Literature Report III

Ni-Catalyzed Regiodivergent and Stereoselective Hydroalkylation of Acyclic Branched Dienes with Unstabilized C(sp³) Nucleophiles

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Mazet, C. et al. J. Am. Chem. Soc. 2020, 142, 16486.

CV of Prof. Clement Mazet



Education:

- > 1998-1999 B.S., University of Strasbourg
- > 1999-2002 B.A., University of Strasbourg
- 2003-2005 Postdoctoral fellow, University of Basel
- > 2006-2007 Postdoctoral fellow, Harvard University
- > 2007-2011 Lecturer, University of Geneva
- 2011-2014 Swiss National Foundation Assistant Professor, University of Geneva
- 2014-present Associate Professor, University of Geneva

Research:

- Isomerizations
- **Cross-Couplings**
- Ligand and Catalyst Design
- Mechanistic Studies



Hartwig, J. F. et al. J. Org. Chem. 2004, 69, 7552.



Hartwig, J. F. et al. J. Org. Chem. 2004, 69, 7552.



Malcolmson, S. J. et al. J. Am. Chem. Soc. 2018, 140, 2761.



Malcolmson, S. J. et al. Org. Lett. 2020, 22, 2032.



Zhou, Q.-L. et al. J. Am. Chem. Soc. 2018, 140, 11627.



Zi, W. et al. J. Am. Chem. Soc. 2019, 141, 14554.

Proposed Mechanism



Zi, W. et al. J. Am. Chem. Soc. 2019, 141, 14554.

Summary

Dienes



terminal diene



internal diene

R

branched diene

Nu







stabilized Nu

stabilized Nu

unstabilized Nu

Systems

Pd/L Ni/L Pd,Cu/L

Previous work



Mazet, C. et al. J. Am. Chem. Soc. 2019, 141, 14814.

	$Ar = 4-Ph-C_6H_4$	Ме +	Me 2a [Ni(c) ba Me 2a [Ni(c) ba Me 3a	and (6.0 mol%) od)₂] (5.0 mol%) ase (<i>n</i> equiv.) Solvent [0.5] 23 °C, 16 h	$\frac{Ph}{Me} = 0$ $3,4$ -addition	O NR ₁ R ₂ Ph Me
Entry	Nu	L	base (n equiv)	solvent	Conv. (%) ^b	3,4-/1,4-addition ^b
1	2a	L1	none ^c	mesitylene	<5	-
2 ^d	2a	L2	[#] BuOK (0.2)	EtOH ^e	<5	-
3 ^f	2a	L1	^t BuOK (2.0)	THF	>95	1:19 (<i>E/Z</i> 2.5:1)
4 ^f	2a	L1	MeOK (0.2)	THF	>95	1:>25 (<i>E</i> /Z 9:1)
5 ^g	2a ^h	L1	MeOK (0.2)	THF	>95	1:>25 <i>(E</i> / <i>Z</i> 15:1)
6	3a	L1	MeOK (0.2)	THF	>95	6.2:1 (<i>dr</i> 2.7:1)
7	3a	L1	BTMG (1.0)	THF	>95	6.5:1 <i>(dr</i> 2.8:1)
8	2a	L1	BTMG (1.0)	THF	<5	-

^a Reaction condition: **1a** (0.1 mmol), **2a** or **3a** (0.2mmol). ^b Determined by ¹H NMR using an internal standard. ^c1.0 equiv of trifluoroethanol. ^d1 00 °C. ^e0.2 M. ^f50 °C. ^g48 h. ^h1.2 equiv.



Ar + Me Ph			Ligand (6.0 mol%) [Ni(cod) ₂] (5.0 mol%) base (2.0 equiv.)		A	r Ph	NR ₂ +		
7.4	Al He			Solvent [0.5], 50 ^o C, 16 h		Me O		Pn Me	
1a Ar = 4-P	h-C ₆ H ₄	2a (1.2 equiv.)				4aa 3,4-addition	Аі 1,4-ғ	5 addition	
	Entry	Ligand	base	solvent	1a Conv. (%)	4aa (%)	5aa (%)	<i>E/Z</i> (5aa)	
PCy ₂ N	1	L1	None ^b	mesitylene	>99	5	95	2.5:1	
L1 Cy-Phox	2 ^c	L2	KO ^{<i>t</i>} Bu	EtOH	>99	5	95	2.5:1	
PAr ₂	3	L2	KOʻBu	THF	>99	5	95	2.5:1	
PAr ₂	4	L1	KOMe	THF	>99	0	>95	9:1	
(S)-L2 DTBM-SegPhos	5	L1	Et ₃ N	THF	35	0	0	-	

