

A Journal of the Gesellschaft Deutscher Chemiker A DOCH International Edition Market Chemiker CDCh Chemiker Ch

Accepted Article

- Title: Bifunctional NHC-Catalyzed Remote Enantioselective Mannichtype Reaction of 5-(Chloromethyl)furfural via Trienolate Intermediates
- Authors: Yuan-Yuan Gao, Chun-Lin Zhang, Ming-Lei Jin, Zhong-Hua Gao, and Song Ye

This manuscript has been accepted after peer review and appears as an Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: Angew. Chem. Int. Ed. 2023, e202301126

Link to VoR: https://doi.org/10.1002/anie.202301126

WILEY ... VCH

WILEY-VCH

Bifunctional NHC-Catalyzed Remote Enantioselective Mannichtype Reaction of 5-(Chloromethyl)furfural via Trienolate Intermediates

Yuan-Yuan Gao⁺,^[a,b] Chun-Lin Zhang⁺,^[a] Ming-Lei Jin^[a,c], Zhong-Hua Gao,^{*[a,c]} Song Ye^{*[a,c]}

Dedicated to Professor Lin-Xin Dai on the occasion of his 100th birthday.

Dr. Y.-Y. Gao, Dr. C.-L. Zhang, M.-L. Jin, Dr. Z.-H. Gao, Prof. Dr. S. Ye [a] Beijing National Laboratory for Molecular Sciences, CAS Key Laboratory of Molecular Recognition and Function, CAS Research/Education Center for Excellence in Molecular Sciences, Institute of Chemistry, Chinese Academy of Sciences Beijing 100190 (China) E-mail: gaozh@iccas.ac.cn; songye@iccas.ac.cn Dr. Y.-Y. Gao [b] Henan Engineering Laboratory of Green Synthesis for Pharmaceuticals, College of Chemistry and Chemical Engineering, Shangqiu Normal University Shangqiu 476000 (China) M.-L. Jin, Dr. Z.-H Gao, Prof. Dr. S. Ye [c] University of Chinese Academy of Sciences Beijing 100049 (China) [*] These authors contributed equally to this work. Supporting information for this article is given via a link at the end of the document.

Abstract: The N-heterocyclic carbene (NHC)-catalyzed enantioselective Mannich-type reactions of the biomass-derived platform compound 5-(chloromethyl)furfural (CMF) with imines were developed. A series of high-value-added chiral amines were afforded in good to high yields with excellent regio- and enantioselectivities. The bifunctional NHC derived from L-pyroglutamic acid was efficient to steer the remote addition of trienolate intermediate to imine in a highly stereocontrolled manner. This represents the first example of enantioselective reactions via NHC-bound trienolate intermediate.

Biomass, produced by plants through solar photosynthetic processes, is abundant on the Earth. Chemical conversion of biomass to access useful materials is considered a promising approach to replacing non-renewable fossil resources.^[1] Lignocellulose occupies the largest proportion of biomass, and is mainly composed of cellulose, hemicellulose and lignin.^[2] Degradation of cellulose and hemicellulose can yield carbonyl compounds, such as glucose, xylose, furfural, 5-hydroxymethyl furfural (HMF), 5-(chloromethyl)furfural (CMF), levulinic acid and lactic acid.^[3] Further transformation of biomass-derived platform compounds via asymmetric catalysis to obtain high-value-added chiral chemicals is of great significance.^[4]

In the past decades, N-heterocyclic carbenes (NHCs) have emerged as efficient organocatalysts to enable highly enantioselective transformations via versatile intermediates.^[5] The umpolung of aldehydes,^[6] and $\alpha^{-[7]}$ or β -functionalizations^[8] of aldehydes or carboxylic acid derivatives have been well established. In recent years, the NHC-catalyzed enantioselective γ -functionalization via dienolate I has been well established (Scheme 1, upper left).^[9] In 2011, our group reported the first NHC-catalyzed asymmetric [4+2] annulation reaction via dienolate from β -methyl- α , β -unsaturated acyl chlorides.^[10] Chi et al developed an elegant oxidative strategy for the formation of dienolate from β -methyl enals.^[11] Since then, studies involving NHC-bound dienolates^[12] and (hetero-) *o*-quinodimethane intermediates^[13] have been extensively investigated. In addition, the NHC-catalvzed δ-functionalizations via α.β-ν.δbisunsaturated acyl azolium intermediates were also reported by Lupton,^[14] Chi,^[15] and Zhu^[16] groups. Compared with the welldeveloped NHC-catalyzed α -, β -, γ -, and δ -functionalizations, the NHC-catalyzed ɛ-functionalizations via trienolate have been far less developed (Scheme 1, upper right). During our research on NHC/photo-cocatalyzed alkylation of enals, the *ε*-alkylated products were afforded via NHC-bound trienolates.^[17] Recently, NHC-catalyzed reactions 4-(chloromethyl)the of benzaldehydes^[18] and CMF^[19] via trienolates were also developed. However, asymmetric version of these NHCcatalyzed reactions meets little success,^[20] compared to the achievements on enantioselective remote functionalization via aminocatalysis.[21]



