



中国科学院大连化学物理研究所
DALIAN INSTITUTE OF CHEMICAL PHYSICS, CHINESE ACADEMY OF SCIENCES

C-H Functionalization of Amines through Cooperative Action of Chiral and Achiral Lewis Acids

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Checker: Yang Zhao

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Chan, J. Z.; Wasa, M. *et al.* *J. Am. Chem. Soc.* **2020**, *142*, 16493

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CV of Prof. Masayuki Wasa



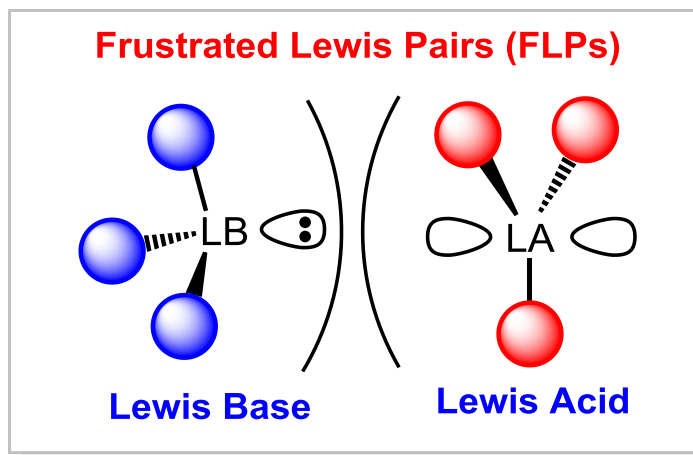
Background:

- ❑ **2013** Ph. D., The Scripps Research Institute
(with Prof. Jin-Quan Yu)
- ❑ **2013-2015** JSPS Postdoctoral Fellow, Harvard University
(with Prof. Eric N. Jacobsen)
- ❑ **2015-present** Assistant Professor of Chemistry,
Boston College

Research Interests:

- ✓ Development of polyfunctional Lewis pair catalysis for practical synthetic transformations with applications in drug discovery and development, and alternative energy.

Frustrated Lewis Pairs (FLPs)

