

Literature Report IX

Intermolecular Anti-Markovnikov Hydroamination of Alkenes with Sulfonamides, Sulfamides, and Sulfamates

Reporter: Qing-Xian Xie
Checker: Gao-Wei Wang
Date: 2025-1-20

Lin, A.; Karrasch, M. J.; Ganley, J. M.; Hejna, B. G.; [Knowles, R. R.](#) *ACS Catal.* **2024**, 14, 13098-13104

CV of Prof. Robert R. Knowles



Background:

- ❑ **1999-2003** B. S., William & Mary
 - ❑ **2003-2008** Ph. D., California Institute of Technology (Prof. David MacMillan)
 - ❑ **2008-2011** Postdoc., Harvard University (Prof. Eric Jacobsen)
 - ❑ **2011-2017** Associate Professor, Princeton University
 - ❑ **2017-Now** Professor, Princeton University
-

Research:

- ❑ Proton-coupled electron transfer in organic synthesis
 - ❑ Enantioselective catalysis with free radical intermediates
 - ❑ Catalytic olefin hydroamination
-

Contents

1 Introduction

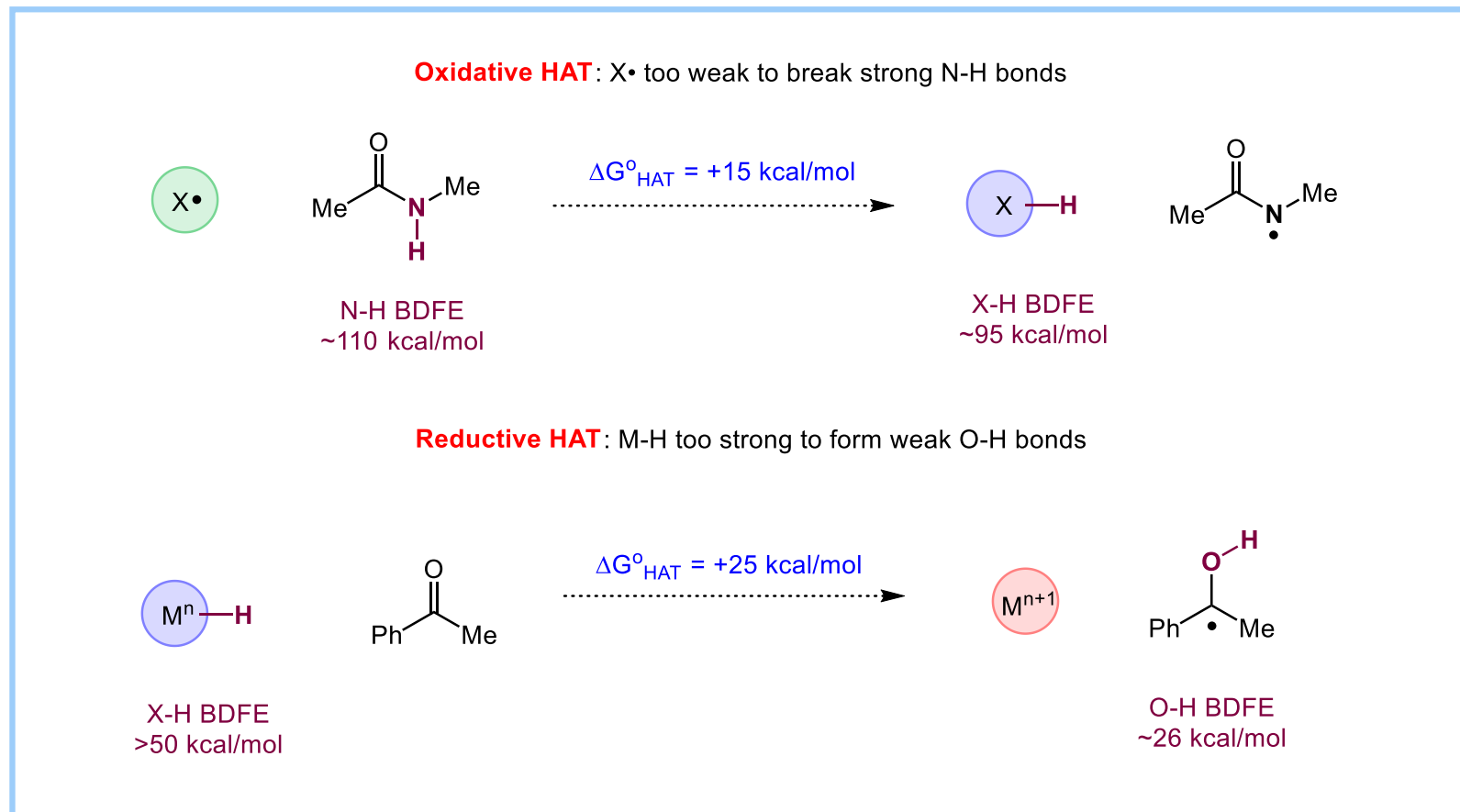
2 Proton-Coupled Electron Transfer in Organic Synthesis

