

www.angewandte.de

# **Akzeptierter Artikel**

Titel: Copper-Catalyzed Enantioselective Hydrophosphorylation of **Unactivated Alkynes** 

Autoren: Jie Kang, Kang Ding, Si-Mu Ren, Wen-Jun Yang, and Bo Su

Dieser Beitrag wurde nach Begutachtung und Überarbeitung sofort als "akzeptierter Artikel" (Accepted Article; AA) publiziert. Die deutsche Übersetzung wird gemeinsam mit der endgültigen englischen Fassung erscheinen. Die endgültige englische Fassung (Version of Record) wird ehestmöglich nach dem Redigieren und einem Korrekturgang als Early-View-Beitrag erscheinen und kann sich naturgemäß von der AA-Fassung unterscheiden. Leser sollten daher die endgültige Fassung, sobald sie veröffentlicht ist, verwenden. Für die AA-Fassung trägt der Autor die alleinige Verantwortung.

Zitierweise: Angew. Chem. Int. Ed. 2024, e202415314

Link zur VoR: https://doi.org/10.1002/anie.202415314

WILEY VCH

## Copper-Catalyzed Enantioselective Hydrophosphorylation of Unactivated Alkynes

Jie Kang,<sup>#[a]</sup> Kang Ding,<sup>#[a]</sup> Si-Mu Ren,<sup>[a]</sup> Wen-Jun Yang,<sup>[a]</sup> Bo Su<sup>\*[a]</sup>

<sup>#</sup>These authors contributed equally to this work.

This work is dedicated to John F. Hartwig on the occasion of his 60th birthday.

J. Kang, K. Ding, S.-M. Ren, W.-J. Yang, Prof. B. Su
 State Key Laboratory of Medical Chemical Biology, College of Pharmacy, Nankai University
 38 Tongyan Road, Jinnan District, Tianjin, 300350 (P. R. China)
 E-mail: subo@nankai.edu.cn

Supporting information for this article is given via a link at the end of the document.

Abstract: P-stereogenic phosphorus compounds are essential across various fields, yet their synthesis via enantioselective P-C bond formation remains both challenging and underdeveloped. We copper-catalyzed first enantioselective report the hydrophosphorylation of alkynes, facilitated by a newly designed chiral 1,2-diamine ligand. Unlike previous methods that rely on kinetic resolution with less than 50% conversion, our approach employs a distinct dynamic kinetic asymmetric transformation mechanism, achieving complete conversion of racemic starting materials. This reaction is compatible with a broad range of aromatic and aliphatic terminal alkynes, producing products with high yields (up to 95%), exclusive cis selectivity, and exceptional regio- and enantioselectivity (> 20:1 r.r. and up to 96% ee). The resulting products were further transformed into a diverse array of enantioenriched P-stereogenic scaffolds. Preliminary mechanistic studies were conducted to elucidate the reaction details.

P-chiral phosphorus compounds play pivotal roles across various fields,<sup>[1]</sup> including agrochemicals, pharmaceuticals,<sup>[2]</sup> material sciences, and asymmetric catalysis.<sup>[3]</sup> Their broad utility has sustained significant interest in their synthesis, making this a vibrant and enduring area of research.<sup>[4]</sup> Traditionally, these compounds have been synthesized using chiral auxiliary-based diastereoselective methods,<sup>[5],[6]</sup> which require stoichiometric amounts of chiral reagents and involve multiple synthetic steps. Although catalytic approaches, such as the desymmetrization of prochiral phosphorus compounds<sup>[7]</sup> and enantioselective transformations of secondary phosphines (SPs; H-PR<sup>1</sup>R<sup>2</sup>),<sup>[6b, 7i, 8]</sup> have been developed, these methods often face limitations, including the need for structurally constrained substrates and the use of highly toxic, volatile, and air- and moisture-sensitive P<sup>(III)</sup>, reagents.

Racemic pentavalent hydrophosphoryl compounds (H- $P^{(V)}(O)R^1R^2$ ) have recently emerged as highly attractive substrates for enantioselective P-C bond-forming reactions due to their ready availability, non-toxicity, bench stability, and ease of handling.<sup>[4f]</sup> Notable progress has been made in the transition metal-catalyzed enantioselective P-C cross-coupling of these compounds with aryl and alkyl (pseudo) halides,<sup>[9],[10]</sup> as well as in their asymmetric conjugate addition to activated alkenes, such as  $\alpha,\beta$ -unsaturated carbonyl compounds, using both transition-metal

catalysis or organocatalysis.<sup>[9d, 11]</sup> Despite these promising developments, enantioselective P-C bond-forming reactions with  $H-P(O)R^1R^2$  remains limited, highlighting the need for further exploration and innovations in this field.

