# **Literature Report 9**

# Enantioselective Type II Cycloaddition of Alkynes via C-C Activation of Cyclobutanones: Rapid and Asymmetric Construction of [3.3.1] Bridged Bicycles

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2 Enantioselective Type II Cycloaddition of Alkynes via C-C Activation

3 Summary

# **CV of Prof. Guangbin Dong**

#### **Background:**



□ 1999-2003	B.S., Peking University
□ 2004-2009	Ph.D., Stanford University (Barry M. Trost)
□ 2009-2011	Postdoctoral Fellow, California Institute of Technology
	(Robert H. Grubbs)

Guangbin Dong

(Robert H. Grubbs)
 **2011-2016** Assistant Professor, University of Texas at Austin
 **2016** Professor, University of Texas at Austin

**2016-now** Professor, University of Chicago

#### **Research Interests:**

- Developing novel catalytic C-H and C-C bond activation methods for efficient small-molecule agents synthesis
- Developing new transition-metal catalysts based on supramolecular chemistry for chemoselective C-H bond activation of small molecules
- Establishing efficient synthetic routes to access natural products



#### **Intramolecular Diels-Alder Reaction**



#### Type II



MacMillan, D. W. C.\* et al. J. Am. Chem. Soc. 2005, 127, 11616



Shea, K. J.\* et al. J. Org. Chem. 2007, 72, 9402

#### Another Way Transition-Metal-Catalyzed C–C Activation



#### **Constructing Various Ring Systems**



Ito, Y.\* et al. J. Am. Chem. Soc. 2002, 124, 13976



Yu, Z.-X.\* et al. J. Am. Chem. Soc. 2008, 130, 7178



Dong, G.\* et al. Angew. Chem. Int. Ed. 2012, 51, 7567



Yu, Z.-X.\* et al. Angew. Chem. Int. Ed. 2017, 56, 8667

# Asymmetric Intramolecular Type-II Cyclobutanone/2π Couplings









[3.2.1] bicycles

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Murakami, M.\* et al. Angew. Chem. Int. Ed. 2012, 51, 2485



Cramer, N.\* et al. Organometallics 2014, 33, 780



Cramer, N.\* et al. Angew. Chem. Int. Ed. 2014, 53, 3001



Cramer, N.\* et al. Angew. Chem. Int. Ed. 2014, 53, 9640



Dong, G.\* et al. Nat. Chem. 2014, 6, 739



Dong, G.\* et al. J. Am. Chem. Soc. 2015, 137, 13715

#### **Proposed Mechanism**



#### **Enantioselective Type II Cycloaddition of Alkynes**



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#### **Optimization of the Reaction Parameters**



## **Optimization of the Reaction Parameters**

entry <sup>a</sup>	Change from the "standard condition"	yield (%) <sup><i>b</i></sup> of 2a	er <sup>b</sup> of 2a	yield (%) <sup><i>b</i></sup> of 2a'
1	none	70	96.5:3.5	<b>11</b> °
2	[Rh(COD) <sub>2</sub> ]BF <sub>4</sub> instead of [Rh(COD) <sub>2</sub> ]NTf <sub>2</sub>	64	84:16	10
3	L2 instead of L1	33	89:11	6
4	L3 instead of L1	24	97:3	6
5	L4 instead of L1	51	98.5:1.5	9
6	L5 instead of L1	27	80:20	7
7	L6 instead of L1	20	85.5:14.5	4
8	1,3-DFB instead of 1,4-DFB	62	96:4	11
9	1,2-DFB instead of 1,4-DFB	66	96:4	11
10	1,4-dioxane instead of 1,4-DFB	64	97.5:2.5	14
11	toluene instead of 1,4-DFB	67	96.5:3.5	14
12	1 mL 1,4-DFB instead of 0.5 mL	70	96.5:3.5	11

<sup>a</sup>All reactions were run on a 0.1 mmol scale for 48 h. <sup>b</sup>Yields are isolated yields; the er was determined by chiral HPLC. <sup>c</sup>The er of 2a' was 97:3. DFB = difluorobenzene.