3 Intermolecular Hydroamination of Alkenes with Sulfonamides

4 Summary

Introduction

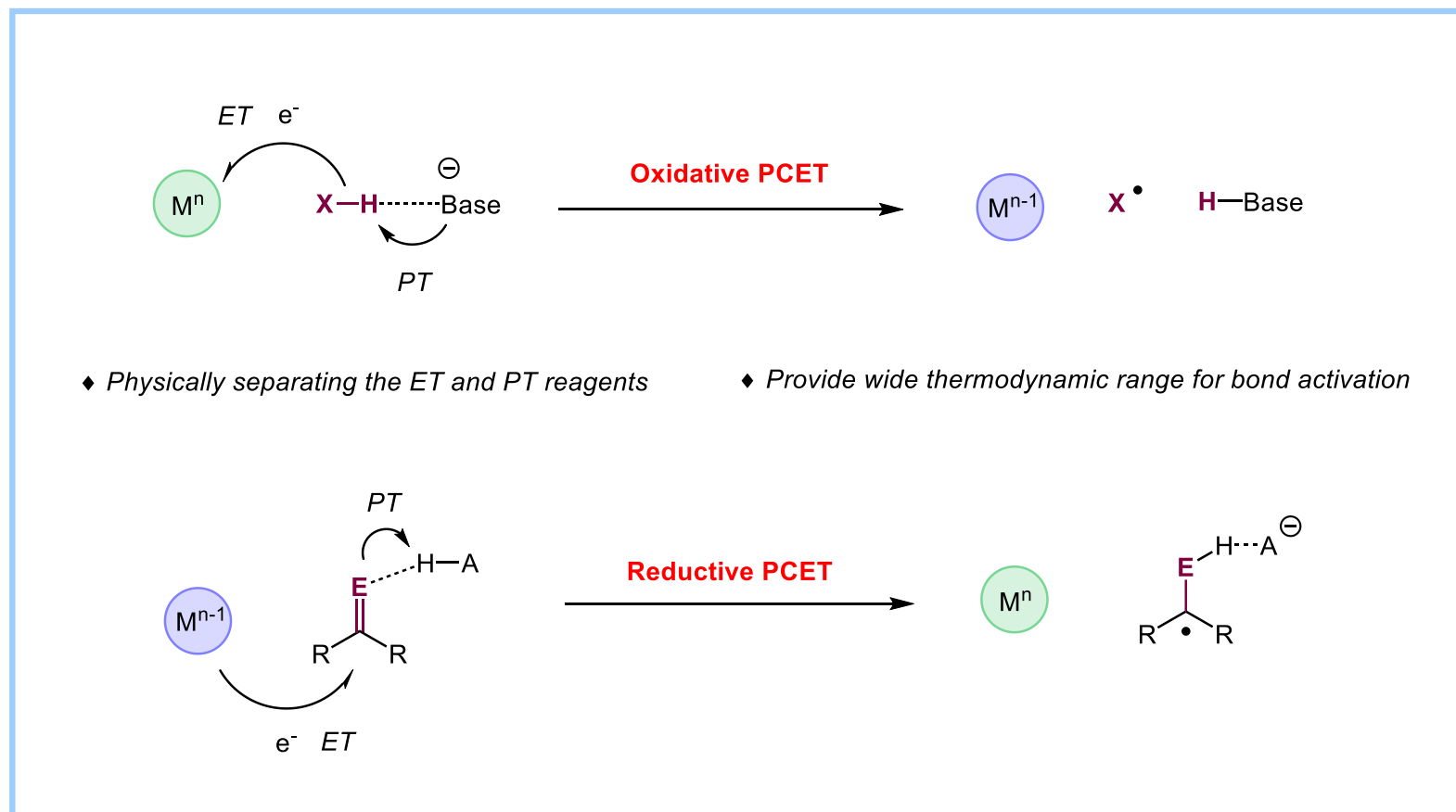
Thermodynamically limited HAT reactions



Gentry, E. C.; Knowles, R. R.* *Acc. Chem. Res.* **2016**, *49*, 1546-1556

Introduction

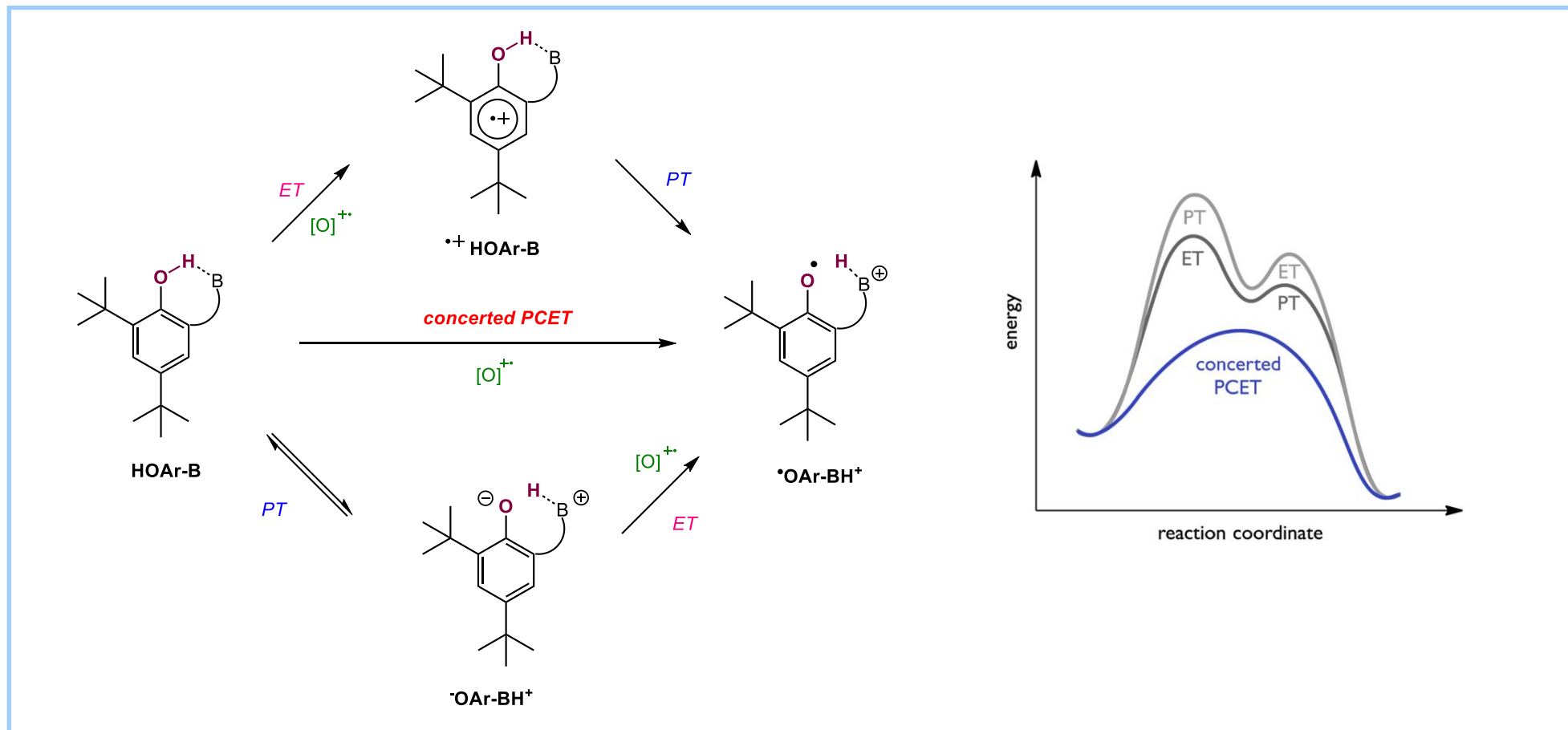
Multi-site proton-coupled electron transfer



Gentry, E. C.; Knowles, R. R.* *Acc. Chem. Res.* **2016**, *49*, 1546-1556

Introduction

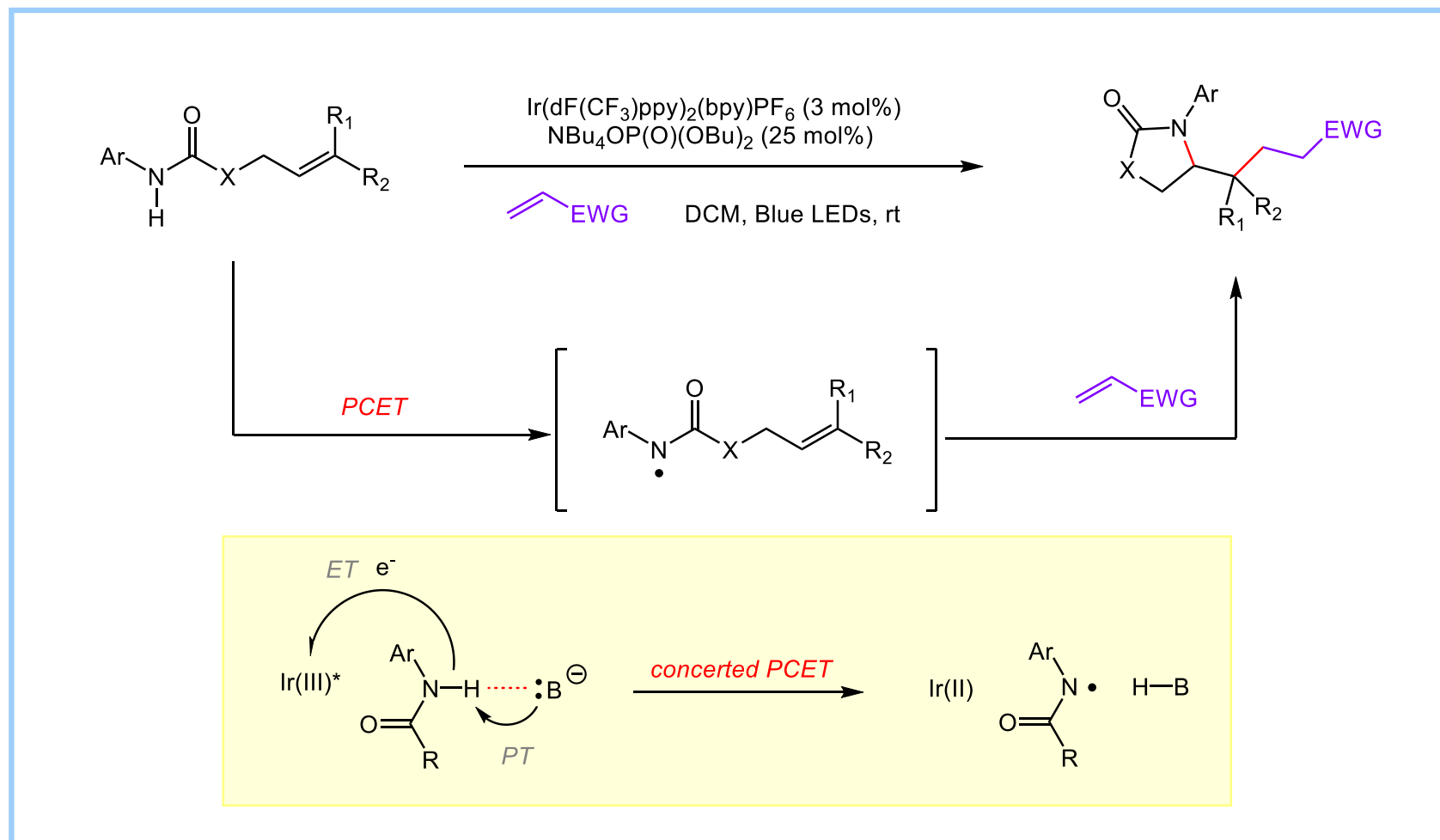
Three pathways of PCET mechanism



Rhile, I. J.; Markle, T. F.; Nagao, H.; Rotter, K; Mayer, J. M.* *J. Am. Chem. Soc.* **2006**, 128, 6075-6088