The enantioselective addition of racemic H-P(O)R<sup>1</sup>R<sup>2</sup> alkynes-generally termed hydrophosphorylationacross provides a direct and atom-economical route to P-stereogenic compounds. Although Tanaka and Han first reported the hydrophosphorylation of alkynes in 1996,<sup>[12]</sup> the development of enantioselective variants has been impeded by significant challenges.<sup>[13]</sup> These challenges include limited reactivity<sup>[14]</sup> and the need to simultaneously control multiple selectivities, such as regioselectivity,<sup>[15]</sup> stereoselectivity (cis vs trans),<sup>[16]</sup> and enantioselectivity. It wasn't until 2020 that Wang and co-workers reported the first enantioselective addition of H-P(O)Ph(OR) to aromatic alkynes using palladium complexes with QuinoxP\*, achieving Markovnikov adducts with moderate to good enantioselectivity (30-86% ee) (Scheme 1a, top).[15a] In the same year, the Zhang group achieved the anti-Markovnikov, highly enantioselective addition of H-P(O)RPh to alkynes using a



Scheme 1. Enantioselective hydrophosphorylation of unactivated alkynes

palladium catalyst derived from their chiral sulfinamide phosphine ligand, Xiao-Phos (Scheme 1a, bottom).[15b] To date, these remain the only two examples of enantioselective hydrophosphorylation of unactivated alkynes, both relying on noble palladium catalysts.<sup>[15]</sup> Therefore, developing catalyst systems based on inexpensive, earth-abundant metals for these reactions remains a highly desirable and significant goal.

Beyond the challenges of reactivity and selectivity, another significant obstacle in developing practical enantioselective hydrophosphorylations is the difficulty in racemizing hydrophosphoryl compounds due to the high energy barrier required for interconversion between their two enantiomers. Consequently, this limitation restricts the reported hydrophosphorylations to proceed via kinetic resolution,[15],[17] only forming P-stereogenic products in less than 50% yield, as the hydrophosphoryl substrates act as the limiting reagents. In contrast, dynamic kinetic asymmetric transformation (DYKAT)<sup>[18]</sup> of hydrophosphoryl compounds can theoretically achieve up to 100% yield,<sup>[19],[20]</sup> making it a more efficient and attractive strategy. However, DYKAT processes are significantly more complex and rare, as they require a catalyst capable of not only delievering high reactivity and selectivity but also promoting the racemization.<sup>[18],[21]</sup> Although various transition metal catalysts have been developed for P-C bond-forming reactions, only two catalyst systemsnickel<sup>[9a]</sup> and copper<sup>[10g, 10i]</sup> <sup>[10d, 10e, 10j]</sup>—have been shown to facilitate racemization, and both have been exclusively applied to P-C cross-coupling reactions. Besides the catalyst, other reaction parameters, such as base, solvent, temperature, and substituent variations of phosphorus substrate,[9c] also play crucial roles in developing such DYKATs.<sup>[19b, 20]</sup> For instance, while the nickel

```
catalyst system with KOAc could promote the dynamic kinetic
asymmetric allylation of hydrophosphoryl compounds, the same
system with K<sub>3</sub>PO<sub>4</sub> failed to do so;<sup>[9a]</sup> similarly, nickel catalysts
were ineffective in catalyzing enantioselective arylation and
benzylation under dynamic kinetic conditions.<sup>[9f, 10f]</sup> Moreover,
side-reactions associated with hydrophosphoryl substrates often
present additional obstacles to achieving DYKATs with these
compounds.<sup>[10a]</sup>
                   These
                            factors
                                       collectively
                                                    render
                                                                the
development of a novel DYKAT with hydrophosphoryl compounds
highly complex and elusive.
```

Building on our ongoing work in developing enantioselective P-C bond-forming reactions,<sup>[10g, 10i]</sup> we report here the first earthabundant copper-catalyzed hydrophosphorylation of unactivated terminal alkynes (Scheme 1b), which also represents the inaugural dynamic kinetic asymmetric hydrophosphorylation of unsaturated C-C bonds. By utilizing a newly developed chiral 1,2diamine ligand, this hydrophosphorylation of terminal aryl and alkyl alkynes produces P-stereogenic products with high yields, exclusive cis- and anti-Markovnikov selectivity, and excellent enantioselectivity (up to 96% ee).

We initiated our investigation by examining various chiral ligands for the reaction between racemic 1a and phenylacetylene 2a, using Cul as the precatalyst and KOAc as the base in toluene (0.2 M) at room temperature for 24 hours (Table 1). Several commercially available ligands were tested, including bisphosphines L1 and L2, bisoxazoline L3, pyridyl oxazolines L4 and L5, pyridyl bisoxazoline L6, amino acid L7, and cyclic and acyclic 1,2-diamines L8 and L9. Among these, 1,2-diphenylethylene-1,2-diamine L9 led to the anti-Markovnikov hydrophosphorylated product 3a with a 63% yield and 64% ee.