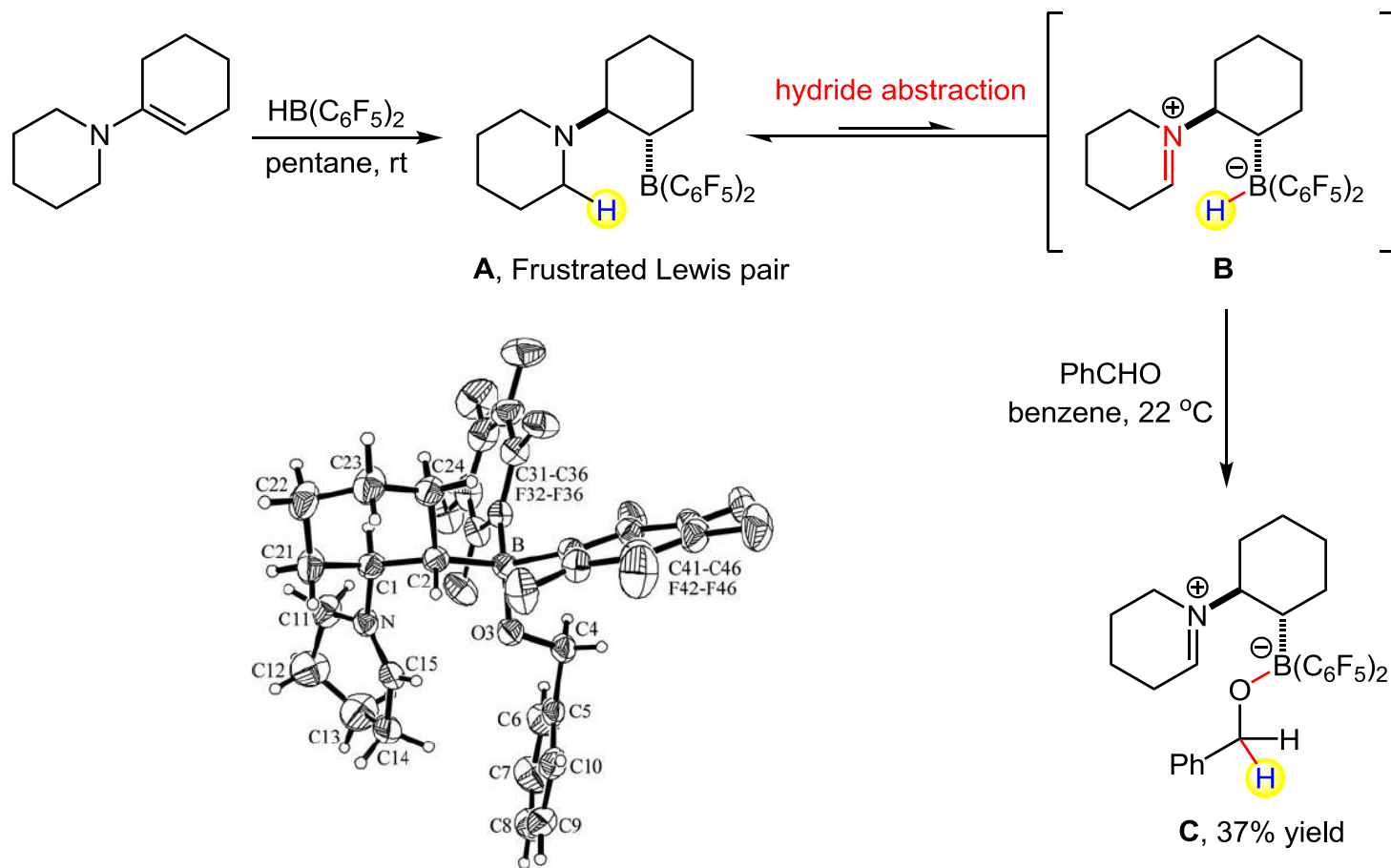


**Small Molecule
Activation**

**Catalytic
Hydrogenation**

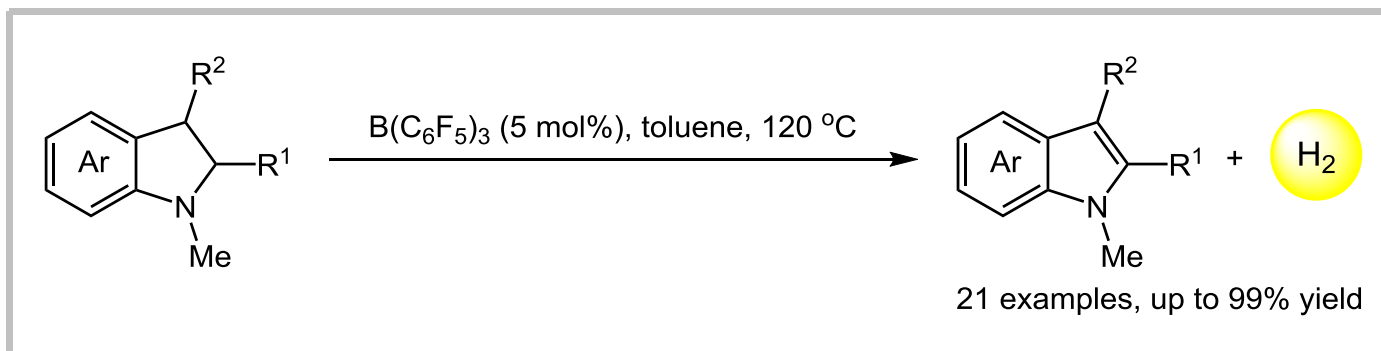
Polymerization

Intramolecular Hydride Transfer within FLPs

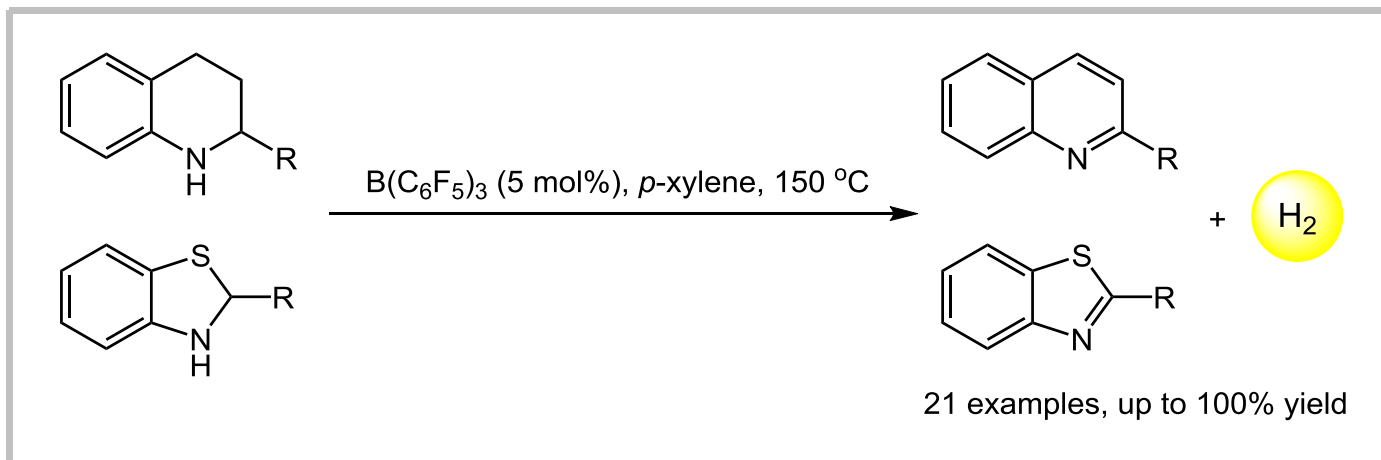


Erker, G. *et al. Chem. Sci.* **2011**, *2*, 1842

Dehydrogenation of *N*-Heterocycles

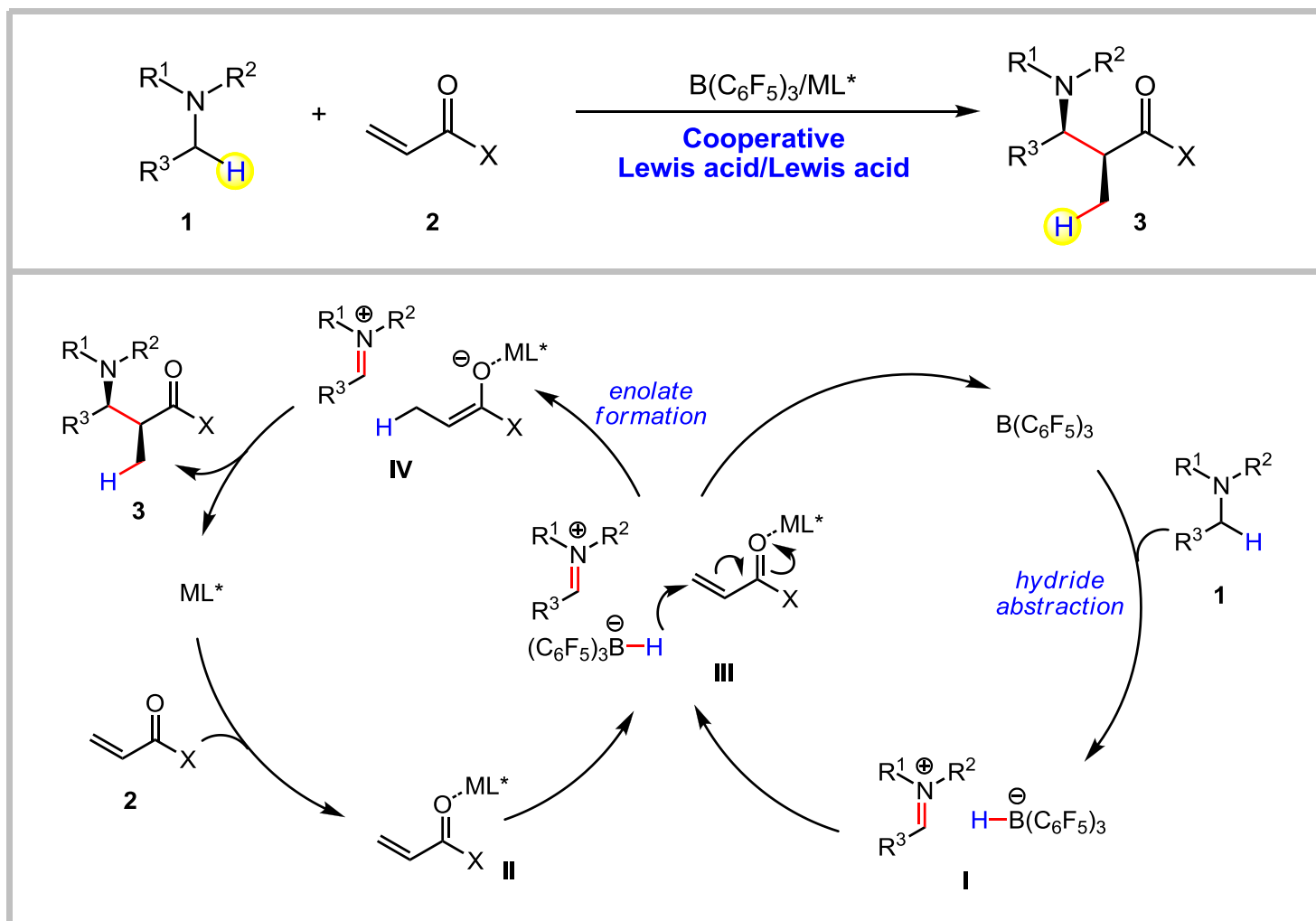


Paradies, J. *et al. Angew. Chem. Int. Ed.* **2016**, *55*, 12219



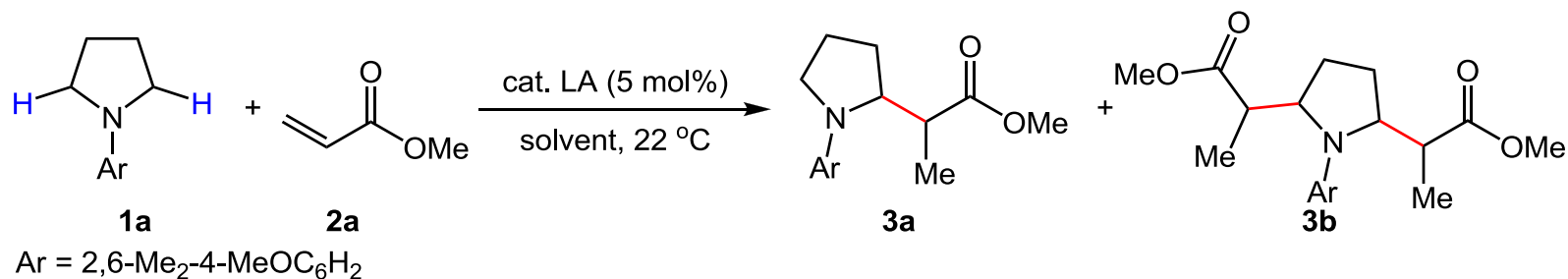
Kanai, M. *et al. Angew. Chem. Int. Ed.* **2016**, *55*, 12224

Enantioselective C-H Functionalization of Amines



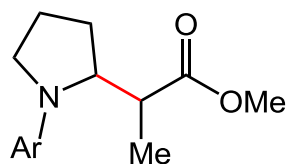
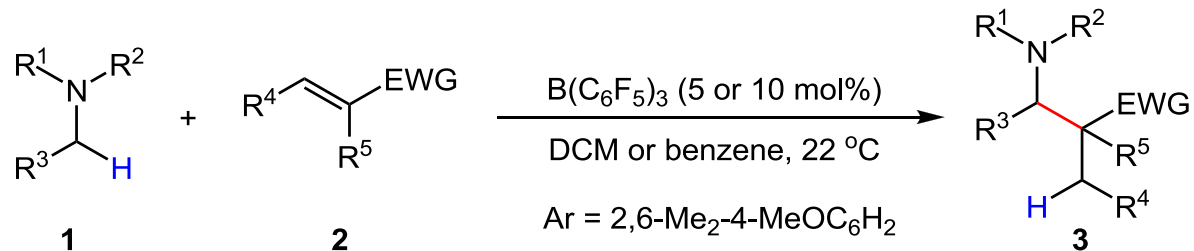
Wasa, M. et al. *J. Am. Chem. Soc.* **2018**, *140*, 10593

C-H Functionalization of Amines

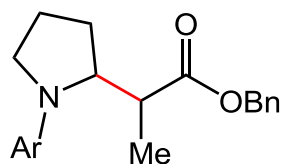


Entry	LA	Solvent	Time (h)	Yield of 3a (%) <i>anti/syn</i>	Yield of 3b (%)
1	B(C ₆ F ₅) ₃	DCM	12	>95, 2.3:1	<5
2	B(C ₆ F ₅) ₃	Et ₂ O	12	94, 2.3:1	<5
3	B(C ₆ F ₅) ₃	Toluene	12	81, 2.4:1	5
4	B(C₆F₅)₃	Benzene	0.5	>95, 2.4:1	<5
5	none	Benzene	12	0	0
6	BF ₃ ·OEt ₂	Benzene	12	0	0
7	BPh ₃	Benzene	12	0	0

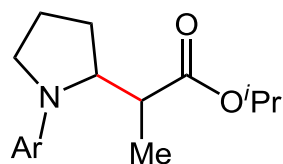
Substrate Scope



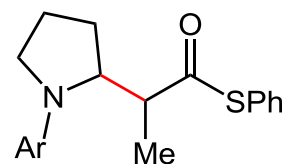
3a: 97% yield
2.3:1 *anti:syn*



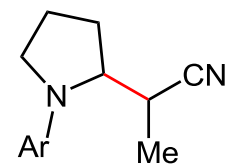
3c: 91% yield
2.5:1 *anti:syn*



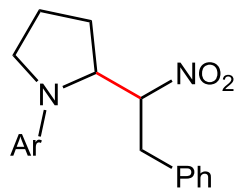
3d: 78% yield
2.1:1 *anti:syn*



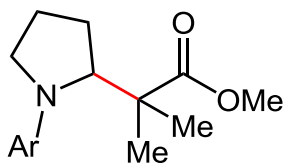
3e: 53% yield
1.8:1 *anti:syn*



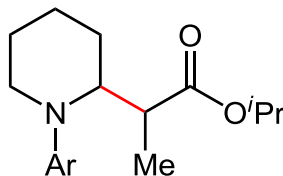
3f: 50% yield
1.3:1 *anti:syn*



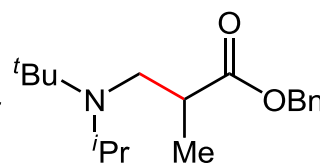
3g: 84% yield
1.2:1 *anti:syn*



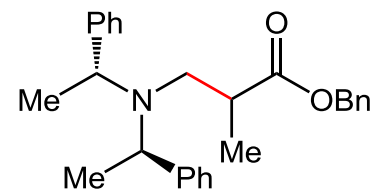
3h: 97% yield



3i: 63% yield
10:1 *anti:syn*

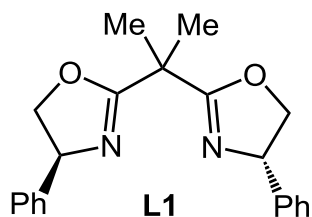
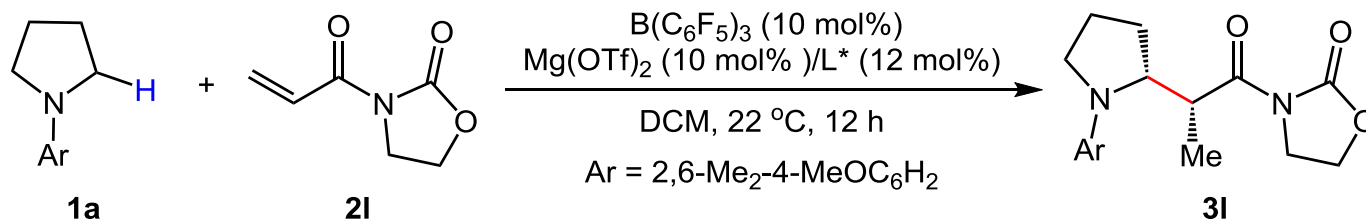