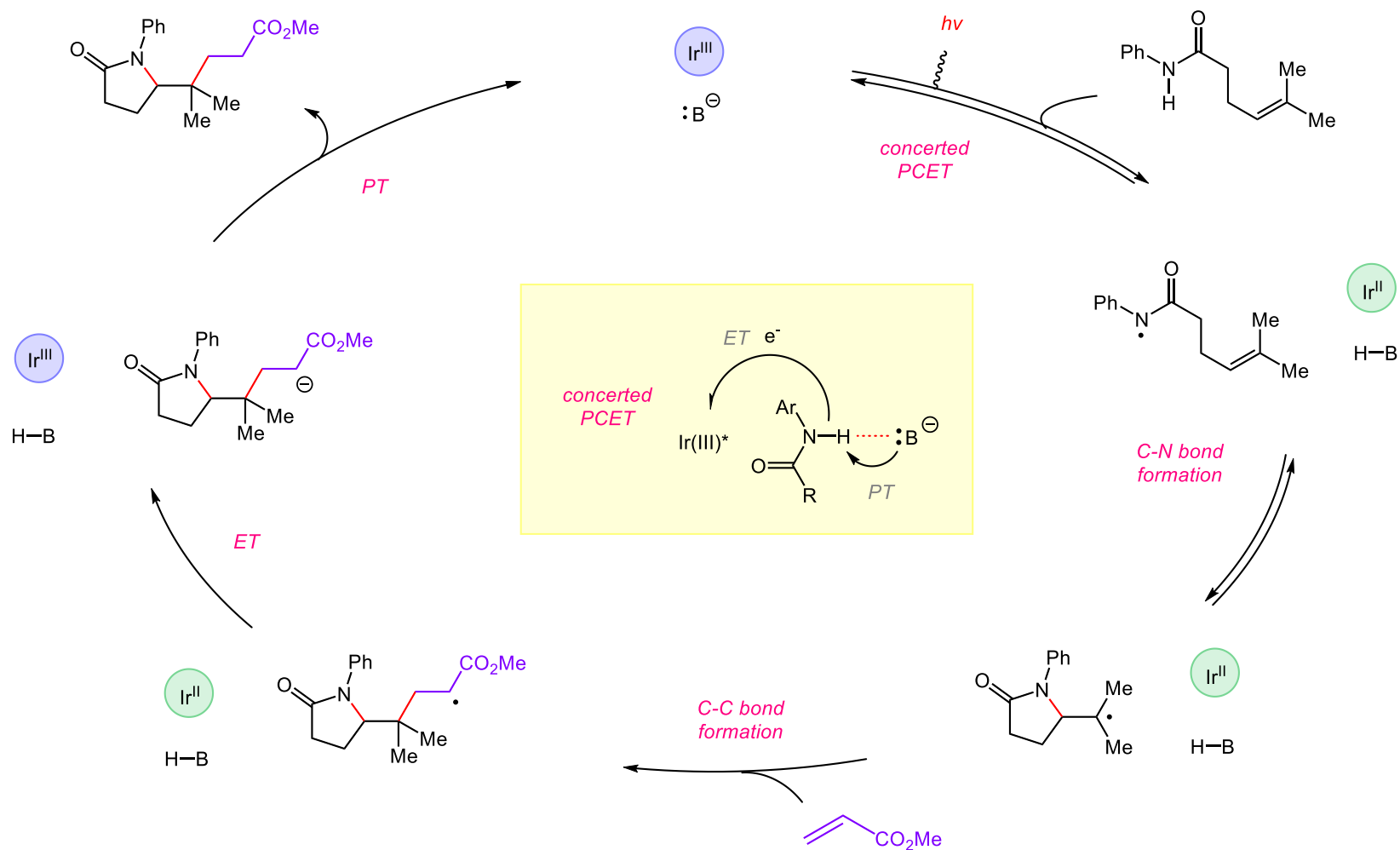
Proton-Coupled Electron Transfer in Organic Synthesis

PCET promoted N-H bond-weakening



Choi, G. J.; Knowles, R. R.* *J. Am. Chem. Soc.* **2015**, *137*, 9226-9229

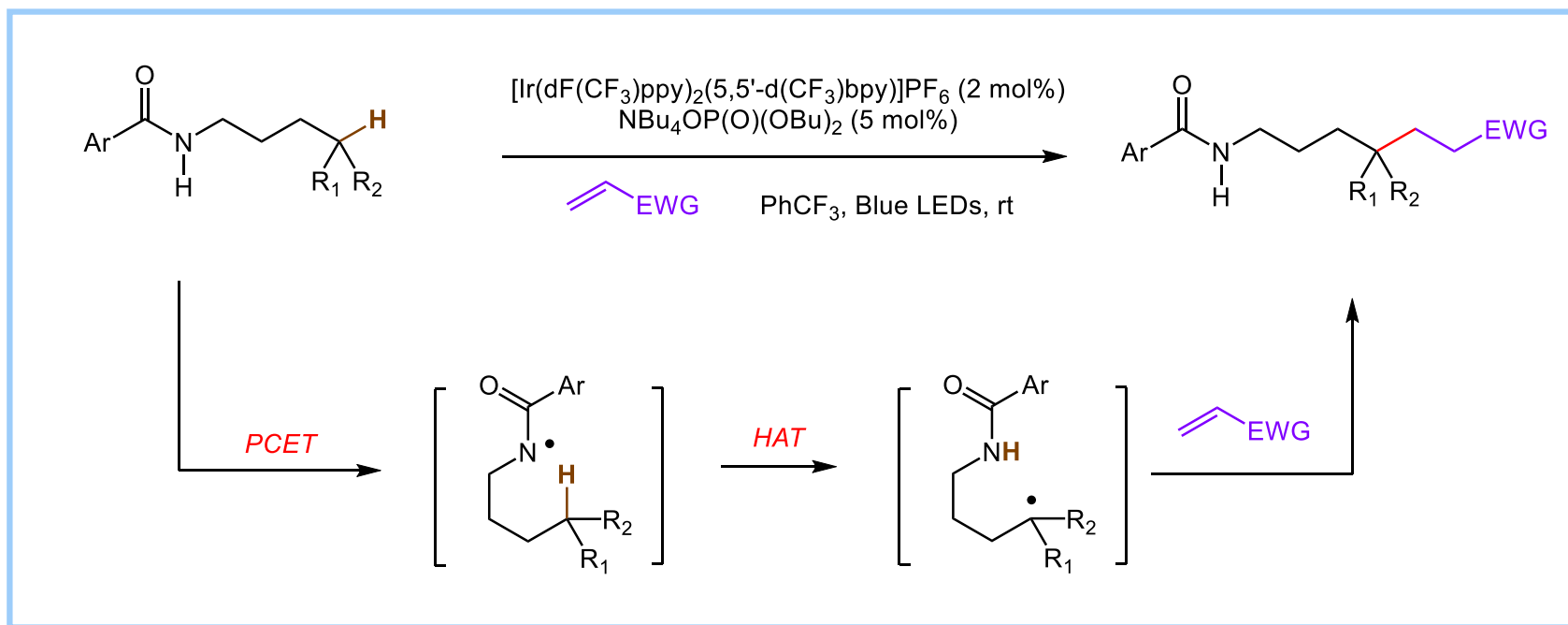
Proton-Coupled Electron Transfer in Organic Synthesis



Choi, G. J.; Knowles, R. R.* *J. Am. Chem. Soc.* **2015**, *137*, 9226-9229

Proton-Coupled Electron Transfer in Organic Synthesis

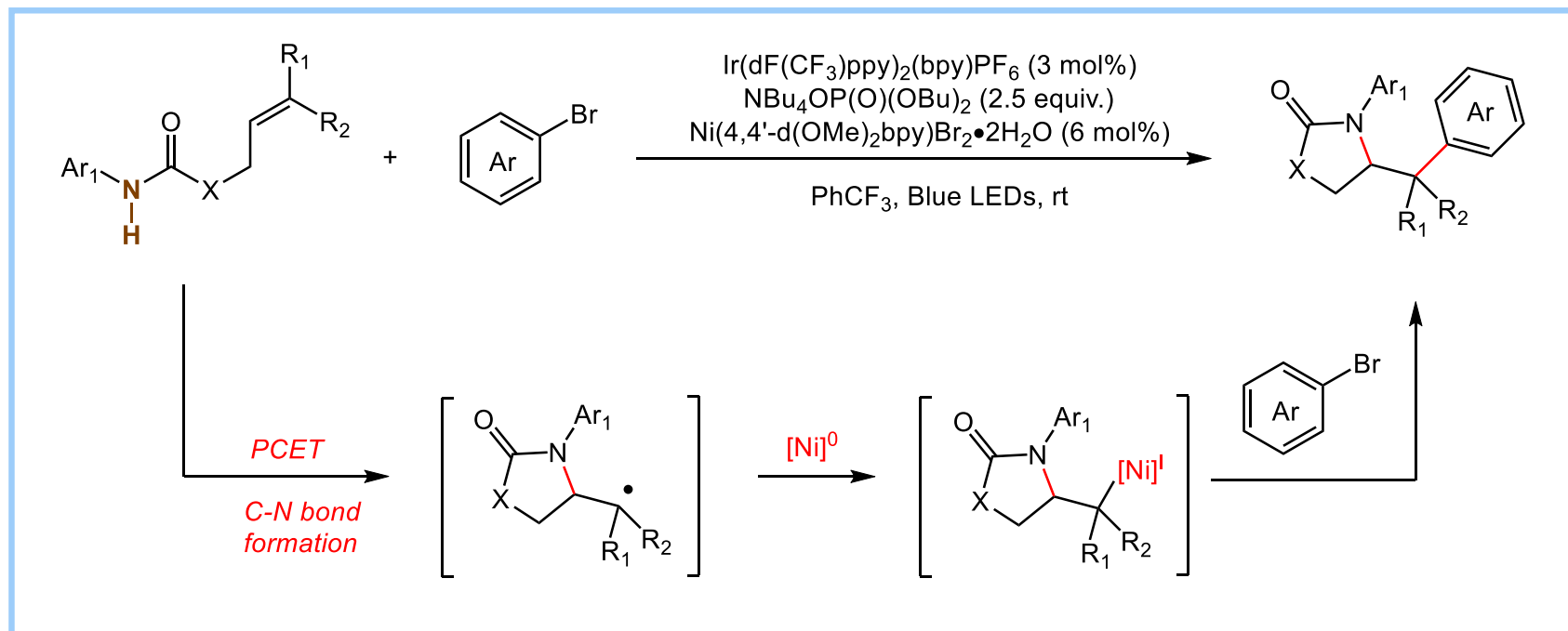
PCET promoted N-H bond-weakening



Choi, G. J.; Zhu, Q.; Miller, D. C.; Gu, C. J.; Knowles, R. R.* *Nature* **2016**, 539, 268-271