Scheme 1. NHC-bound dienolate and trienolate intermediate, and asymmetric remote Mannich reaction via trienolate.

Asymmetric Mannich reaction is one of the most important methodologies to access chiral amine building blocks. The metal- and organo-catalyzed enantioselective classical and vinylogous Mannich reactions have witnessed great success.^[22] Given our ongoing interests in asymmetric NHC catalysis,^[23] we herein report a bifunctional NHC-catalyzed enantioselective

10.1002/anie.202301126

WILEY-VCH

remote Mannich-type reaction of CMF via trienolate with imines for the construction of chiral amines (Scheme 1, down). Hbonding effect between azolium trienolate and the imine was believed to promote the reaction in an efficient and highly enantiocontrolled manner.

Table 1. Optimization of reaction conditions^[a]



[a] General conditions: 1a (0.1 mmol), 2a (2.0 equiv.), preNHC (20 mol%), Cs₂CO₃ (2.2 equiv.), toluene 1 mL (c = 0.1 mol/L), 15 h, RT. [b] Isolated yields. [c] Determined by HPLC analysis using a chiral stationary phase. [d] toluene 2 mL (c = 0.05 mol/L). Mes = mesityl; RT = room temperature.

The model reaction of imine 1a and CMF 2 was investigated under NHC catalysis (Table 1). After initial assessment of base and solvent, the desired amine 3a could be afforded in 64% yield albeit with 6% ee when 20 mol% of preNHC A^[7a] was used, with Cs₂CO₃ as the base and toluene as the solvent (entry 1, for details see the Supporting Information, Table S1). The preNHC $\mathbf{B}^{[24]}$ with a bulky α -naphthyl was then used, which resulted in a slight increase of the yield but without expected enhancement of (entry 2). The reaction using the enantioselectivity morpholinone-derived triazolium preNHC $C^{[25]}$ gave amine 3a in 49% yield with 11% ee (entry 3). A series of bifunctional Lpyroglutamic acid-derived preNHCs with free hydroxyl group were then employed (entries 4-6).^[26] We were happy to find that the usage of N-Mes bifunctional preNHC D dramatically improved the enantioselectivity to 57% ee (entry 4). The N-C₆F₅ substituted preNHC E showed better enantioselectivity but with decreased yield (entry 5). The reaction using preNHC F bearing N-2,4,6-tribromophenyl afforded amine 3a in 45% yield with 80% ee (entry 6). Interestingly, the reaction performed much better to give 3a in 73% yield with 92% ee when concentration of the

reaction was reduced from 0.1 to 0.05 mol/L (entry 7). Decreasing the loading of preNHC F to 10 mol% showed no change for the reaction (entry 8 vs 7). The yield of amine 3a was further increased to 93% when the reaction temperature was lowered to -20 °C (entry 9). Notably, the reaction performed as well with respect to both yield and enantioselectivity when the loading of preNHC F was further decreased to 5 mol% (entry 10), while some decreased yield was observed albeit with high ee unchanged when 2 mol% of preNHC F was used (entry 11).



Scheme 2. Scope of N-Ts imines.

With the optimized reaction conditions in hand (Table 1, entry 10), we next examined the scope of N-Ts imines 1 (Scheme 2). It was found that imines bearing both electrondonating groups (Ar = 4-MeC₆H₄ and 4-OMeC₆H₄) and electronwithdrawing groups (Ar = 4-BrC₆H₄, 4-ClC₆H₄) at the paraposition of the phenyl ring worked well to give products 3b-3e in excellent yields and enantioselectivities. Functional groups, such as ester, cyano and nitro groups, could be well tolerated (3f-3h). The reactions of meta-substituted imines also proceeded well to give products 3i-3k in excellent yields and enantioselectivities. ortho-Substituted aryl imines (Ar = 2-MeC₆H₄, 2-MeOC₆H₄, and 2-BrC₆H₄) performed well under the standard reaction conditions (3I-3n). 2-Furanyl-substituted and 2-thioenyl-substituted imines resulted in some decreased but still good yields under standard conditions (30-3p). High yield and enantioselectivity were achieved for the reaction of α -naphthaldehyde (3g) and β naphthaldehyde (3r). Unfortunately, aliphatic imine did not work for the reaction under current conditions.