<sup>[a]</sup>The yield refers to NMR yield with 2,4,6-trimethoxybenzene as internal standard; <sup>[b]</sup>Reaction was conducted in 2 mL of xylene at -20 °C for 96 h. <sup>[c]</sup>5 equiv of degassed water was added.

on [23/10/2024]

. See the Te

s of use; OA articles are governed by the applicable Creative Commons

i Instit

#### WILEY VCH

#### COMMUNICATION

To improve enantioselectivity, we systematically modified the structure of L9. Removing the methyl groups from the nitrogen atoms (L10) or replacing them with a bulkier ethyl or isopropyl groups (L11 and L12) resulted in no detectable ee, highlighting the importance of the *N*HMe groups in L9 for stereocontrol. We then explored electronic and steric effects by introducing substituents on the phenyl groups of L9. While *ortho-* and *meta*-substituents (L13 to L16) slightly reduced enantioselectivity, *para*-substituents (L17 to L19) increased it. Further modifications at the 3,5-positions of the aryl groups (L20 to L26) significantly increased both yield and enantioselectivity, with L26, bearing 3,4,5-OMe<sub>3</sub>-substituent aryl groups, providing the best results (95% yield and 81% ee).

We further optimized other conditions, including base, solvent, concentration, and temperature (see SI for details). Conducting the reaction in xylene at a lower concentration (0.05 M) and

temperature (-20 °C) increased the enantioselectivity to 93%, but the yield dropped to 51%, even after 96 hours. We attributed this low yield partially to the poor solubility of KOAc in xylene and therefore added 5 equivalents of water to the reaction. Gratifyingly, this adjustment led to a practical yield (82%) with high ee (94%) for product **3a**.

Having established conditions to form tertairy phosphine oxide (TPO) **3a** with high yield and enantioselectivity, we explored the scope of the alkyne component (Table 2). Both aromatic and aliphatic alkynes underwent hydrophosphorylation exclusively in a *cis*- and *anti*-Markovnikov fashion, yielding structurally diverse TPO products with high yields and enantioselectivity (up to 96% ee). The absolute configuration of **3a** was determined to be *S* by X-ray single-crystal diffraction analysis,<sup>[22]</sup> and other products were assigned by analogy.



Table 2. Scope of alkynes for the copper-catalyzed enantioselective hydrophosphorylation<sup>[a]</sup>

<sup>[</sup>a]Reactions were conducted with 1a (0.1 mmol), 2a (0.12 mmol) for 96 h, the yields refer to isolated yields, and the ee was determined by HPLC analysis.

#### WILEY VCH

#### COMMUNICATION

We first investigated aromatic alkynes with various *para*substituents, including alkyl (2b-2d), phenyl (2e), halides (F, Cl, Br; 2i-2k), electron-donating methoxyl (2f) and amino groups (2g and 2h), and electron-withdrawing trifluoromethyl (2l), ester (2m) and nitrile (2n). These alkynes efficiently reacted with SPO 1a, producing TPOs (3b-3n) with high enantioselectivity. Notably, the reaction tolerated a free amine (2h), yielding product 3h with 90% ee and a moderate yield (58%). *Meta-* and *ortho*-substituted aryl alkynes also performed well, affording TPOs 3o-3u with excellent enantioselectivities ( $\geq$  90% ee). Additionally, naphthyl acetylene (2v) and heteroaryl-containing alkynes (2w to 2z) were suitable substrates, leading to highly selective products.

We then studied aliphatic alkynes with various functionalities, including alkyl (**2aa** and **2ab**), phenyl (**2ac**), alkyl chloride (**2ad**), ethers (**2ae** to **2ag**), and ester (**2ah**). These alkynes reacted smoothly and selectively, yielding TPOs in high yields and ee values comparable to those from aromatic alkynes. When enyne **2ai** was used, hydrophosphorylation chemo- and regioselectively occurred at the alkyne moiety, producing TPO **3ai** with 90% ee, albeit in a low yield (21%). Internal alkynes, such as diphenylacetylene and 1-butynylbenzene, did not react under the standard conditions.

Next, we explored the scope of the SPO component for the enantioselective hydrophosphorylation reaction (Table 3). Both alkyl and aryl groups in SPOs were varied. SPOs with either primary (**1b-1d**) or secondary (**1e-1g**) alkyl groups reacted regioand stereoselectively with phenylacetylene, yielding TPOs (**4b-4g**) with 85%-96% ee. Reactions with SPOs bearing bulkier secondary alkyl groups showed decreased reactivity, as seen with the slow formation of **4e** (120 hours, 70% yield). SPOs with either *para*- or *meta*-substituted aryl groups (**1h-1o**) were also

Table 3. Scope of SPOs for the enantioselective hydrophosphorylation<sup>[a]</sup>