Proton-Coupled Electron Transfer in Organic Synthesis

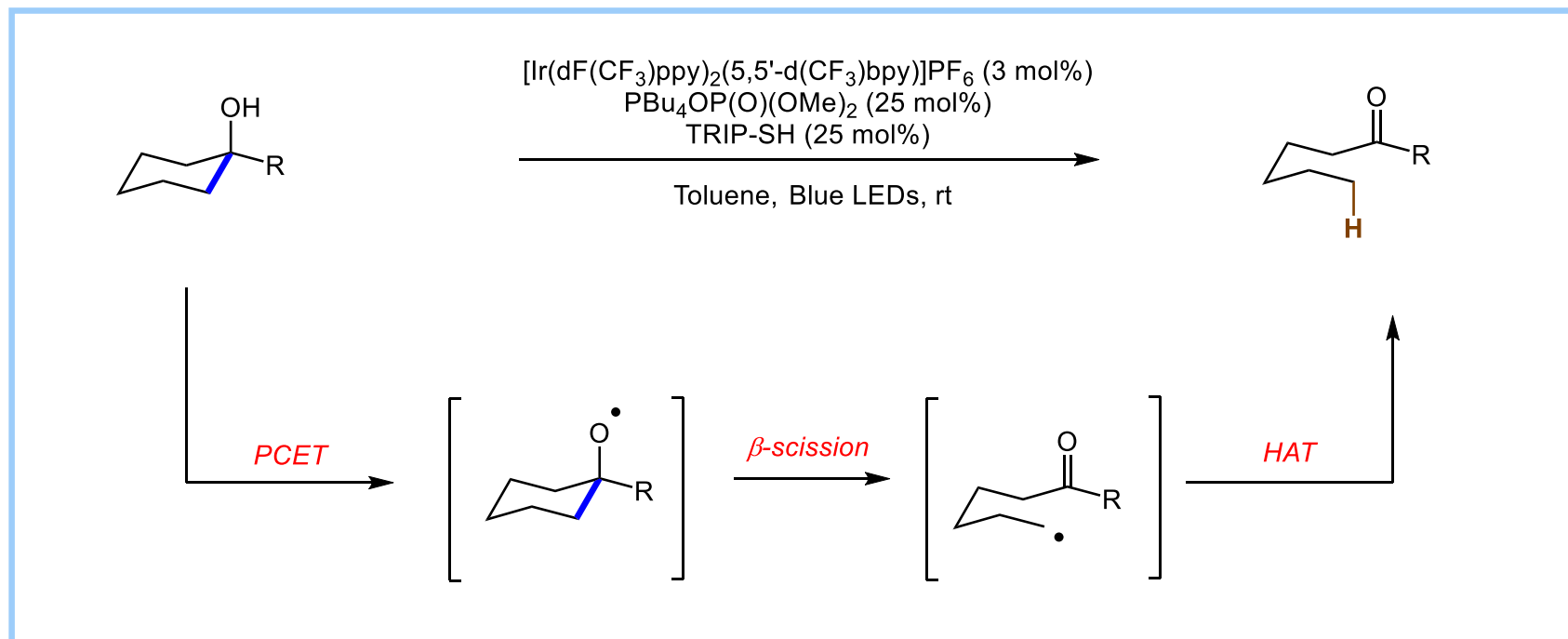
PCET promoted N-H bond-weakening



Zheng, S.; Gutierrez-Bonet, A.; Molander, G. A.* *Chem* **2019**, *5*, 339-352

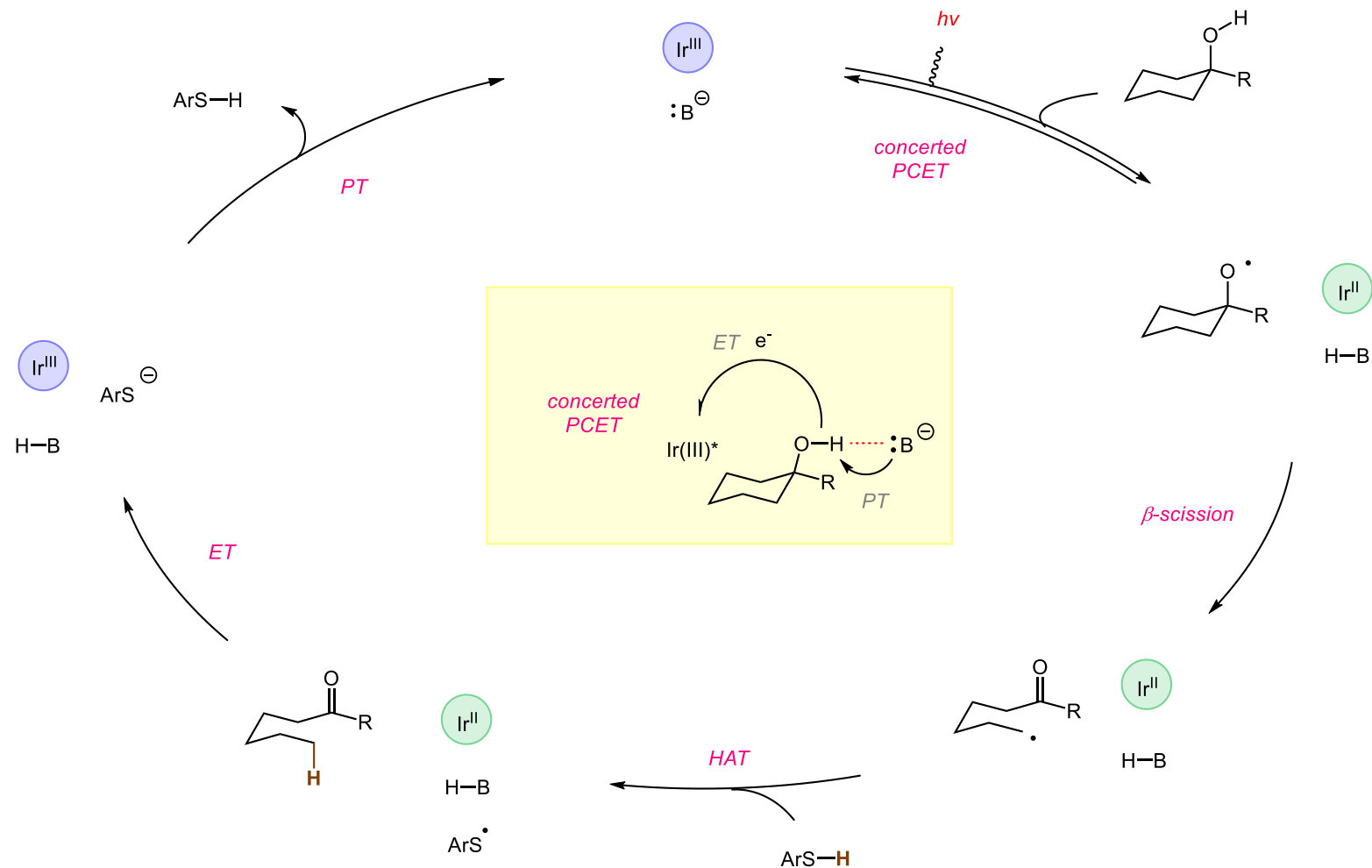
Proton-Coupled Electron Transfer in Organic Synthesis

PCET promoted O-H bond-weakening



Ota, E.; Wang, H.; Frye, N. L.; Knowles, R. R.* *J. Am. Chem. Soc.* **2019**, *141*, 1457-1462

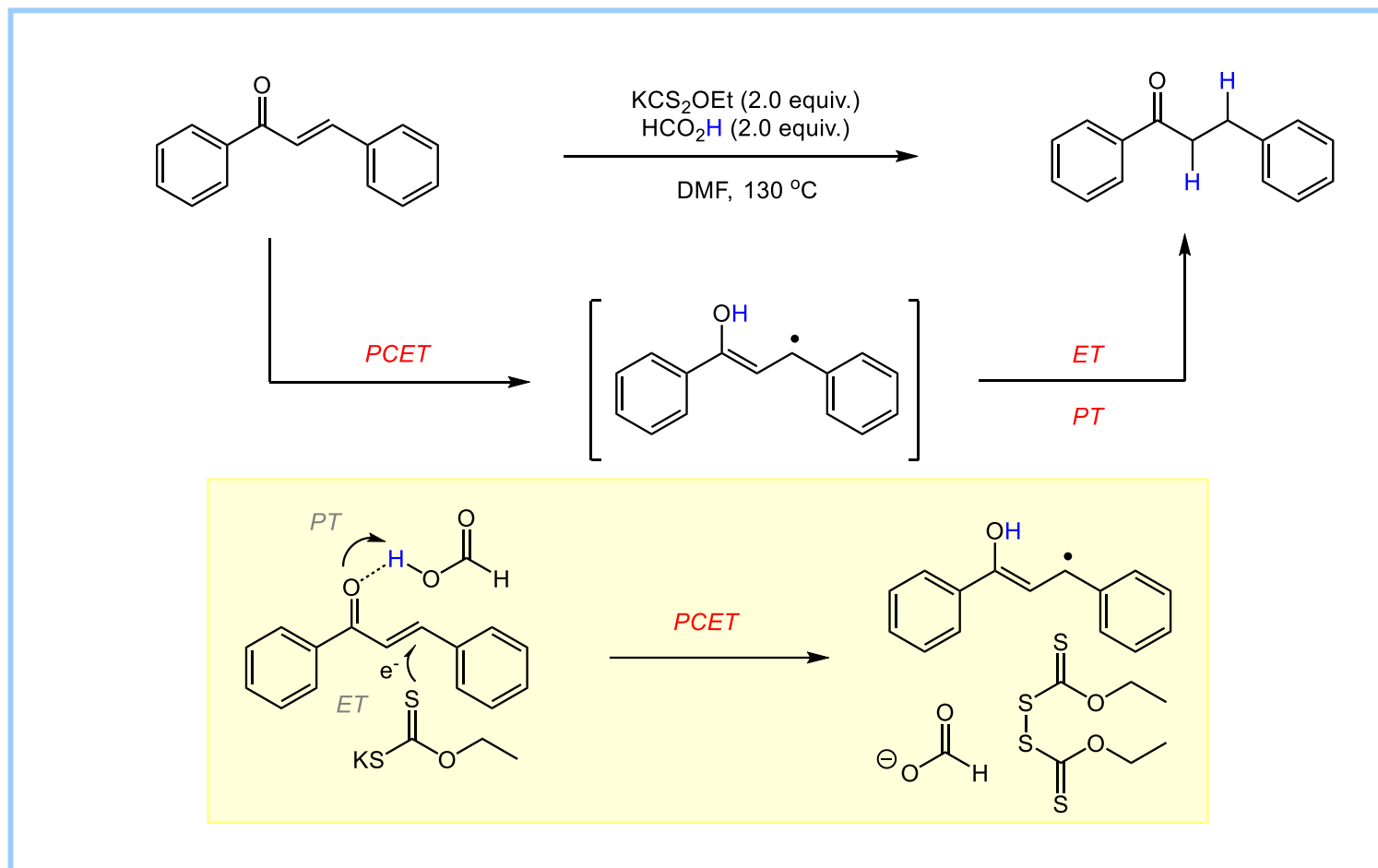
Proton-Coupled Electron Transfer in Organic Synthesis



