

Literature Report 5

Catalytic Kinetic Resolution of Monohydrosilanes *via* Rhodium-Catalyzed Enantioselective Intramolecular Hydrosilylation

Reporter: Hao-Dong Chen

Checker: Yan-Jiang Yu

Date: 2024-06-24

Guo, F.-H.; Ren, F.; Wu, Y.; Wang, P. *Angew. Chem. Int. Ed.* **2024**, 63, e202404732

CV of Prof. Peng Wang (王鹏)



Education & Experiment:

- 2003-2007 B.S., Tianjin University
- 2008-2013 Ph.D., SIOC (Prof. Yong Tang)
- 2013-2017 Research Associate, The Scripps Research Institute
- 2018-Present Professor, SIOC

Research

- Chiral Ligand Development
- Main Group Chemistry
- Reaction Discovery

Contents

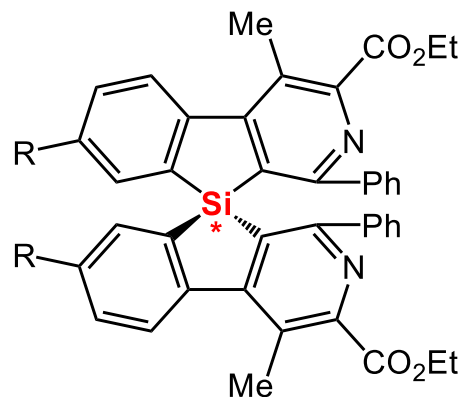
1 Introduction

2 Rh-Catalyzed Catalytic Kinetic Resolution of Monohydrosilanes

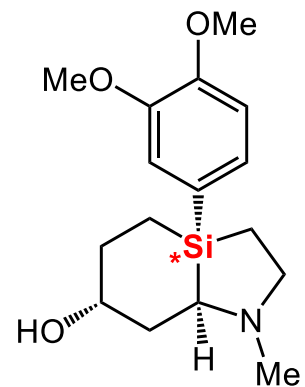
3 Summary

Introduction

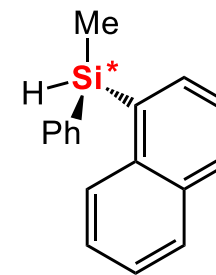
Function Organosilanes Bearing Enantioenriched Stereogenic Silicon Centers



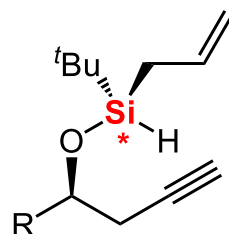
CPL-active molecule
(CPL = circularly polarized luminescence)



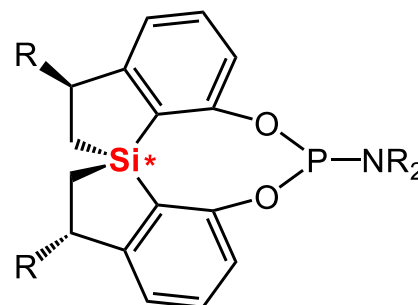
(-)-mesembranol
(antidepressant)



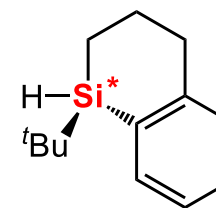
mechanistic probe



chiral auxiliaries



chiral ligand

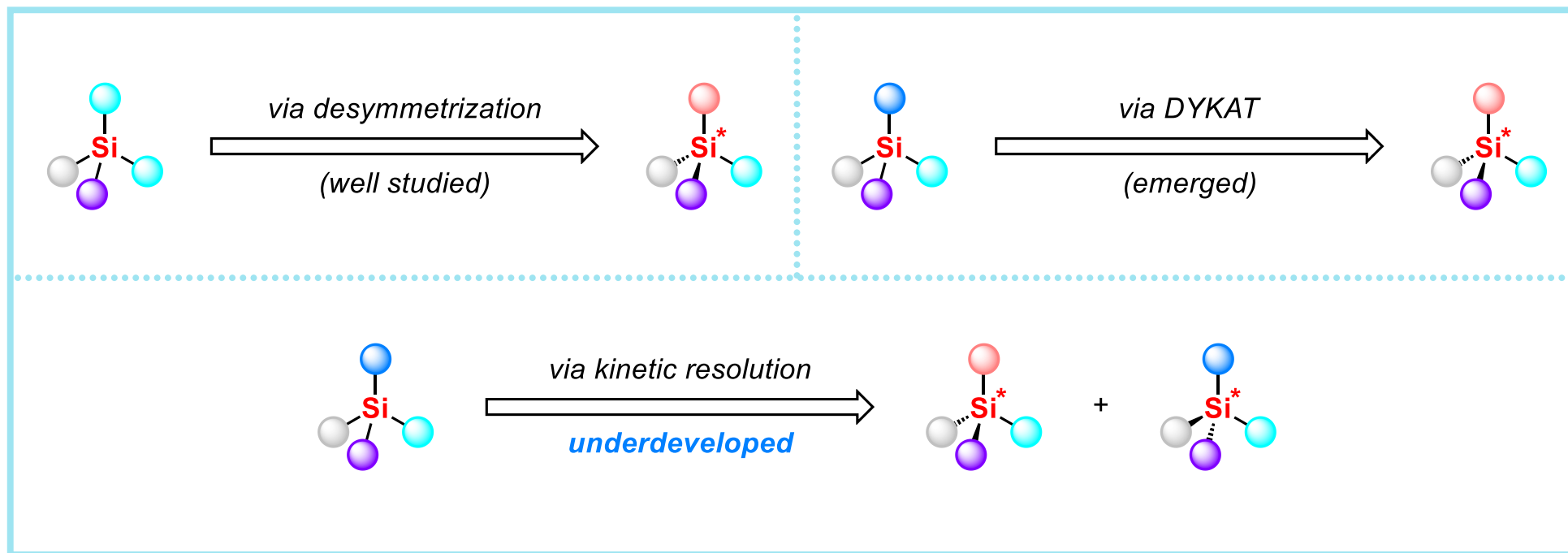


chiral reagent

Hu, T.; Zhao, C.; Zhang, Y.; Kuang, Y.; Gao, L.; Wang, W.; Su, Z.; Song, L. *Nat. Commun.* **2023**, *14*, 4900

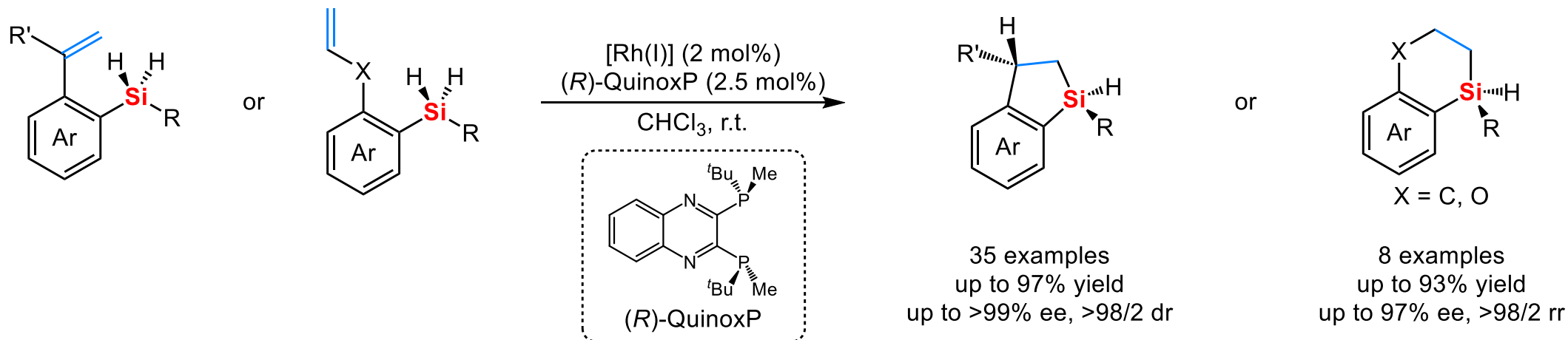
Introduction

Approaches for Catalytic Construction of Si-Stereogenic Organosilanes

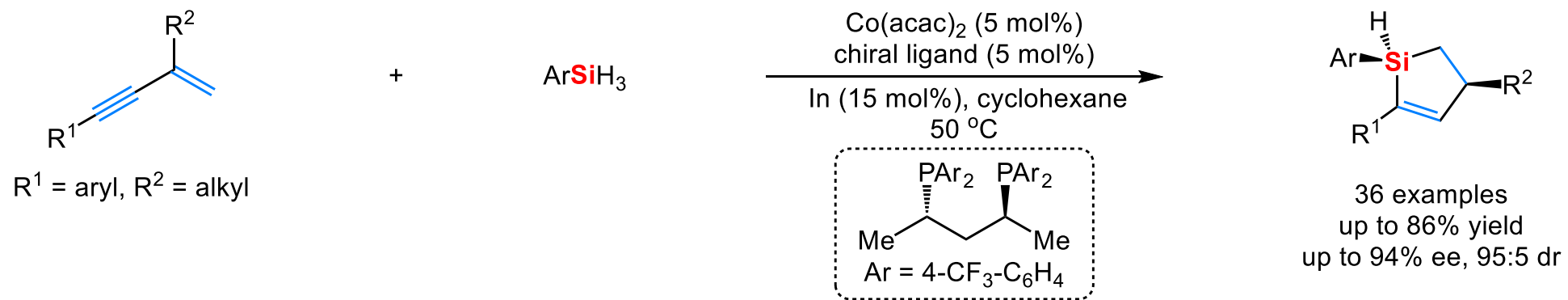


Introduction

Catalytic Construction of Si-Stereogenic Organosilanes via Desymmetrization



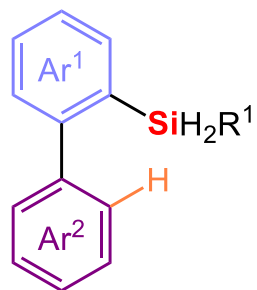
Huang, Y.-H.; Wu, Y.; Wang, P. *et al. Angew. Chem. Int. Ed.* **2022**, 61, e202113052