The reaction of imines with different sulfonyl groups were next explored (Scheme 3). Imines with N-benzenesulfonyl, N-(2-

thienyl)sulfonyl, *N*-(2-thienyl)sulfonyl, and 2-(4nitrophenyl)sulfonyl all furnished the desired products in high yields and excellent enantioselectivities (**3s-3u**). Imine with *N*mesitylenesulphonyl was also tested to afford **3v** in 77% yield with 95% ee. However, the enantioselectivity decreased when *N*-methylsulfonyl protected imine was used (**3w**), possibly due to the less sterically demanded property. The attempt of using 4-(chloromethyl)benzaldehyde instead of CMF failed under current conditions.



Scheme 3. Variation of N-sulfonyl groups of imines.

A variety of nucleophiles were then briefly investigated (Scheme 4). Both primary and secondary alcohols worked well for the reaction (3x-3z). Benzyl alcohol and 2-thiophenemethanol were tolerated to give the desired products (3aa-3ab) in high yields. Furthermore, high yield was also achieved for the reaction with cinnamic alcohol (3ac). Notably, all of the alcohols furnished the corresponding products in excellent enantioselectivities.



Scheme 4. Scope of the nucleophiles.

The reaction could be easily scaled up to 3.0 mmol to give 1.1 g of the desired amine **3a** in 90% yield with 94% ee under standard conditions (Scheme 5).



Scheme 5. Gram-scale synthesis.

The reaction products were amenable to further transformations through simple procedures (Scheme 6). Removal of the 4-nitrophenylsulfonyl with *p*-toluenethiol afforded chiral amine 4 in 80% yield. Reaction of chiral amine 4 with isothiocyanatoarene gave the corresponding thiourea 5 in excellent yield without apparent erosion of the enantioselectivity (94% ee), which was readily available for potential application in asymmetric transformations as a hydrogen-bonding organocatalyst.





The absolute (*S*)-configuration of product **3a** was assigned via single-crystal X-ray analysis (see the Supporting Information).^[27]

A plausible mechanism for this Mannich-type reaction of CMF is proposed as shown in Scheme 7. Addition of the in situ generated NHC to **2** forms the Breslow intermediate **III**, which undergoes 1,6-elimination of HCI to afford the key trienolate azolium **IV**. The following nucleophilic ε -addition of trienolate **IV** to aldimine **1a** gives adduct **V**, which is trapped by methanol to afford the chiral amine product **3a** and regenerates the NHC catalyst. The possible pathway involving the S_{N2} attack of NHC at CI-C bond of CMF was ruled out by a control experiment using methyl 5-(chloromethyl)furan-2-carboxylate instead of CMF, which resulting no reaction and full recovery of methyl 5-(chloromethyl)furan-2-carboxylate (see Supporting Information).





A weak interaction between the free OH of bifunctional NHC catalyst and aldimine was observed by ¹H NMR titration in CD₃CN (see Supporting Information, Figure S1). Based on the control experiments and the stereochemical outcome, a H-bonding directed stereochemical mode is proposed (Figure 1). The imine is pre-positioned with the NHC-CMF adduct via H-bonding, followed by addition of trienolate to the *Si* face of aldimine.



Figure 1. H-Bonding directed stereochemical mode.

In summary, the bifunctional NHC-catalyzed enantioselective Mannich-type reactions of the biomass-derived platform compound CMF with imines were developed, giving a series of high-value-added chiral amines in good to high yields with excellent regio- and enantioselectivities. The L-pyroglutamic acid derived NHC catalyst bearing a free hydroxy group was the key to steer the remote *ɛ*-addition of trienolate in a highly stereocontrolled manner. This protocol exemplifies the powerful activities catalytic of NHCs in remote asymmetric transformations. NHC-catalyzed other asymmetric transformations of biomass-derived platform compounds are underway in our laboratory.

Acknowledgements

Financial support from the National Natural Science Foundation of China (Nos. 22171269, 22071253, 21831008), and the Beijing Natural Science Foundation (2192063) is greatly acknowledged.