Ota, E.; Wang, H.; Frye, N. L.; Knowles, R. R.* *J. Am. Chem. Soc.* **2019**, *141*, 1457-1462

Proton-Coupled Electron Transfer in Organic Synthesis

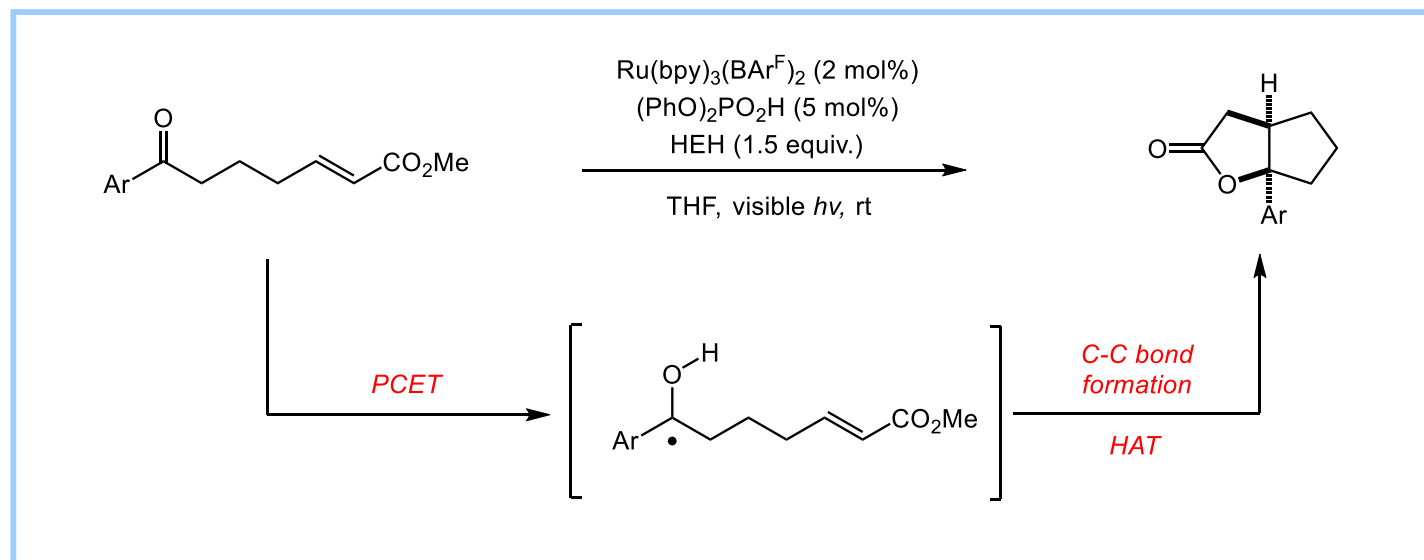
PCET promoted reduction of C=C bond



Prasanna, R.; Guha, S.; Sekar, G.* *Org. Lett.* **2019**, *21*, 2650-2653

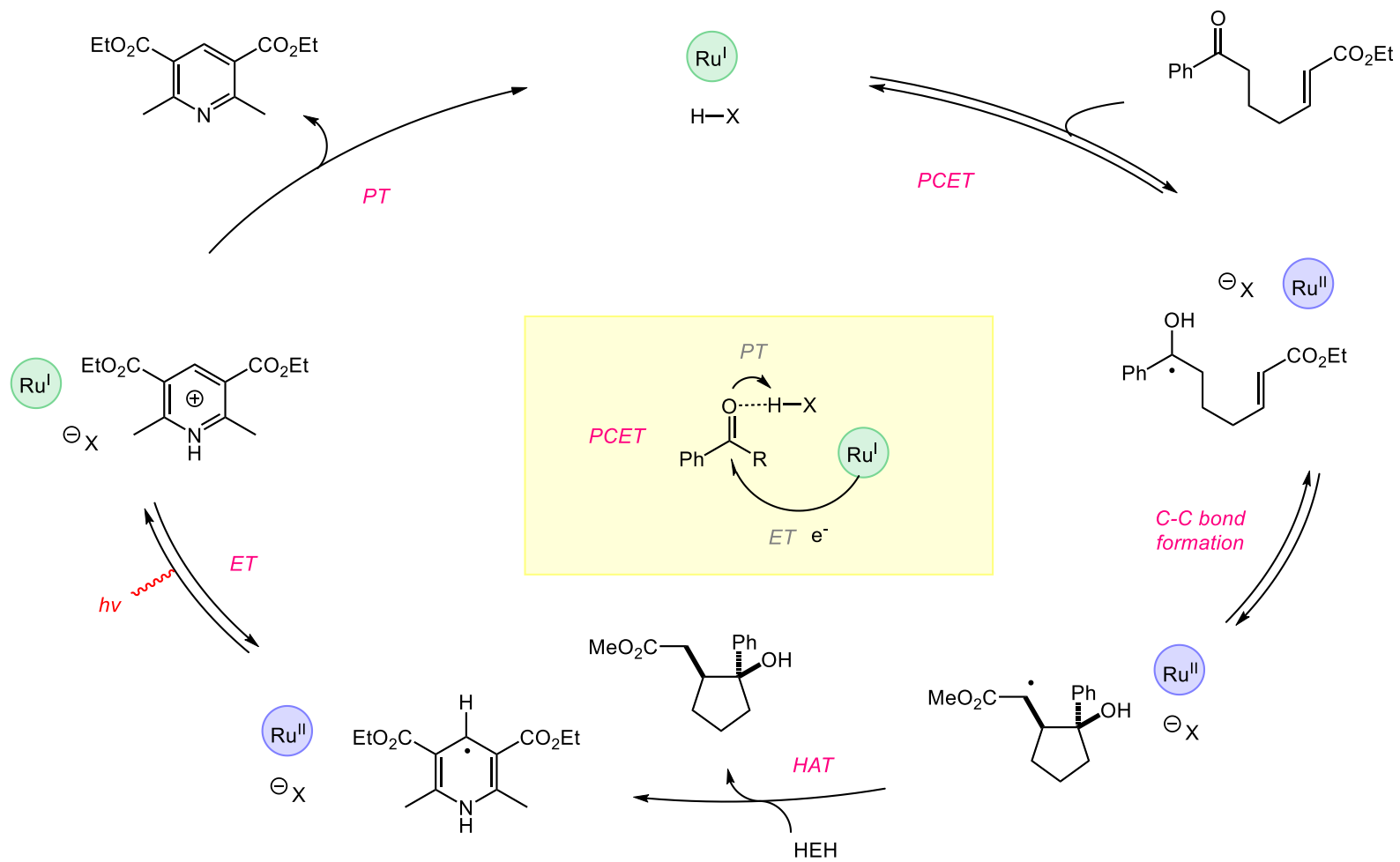
Proton-Coupled Electron Transfer in Organic Synthesis