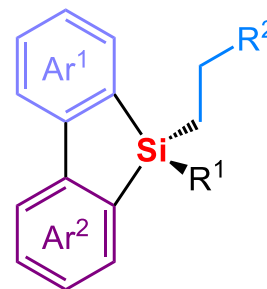
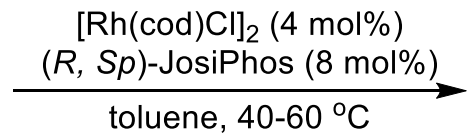
Lu, W.; Zhao, Y.; Meng, F. *J. Am. Chem. Soc.* **2022**, 144, 5233

Introduction

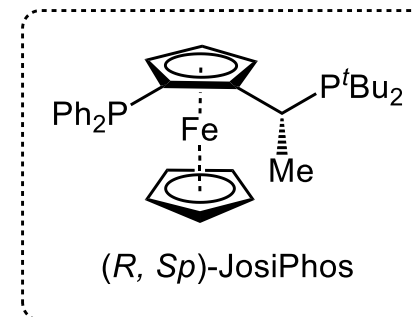
Catalytic Construction of Si-Stereogenic Organosilanes via Desymmetrization



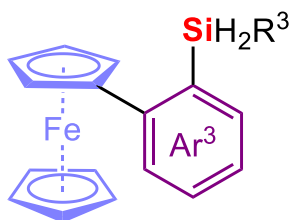
+



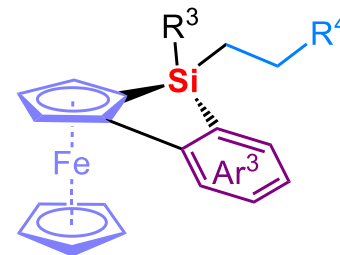
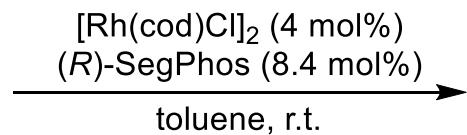
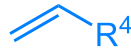
28 examples
up to 88% yield, >99% ee