Keywords: Mannich-type reactions • N-heterocyclic carbenes • asymmetric catalysis • biomass-derived platform compounds • trienolates

- [1] a) J. J. Bozell, Science 2010, 329, 522; b) H. Kopetz, Nature 2013, 494, 29; c) Y. Queneau, B. Han, The Innovation 2022, 3, 100184.
- [2] a) C.-H. Zhou, X. Xia, C.-X. Lin, D.-S. Tong, J. Beltramini, *Chem. Soc. Rev.* 2011, 40, 5588; b) R. Rinaldi, *Angew. Chem. Int. Ed.* 2014, 53, 8559; *Angew. Chem.* 2014, 126, 8699; c) Z. Zhang, J. Song, B. Han, *Chem. Rev.* 2017, 117, 6834.
- [3] a) R.-J. van Putten, J. C. van der Waal, E. de Jong, C. B. Rasrendra, H. J. Heeres, J. G. de Vries, *Chem. Rev.* 2013, *113*, 1499; b) J. Wang, J. Xi, Y. Wang, *Green Chem.* 2015, *17*, 737; c) Z. Zhang, K. Deng, ACS Catal. 2015, *5*, 6529; d) M. Mascal, *ChemSusChem* 2015, *8*, 3391; e) S. Chen, R. Wojcieszak, F. Dumeignil, E. Marceau, S. Royer, *Chem. Rev.* 2018, *118*, 11023; f) M. Mascal, ACS Sustain. Chem. Eng. 2019, *7*, 5588; g) B. Wozniak, S. Tin, J. G. de Vries, *Chem. Sci.* 2019, *10*, 6024.
- [4] a) A. Corma, S. Iborra, A. Velty, *Chem. Rev.* 2007, *107*, 2411; b)
 C. O. Tuck, E. Pérez, I. T. Horváth, R. A. Sheldon, M. Poliakoff, *Science* 2012, *337*, 695; c) M. Besson, P. Gallezot, C. Pinel, *Chem. Rev.* 2014, *114*, 1827; d) L. Wu, T. Moteki, Amit A. Gokhale, David W. Flaherty, F. D. Toste, *Chem* 2016, *1*, 32; e) T.