PCET promoted reduction of C=O bond



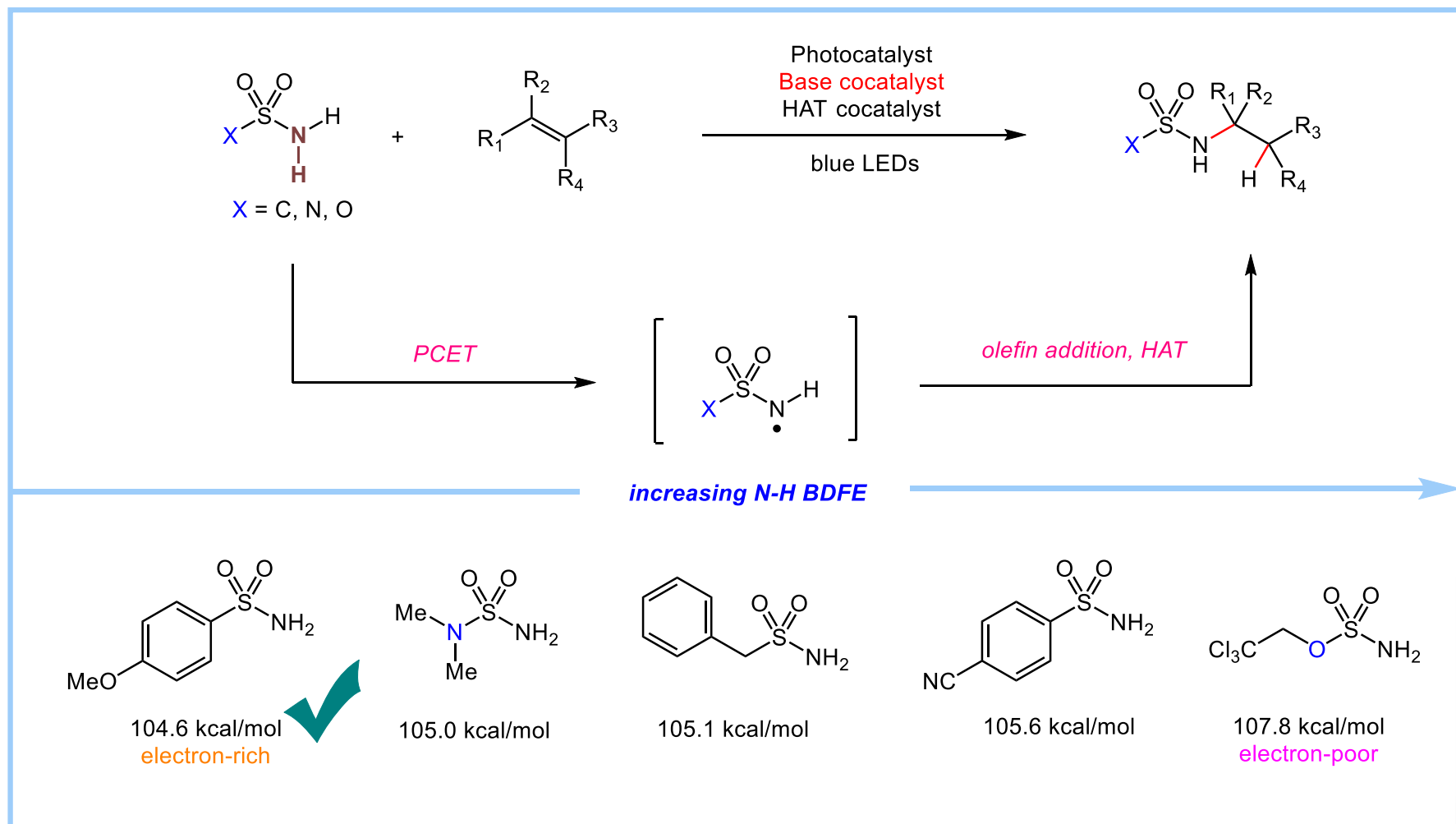
Tarantino, K. T.; Liu, P.; Knowles, R. R.* *J. Am. Chem. Soc.* **2013**, *135*, 10022-10025

Proton-Coupled Electron Transfer in Organic Synthesis



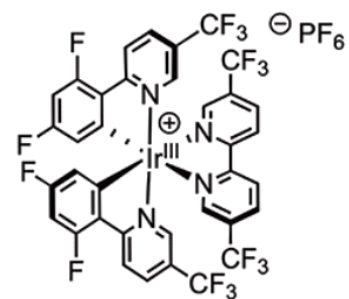
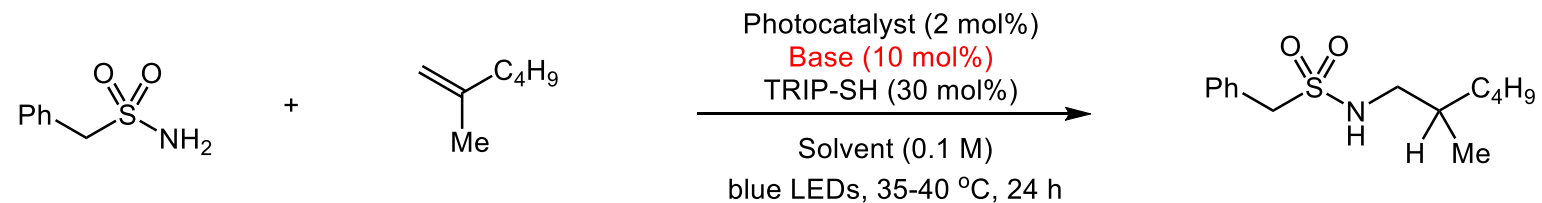
Choi, G. J.; Knowles, R. R.* *J. Am. Chem. Soc.* **2015**, *137*, 9226-9229

Project Synopsis

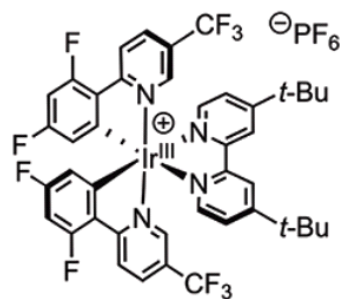


Zhu, Q.; Graff, D. E.; Knowles, R. R.* *J. Am. Chem. Soc.* **2018**, *140*, 741-747

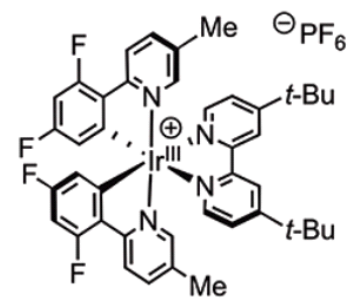
Optimization of Reaction Conditions



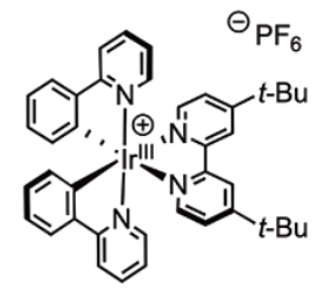
[Ir-A]PF₆
 $E_{1/2}(*\text{Ir}^{\text{III}}/\text{Ir}^{\text{II}}) = +1.30 \text{ V}^b$



[Ir-B]PF₆
 $E_{1/2}(*\text{Ir}^{\text{III}}/\text{Ir}^{\text{II}}) = +0.81 \text{ V}^b$



[Ir-C]PF₆
 $E_{1/2}(*\text{Ir}^{\text{III}}/\text{Ir}^{\text{II}}) = +0.59 \text{ V}^b$

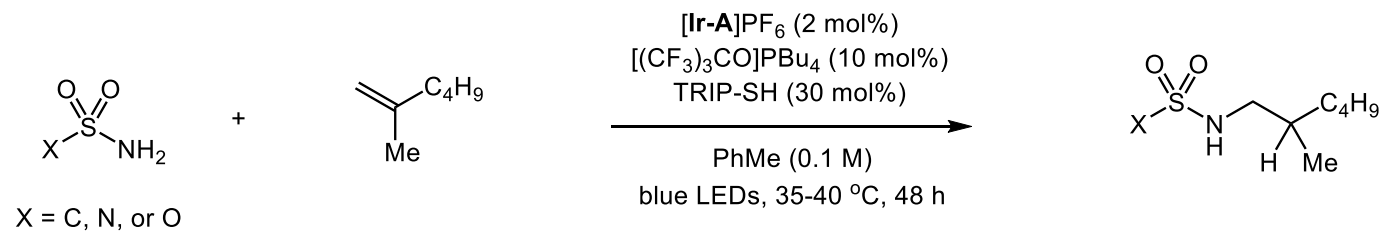


[Ir-D]PF₆
 $E_{1/2}(*\text{Ir}^{\text{III}}/\text{Ir}^{\text{II}}) = +0.28 \text{ V}^b$

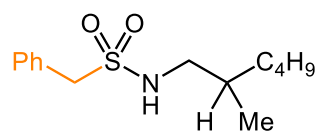
Optimization of Reaction Conditions

Entry	Base	Photocatalyst	Solvent	Yield (%)
1	NBu ₄ [(BuO) ₂ OPO]	[Ir-A]PF ₆	PhMe	2
2	NBu ₄ [(CF ₃) ₃ CO]	[Ir-A]PF ₆	PhMe	85
3	PBu ₄ [(CF ₃) ₃ CO]	[Ir-A]PF ₆	PhMe	97
4	PBu ₄ [BzO]	[Ir-A]PF ₆	PhMe	41
5	PBu ₄ [BuO]	[Ir-A]PF ₆	PhMe	6
6	2,6-lutidine	[Ir-A]PF ₆	PhMe	0
7	PBu ₄ [(CF ₃) ₃ CO]	[Ir-B]PF ₆	PhMe	92
8	PBu ₄ [(CF ₃) ₃ CO]	[Ir-C]PF ₆	PhMe	51
9	PBu ₄ [(CF ₃) ₃ CO]	[Ir-D]PF ₆	PhMe	2
10	PBu ₄ [(CF ₃) ₃ CO]	[Ir-A]PF ₆	PhCF ₃	74
11	PBu ₄ [(CF ₃) ₃ CO]	[Ir-A]PF ₆	DCM	19
12	PBu ₄ [(CF ₃) ₃ CO]	[Ir-A]PF ₆	MeCN	10
13	PBu ₄ [(CF ₃) ₃ CO]	[Ir-A]PF ₆	THF	4