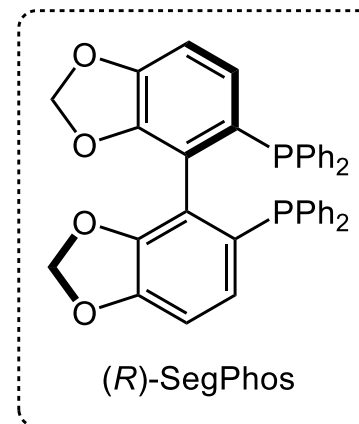
(*R, Sp*)-JosiPhos



+



35 examples
up to 94% yield, 99% ee

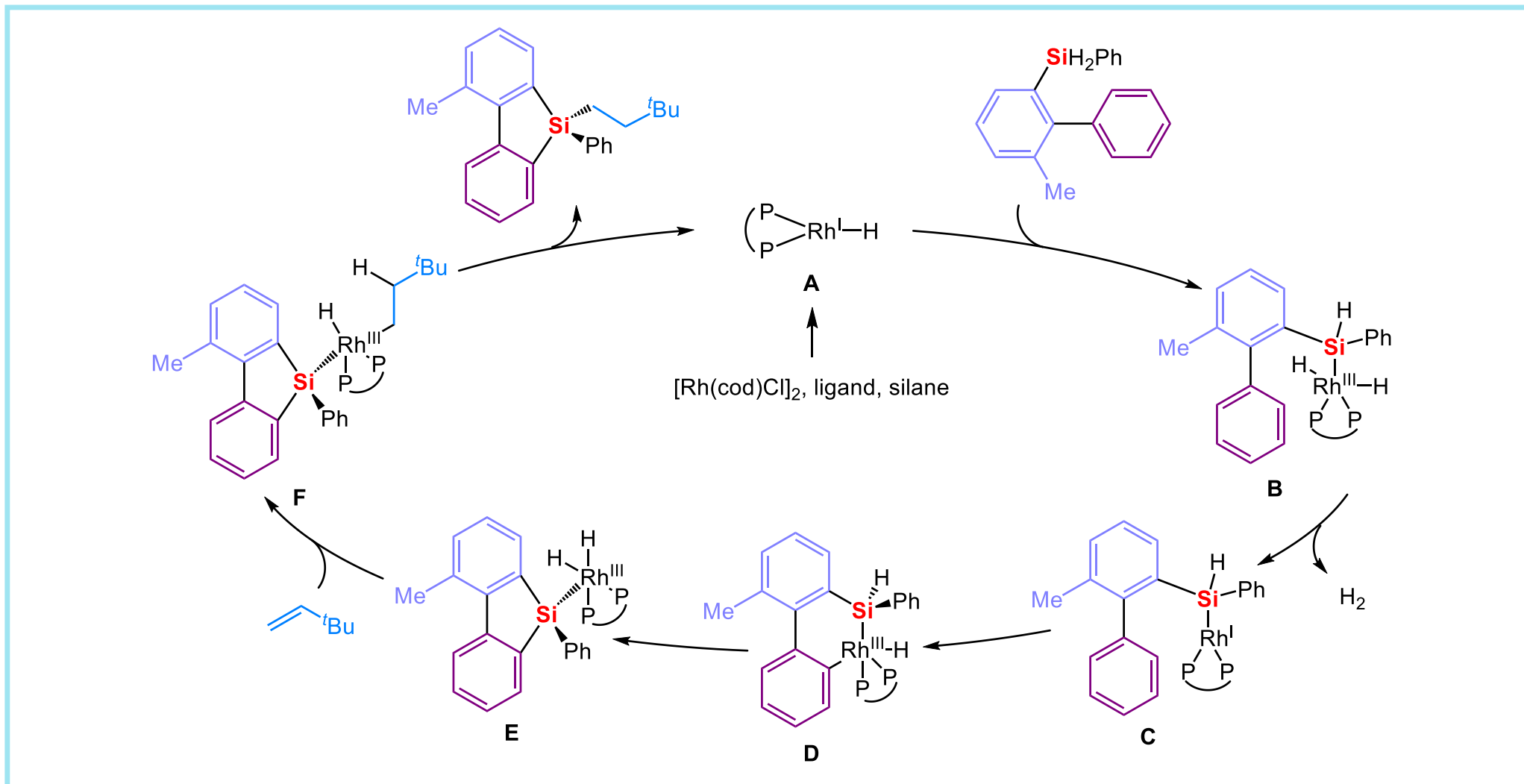


(*R*)-SegPhos

Mu, D.; Yuan, W.; He, C. *et al. J. Am. Chem. Soc.* **2020**, *142*, 13459

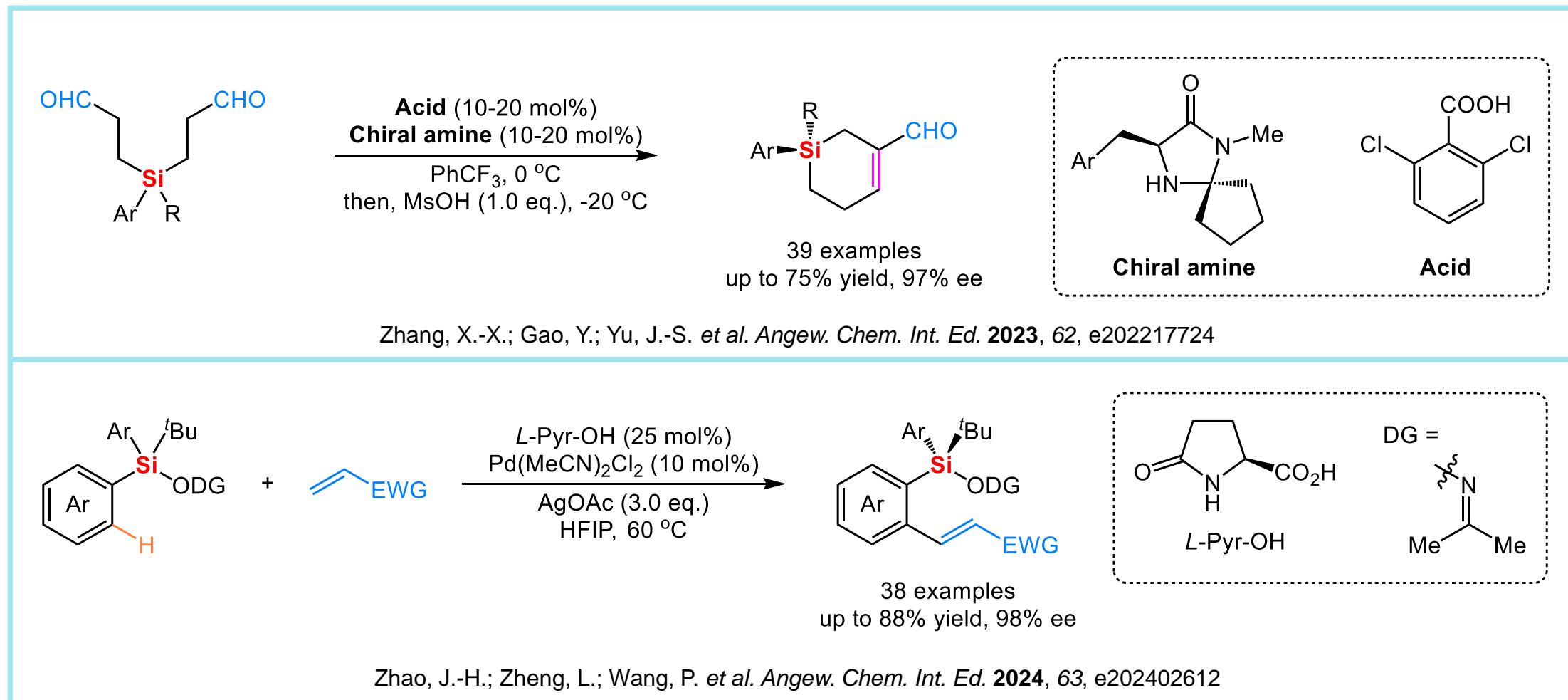
Introduction

Catalytic Construction of Si-Stereogenic Organosilanes via Desymmetrization



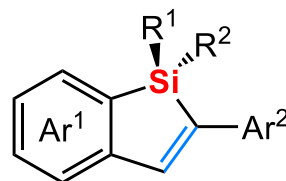
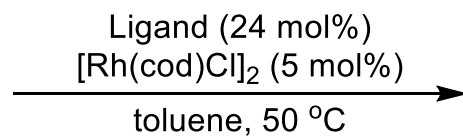
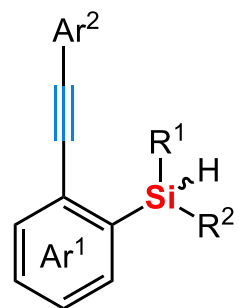
Introduction

Catalytic Construction of Si-Stereogenic Organosilanes via Desymmetrization

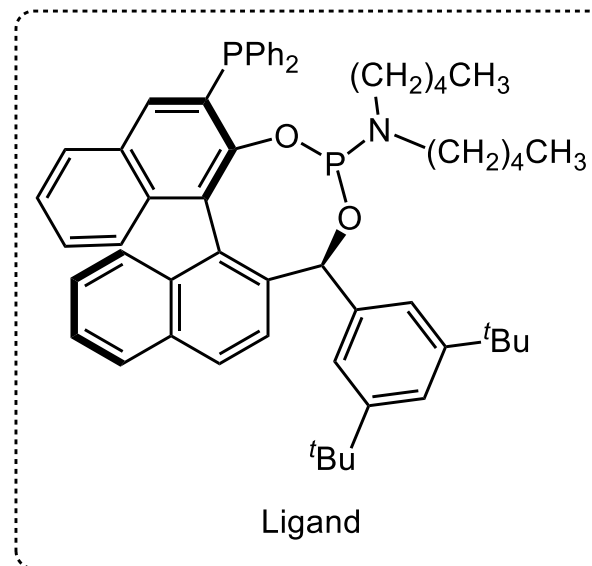


Introduction

Catalytic Construction of Si-Stereogenic Organosilanes via DYKAT



22 examples
up to 89% yield, 92% ee

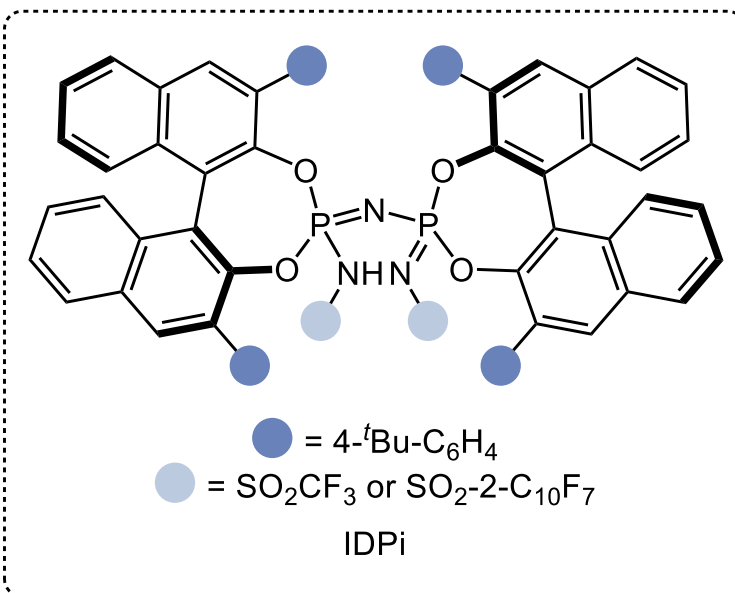


Ligand

Zeng, Y.; Fang, X.-J.; Xu, L.-W. *et al. Angew. Chem. Int. Ed.* **2022**, 61, e202214147

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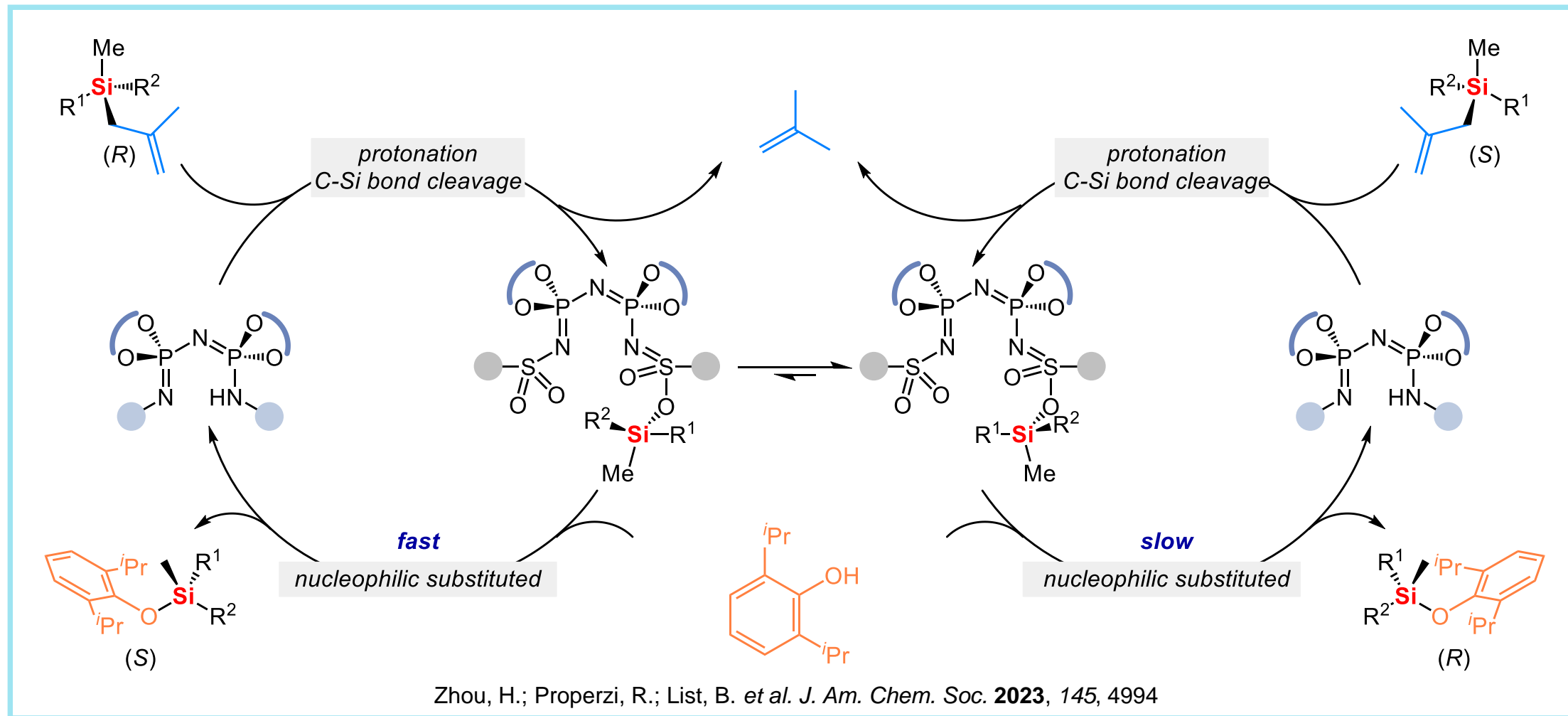
Catalytic Construction of Si-Stereogenic Organosilanes via DYKAT