3j: 85% yield

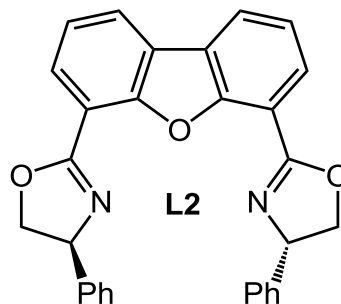


3k: 91% yield
(*R,S*)/(*R,R*) = 1.5:1

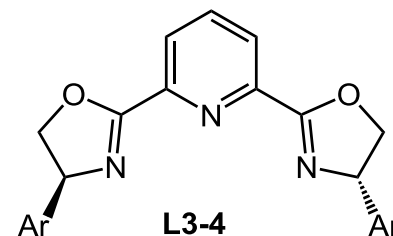
Enantioselective C-H Functionalization of Amines



7% yield
1.3:1 *anti:syn*
56:44 e.r. (*anti*)
52:48 e.r. (*syn*)



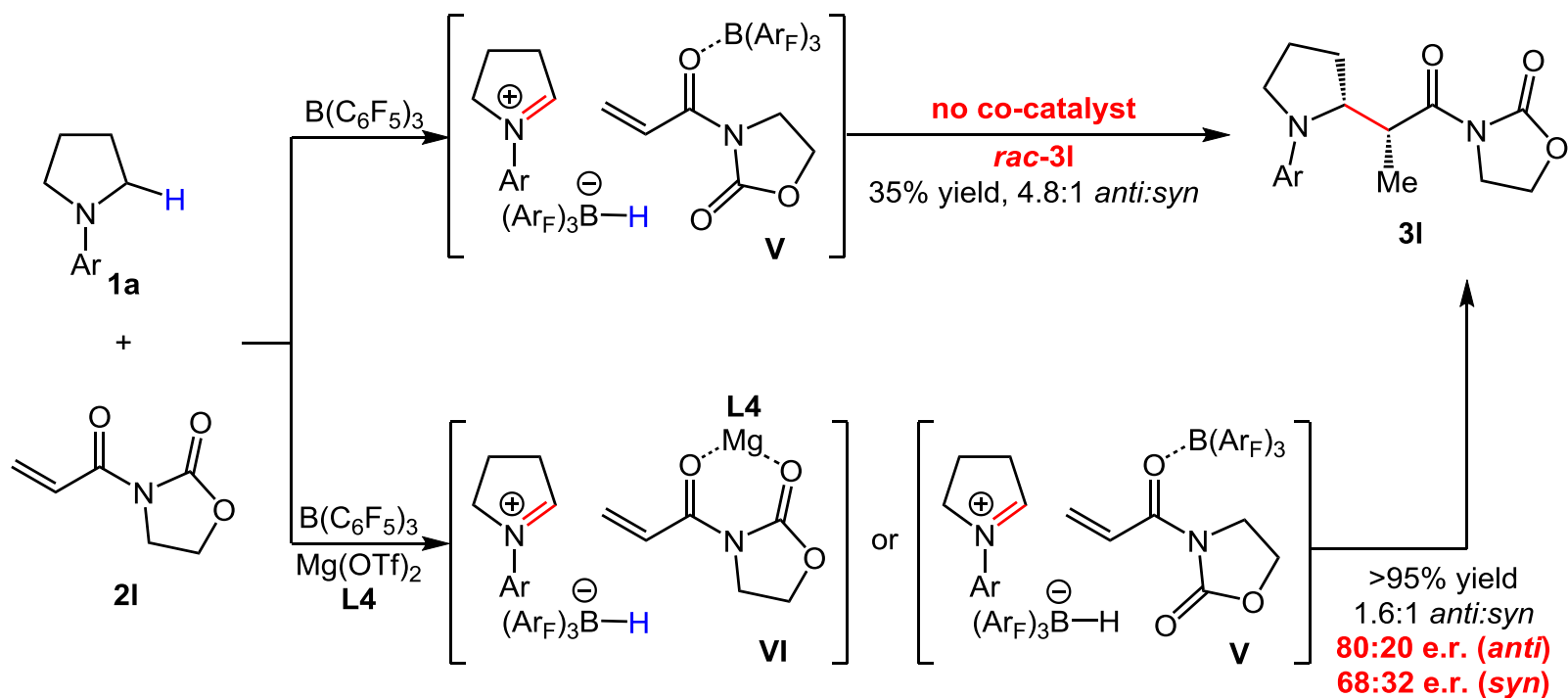
<5% yield
2.0:1 *anti:syn*
50:50 e.r. (*anti*)
50:50 e.r. (*syn*)



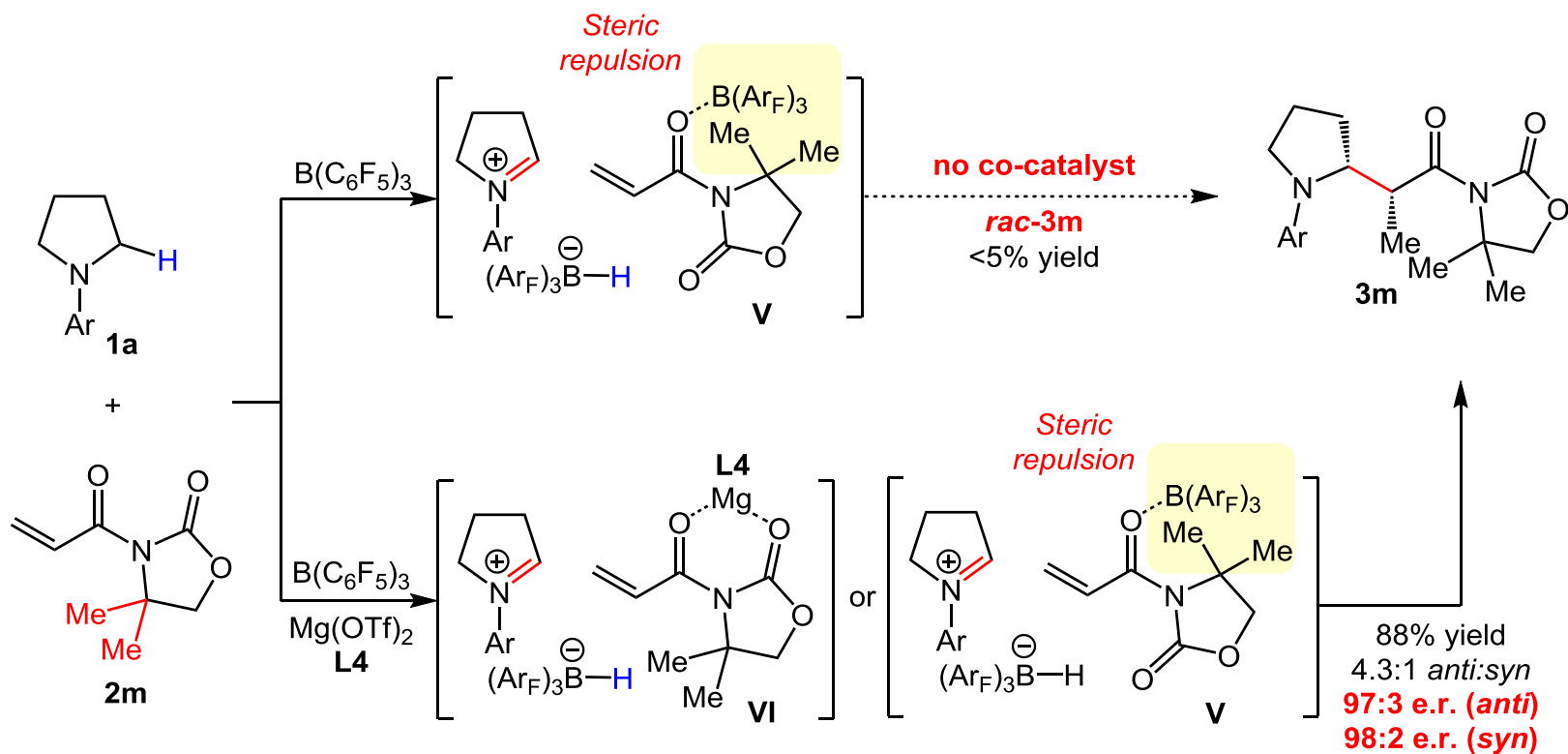
L3: $Ar = Ph$
35% yield
1.1:1 *anti:syn*
71:29 e.r. (*anti*)
80:20 e.r. (*syn*)

L4: $Ar = 3-ClC_6H_4$
>95% yield
1.6:1 *anti:syn*
80:20 e.r. (*anti*)
68:32 e.r. (*syn*)

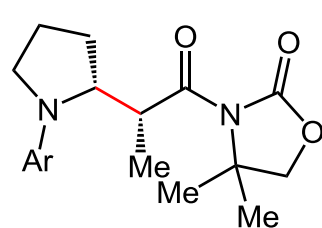
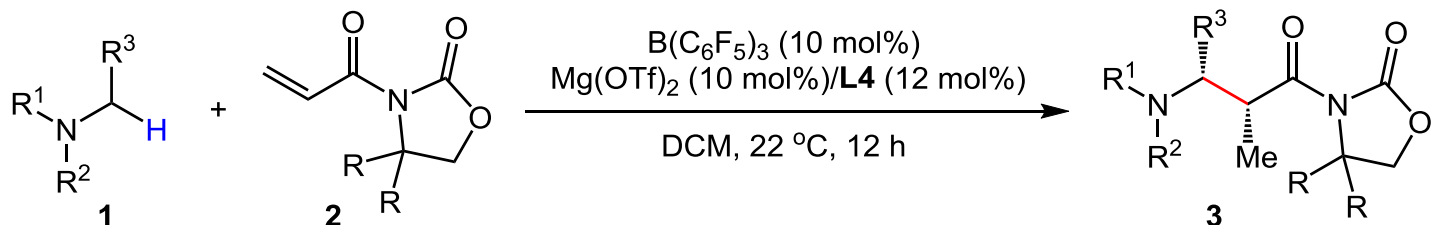
Enantioselective C-H Functionalization of Amines



Enantioselective C-H Functionalization of Amines

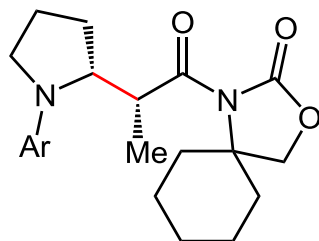


Substrate Scope



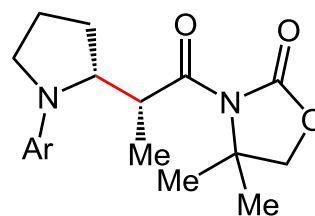
Ar = 2,6-Me₂-4-MeOC₆H₂

3m: 88% yield
anti:syn: 4.3:1
anti: 97:3 e.r.
syn: 98:2 e.r.