Cul (10 mol%), **L26** <sup>(15 mol%)</sup> KOAc (1.0 equiv), H<sub>2</sub>O (5.0 equiv) Δr Ph '//Ar . C. Ar. 96 h k 2a (0.(<sup>±</sup>)n1nol) 4 (1.2 equiv) Variation of R group 0 0 ''Ph ′′′Ph Ph ''Ph Ph Ph Me Et Řu 4b, 83%, 93% ee 4c, 94%, 96% ee 4d, 80%, 94% ee "′Ph ۰Ph Ph Me Me 4e<sup>,b</sup> 70%, 88% ee 4f, 73%, 90% ee 4g, 55%, 85% ee Variation of Ar group ОЦ 0 Ph Ēt Ł Ēt fBu 4h, 85%, 96% ee 4i, 81%, 94% ee **4j**, 93%, 95% ee Ph Ph Ph Et Ét Ēt NMe<sub>2</sub> 4k, 67%, 84% ee 4l, 85%, 96% ee 95%, 94% ee 4m. 0 Me Ph Ēt Ēt 4**p**, R==**b**h, 27%, 32% ee 40, 75%, 96% ee 4n, 67%, 94% ee

<sup>[a]</sup>Reactions were conducted with **1a** (0.1 mmol) and **2a** (0.12 mmol). <sup>[b]</sup>Reaction was conducted at 0 °C for 120 h.

compatible. However, SPOs with *para*-methyl substituted aryl group (**1p** and **1q**) or with *tert*-butyl and methyl groups (structure not shown) did not undergo hydrophosphorylation under standard conditions. While these sterically hindered SPOs could react at elevated temperatures, both the yield and enantioselectivity decreased significantly (**1q**: 27% yield, 32% ee at 0 °C). Additionally, SPO bearing an ethoxyl group (ethyl phenylphosphinate) exhibited lower reactivity, yielding the desired product in 15% yield with 57% ee at room temperature.

To showcase the synthetic utility of the newly developed enantioselective hydrophosphorylation, we conducted scaled-up reactions and transformed the resulting enantioenriched products into structurally diverse P-chiral scaffolds (Scheme 2). Hydrophosphorylation reactions with SPOs 1a and 1b on a 2 mmol scale formed products with yields and ee values comparable to those obtained on a 0.1 mmol scale (Scheme 2a). We then explored transformations to further diversify the alkenylsubstituted TPO products (Scheme 2b). The P(V)-3a was stereoinversely reduced under conditions with MeOTf and LiAIH<sub>4</sub>,<sup>[23]</sup> followed by boron complexation, producing P<sup>(III)</sup>-5 in 76% yield and 90% enantiopurity. The electron-withdrawing phosphoryl functionality enabled the C-C double bonds in the alkenyl-substituted TPO to act as Michael receptors.<sup>[24]</sup> For instance, TPO 3a underwent 1,4-addition with nucleophilic diphenyl phosphine, followed by oxidation, forming 1,2bisphosphine oxides 6 in 68% yield as a mixture of two diastereomers (d.r. = 4.2:1). Similarly, reaction with nucleophilic N-methylhydroxylamine produced 1,3-N, P-containing scaffold 7. The alkenyl group in 4b was efficiently reduced to an alkyl group via palladium-catalyzed hydrogenation, yielding dialkyl aryl TPO 8 in 93% yield while maintaining enantiopurity. Deprotonation of



Scheme 2. Scalability and diverse transformations

. See the Te

for

rules of use; OA articles are governed by the applicable Creative Commons

TPO **8** with *n*BuLi preferentially occurred at the methyl group, and subsequent trapping with cyclohexanone afforded TPO **9** in 85% yield and 89% ee. Bisphosphine oxide **10** was efficiently synthesized from TPO **4q** and diphenyl phosphine oxide using palladium-catalyzed C-P cross-coupling conditions. These transformations yield P-stereogenic compounds that are difficult to produce using conventional methods, making them valuable precursors for the development of novel chiral ligands or organocatalysts.<sup>[3a, 3b, 25]</sup>

We conducted various reactions to gain preliminary insights into the mechanism of this enantioselective hydrophosphorylation (Scheme 3). First, we examined the variations in enantiomeric excess (ee) of the resulting TPO **3a** and the unreacted SPO **1a** during the reaction. The ee of **3a** remained consistently high (92%-94%) throughout the reaction (Scheme 3a), indicating that racemic SPOs reacted *via* a dynamic kinetic asymmetric transformation mechanism. Meanwhile, the ee of the recovered unreacted **1a**, with (S)-**1a** as the major enantiomer, gradually increased up to 46%, suggesting that the interconversion between (S)-**1a** and (*R*)-**1a** (racemization) was slower than the formation of (S)-**3a**. Parallel reactions with (*R*)-**1a** and (*S*)-**1a** showed that both led to the formation of (S)-**3a** as the major component, but (S)-**1a** reacted much slower and significantly less enantioselectively than (*R*)-**1a** (Scheme 3b).<sup>[26]</sup> These results suggest that (*R*)-**1a** is the preferred enantiomer in the reaction and reacts in a stereoretentive fashion.