Zhou, H.; Properzi, R.; List, B. *et al. J. Am. Chem. Soc.* **2023**, *145*, 4994

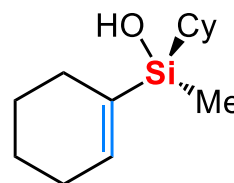
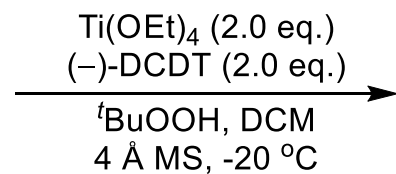
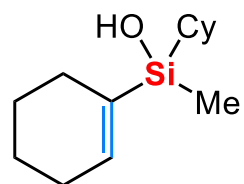
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Catalytic Construction of Si-Stereogenic Organosilanes via DYKAT



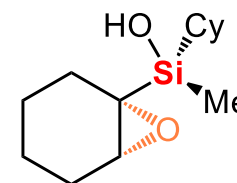
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Construction of Si-Stereogenic Organosilanes via KR



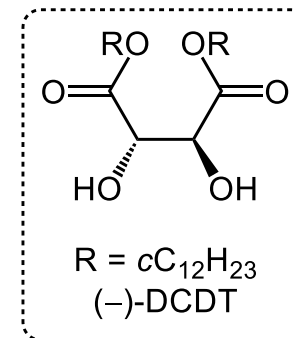
99% ee

+

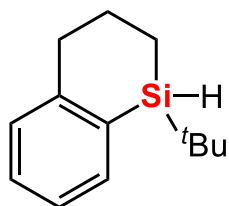


95:5 dr

$s > 11$

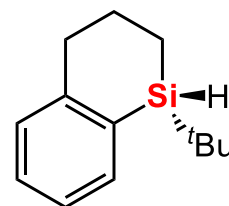
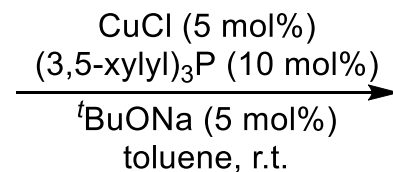


Yamamoto, K.; Kawanami, Y.; Miyazawa, M. *J. Chem. Soc. Chem. Commun.* **1993**, 436



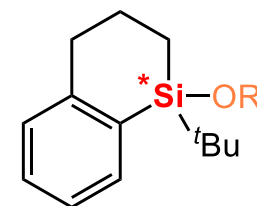
+

R*OH



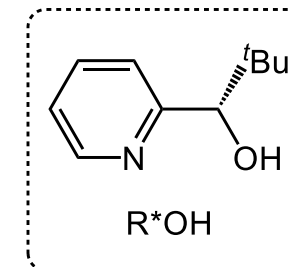
52% ee

+



76:24 dr

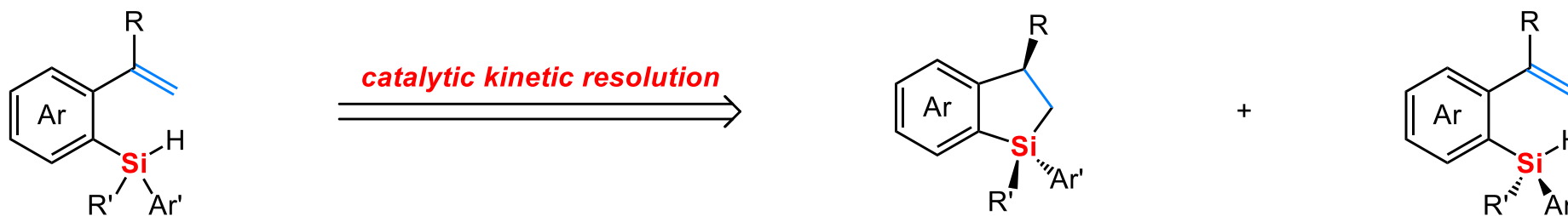
$s = 4.9$



Rendler, S.; Auer, G.; Oestreich, M. *et al. Adv. Synth. Catal.* **2006**, 348, 1171

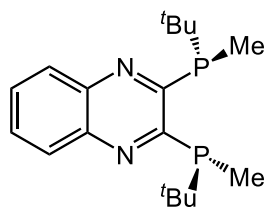
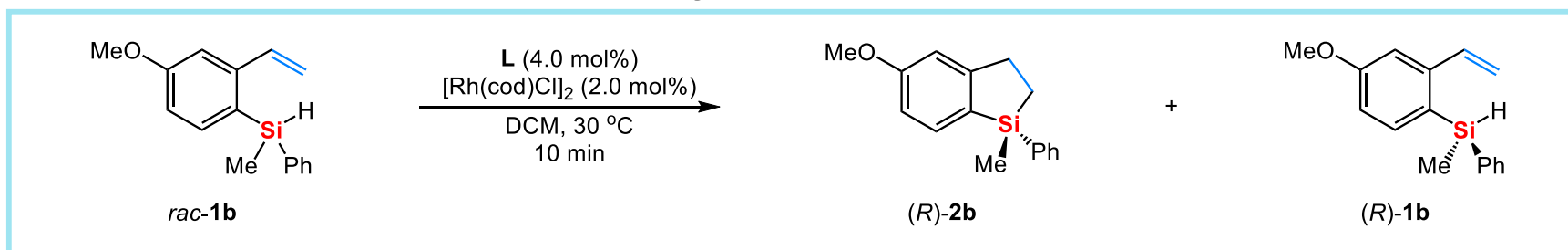
Project Synopsis

This Work: Access of Si-Stereogenic Silane via Catalytic Kinetic Resolution of Monohydrosilane

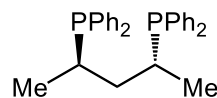


Optimization of the Reaction Conditions

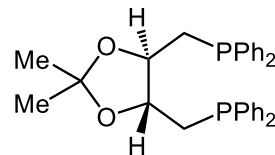
Ligand Evaluation



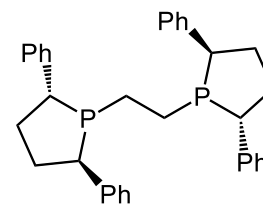
L1, (R)-QuinoxP
2b, 79%, 7% ee
1b, 10%, -0.4% ee
c = 89%, *s* = 1



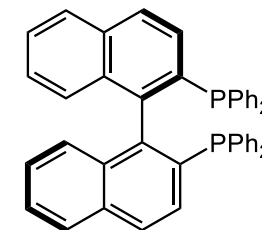
L2, (R)-BDPP
2b, 20%, -8% ee
1b, 63%, -5% ee
c = 24%, *s* = 2



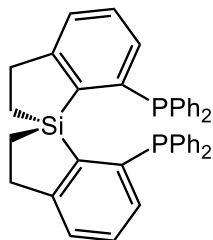
L3, (R)-DIOP
2b, 9%, 28% ee
1b, 74%, 4% ee
c = 11%, *s* = 2



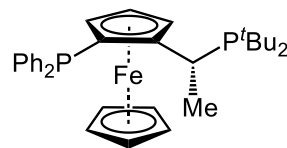
L4, (R,R)-Ph-BPE
2b, 5%, 45% ee
1b, 71%, -14% ee
c = 7%, *s* = 4



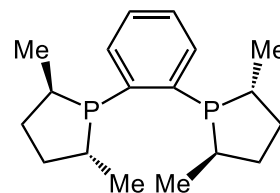
L5, (R)-BINAP
2b, 4%, 7% ee
1b, 79%, 4% ee
c = 5%, *s* = 8



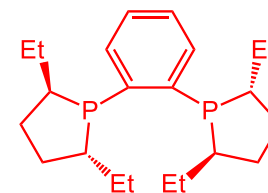
L6, (S)-SPSIP
2b, 16%, -5% ee
1b, 84%, 1% ee
c = 16%, *s* = 1



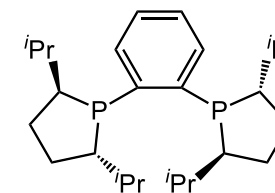
L7
2b, 5%, -5% ee
1b, 72%, 25% ee
c = 6%, *s* = 2