OMe

^a Reaction conditions: all reactions were performed with **1a** (0.10 mmol, 1.0 equiv.), **2a** (0.2 mmol, 2.0 equiv.), Ni(cod)₂ (0.005 mmol, 5 mol%), Ligand (0.006 mmol, 6 mol%), base (0.20 mmol, n equiv) in THF (2.0 mL, 0.5 M) at 50 °C. Consumption of 1a, conversions and E/Z ratio were assessed by ¹H NMR. ^b 1.0 equiv. of trifluoroethanol. ^c Using 0.2 equiv. of KO^tBu and EtOH [0.2] at 100 °C. ^dObtained as a racemic mixture.

	Δr	Ar He N Ph -		Ligand (6.0 mol%) [Ni(cod) ₂] (5.0 mol%) base (2.0 equiv.) Solvent [0.5], 50 °C, 16 h					
	7.4						I II Me O		Me
	A	1a	2a				4aa	5	
	Ar = 4	Рп-С ₆ Н ₄	(1.2 equiv.)				3,4-addition	1,4-add	dition
		Entry	Ligand	base	solvent	1a Conv. (%)	4aa (%)	5aa (%)	<i>E/Z</i> (5aa)
Me	−Me	6	dppe	KOMe	THF	>99	0	>95	9:1
		7	dppf	KOMe	THF	>99	<1	>95	8:1
PCy ₂ N	Ph	8	Rac-BINAP	KOMe	THF	>99	1	>95	9:1
L5		9 ^d	(S) -L4	KOMe	THF	>99	0	>95	8:1
$\sum_{i=1}^{n}$		10 ^d	(S) -L5	KOMe	THF	35	0	>95	8:1
PPh ₂ N	PPh ₂ N	11 <i>^d</i>	(S) -L13	KOMe	THF	35	0	81	9:1
L13		12 ^d	(<i>S,Sp</i>) -L14	KOMe	THF	35	0	>95	9:1

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^a Reaction conditions: all reactions were performed with **1a** (0.10 mmol, 1.0 equiv.), **2a** (0.2 mmol, 2.0 equiv.), Ni(cod)₂ (0.005 mmol, 5 mol%), Ligand (0.006 mmol, 6 mol%), base (0.20 mmol, n equiv) in THF (2.0 mL, 0.5 M) at 50 °C. Consumption of **1a**, conversions and E/Z ratio were assessed by ¹H NMR. ^b1.0 equiv. of trifluoroethanol. ^cUsing 0.2 equiv. of KO*t*Bu and EtOH [0.2] at 100 °C.^d Obtained as a racemic mixture.

Ar Ar Ar = 4-F	a Ph-C ₆ H ₄ (1	O Ph Ae 2a .2 equiv.)	[Ni(cod) L (6 KOM	9 ₂] (5.0 mol%) 6.0 mol%) e (x equiv.) → Ar HF [0.5], T °C	H Me 3,4-addition	O Ph Ar 5 1,4-addition	$ \begin{array}{c} $
Entry	T (°C)	t (h)	х	1a Conv. (%)	4aa (%)	5aa (%)	<i>E/Z</i> (5aa)
1	50	16	2.0	>99	0	95	9:1
2	23	16	1.0	69	0	66	13:1
3	23	16	0.2	78	0	75	15:1
4	23	48	0.2	96	0	95	15:1
5 ^c	23	16	0.2	2	0	0	-

^a Reaction conditions: all reactions were performed with **1a** (0.10 mmol, 1.0 equiv.), **2a** (0.12 mmol, 1.2 equiv.), Ni(cod)₂ (0.005 mmol, 5 mol%), **L1** (0.006 mmol, 6 mol%), KOMe (0.10 x mmol, x equiv) in THF (0.2 mL, 0.5 M) at indicated temperature. Consumption of **1a**, conversions and E/Z ratio were assessed by ¹H NMR. ^b Using 2.0 equiv. of **2a**. ^c Using BTMG as base.