To understand to role of added water in the dynamic kinetic asymmetric hydrophosphorylation, we compared reactions with



Scheme 3. Mechanistic studies and proposed catalytic cycle

and without added water (Scheme 3c vs Scheme 3a). In the reaction without water, product **3a** was also formed with consistently high ee values (90%-94% ee), and racemization of SPO **1a** occurred as well, indicating that water is not essential for the stereocontrol of the hydrophosphorylation and the racemization of SPOs. However, water significantly influenced the reaction rate: the reaction with water yielded product **3a** in 74% yield in 4 days, whereas the reaction without water only achieved 51% yield in the same period and less than 60% yield even after 7 days. This rate-accelerating effect of water is likely due to the increased solubility of KOAc in the reaction system.

We further investigated the factors causing SPO racemization by subjecting (*S*)-**1a** to various control conditions (Scheme 3d). (*S*)-**1a** did not racemize over 48 hours with either KOAc or copper complexes alone (Scheme 3d, entries 1 and 2). In contrast, nearly complete racemization occurred in the presence of both KOAc and copper complexes, regardless of the presence of water (Scheme 3d, entries 3 and 4). These results suggest that both the base and copper complexes are essential for SPO racemization. While the exact mechanism behind the racemization remains unclear, it is likely driven by a base-promoted pyramidal inversion of the trivalent phosphinous acid (the tautomer of SPO) or its conjugate base.<sup>[27]</sup> In this process, the copper complex is thought to coordinate with the oxygen of SPO, thereby increasing its acidity and facilitating the inversion.

Based on these preliminary results and related literature,<sup>[16],[28]</sup> we proposed a mechanism for the copper-catalyzed enantioselective hydrophosphorylation of alkynes, exemplified with reaction between **1a** and **2a** (Scheme 3e). Tautomerization of pentavalent racemic SPO (+/-)-**1a** leads to two trivalent enantiomers, phosphinous acids (*S*)-**PA** and (*R*)-**PA**. (*R*)-**PA** reacts with in situ generated chiral copper catalysts [Cu\*]-I in the presence of base to form a phosphoryl copper intermediate (*S*)-P(O)-[Cu\*], which then reacts with alkyne **2a** through coordination and *cis*-addition to yield (*S*)-alkenyl-[Cu\*] species. This species subsequently forms the *anti*-Markovnikov product (*S*)-**3a** upon protonation. This process, forming (*S*)-**3a**, is significantly faster than the parallel formation of (*R*)-**3a** from (*S*)-**PA**.

In conclusion, we have successfully developed the first copper-catalyzed enantioselective hydrophosphorylation of alkynes, utilizing a newly designed chiral 1,2-diamine ligand. This work represents the first example of dynamic kinetic asymmetric hydrophosphorylation of unsaturated carbon-carbon bonds, driven by a unique copper catalyst capable of racemizing of SPOs. The reaction exhibits broad tolerance for unactivated terminal alkynes, consistently producing structurally diverse P-stereogenic phosphorus compounds with exclusive *cis*-selectivity and high regio- and enantioselectivity (up to >20:1 r.r. and 96% ee). Mechanistic studies further elucidate the pivotal roles of the copper catalyst and base in facilitating SPO racemization, while water acts as a rate-accelerating factor by improving the solubility of the inorganic base in the reaction medium.

#### Acknowledgements

We thank the National Natural Science Foundation of China (22271161, 22188101), the Tianjin Science Fund for Distinguished Young Scholars (23JCJQJC00180), the Fundamental Research Funds for the Central Universities (63233165; 63243134), Nankai University, for financial support.

**Keywords:** *P*-stereogenic compounds • asymmetric synthesis • dynamic kinetic resolution • copper catalysis• transition metal catalysis