A. Bender, J. A. Dabrowski, M. R. Gagné, *Nat. Rev. Chem.* **2018**, 2, 35.

- [5] a) D. Enders, O. Niemeier, A. Henseler, *Chem. Rev.* 2007, 107, 5606; b) S. De Sarkar, A. Biswas, R. C. Samanta, A. Studer, *Chem. Eur. J.* 2013, 19, 4664; c) S. J. Ryan, L. Candish, D. W. Lupton, *Chem. Soc. Rev.* 2013, 42, 4906; d) M. N. Hopkinson, C. Richter, M. Schedler, F. Glorius, *Nature* 2014, 510, 485; e) D. M. Flanigan, F. Romanov-Michailidis, N. A. White, T. Rovis, *Chem. Rev.* 2015, 115, 9307; f) M. H. Wang, K. A. Scheidt, *Angew. Chem. Int. Ed.* 2016, 55, 14912; *Angew. Chem.* 2016, 128, 15134; g) K. J. R. Murauski, A. A. Jaworski, K. A. Scheidt, *Chem. Soc. Rev.* 2018, 47, 1773; h) P. Bellotti, M. Koy, M. N. Hopkinson, F. Glorius, *Nat. Rev. Chem.* 2021, 5, 711.
- [6] a) J. C. Sheehan, D. H. Hunneman, J. Am. Chem. Soc. 1966, 88, 3666; b) D. Enders, U. Kallfass, Angew. Chem. Int. Ed. 2002, 41, 1743; Angew. Chem. 2002, 114, 1822; c) S. M. Mennen, J. D. Gipson, Y. R. Kim, S. J. Miller, J. Am. Chem. Soc. 2005, 127, 1654; d) J. Read de Alaniz, T. Rovis, J. Am. Chem. Soc. 2005, 127, 6284; e) H. Takikawa, Y. Hachisu, J. W. Bode, K. Suzuki, Angew. Chem. Int. Ed. 2006, 45, 3492; Angew. Chem. 2006, 118, 3572; f) X. Huang, S. Ye, Chin. Sci. Bull. 2010, 55, 1753; g) I. Piel, M. Steinmetz, K. Hirano, R. Fröhlich, S. Grimme, F. Glorius, Angew. Chem. Int. Ed. 2011, 50, 4983; Angew. Chem. 2011, 123, 5087; h) L.-H. Sun, Z.-Q. Liang, W.-Q. Jia, S. Ye, Angew. Chem. Int. Ed. 2013, 52, 5803; Angew. Chem. 2013, 125, 5915; i) M.-Q. Jia, S.-L. You, ACS Catal. 2013, 3, 622.
- [7] a) M. He, J. R. Struble, J. W. Bode, J. Am. Chem. Soc. 2006, 128, 8418; b) Y.-R. Zhang, L. He, X. Wu, P.-L. Shao, S. Ye, Org. Lett. 2008, 10, 277; c) N. Duguet, C. D. Campbell, A. M. Z. Slawin, A. D. Smith, Org. Biomol. Chem. 2008, 6, 1108; d) X. Zhao, K. E. Ruhl, T. Rovis, Angew. Chem. Int. Ed. 2012, 51, 12330; Angew. Chem. 2012, 124, 12496; e) L. Hao, Y. Du, H. Lv, X. Chen, H. Jiang, Y. Shao, Y. R. Chi, Org. Lett. 2012, 14, 2154.
- [8] a) C. Burstein, F. Glorius, Angew. Chem. Int. Ed. 2004, 43, 6205; Angew. Chem. 2004, 116, 6331; b) S. S. Sohn, E. L. Rosen, J. W. Bode, J. Am. Chem. Soc. 2004, 126, 14370; c) A. Chan, K. A. Scheidt, J. Am. Chem. Soc. 2008, 130, 2740; d) S. De Sarkar, A. Studer, Angew. Chem. Int. Ed. 2010, 49, 9266; Angew. Chem. 2010, 122, 9452; e) F.-G. Sun, L.-H. Sun, S. Ye, Adv. Synth. Catal. 2011, 353, 3134; f) L. Candish, D. W. Lupton, J. Am. Chem. Soc. 2013, 135, 58; g) J. Cheng, Z. Huang, Y. R. Chi, Angew. Chem. Int. Ed. 2013, 52, 8592; Angew. Chem. 2013, 125, 8754; h) X.-Y. Chen, Z.-H. Gao, C.-Y. Song, C.-L. Zhang, Z.-X. Wang, S. Ye, Angew. Chem. Int. Ed. 2014, 53, 11611; Angew. Chem. 2014, 126, 11795.
- [9] a) X.-Y. Chen, Q. Liu, P. Chauhan, D. Enders, *Angew. Chem. Int. Ed.* 2018, 57, 3862; *Angew. Chem.* 2018, 130, 3924; b) J. Gao, J. Feng, D. Du, *Org. Chem. Front.* 2021, *8*, 6138.
- [10] L.-T. Shen, P.-L. Shao, S. Ye, *Adv. Synth. Catal.* **2011**, 353, 1943.
- [11] J. Mo, X. Chen, Y. R. Chi, J. Am. Chem. Soc. 2012, 134, 8810.
 [12] a) X.-Y. Chen, F. Xia, J.-T. Cheng, S. Ye, Angew. Chem. Int. Ed. 2013, 52, 10644; Angew. Chem. 2013, 125, 10838; b) Z. Xiao, C. Yu, T. Li, X.-S. Wang, C. Yao, Org. Lett. 2014, 16, 3632; c) B.-S. Li, Y. Wang, Z. Jin, P. Zheng, R. Ganguly, Y. R. Chi, Nat. Commun. 2015, 6, 6207; d) Z. Wu, F. Li, J. Wang, Angew. Chem. Int. Ed. 2015, 54, 1629; Angew. Chem. 2015, 127, 1649.
- [13] a) X. Chen, S. Yang, B.-A. Song, Y. R. Chi, Angew. Chem. Int. Ed. 2013, 52, 11134; Angew. Chem. 2013, 125, 11340; b) D. Janssen-Müller, S. Singha, T. Olyschläger, C. G. Daniliuc, F. Glorius, Org. Lett. 2016, 18, 4444; c) D.-F. Chen, T. Rovis, Synthesis 2017, 49, 293; d) A. Przydacz, A. Topolska, A. Skrzyńska, Ł. Albrecht, Adv. Synth. Catal. 2022, 364, 1434.
- [14] a) M. Kowalczyk, D. W. Lupton, Angew. Chem. Int. Ed. 2014, 53, 5314; Angew. Chem. 2014, 126, 5418; b) R. M. Gillard, J. E. M. Fernando, D. W. Lupton, Angew. Chem. Int. Ed. 2018, 57, 4712; Angew. Chem. 2018, 130, 4802.
- [15] T. Zhu, C. Mou, B. Li, M. Smetankova, B.-A. Song, Y. R. Chi, J. Am. Chem. Soc. 2015, 137, 5658.
- [16] a) K. Xu, W. Li, S. Zhu, T. Zhu, Angew. Chem. Int. Ed. 2019, 58, 17625; Angew. Chem. 2019, 131, 17789; b) K. Xu, Z. Wang, T. Zhu, Synlett 2020, 31, 925.
- [17] a) L. Dai, Z.-H. Xia, Y.-Y. Gao, Z.-H. Gao, S. Ye, Angew. Chem. Int. Ed. 2019, 58, 18124; Angew. Chem. 2019, 131, 18292; b) Y.-Y. Xu, L. Dai, Z.-H. Gao, S. Ye, J. Org. Chem. 2022, 87, 14970.