L8, (R,R)-Me-DuPhos
2b, 62%, 46% ee
1b, 27%, >99% ee
c = 70%, *s* = 11

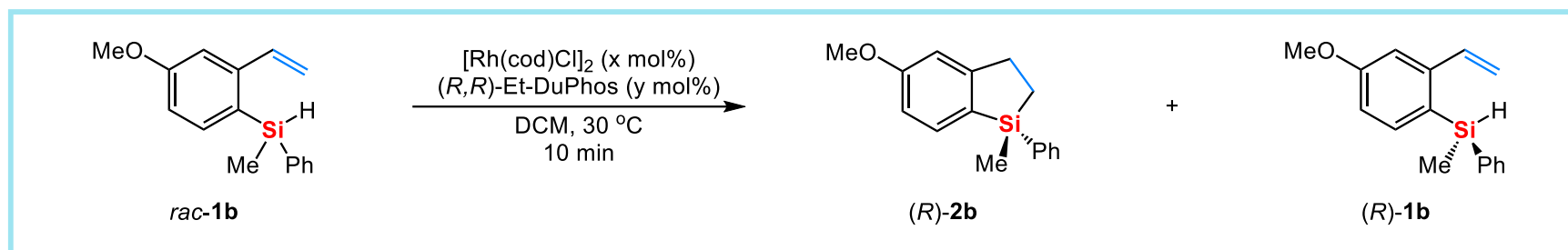


L9, (R,R)-Et-DuPhos
2b, 50%, 85% ee
1b, 45%, 95% ee
c = 53%, *s* = 43



L10, (R,R)-iPr-DuPhos
2b, 3%, 73% ee
1b, 92%, 2% ee
c = 3%, *s* = 5

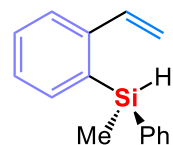
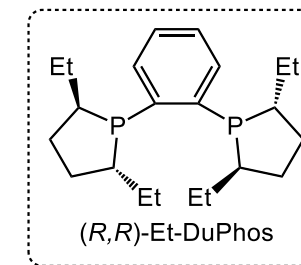
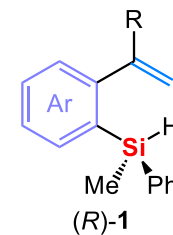
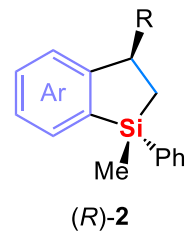
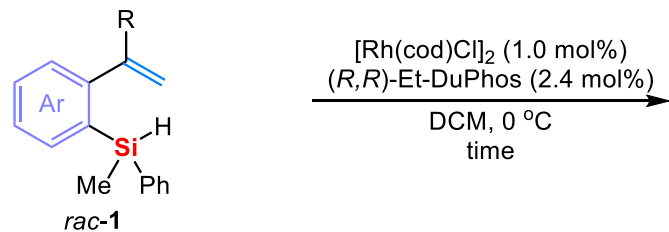
Optimization of the Reaction Conditions



Entry	Solvent	X	Y	<i>(R)</i> -2b		<i>(S)</i> -1b		C (%)	S
				Yield (%)	Ee (%)	Yield (%)	Ee (%)		
1	DCM	2.0	4.0	50	85	45	95	53	43
2	CHCl ₃	2.0	4.0	16	83	77	14	17	6
3	DCE	2.0	4.0	43	83	47	77	48	26
4	THF	2.0	4.0	12	74	81	10	13	6
5	PhMe	2.0	4.0	10	66	83	7	11	4
6	DCM	2.0	4.0	50	85	45	95	53	43
7	DCM	1.0	2.0	42	93	53	75	44	97
8	DCM	1.0	2.4	43	94	57	71	43	70
9 ^[a]	DCM	1.0	2.4	40	95	60	64	40	95

Reaction conditions: *rac*-1b (0.1 mmol), $[\text{Rh}(\text{cod})\text{Cl}]_2$ (x mol%), (R,R) -Et-DuPhos (y mol%), DCM (1.0 mL), 30 °C, N₂, 10 min. The yield was determined by ¹H NMR using dibromomethane as the internal standard. The ee value was determined by chiral HPLC. Calculated conversion, $c = \text{yield}_{\text{PR}} / (\text{yield}_{\text{SM}} + \text{yield}_{\text{PR}})$. Selectivity, $s = \text{Ln}[(1-c)(1-\text{ee}_{\text{SM}})] / \text{Ln}[(1-c)(1+\text{ee}_{\text{SM}})]$. ^[a]The reaction was conducted at 0 °C, for 20 min.