- [1] For reviews, see: a) C. A. Busacca, C. H. Senanayake, in *Comprehensive Chirality* (Eds.: E. M. Carreira, H. Yamamoto), Elsevier, Amsterdam, **2012**, pp. 167-216; b) M. Dutartre, J. Bayardon, S. Juge, *Chem. Soc. Rev.* **2016**, *45*, 5771-5794.
- [2] For reviews, see: a) J. B. Rodriguez, C. Gallo-Rodriguez, *ChemMedChem.* 2019, 14, 190-216; b) H. Yu, H. Yang, E. Shi, W. Tang, *Med. Drug. Discov.* 2020, *8*, 100063.
- [3] For reviews, see: a) W. Tang, X. Zhang, *Chem. Rev.* 2003, *103*, 3029-3070; b) H. Guo, Y. C. Fan, Z. Sun, Y. Wu, O. Kwon, *Chem. Rev.* 2018, *118*, 10049-10293; c) H. Ni, W. L. Chan, Y. Lu, *Chem. Rev.* 2018, *118*, 9344-9411.
- [4] For reviews and perspectives, see: a) S. Jugé, *Phosphorus, Sulfur, and Silicon and the Related Elements* 2008, *183*, 233-248; b) J. S. Harvey, V. Gouverneur, *Chem. Commun.* 2010, *46*, 7477-7485; c) S. Lemouzy, L. Giordano, D. Herault, G. Buono, *Eur. J. Org. Chem.* 2020, *2020*, 3351-3366; d) C. Luan, C.-J. Yang, L. Liu, Q.-S. Gu, X.-Y. Liu, *Chem Catalysis* 2022, *2*, 2876-2888; e) C. Luo, Y. Yin, Z. Jiang, *Chin. J. Org. Chem.* 2023, *43*, 1963-1976; f) K. Ding, B. Su, *Eur. J. Org. Chem.* 2024, *27*, e202301160.
- [5] For reviews, see: a) K. M. Pietrusiewicz, M. Zablocka, *Chem. Rev.* 1994, 94, 1375-1411; b) A. Grabulosa, J. Granell, G. Muller, *Coord. Chem. Rev.* 2007, 251, 25-90; c) P. Bagi, V. Ujj, M. Czugler, E. Fogassy, G. Keglevich, *Dalton Trans.* 2016, 45, 1823-1842.
- [6] For selected examples, see: a) Z. X. S. Han, N. Goyal, M. A. Herbage, J. D. Sieber, B. Qu, Y. B. Xu, Z. B. Li, J. T. Reeves, J. N. Desrosiers, S. L. Ma, N. Grinberg, H. Lee, H. P. R. Mangunuru, Y. D. Zhang, D. Krishnamurthy, B. Z. Lu, J. H. J. Song, G. J. Wang, C. H. Senanayake, *J. Am. Chem. Soc.* 2013, *135*, 2474-2477; b) O. Berger, J. L. Montchamp, *Angew. Chem. Int. Ed.* 2013, *52*, 11377-11380; c) S. Rast, B. Mohar, M. Stephan, *Org. Lett.* 2014, *16*, 2688-2691; d) K. Nikitin, K. V. Rajendran, H. Müller-Bunz, D. G. Gilheany, 2014, *53*, 1906-1909; e) D. Xu, N. Rivas-Bascón, N. M. Padial, K. W. Knouse, B. Zheng, J. C. Vantourout, M. A. Schmidt, M. D. Eastgate, P. S. Baran, *J. Am. Chem. Soc.* 2020, *142*, 5785-5792.
- [7] For selected examples, see: a) Y. Toda, M. Pink, J. N. Johnston, J. Am. Chem. Soc. 2014, 136, 14734-14737; b) G. Q. Xu, M. H. Li, S. L. Wang, W. J. Tang, Org. Chem. Front. 2015, 2, 1342-1345; c) Y. Sun, N. Cramer, Angew. Chem. Int. Ed. 2017, 56, 364-367; d) Z. Wang, T. Hayashi, Angew. Chem. Int. Ed. 2018, 57, 1702-1706; e) B. M. Trost, S. M. Spohr, A. B. Rolka, C. A. Kalnmals, J. Am. Chem. Soc. 2019, 141, 14098-14103; f) R. Y. Zhu, L. Chen, X. S. Hu, F. Zhou, J. Zhou, Chem. Sci. 2020, 11, 97-106; g) S.-Y. Song, Y. Li, Z. Ke, S. Xu, ACS Catal. 2021, 11, 13445-13451; h) K. C. Forbes;, E. N. Jacobsen, Science 2022, 376, 1230-1236; i) C.-W. Zhang, X.-Q. Hu, Y.-H. Dai, P. Yin, C. Wang, W.-L. Duan, ACS Catal. 2022, 12, 193-199; j) S. B. Yan, R. Wang, Z. G. Li, A. N. Li, C. Wang, W. L. Duan, Nat. Commun. 2023, 14, 2264; k) M. Formica, T. Rogova, H. Shi, N. Sahara, B. Ferko, A. J. M. Farley, K. E. Christensen,

F. Duarte, K. Yamazaki, D. J. Dixon, Nat. Chem. 2023, 15, 714-721; I) M. Formica, B. Ferko, T. Marsh, T. A. Davidson, K. Yamazaki, D. J. Dixon, Angew. Chem. Int. Ed. 2024, 63, e202400673.