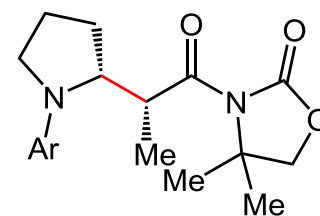
Ar = 2,6-Me₂-4-MeOC₆H₂

3n: 79% yield
anti:syn: 5.6:1
anti: 95:5 e.r.
syn: 98:2 e.r.



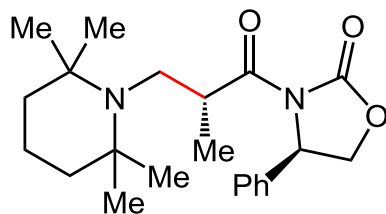
Ar = 2,6-Me₂C₆H₃

3o-*anti*
 56% yield, 89:11 e.r.
3o-*syn*
 15% yield, 93:7 e.r.

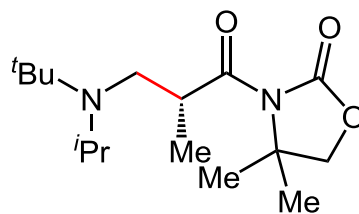


Ar = 2,6-Me₂-4^tBuC₆H₂

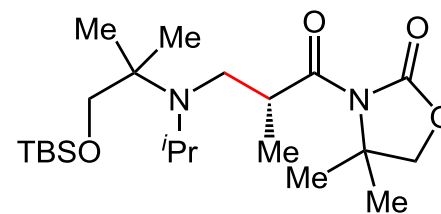
3p-*anti*
 57% yield, 91:9 e.r.
3p-*syn*
 13% yield, 98:2 e.r.



3r-major: 47% yield, >99:1 e.r.
3r-minor: 24% yield, >99:1 e.r.

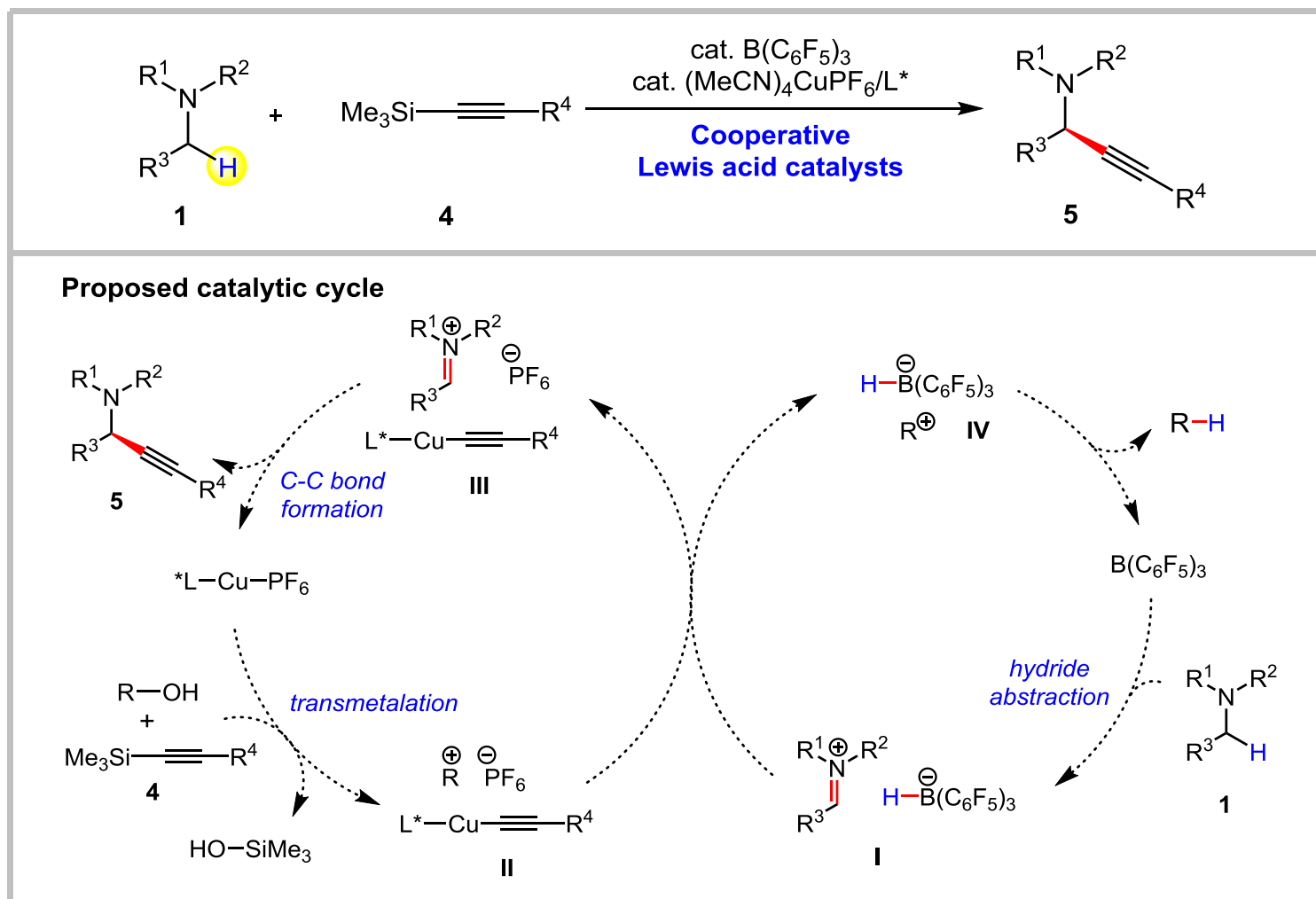


3s: 52% yield
 88:12 e.r.



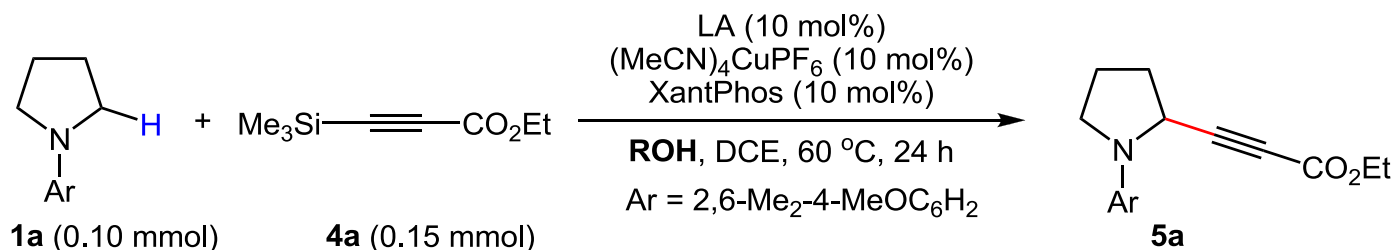
3t: 46% yield
 96:4 e.r.

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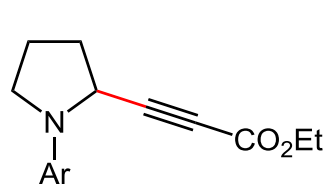
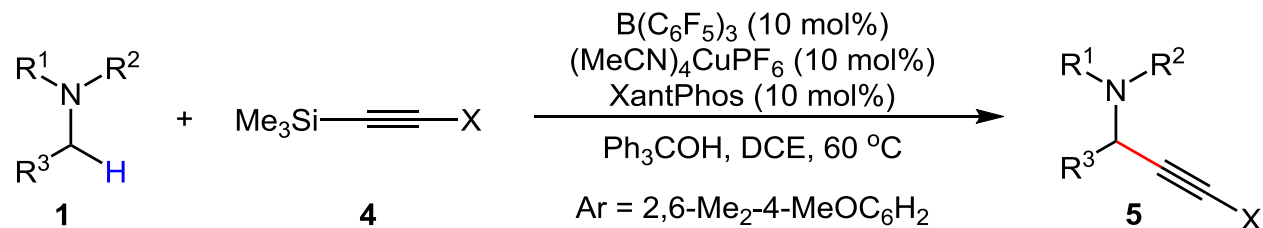
C-H Functionalization of Amines



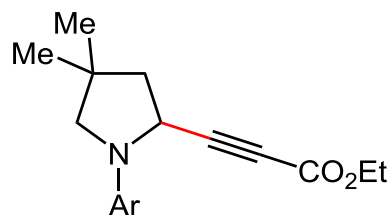
Entry	LA	ROH (x mmol)	Yield of 5a (%)
1	B(C ₆ F ₅) ₃	none	7
2	B(C ₆ F ₅) ₃	<i>i</i> PrOH (0.20)	0
3	B(C ₆ F ₅) ₃	<i>t</i> BuOH (0.20)	17
4	B(C ₆ F ₅) ₃	Ph ₃ COH (0.20)	52
5	B(C ₆ F ₅) ₃	Ph ₃ COH (0.10)	83
6^a	B(C₆F₅)₃	Ph₃COH (0.10)	90
7	none	Ph ₃ COH (0.10)	0
8	BPh ₃	Ph ₃ COH (0.10)	0