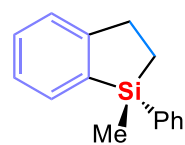
Substrate Scope



1a

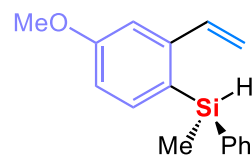
44%, 96% ee

50 min, *c* = 51%, *s* = 97



2a

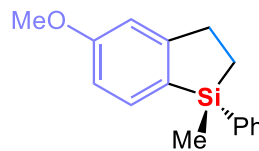
49%, 84% ee



1b

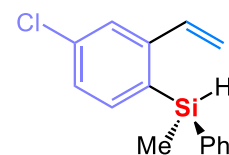
45%, 98% ee

90 min, *c* = 53%, *s* = 65



2b

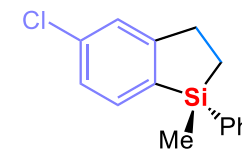
49%, 88% ee



1c

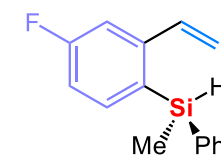
48%, 94% ee

45 min, *c* = 52%, *s* = 50



2c

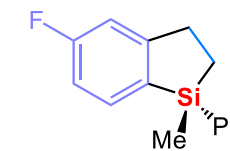
46%, 77% ee



1d

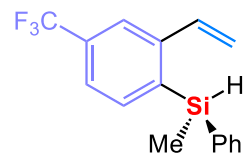
51%, 90% ee

90 min, *c* = 49%, *s* = 95



2d

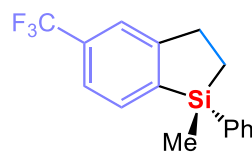
46%, 89% ee



1e

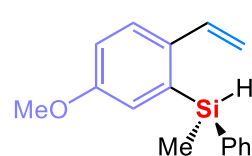
45%, 81% ee

50 min, *c* = 50%, *s* = 29



2e

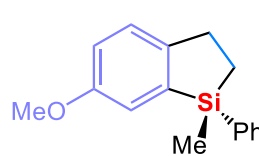
52%, 76% ee



1f

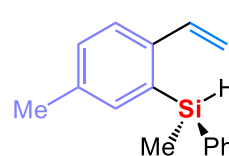
50%, 89% ee

90 min, *c* = 49%, *s* = 78



2f

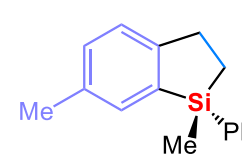
50%, 92% ee



1g

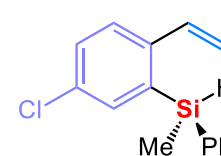
48%, 94% ee

20 min, *c* = 50%, *s* = 115



2g

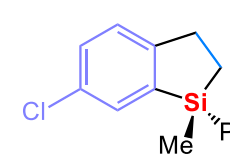
46%, 92% ee



1h

50%, 91% ee

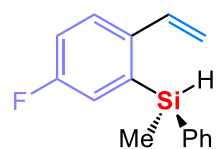
50 min, *c* = 50%, *s* = 67



2h

46%, 85% ee

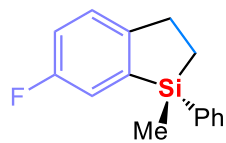
Substrate Scope



1i

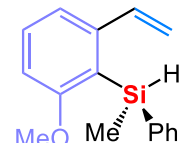
49%, 84% ee

50 min, $c = 49\%$, $s = 39$



2i

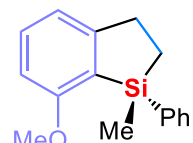
47%, 87% ee



1j

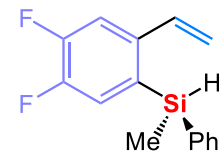
47%, 73% ee

45 min, $c = 49\%$, $s = 16$



2j

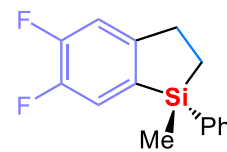
47%, 79% ee



1k

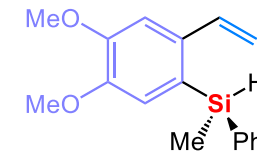
47%, 87% ee

55 min, $c = 50\%$, $s = 41$



2k

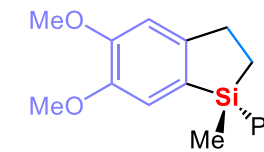
49%, 80% ee



1l

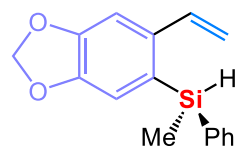
48%, 88% ee

50 min, $c = 48\%$, $s = 122$



2l

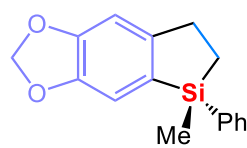
43%, 91% ee



1m

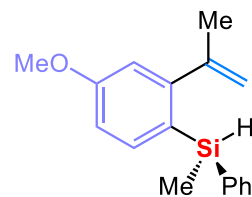
48%, 92% ee

40 min, $c = 50\%$, $s = 79$



2m

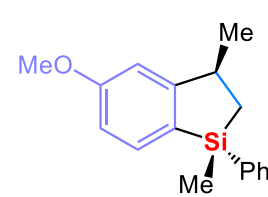
52%, 91% ee



1n

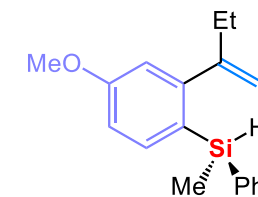
51%, 84% ee

30 °C, 5 h, $c = 50\%$, $s = 30$



2n

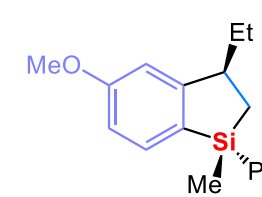
44%, 90% ee, 21:1 dr



1o

49%, 68% ee

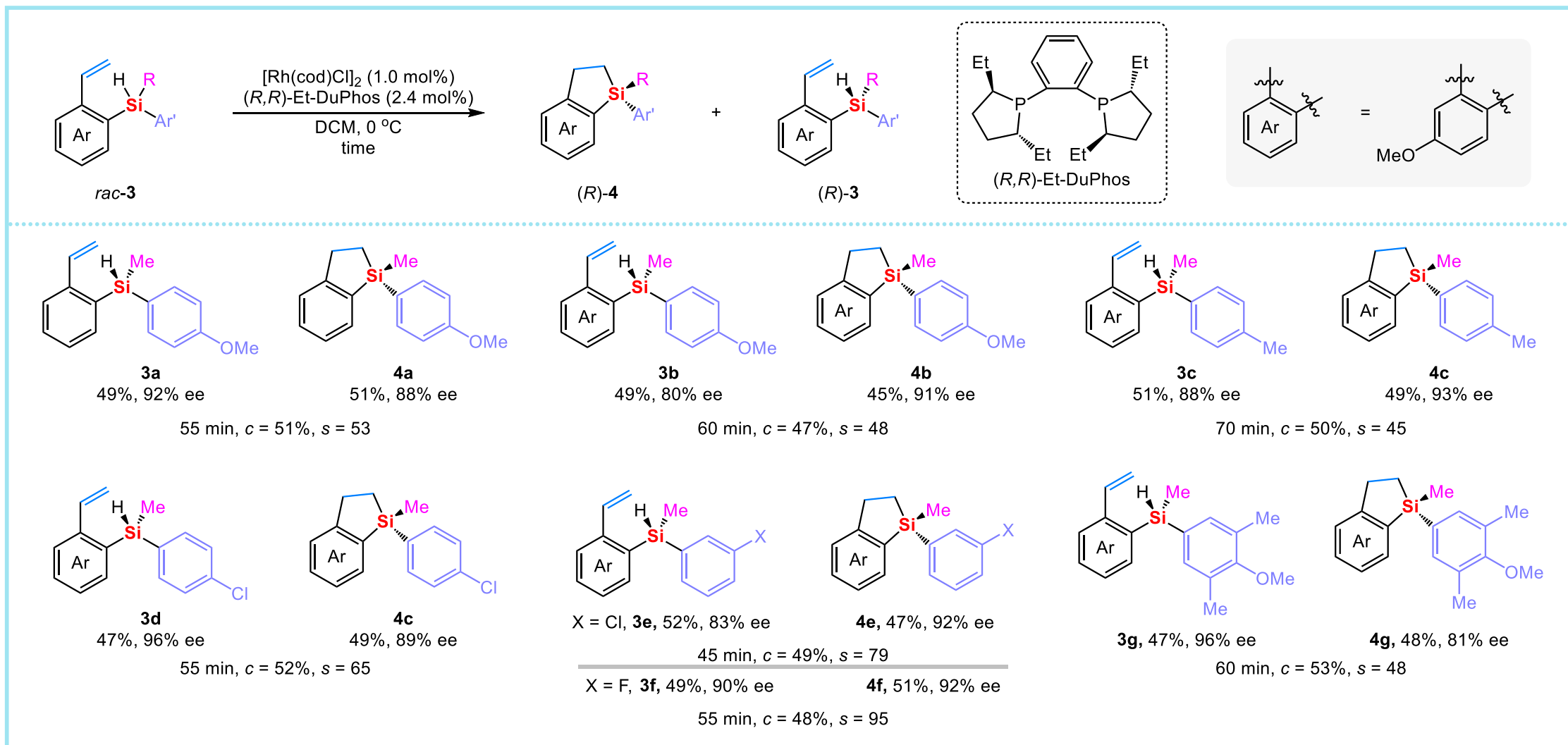
30 °C, 9 h, $c = 47\%$, $s = 15$



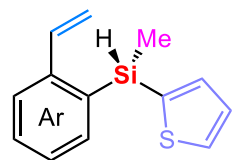
2o

51%, 86% ee, 12:1 dr

Substrate Scope

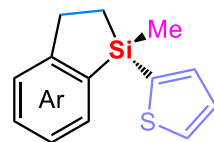


Substrate Scope



3h

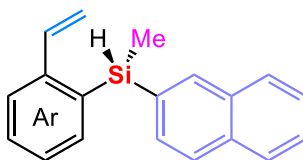
46%, 87% ee



4h

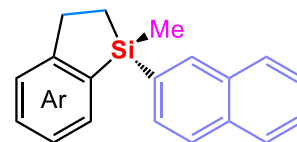
50%, 87% ee

50 min, $c = 50\%$, $s = 41$



3i

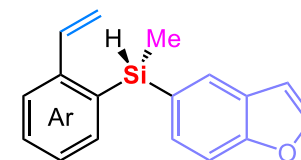
52%, 81% ee



4i

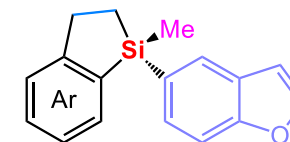
47%, 83% ee

60 min, $c = 50\%$, $s = 24$



3j

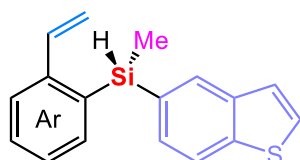
49%, 93% ee



4j

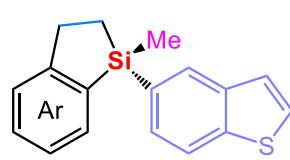
50%, 89% ee

75 min, $c = 51\%$, $s = 60$



3k

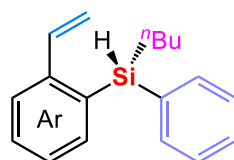
49%, 91% ee



4k

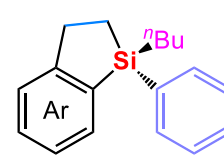
49%, 88% ee

80 min, $c = 51\%$, $s = 47$



3l

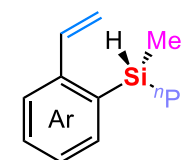
48%, 74% ee



4l

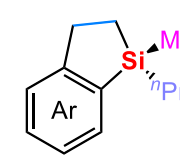
45%, 75% ee

6 h, $c = 48\%$, $s = 17$



3m

52%, 53% ee



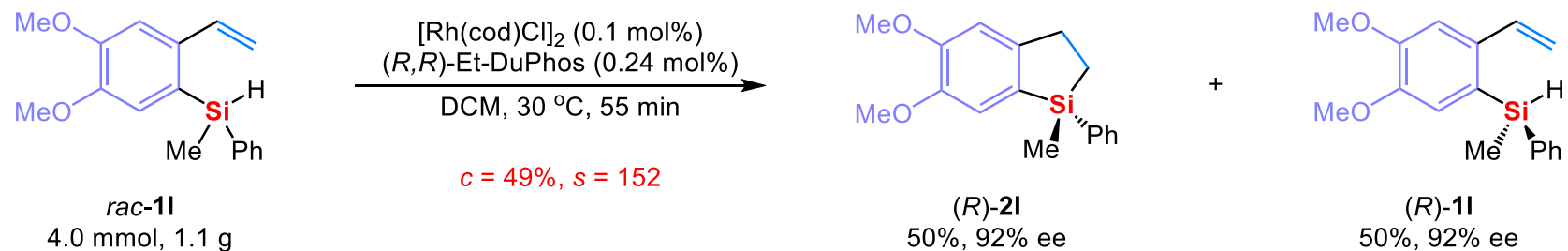
4m

46%, 56% ee

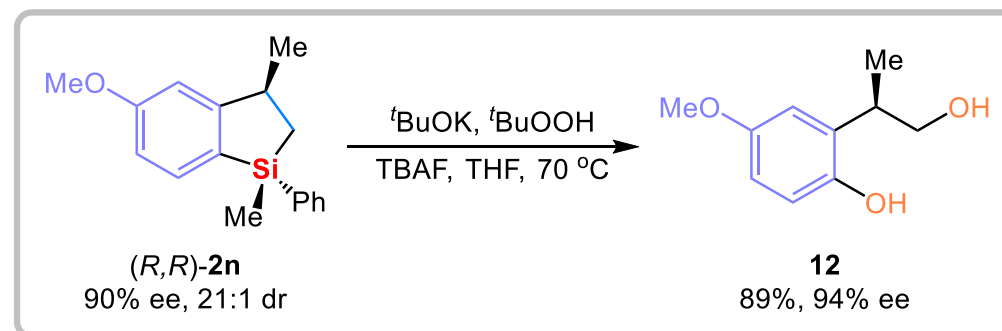
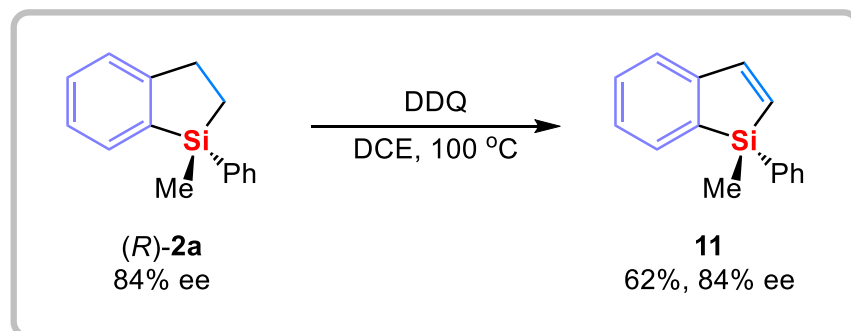
90 min, $c = 47\%$, $s = 7$

