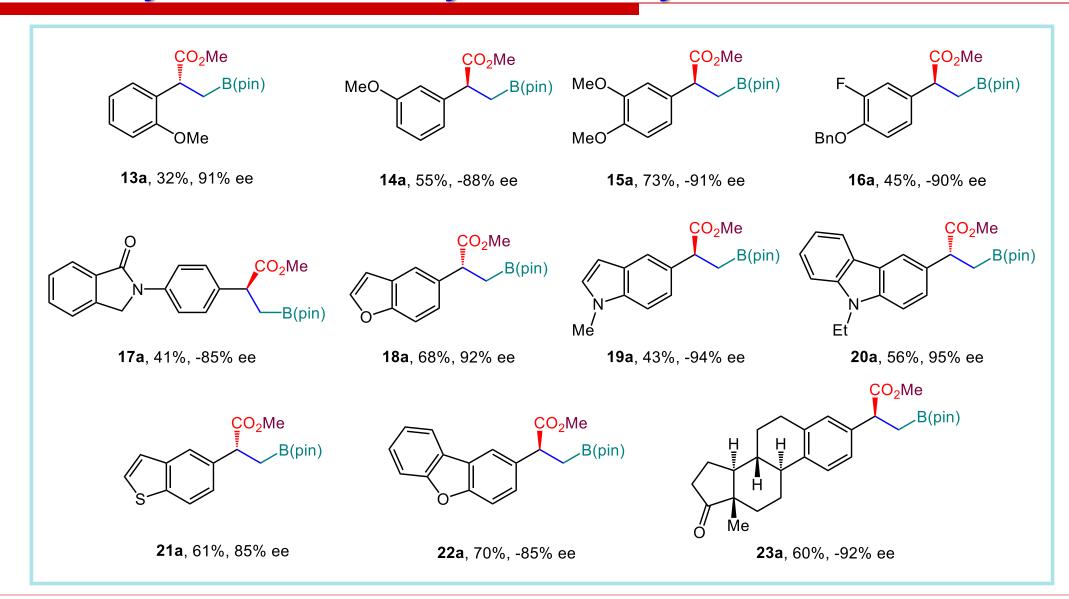
9a, 65%, 94% ee

11a, 65% (*E/Z* = 77:23) -83% ee (*E*); -84% ee (*Z*)