^a 12 h

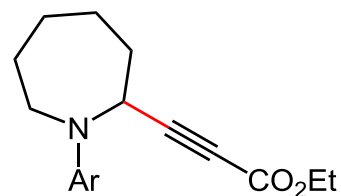
Substrate Scope



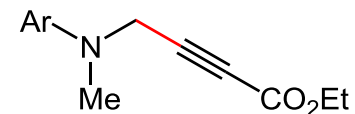
5a: 90% yield



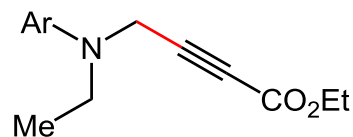
5b: 77% yield



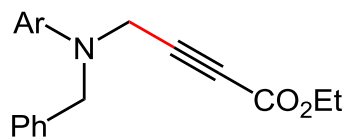
5c: 77% yield



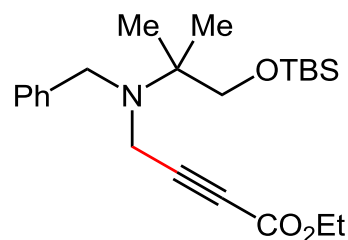
5d: 90% yield



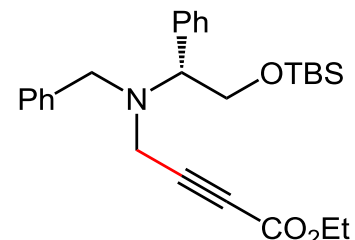
5e: 42% yield



5f: 70% yield



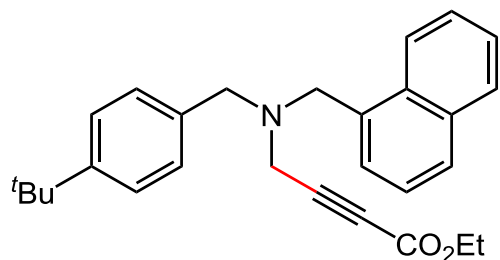
5g: 76% yield^a



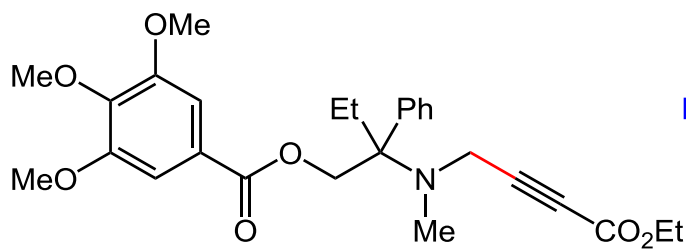
5h: 86% yield^a

^aDppe (10 mol %), Ph_3COH (2.0 equiv.), 80 °C.

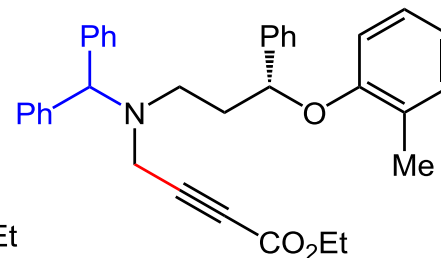
Substrate Scope



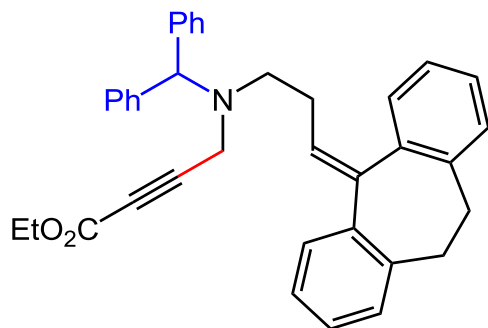
5i: 76% yield^a
butenafine derivative



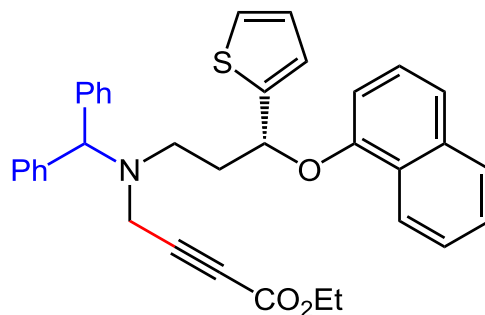
5j: 71% yield^a
trimebutine derivative



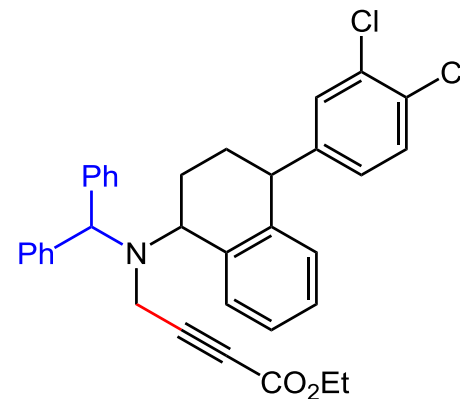
5k: 56% yield^a
atomoxetine derivative



5l: 74% yield^a
nortriptyline derivative



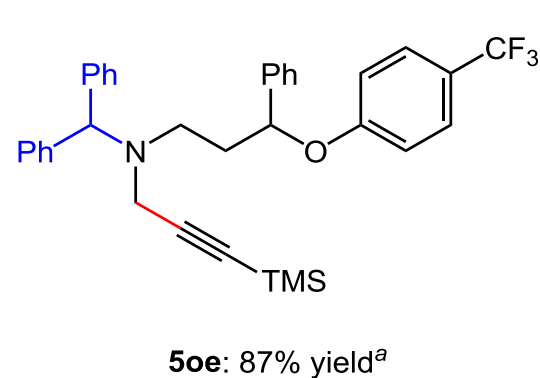
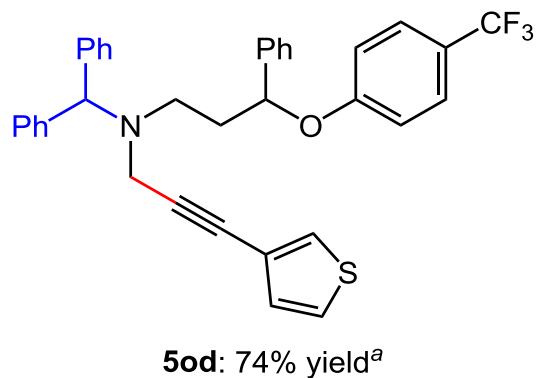
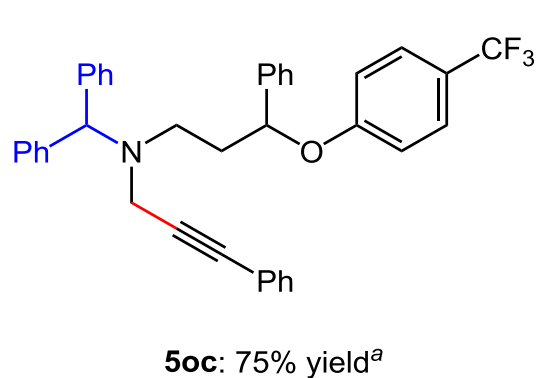
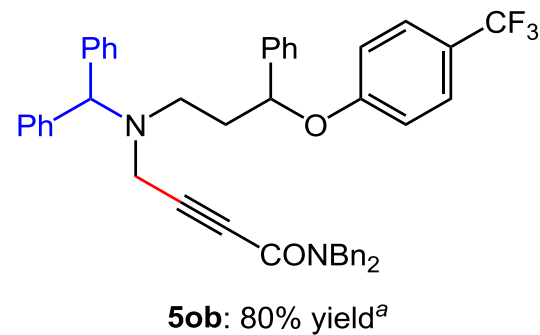
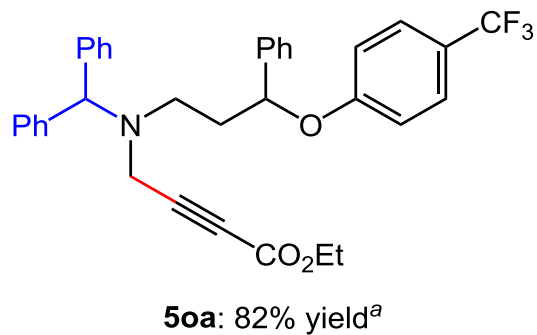
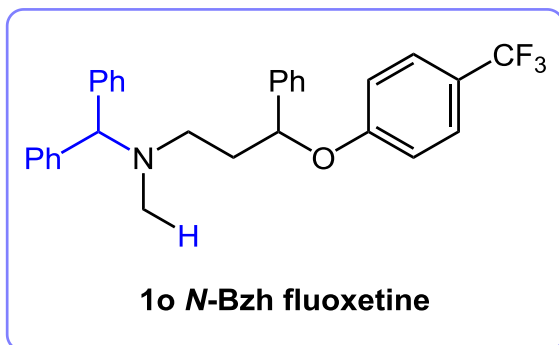
5m: 68% yield^a
duloxetine derivative



5n: 67% yield^a
sertraline derivative

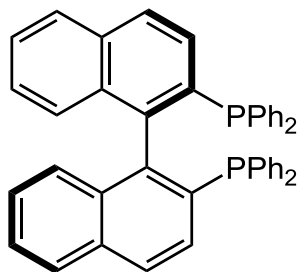
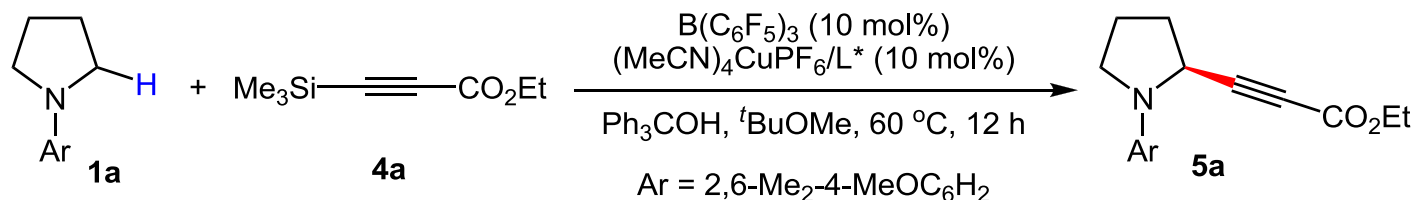
^a Dppe (10 mol %), Ph₃COH (2.0 equiv.), 80 °C.

Substrate Scope

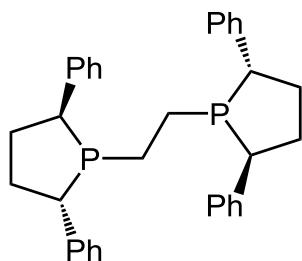


^aDppe (10 mol %), Ph₃COH (2.0 equiv.), 80 °C.