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Substrate Scope



Substrate Scope





Ar Ar = 4-	1a Ph-C ₆ H ₄	+	OOO N 3a (m equiv)	L9 (6 [Ni(cod) BTMC THF [5.0 mol%) ₂] (5.0 mol%) 6 (n equiv.) 0.5], 23 ⁰C Ar	Ph Me 7aa 1,4-addition	h NR ₂ 6aa 3,4-add Me Ar Ph 8aa 4,1-addition	ition D NR ₂ PCy (4	² N Ph Ph <i>R</i> ,5 <i>S</i>)- L9
Entry	t (h)	m	n	1a Conv. (%)	6aa (%)	7aa (%)	8aa (%)	dr (6aa)	ee(6aa) ^b (%)
1	16	2.0	1.0	99	78	7	13	5.3:1	92;80
2	24	1.2	1.0	99	81	6	13	7.0:1	92;79
3	48	1.2	0.2	99	78	6	13	7.0:1	92;79
4 ^c	48	2.0	0.2	99	75	8	2	5.3:1	91;80

^a Reaction conditions: all reactions were performed with **1a** (0.10 mmol, 1.0 equiv.), **3a** (0.1 m mmol, m equiv.), Ni(cod)₂ (0.005 mmol, 5 mol%), **L9** (0.006 mmol, 6 mol%), BTMG (0.10 n mmol, n equiv) in THF (2.0 mL, 0.5 M) at room temperature. Consumption of **1a**, conversions of **6aa**, **7aa**, **8aa** and diastereoisomeric ratio were assessed by ¹H NMR. Enantiomeric excess determined by HPLC equipped with chiral columns. ^b Representing the ee of major diastereoisomer and minor diastereoisomer.^c Using KOMe as base.

Substrate Scope



Substrate Scope



Product Transformation





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The First Paragraph

Writing Strategy



The development of novel methods for the diastereoselective and enantioselective construction of adjacent stereocenters based on C-C bond-forming reaction is a resounding challenge in chemical synthesis. The atom-economical intermolecular hydroalkylation of conjugated dienes has recently emerged as an enabling strategy because it provides access synthetically useful compounds with an allylic stereocenter, compounds that would be difficult to prepare using conventional protocols. The influence of substitution pattern of dienes on reactivity and selectivity and the diversity of insertion modes conceivable for a transition-metal catalyst across the conjugated double bonds pose major difficulties for their selective functionalization.

Consequently, in its most demanding version, the successful development of a metal-catalyzed hydroalkylation of dienes requires addressing altogether chemoselectivity, regioselectivity, diastereoselectivity, and-ultimately-enantioselectivity challenges.



Writing Strategy



In addition to stereoselectivity, regioselectivity is often a major challenge in diene hydrofunctionalization reactions. In this article, we have reported complementary regiodivergent Ni-catalyzed hydroalkylations of two branched dienes with unstabilized C(sp³) nucleophiles. The first system uses an achiral C1-symmetric phosphinooxazoline ligand and simple amides, which once deprotonated in situ, undergo a highly 1,4-selective addition process with excellent stereocontrol of the trisubstituted C=C bond generated. The method displays a broad scope in both the nucleophilic and electrophilic component, enabling the use of sensitive functionalities and heteroaromatic-containing precursors.

Switching to imides as carbon nucleophiles favored formation of 3,4addition products. The constructions of vicinal tertiary stereocenters was achieved with moderate to high diastereoselectivity and excellent enantioselectivity by means of a novel chiral (P,N) ligand. Notably, a wide range of functional groups were compatible with the mild conditions employed and several postcatalytic derivatizations were conducted to measure the synthetic potential of the method. Studies aiming at understanding the factors that determine reactivity and selectivity for both systems are currently ongoing in our laboratories.

Reasoing (考虑到.....) that the use of imides in place of amides in the Ni-catalyzed 3,4-hydroalkylation diene may positively influence the stereoselective outcome of the reaction, we directly set out to (着手于) develop an enantioselective variant of this process.

To date (到目前为止), there is only a handful of (少量) reports on the selective intermolecular hydroalkylation of 1,3-dienes..

Although 5aa was not our initial target, the presence of two stereogenic elements and two functional handles in its structure prompted (促使、激励) us to further explore conditions favoring its formation.

Thanks for your attention