- [8] For selected examples, see: a) J. R. Moncarz, N. F. Laritcheva, D. S. Glueck, J. Am. Chem. Soc. 2002, 124, 13356-13357; b) C. Scriban, D. S. Glueck, J. Am. Chem. Soc. 2006, 128, 2788-2789; c) N. F. Blank, J. R. Moncarz, T. J. Brunker, C. Scriban, B. J. Anderson, O. Amir, D. S. Glueck, L. N. Zakharov, J. A. Golen, C. D. Incarvito, A. L. Rheingold, J. Am. Chem. Soc. 2007, 129, 6847-6858; d) V. S. Chan, R. G. Bergman, F. D. Toste, J. Am. Chem. Soc. 2007, 129, 15122-15123; e) Y.-B. Li, H. Tian, L. Yin, J. Am. Chem. Soc. 2020, 142, 20098-20106; f) L. B. Balázs, Y. Huang, J. B. Khalikuzzaman, Y. Li, S. A. Pullarkat, P.-H. Leung, J. Org. Chem. 2020, 85, 14763-14771; g) S. Zhang, J.-Z. Xiao, Y.-B. Li, C.-Y. Shi, L. Yin, J. Am. Chem. Soc. 2021, 143, 9912-9921; h) X.-T. Liu, X.-Y. Han, Y. Wu, Y.-Y. Sun, L. Gao, Z. Huang, Q.-W. Zhang, J. Am. Chem. Soc. 2021, 143, 11309-11316; i) A. Mondal, N. O. Thiel, R. Dorel, B. L. Feringa, Nat. Catal. 2022, 5, 10-19; j) W.-H. Wang, Y. Wu, P.-J. Qi, Q.-W. Zhang, ACS Catal. 2023, 13, 6994-7001; k) B. Zhang, W. Q. Zhou, X. T. Liu, Y. Sun, Q. W. Zhang, Chem. Sci. 2023, 14, 1286-1290; I) X. B. Chen, D. Padin, C. N. Stindt, B. L. Feringa, Angew. Chem. Int. Ed. 2023, 62, e202307450.
- [9] For selected examples, see: a) X.-T. Liu, Y.-Q. Zhang, X.-Y. Han, S.-P. Sun, Q.-W. Zhang, J. Am. Chem. Soc. 2019, 141, 16584-16589; b) H. Qiu, Q. Dai, J. He, W. Li, J. Zhang, Chem. Sci. 2020, 11, 9983-9988; c) Q. Zhang, X.-T. Liu, Y. Wu, Q.-W. Zhang, Org. Lett. 2021, 23, 8683-8687; d) Z.-H. Wu, A.-Q. Cheng, M. Yuan, Y.-X. Zhao, H.-L. Yang, L.-H. Wei, H.-Y. Wang, T. Wang, Z. Zhang, W.-L. Duan, Angew. Chem. Int. Ed. 2021, 60, 27241-27246; e) Q. Dai, L. Liu, J. Zhang, Angew. Chem. Int. Ed. 2021, 60, 27247-27252; f) W.-Q. Cai, Q. Wei, Q.-W. Zhang, Org. Lett. 2022, 24, 1258-1262; g) Z. H. Wu, H. Y. Wang, H. L. Yang, L. H. Wei, T. Hayashi, W. L. Duan, Angew. Chem. Int. Ed. 2022, 61, e202213904.
- [10] a) R. Beaud, R. J. Phipps, M. J. Gaunt, J. Am. Chem. Soc. 2016, 138, 13183-13186; b) Y. Zhang, H. He, Q. Wang, Q. Cai, Tetrahedron Lett. 2016, 57, 5308-5311; c) Q. Dai, W. Li, Z. Li, J. Zhang, J. Am. Chem. Soc. 2019, 141, 20556-20564; d) Y. Li, X. Jin, P. Liu, H. Zhang, X. Yu, Y. Liu, B. Liu, W. Yang, Angew. Chem. Int. Ed. 2022, 61, e202117093; e) B. Liu, P. Liu, X. Wang, F. Feng, Z. Wang, W. Yang, Org. Lett. 2023, 15, 2178-2183; f) R. Cui, Y. Wang, L. Yuwen, L. Gao, Z. Huang, W. H. Wang, Q. W. Zhang, Org. Lett. 2023, 25, 6139-6142; g) J. Kang, K. Ding, S.-M. Ren, B. Su, Angew. Chem. Int. Ed. 2023, 62, e202301628; h) C. Wang, X. Hu, C. Xu, Q. Ge, Q. Yang, J. Xiong, W. L. Duan, Angew. Chem. Int. Ed. 2023, 62, e202300011; i) J. Kang, S. Ren, B. Su, Synlett 2023, 35, 741-746; j) X. Wang, B. Liu, L. Ge, S. Hou, M. Liu, W. Yang, Adv. Synth. Catal. 2024, 366, 2285-2291.
- a) Y.-Q. Zhang, X.-Y. Han, Y. Wu, P.-J. Qi, Q. Zhang, Q.-W. Zhang, [11] Chem. Sci. 2022, 13, 4095-4102; b) Z. Huang, X.-T. Liu, R. Cui, Q.-W. Zhang, Org. Biomol. Chem. 2023, 21, 3096-3100; c) B. Wang, Y. Liu, C. Jiang, Z. Cao, S. Cao, X. Zhao, X. Ban, Y. Yin, Z. Jiang, Angew. Chem. Int. Ed. 2023, 62, e202216605.
- [12] H. Li-Biao, T. Masato, J. Am. Chem. Soc. 1996, 118, 1571-1572.
- [13] a) C. A. Bange, R. Waterman, Chem. Eur. J. 2016, 12598-12605; b) B. T. Novas, R. Waterman, ChemCatChem 2022, 14, e202200988.
- [14] L. Coudray, J.-L. Montchamp, Eur. J. Org. Chem. 2008, 2008, 3601-3613.
- a) Z. Yang, X. Gu, L. B. Han, J. J. Wang, Chem. Sci. 2020, 11, 7451-[15] 7455; b) Q. Dai, L. Liu, Y. Qian, W. Li, J. Zhang, Angew. Chem. Int. Ed. 2020, 59, 20645-20650.
- [16] M. Niu, H. Fu, Y. Jiang, Y. Zhao, Chem. Commun. 2007, 272-274.