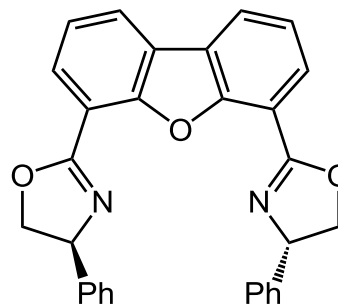
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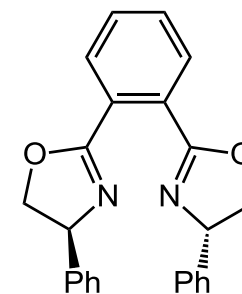
L1: 61% yield, 49:51 e.r.



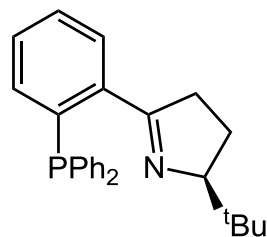
L2: 70% yield, 45:55 e.r.



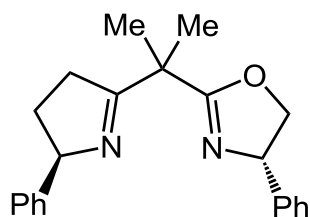
L3: 6% yield, 72:28 e.r.



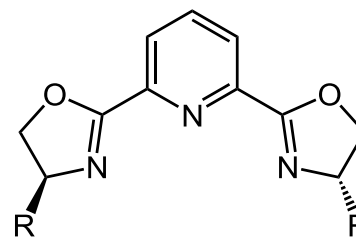
L4: 29% yield, 66:34 e.r.



L5: 15% yield, 41:59 e.r.



L6: 53% yield, 48:52 e.r.

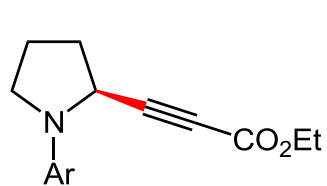
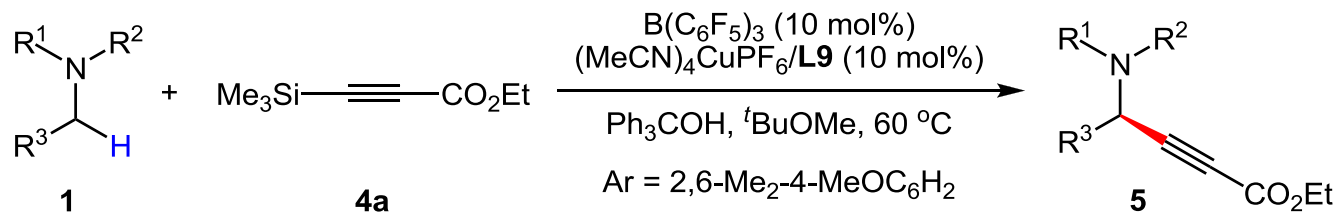


L7: R = ^tBu
81% yield, 52:48 e.r.

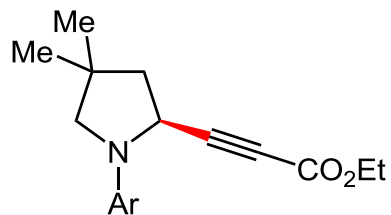
L8: R = Ph
84% yield, 82:18 e.r.

L9: R = 3,5-Me₂C₆H₃
75% yield, 95:5 e.r.

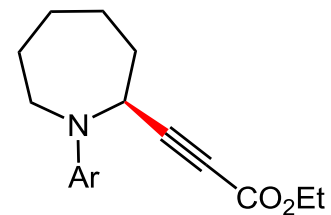
Substrate Scope



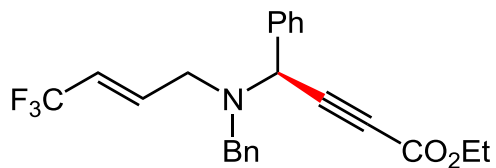
5a: 75% yield, 95:5 e.r.



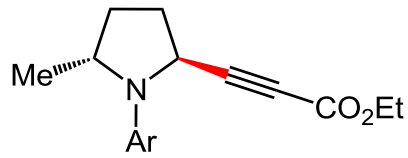
5b: 69% yield, 93:7 e.r.



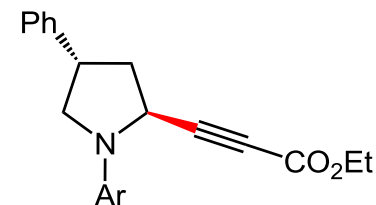
5c: 64% yield, 95:5 e.r.



5p: 64% yield, 84:16 e.r.

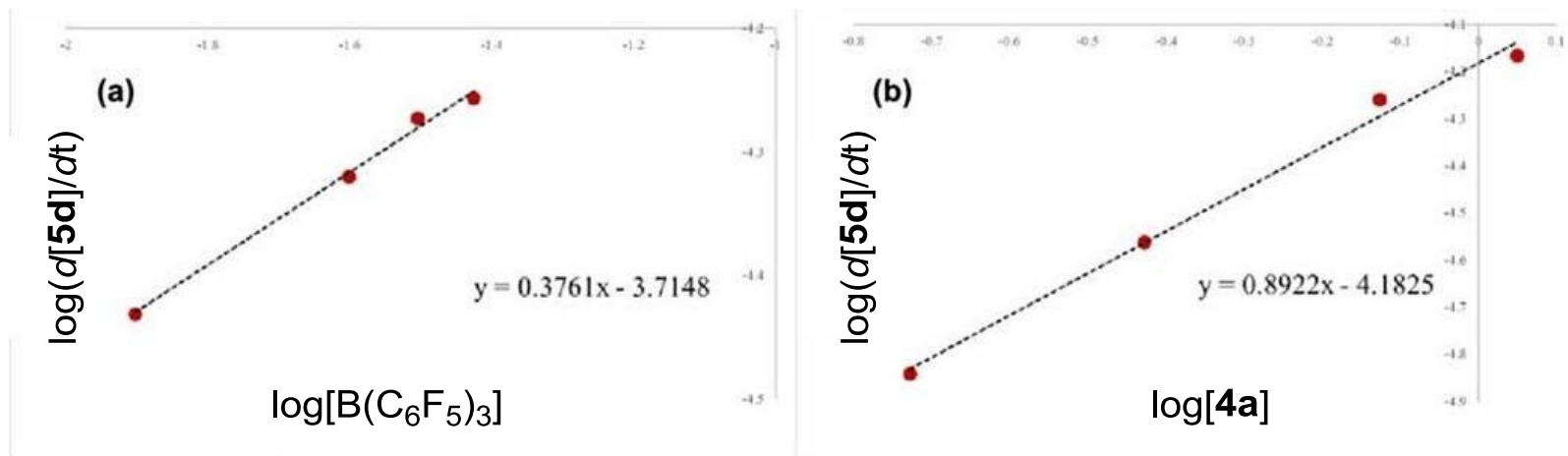
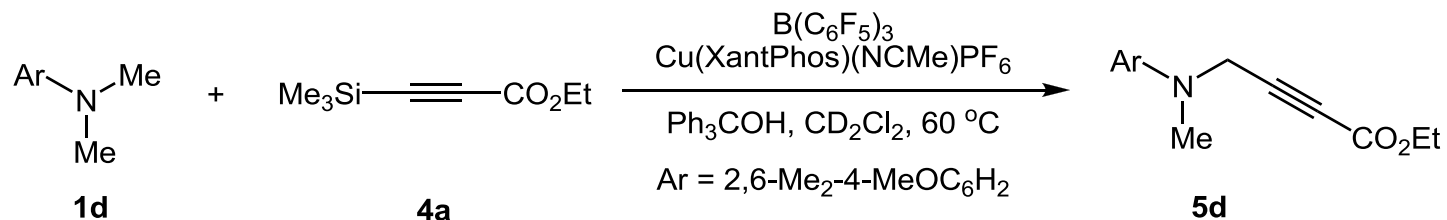


5q: 66% yield
6.3:1 (*trans*:*cis*)
83:17 e.r. (*trans*)



5r: 68% yield
10.1:1 (*trans*:*cis*)
88:12 e.r. (*trans*)

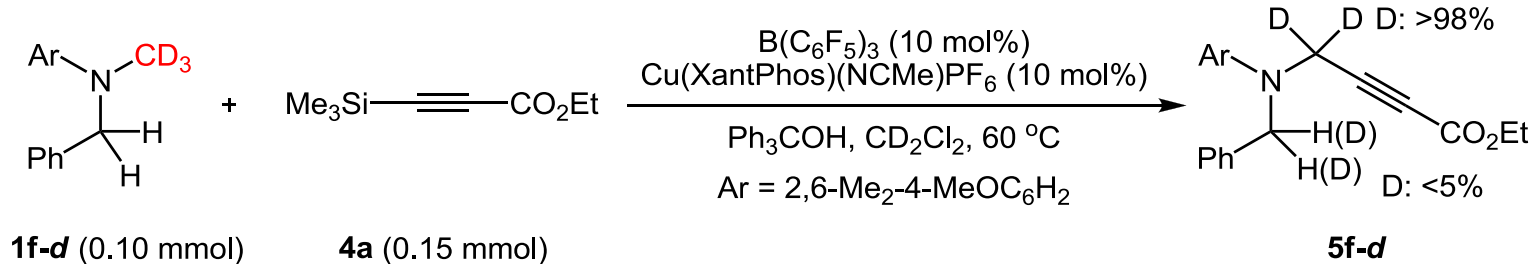
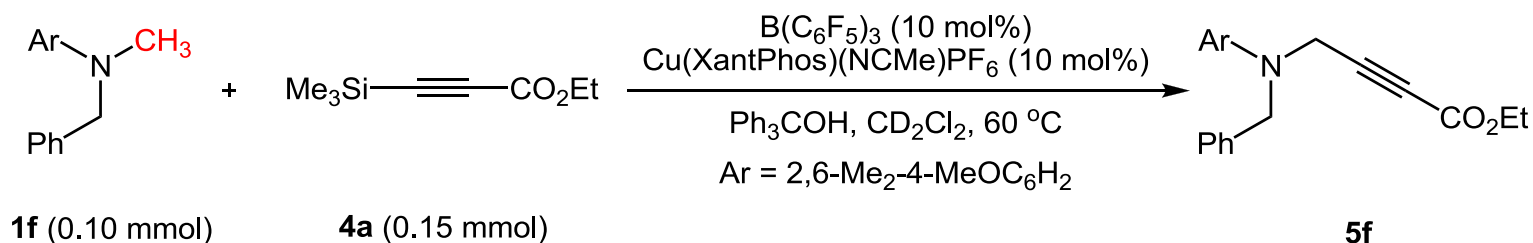
Kinetic Studies



The reaction of **1d** and **4a** to afford propargylamine **5d** was found to be 0.5-order in $B(C_6F_5)_3$ and 1.0-order in alkyne.