Synthetic Applications

Gram-Scale Reaction

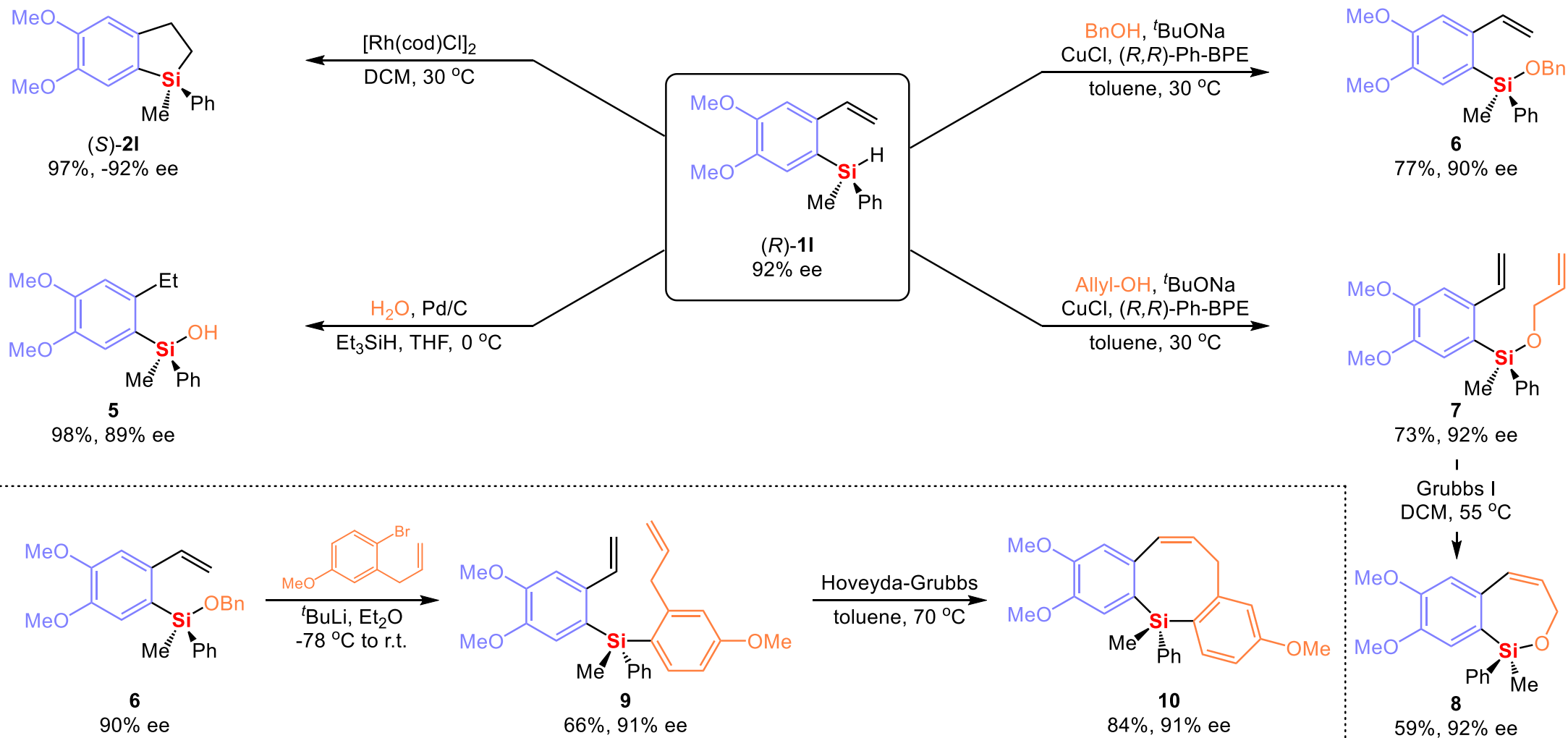


Transformation of Si-Stereogenic Dihydrobenzosilole Compounds



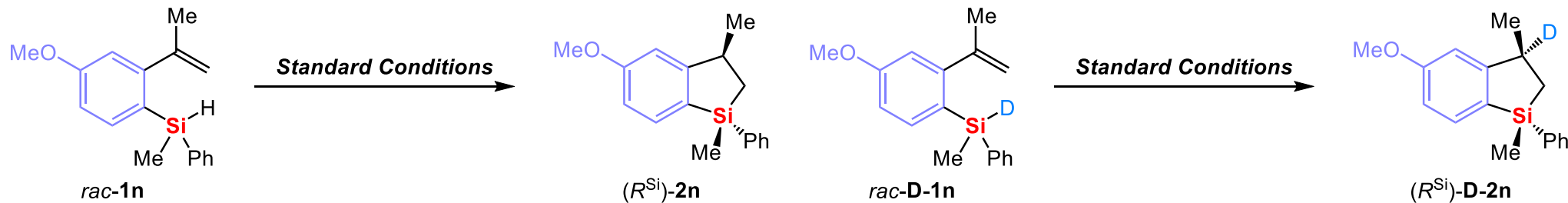
Synthetic Applications

Transformation of Monohydrosilane



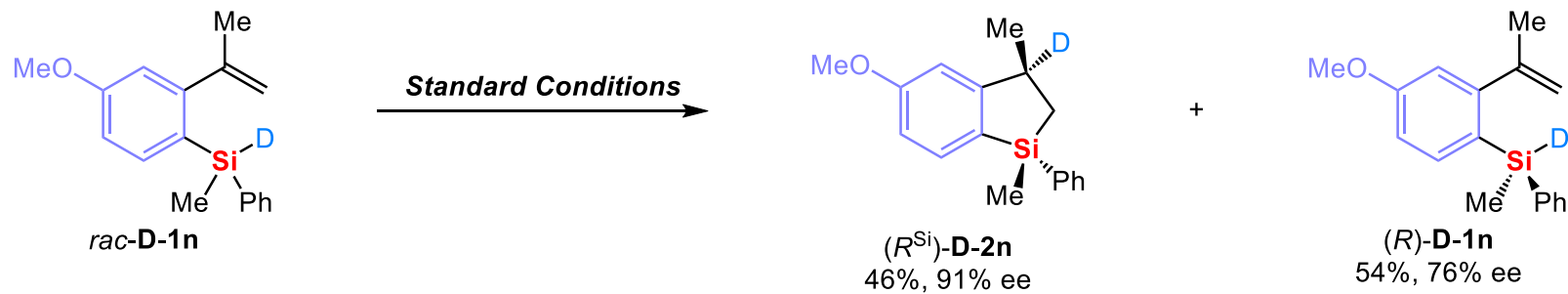
Mechanistic Investigation

Parallel KIE Experiments



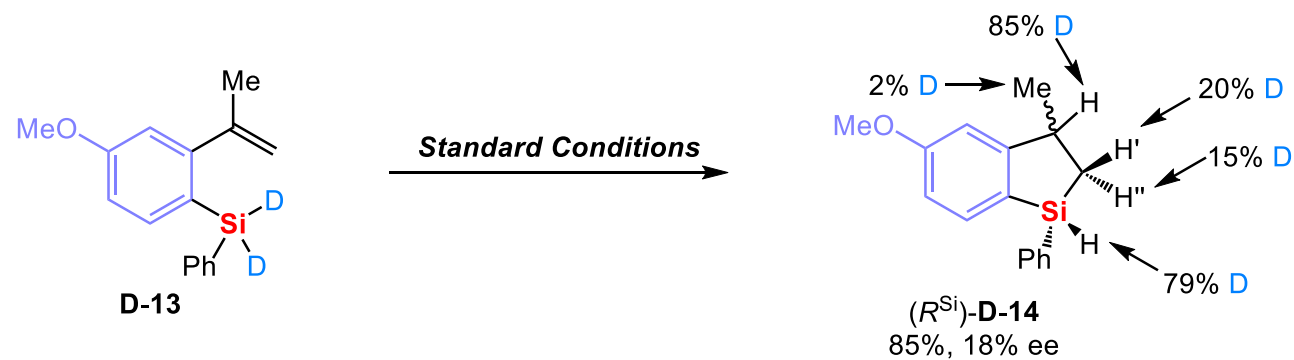
$$\text{KIE} = K_H/K_D = 1.02$$

Deuterium-Labeling Experiments

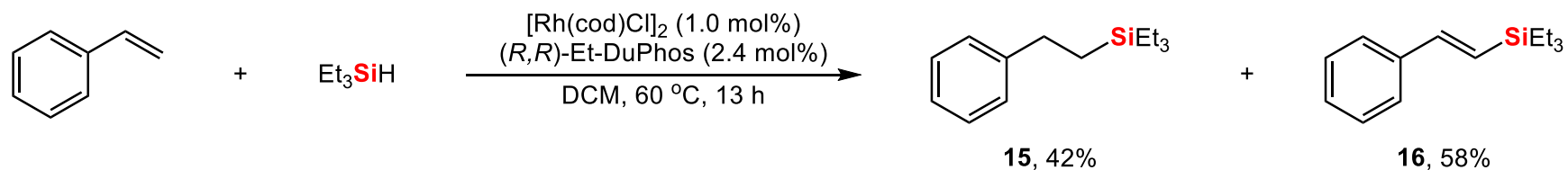


Mechanistic Investigation

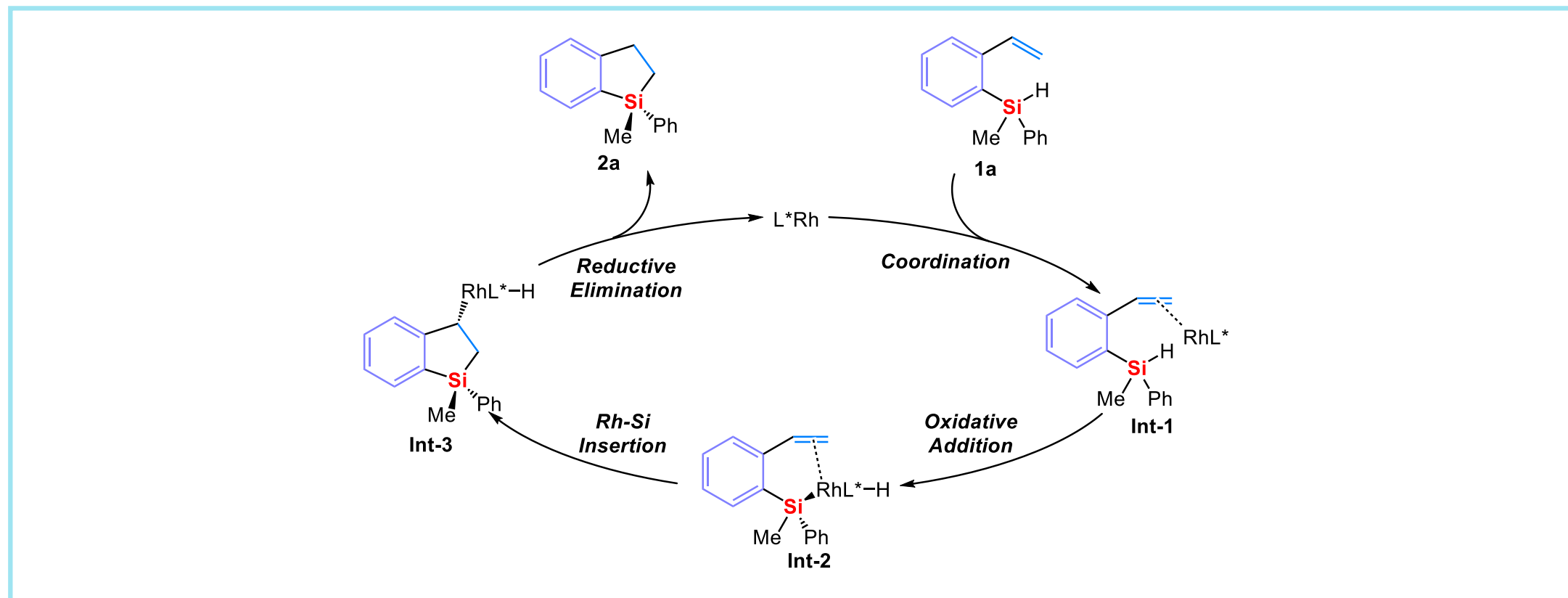
Deuterium-Labeling Experiments



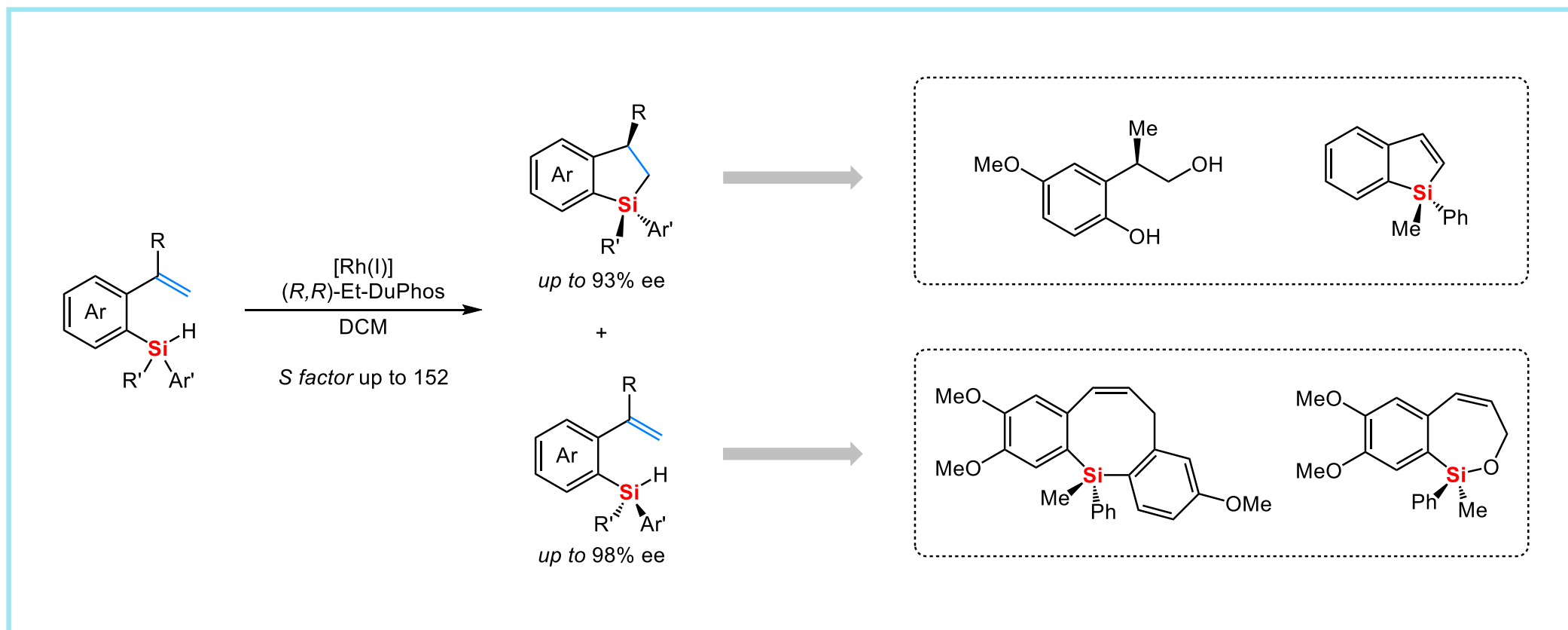
Rh-Catalyzed Hydrosilylation of Styrene with Triethylsilane



Proposed Reaction Mechanism



Summary



➤ **Low Catalyst Loading**

➤ **Excellent Stereocontrol**

➤ **Middle-Sized Silacycle**

Strategy for Writing The First Paragraph

The **Importance** of Silicon-Stereogenic Organosilanes



Construction of Silicon-Stereogenic Organosilanes



The Demand to Develop
New Approach

- ✓ Silicon-stereogenic organosilanes have received increasing attention in various scientific fields, including synthetic chemistry, material science and pharmaceutical chemistry...
- ✓ Currently, the majority of catalytic processes for the preparation of silicon-stereogenic organosilanes relies on enantioselective desymmetrization of prochiral precursors with two identical substitutes attached to the Si(sp^3)-atom center...