- [17] a) H. B. Kagan, J. C. Fiaud, Top. Stereochem. 1998, 18, 249-330; b) J. M. Keith, J. F. Larrow, E. N. Jacobsen, Adv. Synth. Catal. 2001, 343, 5-26.
- V. Bhat, E. R. Welin, X. L. Guo, B. M. Stoltz, Chem. Rev. 2017, 117, [18] 4528-4561.
- a) R. D. Baechler, K. Mislow, J. Am. Chem. Soc. 1970, 92, 3090-3093; [19] b) A. Christiansen, C. Li, M. Garland, D. Selent, R. Ludwig, A. Spannenberg, W. Baumann, R. Franke, A. Börner, Eur. J. Org. Chem. 2010, 2010, 2733-2741; c) K. D. Reichl, D. H. Ess, A. T. Radosevich, J. Am. Chem. Soc. 2013, 135, 9354-9357.
- A. Gallen, A. Riera, X. Verdaguer, A. Grabulosa, Catal. Sci. Technol. [20] 2019, 9, 5504-5561.
- [21] a) P. K. Dornan, K. G. M. Kou, K. N. Houk, V. M. Dong, J. Am. Chem. Soc. 2014, 136, 291-298; b) C. K. Hazra, Q. Dherbassy, J. Wencel-Delord, F. Colobert, Angew. Chem. Int. Ed. 2014, 53, 13871-13875; c) C. Yu, H. Huang, X. Li, Y. Zhang, W. Wang, J. Am. Chem. Soc. 2016, 138, 6956-6959; d) K. M. H. Lim, T. Hayashi, J. Am. Chem. Soc. 2017, 139, 8122-8125; e) X. Huang, W. R. J. J. Oh, J. S. Zhou, Angew. Chem. Int. Ed. 2018, 57, 7673-7677; f) M. Pareek, R. B. Sunoj, ACS Catal. 2020, 10, 4349-4360.
- Deposition Number 2305381 (3a) contains the supplementary [22] crystallographic data for this paper. These data are provided free of charge by the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service.
- K. V. Rajendran, D. G. Gilheany, Chem. Commun. 2012, 48, 817-819. [23]
- [24] M. Oliana, F. King, P. N. Horton, M. B. Hursthouse, K. K. Hii, J. Org. Chem. 2006, 71, 2472-2479.
- H. Fernandez-Perez, P. Etayo, A. Panossian, A. Vidal-Ferran, Chem. [25] Rev. 2011. 111. 2119-2176.
- Q. Xu, C.-Q. Zhao, L.-B. Han, J. Am. Chem. Soc. 2008, 130, 12648-[26] 12655.
- [27] a) T. L. Emmick, R. L. Letsinger, J. Am. Chem. Soc. 1968, 90, 3459-3465; b) W. B. Farnham, R. A. Lewis, R. K. M. Jr., K. Mislow, J. Am. Chem. Soc. 1970, 92, 5808-5809.
- [28] Y. Gao, G. Wang, L. Chen, P. Xu, Y. Zhao, Y. Zhou, L.-B. Han, J. Am. Chem. Soc. 2009, 131, 7956-7957.

### Entry for the Table of Contents



The first copper-catalyzed enantioselective hydrophosphorylation of alkynes is reported, marking the inaugural example of dynamic kinetic asymmetric hydrophosphorylation of unsaturated carbon-carbon bonds. The reaction exhibits a broad scope of unactivated alkynes, consistently yielding structurally diverse P-stereogenic phosphorus compounds with exclusive *cis*-selectivity and high regioand enantioselectivity (up to >20:1 r.r. and 96% ee).

Researcher Twitter usernames: @subohxx