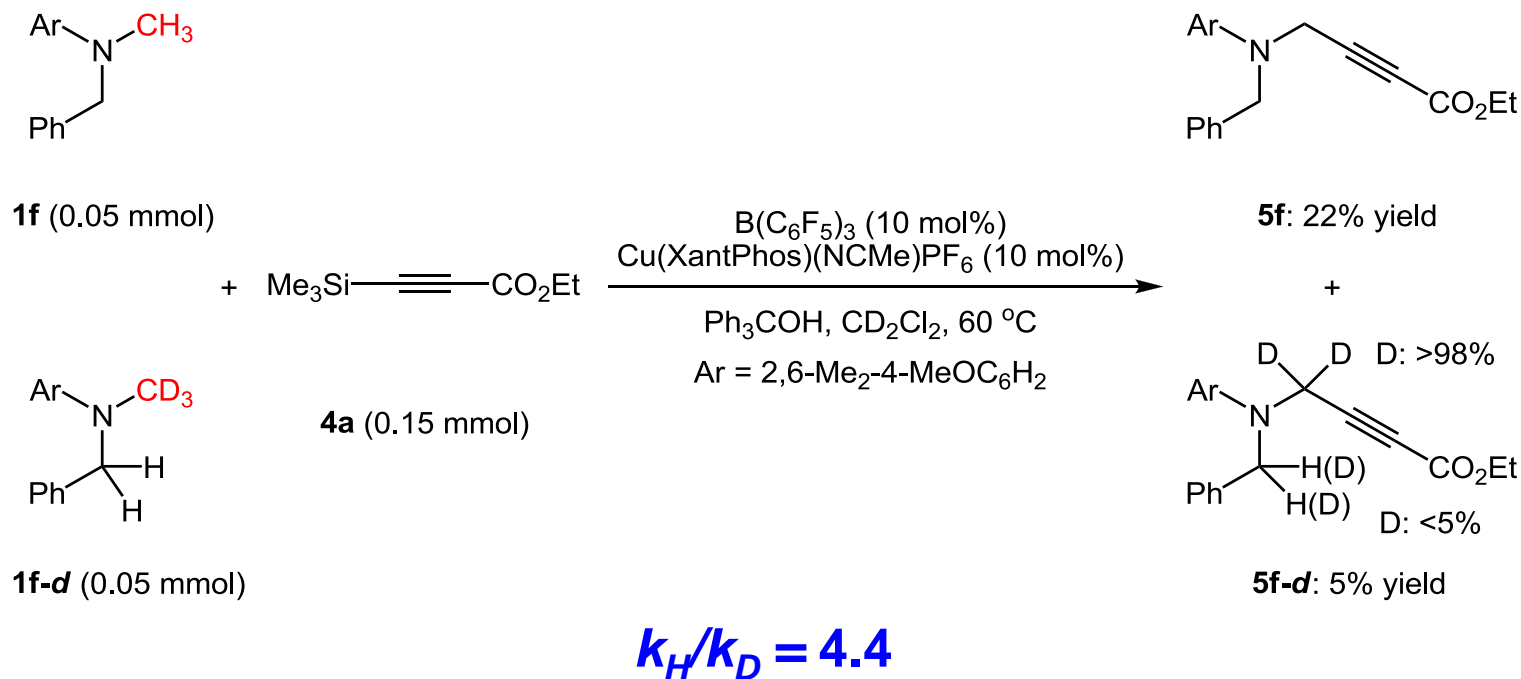
These data imply that C-H bond cleavage through $B(C_6F_5)_3$ catalyzed hydride abstraction occurs after the turnover-limiting step.

Kinetic Isotope Effect Studies



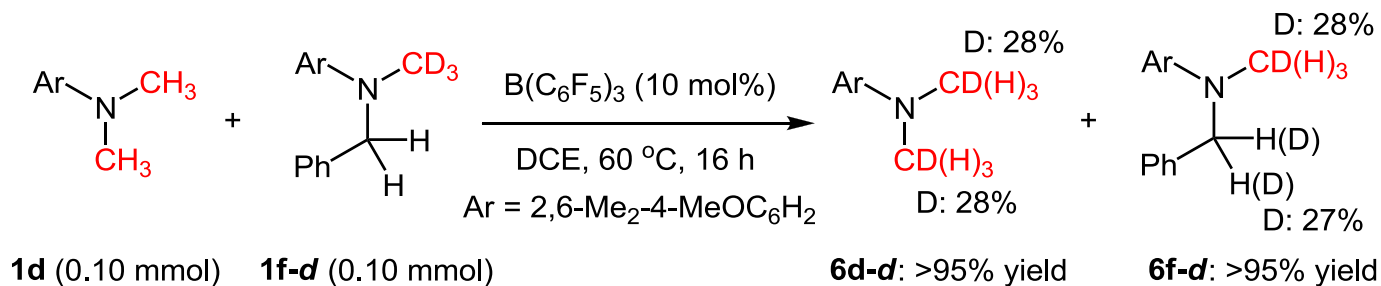
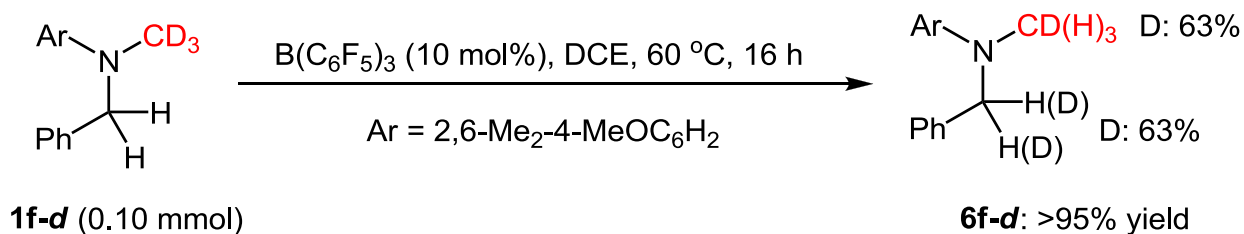
$$k_H/k_D = 1.02 \pm 0.02$$

Kinetic Isotope Effect Studies



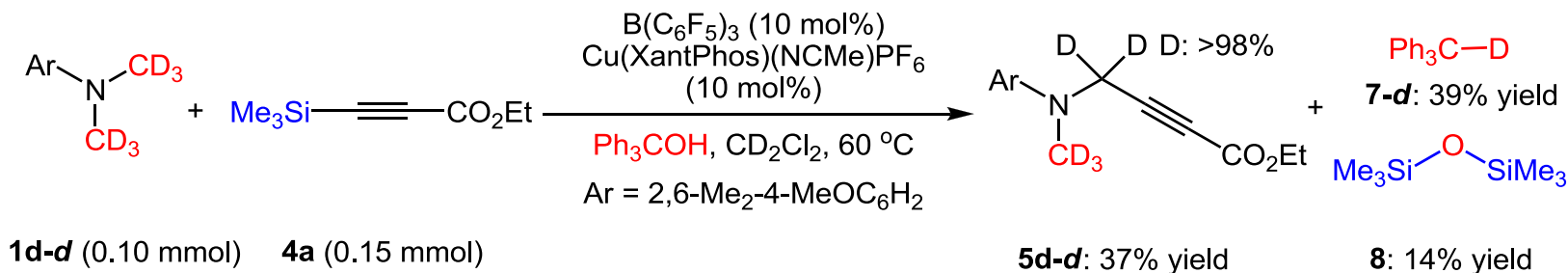
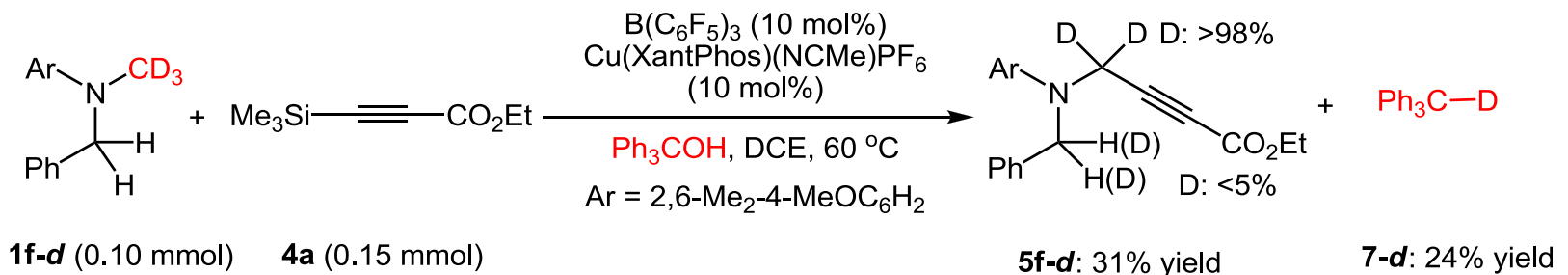
These data imply that the turnover-limiting step is before the (F₅C₆)₃B-catalyzed hydride abstraction and that the C-H bond cleavage step is irreversible.