- ✓ Nevertheless, those early examples remain several limitations, including the use of stoichiometric chiral reagent or substrate, and insufficient chiral induction. To develop catalytic KR process with a chiral catalyst for the enantioselective preparation of silicon-stereogenic organosilanes is still in high demand.

Strategy for Writing The Last Paragraph

**Summary
of this Work**



**Highlights of
this Work**

- ✓ In conclusion, the access of silicon-stereogenic organosilanes has been achieved by catalytic kinetic resolution of monohydrosilanes.
- ✓ With this newly established protocol, both silicon-stereogenic dihydrobenzosiloles and silicon-stereogenic monohydrosilanes could be prepared with high efficiency and good to excellent enantioselectivities in one single operation. Both enantioenriched silacyclic products and monohydrosilanes are synthetically useful, which can be readily further transformed to diverse chiral compounds, including the construction of a seven-membered siloxane and an eight-membered silacycle.

Representative Examples

- Nevertheless, those early examples remain several limitations, including the use of stoichiometric chiral reagent or substrate, and **insufficient** chiral induction. (**insufficient**, adj. 不足的; 不充分的)
- The replacement of the methyl group with n-butyl group led to a slight **diminution** in enantioselectivity, giving **4I** and **3I** in 75% ee and 74% ee, respectively. (**diminution**, n. 减少; 降低; 缩减; 缩小)
- **For comparison**, deuterium-labeled dideuterosilane **D-13** was subjected to the same reaction conditions, delivering the corresponding product (R^{Si})-**D-14** with deuterium distribution in both α - and β -positions to the Si atom. (以作比较; 为了比较)

Acknowledgement

Thanks for your attention