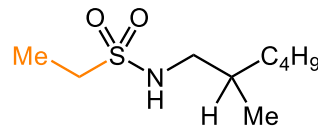
Substrate Scope



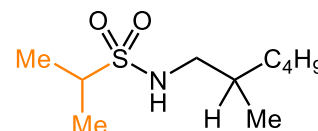
Alkyl sulfonamides



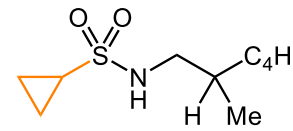
1 86%



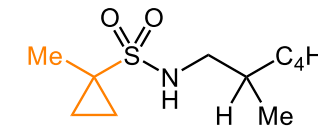
2 66%



3 54%

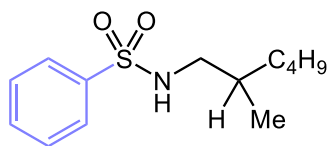


4 79%

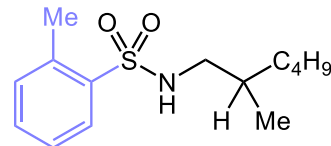


5 63%

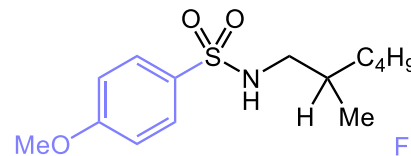
Aryl sulfonamides



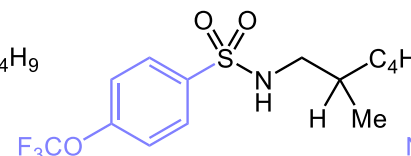
6 98%



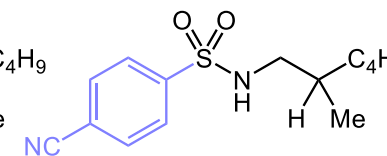
7 94%



8 95%

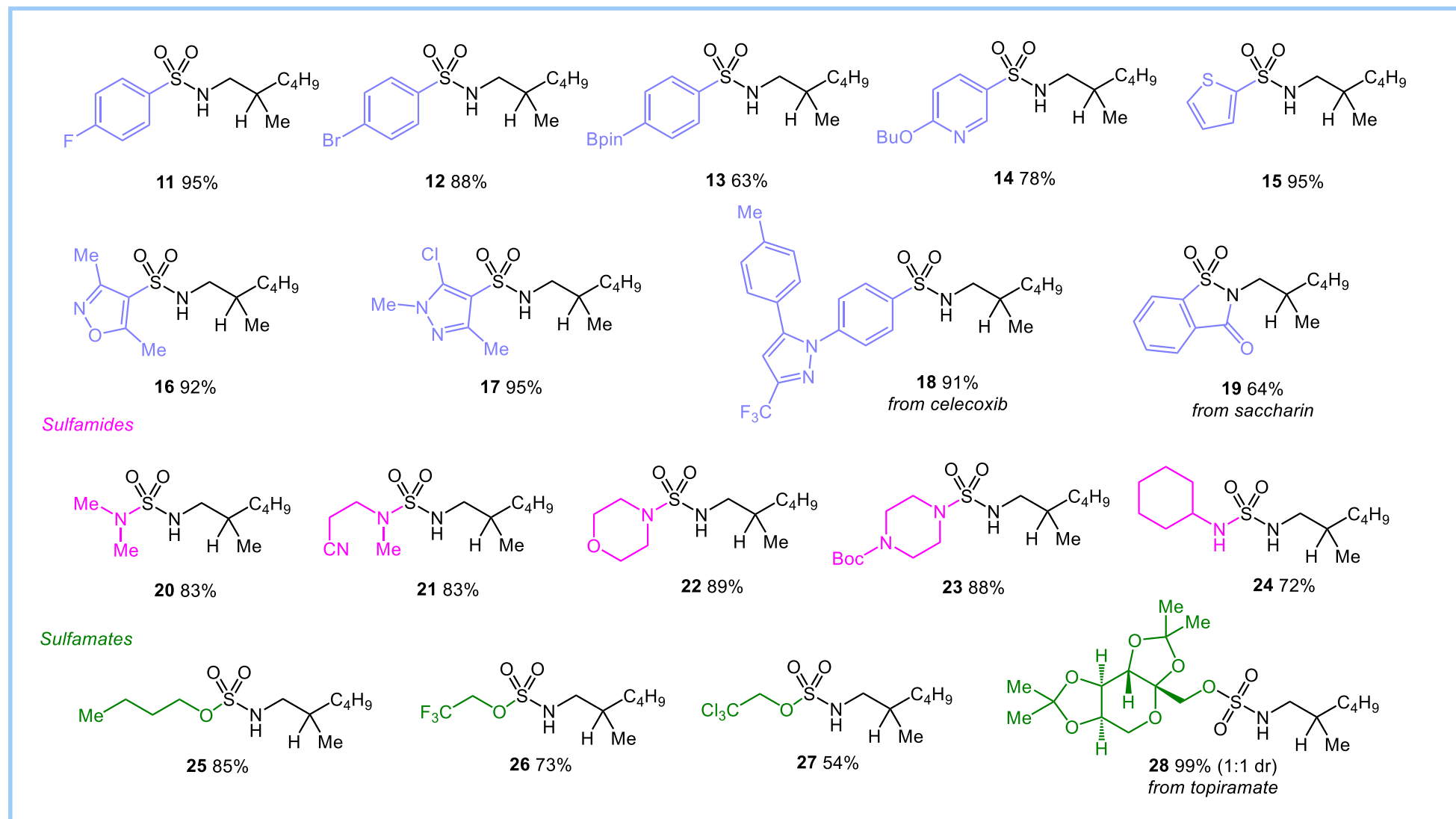


9 92%

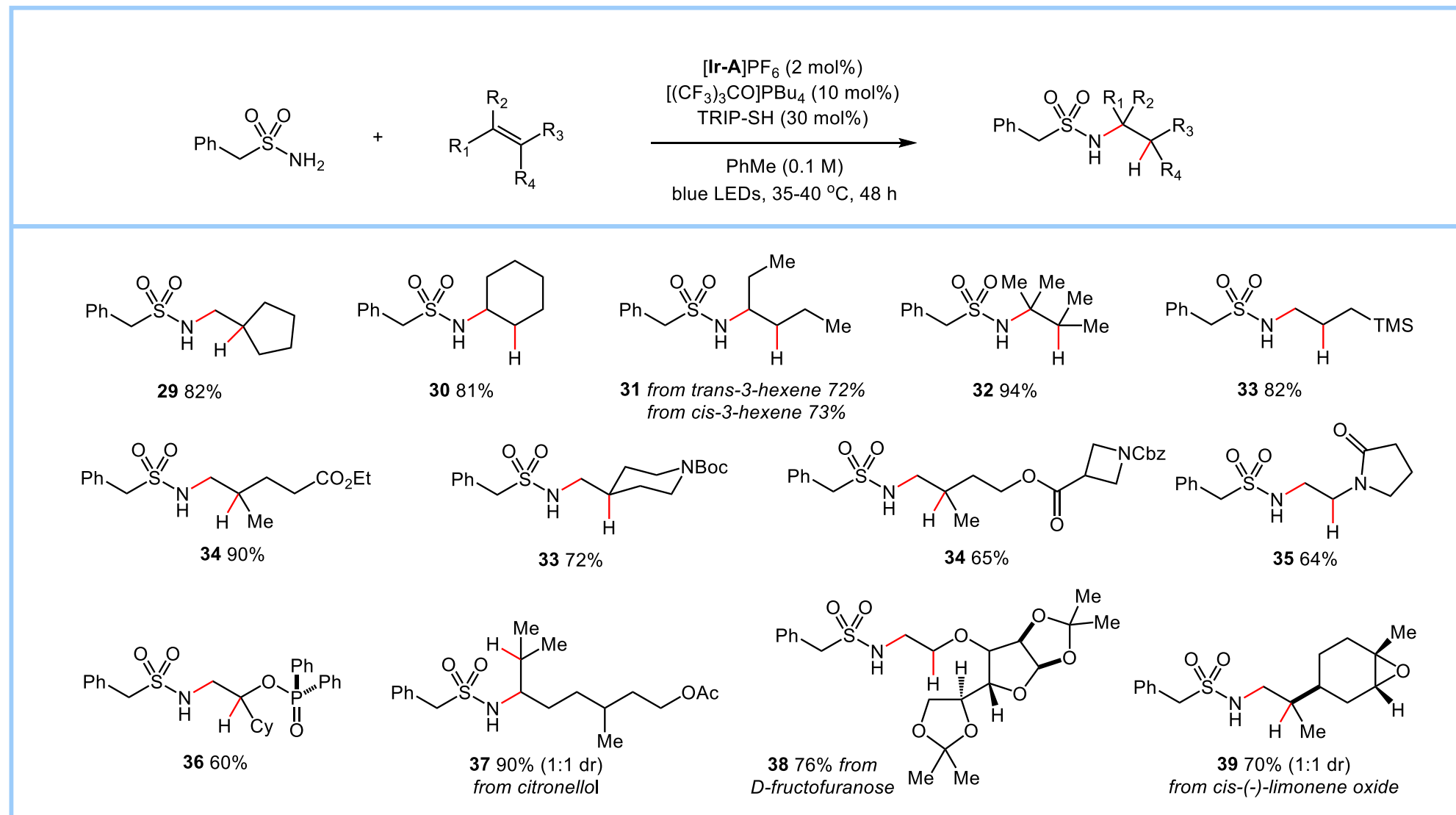


10 82%

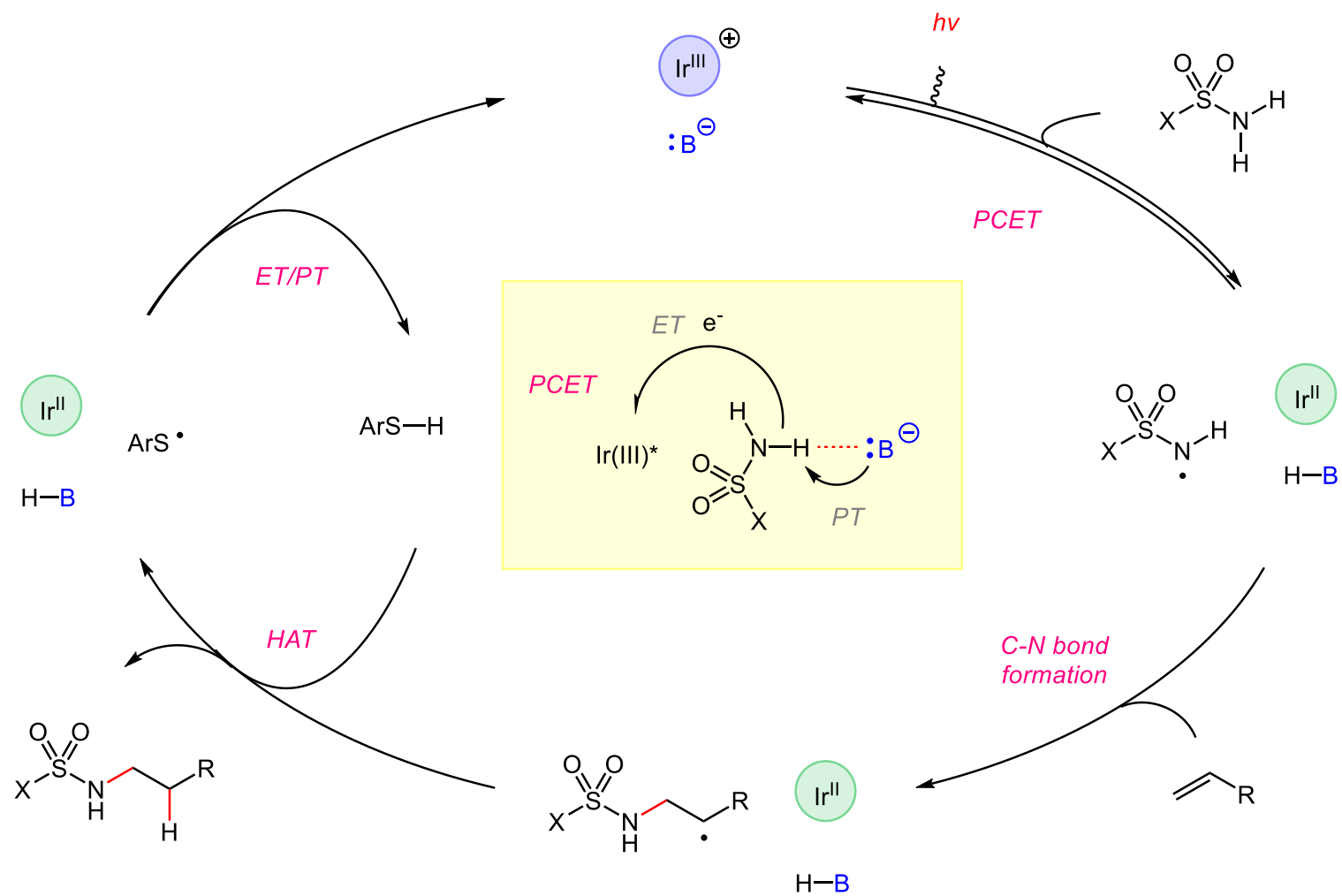
Substrate Scope



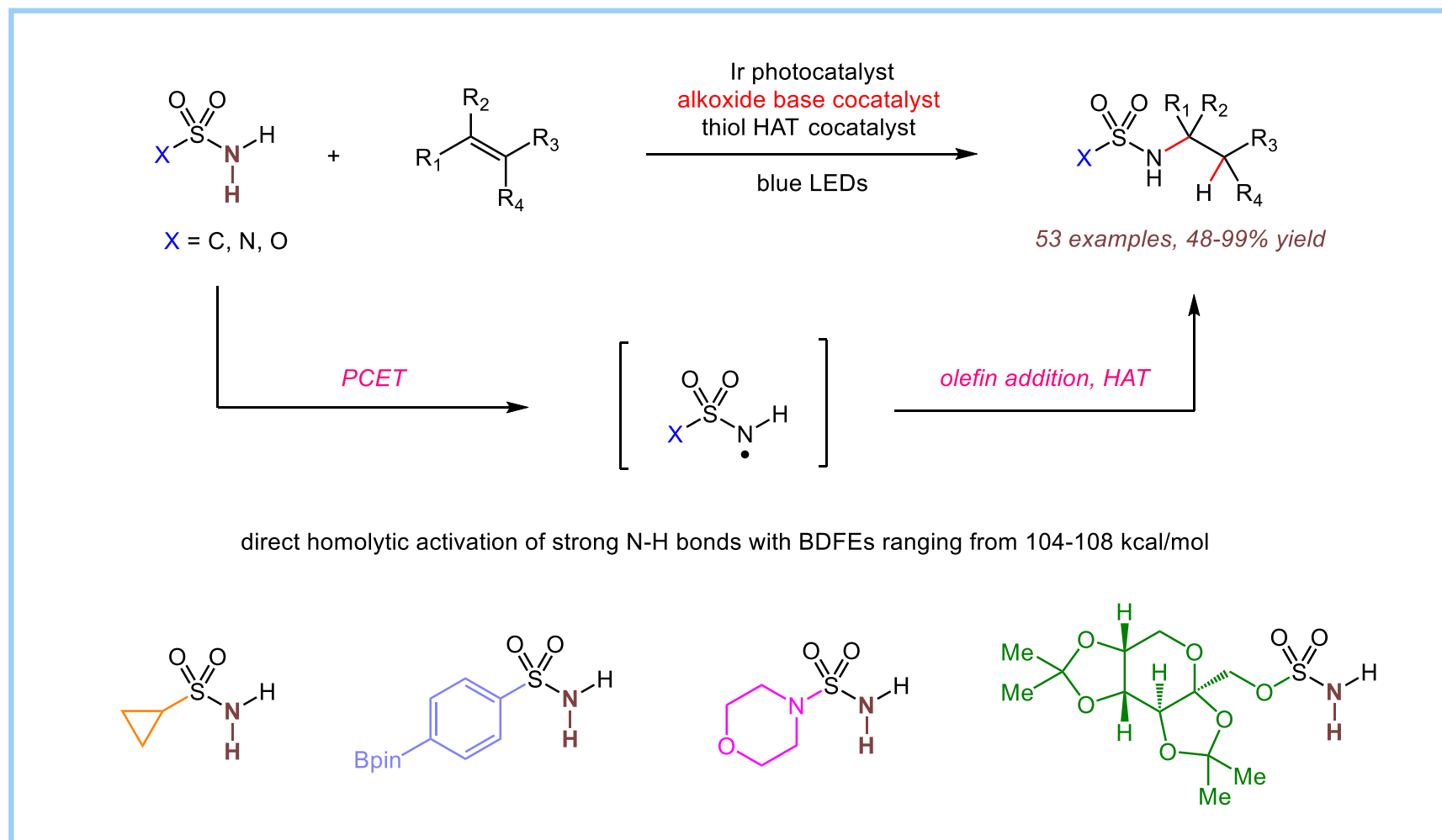
Substrate Scope



Proposed Mechanism

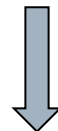


Summary



首段写作思路

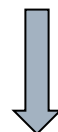
磺胺类化合物在药物中的应用



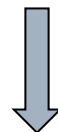
以往合成磺胺类化合物合成策略

末段写作思路

实现多种类型磺胺与烯烃的反马氏氢胺化



成功关键在于碱的选择



PCET在烯烃官能团化中具有较大潜力

Representative examples

- ❑ We previously leveraged this strategy to generate sulfonamidyl radicals under the **joint action of** $[\text{Ir}(\text{dF}(\text{CF}_3)\text{ppy})_2(5,5'\text{-d}(\text{CF}_3)\text{bpy})]\text{PF}_6$ ([Ir-A] PF_6) photocatalyst and tetrabutylammonium dibutyl phosphate base cocatalyst for the....(**联合作用**)
- ❑ While notable, all three protocols are **restricted** in scope with respect to either the alkene or sulfonamide component. (**受限制的**)
- ❑ Additional mechanistic work aimed at elucidating the **precise nature** of the electron and proton transfer steps involved in N-radical formation will require further investigation. (**精确的机制**)

Acknowledgment

***Thanks
for your attention !***