10.1002/anie.202301126

- [18] L. Dai, S. Ye, ACS Catal. 2020, 10, 994.
- [19] L. Dai, Y. Qiu, Y.-Y. Xu, S. Ye, Cell Rep. Phys. Sci. 2020, 1, 100071.
- [20] a) L. Bernardi, J. López-Cantarero, B. Niess, K. A. Jørgensen, J. Am. Chem. Soc. 2007, 129, 5772; b) D. Uraguchi, K. Yoshioka, Y. Ueki, T. Ooi, J. Am. Chem. Soc. 2012, 134, 19370; c) L. Dell'Amico, Ł. Albrecht, T. Naicker, P. H. Poulsen, K. A. Jørgensen, J. Am. Chem. Soc. 2013, 135, 8063; d) R. Deng, S. Wu, C. Mou, J. Liu, P. Zheng, X. Zhang, Y. R. Chi, J. Am. Chem. Soc. 2022, 144, 5441.
- [21] a) A. Przydacz, A. Skrzyńska, Ł. Albrecht, Angew. Chem. Int. Ed. 2019, 58, 63; Angew. Chem. 2019, 131, 64; b) B.-X. Xiao, X.-Y. Gao, W. Du, Y.-C. Chen, Chem. - Eur. J. 2019, 25, 1607.
- [22] a) J. M. M. Verkade, L. J. C. v. Hemert, P. J. L. M. Quaedflieg, F. P. J. T. Rutjes, *Chem. Soc. Rev.* 2008, *37*, 29; b) A. Ting, S. E. Schaus, *Eur. J. Org. Chem.* 2007, 2007, 5797; c) B. Karimi, D. Enders, E. Jafari, *Synthesis* 2013, *45*, 2769; d) M. S. Roselló, C. del Pozo, S. Fustero, *Synthesis* 2016, *48*, 2553; e) S. Saranya, N. A. Harry, K. K. Krishnan, G. Anilkumar, *Asian J. Org. Chem.* 2018, *7*, 613; f) I. Bagheri, L. Mohammadi, V. Zadsirjan, M. M. Heravi, *ChemistrySelect* 2021, *6*, 1008.
- [23] X.-Y. Chen, Z.-H. Gao, S. Ye, Acc. Chem. Res. 2020, 53, 690.
- [24] C. Zhao, F. Li, J. Wang, Angew. Chem. Int. Ed. 2016, 55, 1820; Angew. Chem. 2016, 128, 1852.
- [25] M. Wadamoto, E. M. Phillips, T. E. Reynolds, K. A. Scheidt, J. Am. Chem. Soc. 2007, 129, 10098.
- [26] L. He, Y.-R. Zhang, X.-L. Huang, S. Ye, Synthesis 2008, 2008, 2825.
- [27] CCDC 2237090 (**3a**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.

WILEY-VCH

COMMUNICATION

Entry for the Table of Contents

Insert graphic for Table of Contents here. ((Please ensure your graphic is in one of following formats))



The NHC-catalyzed enantioselective Mannich-type reactions of 5-(chloromethyl)furfural (CMF), an important biomass-derived platform compound, with aldimines were developed. High-value-added chiral amines were afforded in good yields with excellent regio- and enantioselectivities. The bifunctional NHC bearing a free hydroxy group was efficient to steer the remote addition of trienolate intermediate to imine in a highly stereocontrolled manner.

Institute and/or researcher Twitter usernames: ((optional))