10a, 72%, -93% ee

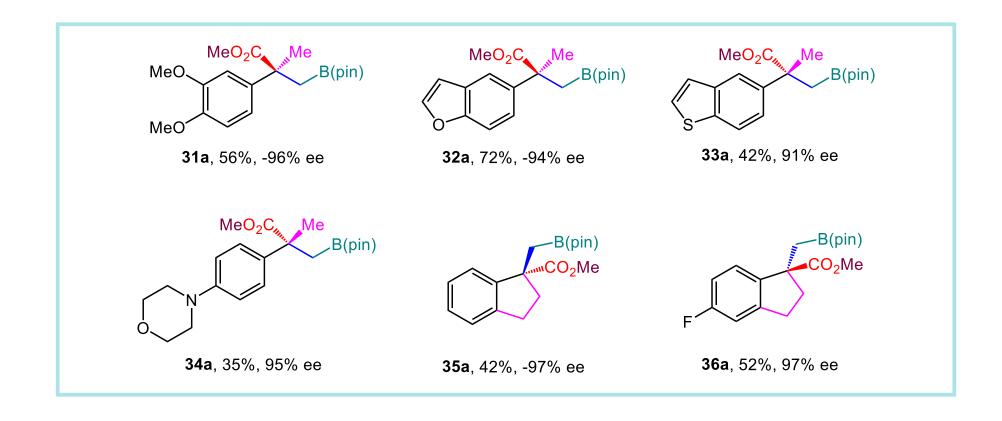
12a, 62%, -78% ee

Cu-Catalyzed Boracarboxylation of Styrenes

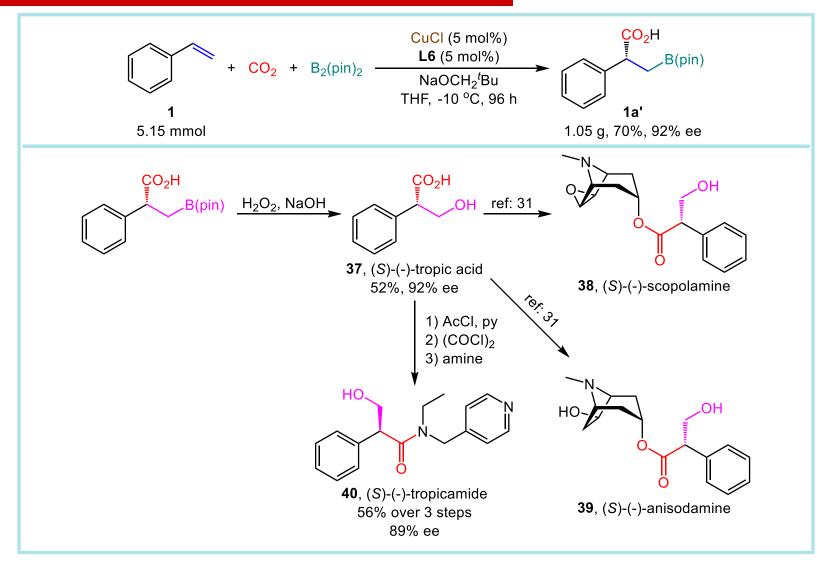


Scope of α-Substituted Styrenes

Scope of α-Substituted Styrenes

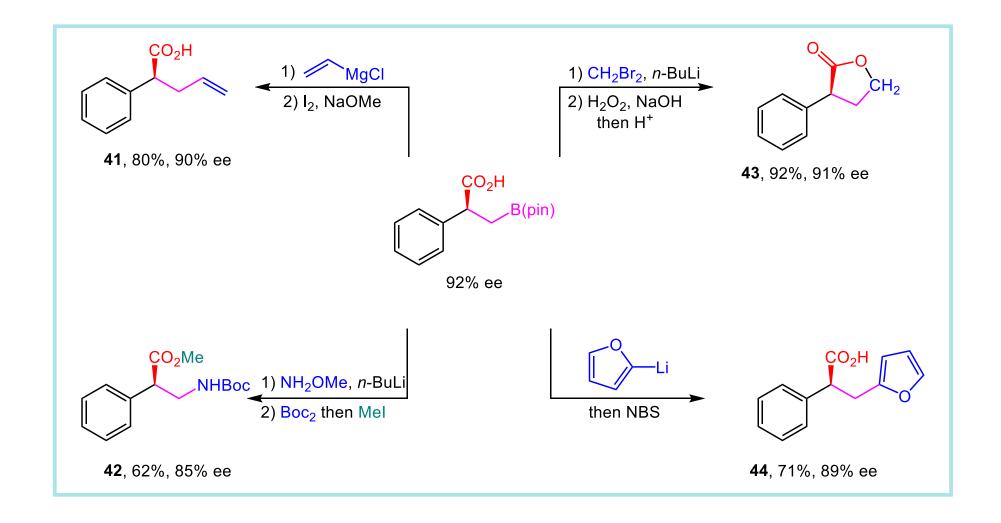


Gram Scale & Transformations of Products

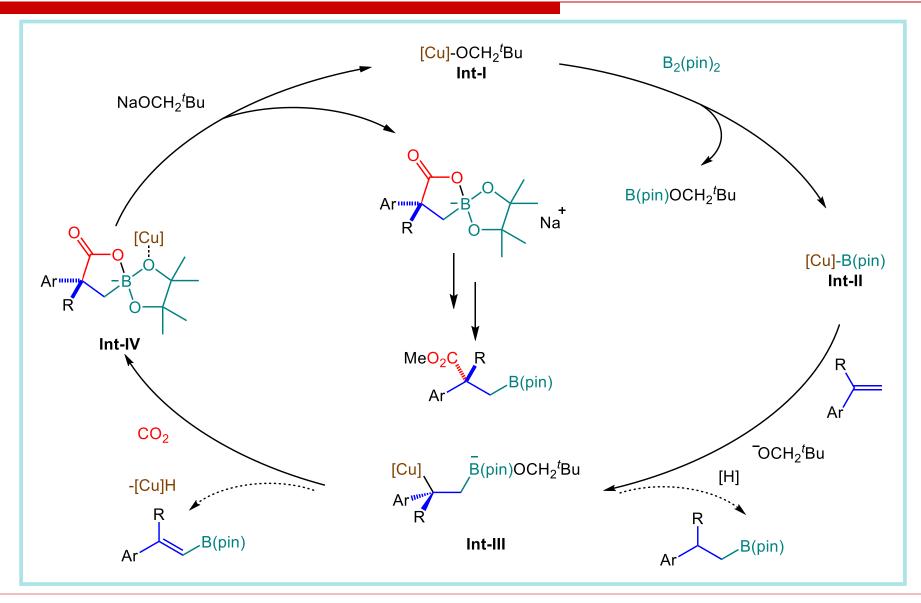


Ushimaru, R.; Ruszczycky, M. W.; Liu, H.-W.* J. Am. Chem. Soc. 2018, 140, 7433

Gram Scale & Transformations of Products



Proposed Mechanism



Summary

$$+ CO_2 + B_2(pin)_2$$

- Enantioselective boracarboxylation of arylalkenes with diboron and CO₂
- ☐ Mild reaction conditions, high catalytic efficacy, and excellent enantioselectivity
- Synthesis of bioactive compounds

Strategy for Writing The First Paragraph

α立体中心手性羧酸的重要性



过去α-羧酸的合成方法



引出本文工作

- Chiral carboxylic acids and derivatives bearing an αstereogenic center are one of the most prevalent functional motifs among natural products and pharmaceuticals and have therefore attracted the synthetic community for decades to develop highly efficient and direct synthetic methods.
- ✓ In recent years, chemists have devoted extensive efforts to the transition-metalcatalyzed carboxylation of suitable nucleophiles with CO₂ for the synthesis of high-value-added carboxylic acids via C-C bond construction.

✓ Among these, boracarboxylation of olefins with CO₂ provides the most profound versatility because of the many established transformations of the boron moiety.

Strategy for Writing The Last Paragraph

总结工作



强调亮点

- In summary, the first enantioselective boracarboxylation protocol of arylalkenes with diboron and 1 atm of CO_2 was developed. Featuring mild reaction conditions, high catalytic efficacy, and excellent enantioselectivity, this strategy provides access to a myriad of β-boron-functioned chiral α-carboxylic acids.
- In addition to a broad range of functional group tolerance, the reaction can also be applied to styrenes with αsubstitution to deliver enantioenriched carboxylic acids with αchiral all-carbon quaternary centers that were previously inaccessible. The synthetic utility was further showcased by gram-scale reactions, product conversion, and the synthesis of bioactive compounds.

Representative Examples

- ... for the synthesis of high-value-added carboxylic acids (high-value-added, 高附加值的: 指产品或服务在生产过程中获得的价值显著增加)
- … has remained a formidable challenge (formidable, 艰难的)
- ... which furnished valuable enantioenriched alcohols with broad functional group tolerance (furnish, 提供).

Acknowledgement

Thanks for your attention