### **Scope of Substrates with Nitrogen and Carbon Linkers**





2a, R<sup>1</sup> = Me, 70% (er 96.5:3.5) + 11% 2a'
2b, R<sup>1</sup> = Et, 71% (er 95:5) + 12% 2b'
2c, R<sup>1</sup> = <sup>i</sup>Pr, 83% (er 98:2) + 16% 2c'
2d, R<sup>1</sup> = CH<sub>2</sub>OTIPS, 69% (er 97.5:2.5) + 11% 2d'



**2e**,  $R^3 = H$ , 76% (er 95.5:4.5) + 12% **2e' 2f**,  $R^3 = F$ , 78% (er 94.5:5.5) + 13% **2f' 2g**,  $R^3 = CF_3$ , 78% (er 95.5:4.5) + 13% **2g' 2h**,  $R^3 = OMe$ , 80% (er 94.5:5.5) + 13% **2h' 2i**,  $R^3 = Me$ , 71% (er 96:4) + 12% **2i'** 



2j, 74% (er 99:1) + 12% 2j'



2k, 56% (er 92.5:7.5) + 11% 2k'



2I, 30% (er 99:1) + 5% 2I'

### **Scope of Substrates with Nitrogen and Carbon Linkers**







Me

2s, 18% (er 61:39) + 8% 2s'

**2m**,  $R^4$  = Bs, 63% (er 95:5) + 10% **2m' 2n**,  $R^4$  = *p*-Ns, 53% (er 94.5:5.5) + 9% **2n' 2o**,  $R^4$  = Ms, 60% (er 96:4) + 10% **2o' 2p**,  $R^4$  = *o*-Ns, 32% (er 90:10) + 5% **2p' 2q**,  $R^4$  = Piv, <2%



**2t**, R<sup>2</sup> = <sup>*n*</sup>Pr, 71% (er 95:5) dr >20:1 **2u**, R<sup>2</sup> = Et, 65% (er 95.5:4.5) dr >20:1 **2v**, R<sup>2</sup> = Bn, 64% (er 94:6) dr >20:1 (Pd/C (20%wt), H<sub>2</sub>, EtOH, rt, 12 h)



### **Scope of Substrates with Nitrogen and Carbon Linkers**



### **Scope of Substrates with Oxygen Linkages**





## **Scope of Cyclobutanones with Existing Stereocenters**



#### **Scope of the Olefin Derived Substrate**



## **Synthetic Utilities**



# **Synthetic Utilities**



#### **Strain Release through Metal Coordination**





# **The First Paragraph**

Chiral bridged scaffolds are commonly found in biologically important natural products and drugs. However, compared to fused rings, much fewer cyclization approaches have been developed for constructing bridged rings, often due to enhanced angular and torsional strains in reaction transition states. For example, while the Type I Diels-Alder reaction the intramolecular  $[4\pi+2\pi]$  cyclization using C1-tethered dienes is powerful for building fused rings with numerous enantioselective versions achieved, the corresponding Type II reaction introduces a twisted alkene at bridgeheads, known as an anti-Bredt olefin, due to the C2 linkage. As exemplified in the formation of 6-6 bridged rings, the corresponding Type II Diels-Alder reaction proves to be highly challenging and requires special and forcing conditions (e.g., highgas-phase reactions). As a consequence, asymmetric temperature synthesis of bridged rings via Type-II cycloadditions has been even rarer.



# **The Last Paragraph**

In summary, an enantioselective Rh-catalyzed intramolecular Type II cyclization between cyclobutanones and alkynes has been developed, which provides a convenient entry to chiral [3.3.1] bridged bicyclic scaffolds. The reaction is redox neutral and strong acid/base-free, thus tolerating a broad range of functional groups. The products bearing a reactive exocyclic enone moiety can be readily functionalized. It is thus this could be useful for preparing anticipated that method bioactive and complex alkaloids or terpenes. The stabilization of an anti-Bredt olefin in a strained [3.3.1] scaffold via metal coordination, as revealed by the mechanistic study, could have broad implications beyond this work. Efforts on extending the substrate scope to construct other bridged rings and exploring other coupling partners besides alkynes are ongoing.

To the best of our knowledge (据我们所知), the challenge of enantioselective preparation of more common [3.3.1] bicycles (6/6 bridged rings), found in numerous bioactive terpenes and alkaloids, remains to be addressed. (有待解决)

It is also known that (阐述常见现象时) binding of strained olefins to low valent metals changes the hybridization of olefin carbons (toward sp<sup>3</sup>) and alters the bond lengths/angles from those in free olefin, resulting in strain release.

As exemplified (例如) in the formation of 6-6 bridged rings, the corresponding Type II Diels-Alder reaction proves to be highly challenging and requires special and forcing conditions.













# Buchwald-Hartwig-Miura α-Arylation



# **Hosomi-Sakurai Reaction**



桥环化合物是指化合物中的任意两个环共用两不直接相连的 碳原子的环烃,根据组成环的数目分为二环烃、三环烃、四环烃 等。两个环或多个环用的碳原子为桥头碳原子,连接桥头碳原子 的键称做桥。