Isotope Experiments

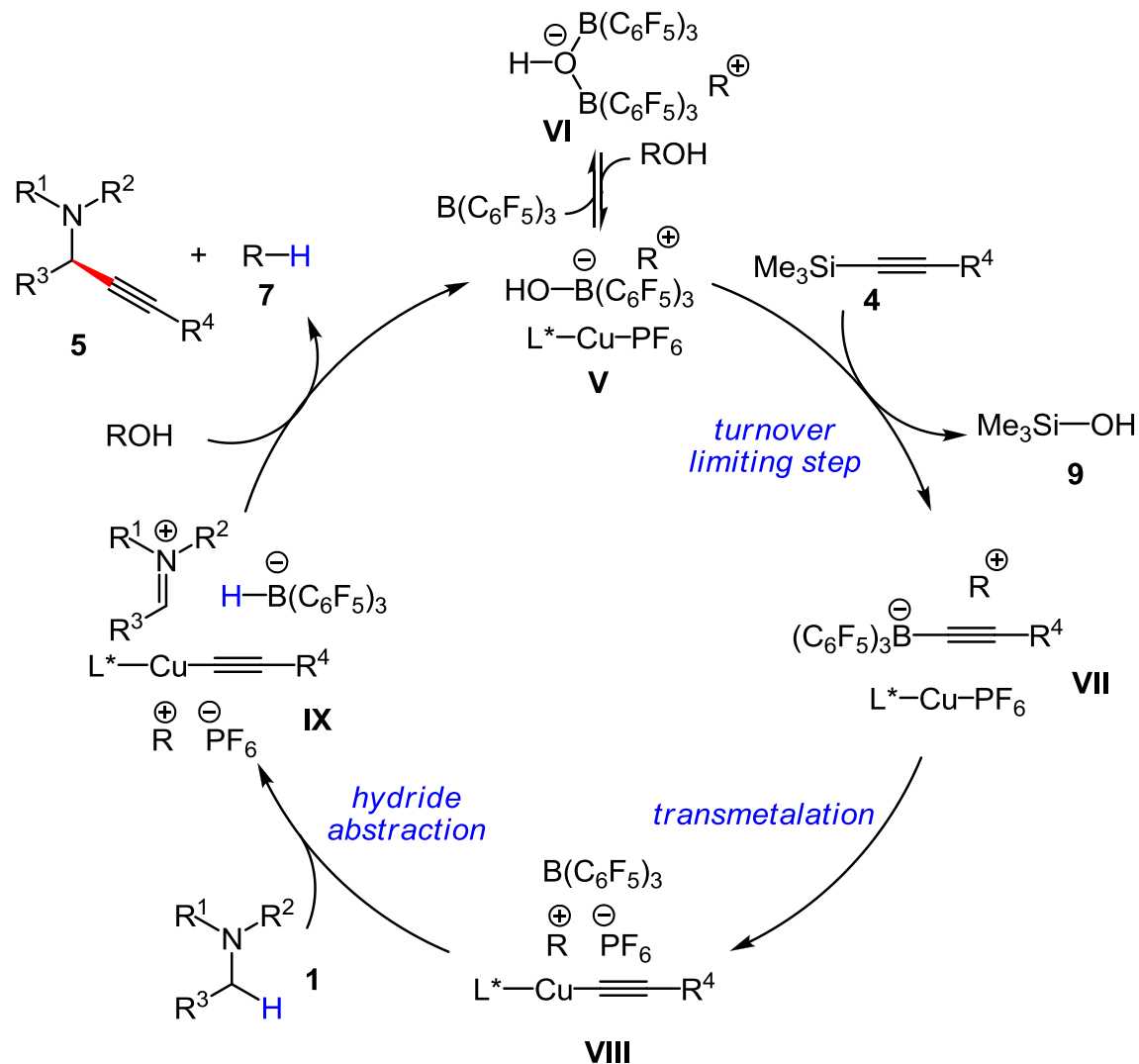


C-H bond cleavage step is irreversible

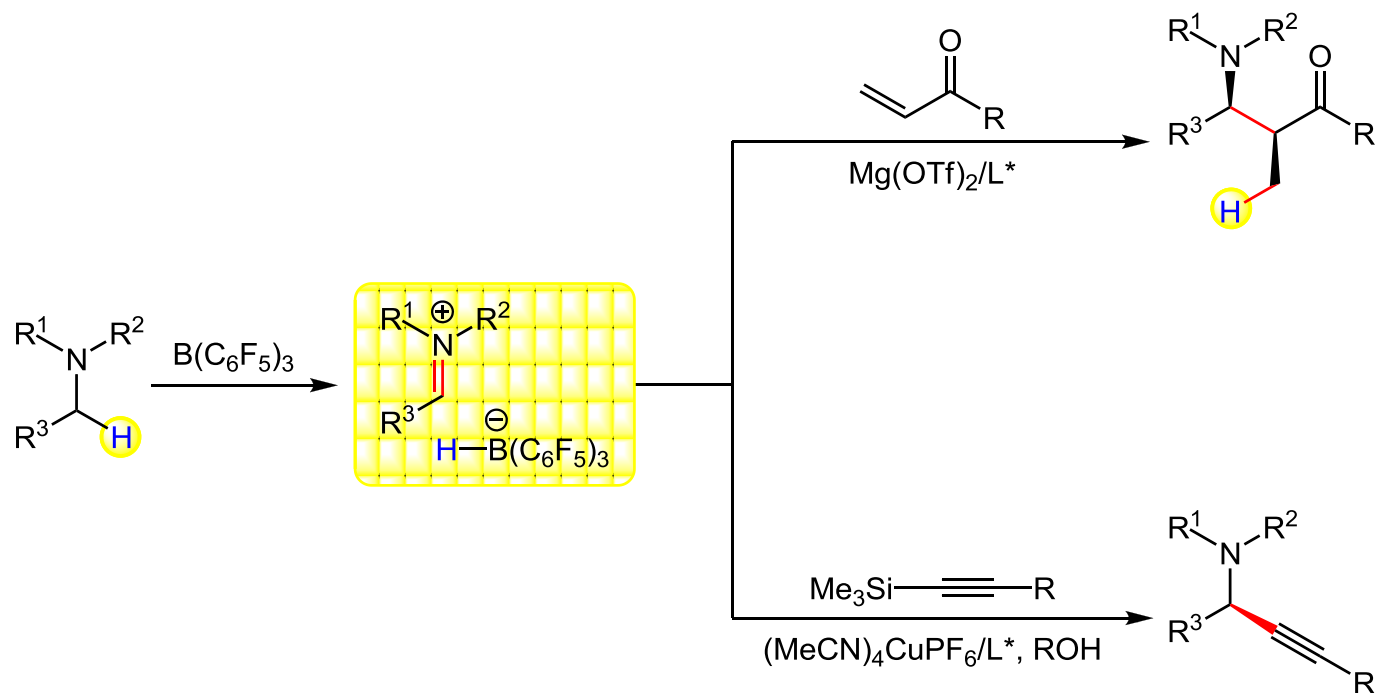
Structure Determination of Products



Mechanism of Catalytic Process



Summary



Wasa, M. *et al.* *J. Am. Chem. Soc.* **2018**, *140*, 10593;
J. Am. Chem. Soc. **2020**, *142*, 16493.

The First Paragraph

Writing Strategy

The importance of propargylamines



**Methods for the preparation
of chiral propargylamines**



**The advantages of direct conversion
of alkylamines to propargylamines**

The First Paragraph

Propargylamines are prevalent in pharmaceuticals and are commonly used as intermediates in the synthesis of bioactive amines. Enantiomerically enriched propargylamines have been prepared by the addition of an alkynylmetal compound to an imine. An attractive alternative would entail the conversion of an α -amino C(sp³)-H bond into an α -C-alkynyl bond. One way to accomplish this would be through in situ generation of an iminium ion intermediate formed from the corresponding amine under oxidative conditions. An illustrative case is enantioselective Cu-PyBOX-catalyzed coupling of a benzylic α -amino C-H bond of *N*-phenyl tetrahydroisoquinoline **1a** with ethynylbenzene **2a** to afford propargylamine **3a**.

The First Paragraph

Still, development of a nonprecious transition metal- and oxidant-free catalytic C-H functionalization process represents a compelling research objective. Particularly noteworthy would be the direct conversion of α -C-H bonds contained in bioactive *N*-alkylamines into α -C-alkynyl bonds, because these entities constitute over 50% of the top-selling drugs; the resulting derivatives of these pharmaceuticals possessing the alkyne unit can serve as modifiable intermediates for late-stage structural diversification that could lead to new leads and/or more effective therapeutics.

The Last Paragraph

Writing Strategy

Summary of this work



Significance of this work

The Last Paragraph

In summary, we developed an efficient and diastereo- and enantioselective method for the activation of α -amino C-H bonds to generate propargylamines. We find that by using a blend of $B(C_6F_5)_3$ and an organocopper complex, it is possible to generate an iminium from a *N*-alkylamine, and a L_nCu -alkynyl complex from an alkynyl-silane. The catalyst system tolerates a wide variety of Lewis acid-sensitive functional groups and is therefore applicable to the late-stage transformation of a complex (and bioactive) trialkyl amine molecule to its derived propargylamine. Mechanistic investigations indicate that the turnover-limiting step occurs prior to $(F_5C_6)_3B$ -catalyzed C-H abstraction and that $(F_5C_6)_3B$ catalyzed C-H abstraction is an irreversible process under the reaction conditions for alkyne incorporation.

The Last Paragraph

The principles outlined here demonstrate that the proper combination of an achiral organoborane and a chiral organometallic catalyst can be used for chemo- and enantioselective C-H bond activation, providing a rational framework for the further development of processes involving the late-stage stereoselective α -functionalization of bioactive amines. **Studies aimed at achieving these objectives are currently underway.**

Representative Examples

An attractive alternative would entail the conversion of an α -amino C(sp³)-H bond into an α -C-alkynyl bond. (阐述合成方法)

Still, development of a nonprecious transition metal- and oxidant-free catalytic C-H functionalization process represents a compelling research objective. (引出方法的重要性)

To begin, we set out to identify a suitable combination of catalysts. (条件优化)

An assortment of cyclic and acyclic *N*-alkylanilines may be used in reaction with 3-(trimethylsilyl)propiolate to generate the corresponding propargylamines. (底物拓展)

We designed and performed studies aimed at shedding light on the mechanism of the catalytic process. (机理研究)

Acknowledgement

***Thanks
for your attention***