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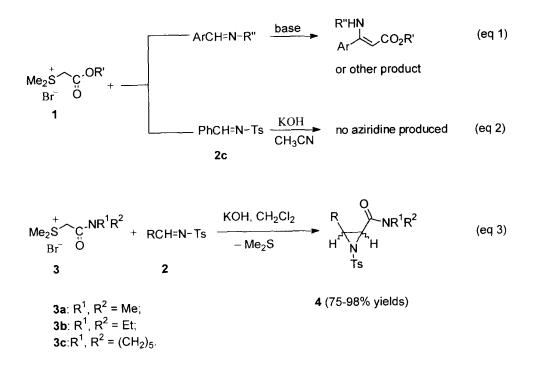
The Aziridination of *N*-Tosylimines with Amide-Stabilized Sulfonium Ylides: A Simple and Efficient Preparation of Aziridinyl Carboxamides

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Abstract:: Reaction of N-sulfonylimines 2 with N.N-dialkylcarbamoylmethyl dimethylsulfonium bromides 3 in the presence of solid KOH gives the aziridinyl carboxamides 4 in good to excellent yields. © 1997 Elsevier Science Ltd.

With the fair demonstration of the versatility of aziridine carboxylic acid derivatives in the synthesis of amino acids.¹ a variety of synthetic methods for this type of specially important synthetic intermediates have been developed.^{1.2} Among them, the direct aziridination of C=C or C=N bonds, includes carbene³ or nitrene⁴ transfer, aza-Darzens reaction,⁵ Lewis acid-catalyzed route,⁶ etc. Previously, we have succeeded in the preparation of vinylaziridines by the reaction of *N*-tosylimine with allylic sulfonium ylides⁷ (semistabilized ylides). However, this reaction failed with a carboxylate-stabilized ylide in preparation of aziridine carboxylate. When an ylide derived from sulfonium salt 1 bearing a carboxylic ester group was reacted with common *N*-alkyl or -arylimines, only enamines (eq 1),^{8a,b} cyclopropanes,^{8b} or a more complex product^{8c} were obtained. Even with the more reactive *N*-tosylimine **2c**, we still failed in obtaining the desired aziridine (eq 2). Fortunately, when we tried to use a carboxamide-bearing ylide derived from sulfonium salt **3** replacing ester sulfonium salt **1** to perform the same reaction, the aziridines **4** were obtained efficiently under very mild conditions (eq 3). Herein, we would like to communicate these results.



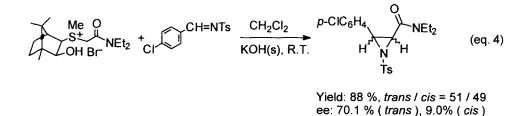
These reactions were carried out simply by mixing three solids of *N*-tosylimines 2^9 (1.0 equiv.), sulfonium salt 3^{10} (1.2 equiv.), and powdered KOH (1.2 equiv.) in CH₂Cl₂ and by stirring at room temperature for about 1 hour. After the reaction was complete (by TLC), column chromatography gave the pure product. The investigation of the effect of solvent and base on the yields and the *trans/cis* selectivity illustrated that under solid-liquid phase transfer conditions, the variation of these factors could only affect the yield and showed almost no effect on the selectivity. The CH₂Cl₂ / KOH system is the most suitable condition for this reaction. Some results are summarized in Table 1. It is easy to find that the present reaction works well with various *N*tosylimines.¹¹ derived from aromatic, heteroaromatic, as well as aliphatic aldehydes. Excellent yields have been achieved in most examples. It should be mentioned that the same ylide (derived from **3**) with the relatively unreactive *N*-alkyl or -arylimines did not produce the desired aziridinyl amide. The aziridinyl carboxamide can also not be prepared by the aza-Darzens reaction using *N*-tosylimines with *N*,*N*-dimethyl bromoacetic amide under the above-mentioned ylide reaction conditions.

We also used camphor-derived chiral sulfonium salt to proceed this reaction (eq. 4) instead of **3**. The ratio of *trans* / *cis* is 51 / 49, the ee value of the *trans* product is 70.1% by chiral HPLC (determined by chiral HPLC Chiralcel OD), of *cis* is 9.0%.¹²

Entry	Sulfonium salt	R	Yield, ^b %	trans / cis °
1	3a	$p-MeC_6H_4(2a)$	91% (4aa)	75 / 25
2	3a	<i>p</i> -ClC ₆ H ₄ (2b)	87% (4ab)	21 / 79
3	3c	p-ClC ₆ H ₄ (2b)	96% (4cb)	42 / 58
4	3b	<i>p</i> -ClC ₆ H ₄ (2b)	98% (4bb)	73 / 27
5	3a	C_6H_5 (2c)	88% (4ac)	72 / 28
6	3a	p-CF ₃ C ₆ H ₄ (2d)	81% (4ad)	47 / 53
7	3a	3-pyridyl (2e)	85% (4ae)	59 / 41
8	3a	p-O ₂ NC ₆ H ₄ (2f)	83% (4af)	33 / 67
9	3a	cyclohexyl (2g)	80% (4ag)	79 / 21
10	3a	tert-butyl (2h)	75% (4ah)	66 / 34

Table 1. Preparation of aziridine amides 4 by the reaction of N-sulfonylimines 2 and sulfonium salts 3 under solid-liquid phase transfer conditions.⁴

* All reactions were carried out under phase transfer conditions at room temperature in a ratio of imines/sulfonium salt/KOH(s) = 1/1.2/1.2 at a 0.5 mmol scale in dichloromethane. ^b Isolated yields based on the imine. ^c Determined by 300 MHz ¹H NMR analysis.



The compatibility of a wide range of substrates, very mild reaction condition and simple manipulations, excellent yields, and the potential use of the products may make the present reaction an efficient entry to a

type of very important functionalized aziridines.

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- 11. Selected characterisation data for compound 4aa: trans: ¹H NMR (CDCl₃) δ 2.29 (s, 3 H), 2.40 (s, 3 H), 3.04 (s, 3 H), 3.24 (s, 3 H), 3.58 (d, J = 4.3 Hz, 1 H), 4.41 (d, J = 4.2 Hz, 1 H), 7.10 (m, 4 H), 7.24 (d, J = 8.2 Hz, 2 H). 7.80 (d, J = 8.2 Hz, 2 H); MS m/z 359 (M⁺+1, 0.64), 203 (100), 158 (18.7), 91 (22.3), 72 (79); Anal. Cacld for C₁₉H₂₂N₂O₃S: C, 63.66; H, 6.19; N, 7.82. Found: C, 63.44; H, 6.18; N, 7.57. *cis*: δ 2.28 (s, 3 H), 2.41 (s, 3 H), 2.72 (s, 3 H), 2.91 (s, 3 H), 3.76 (d, J = 7.5 Hz, 1 H), 4.09 (d, J = 7.6 Hz, 1 H), 7.10 (m, 4 H), 7.33 (d, J = 8.0 Hz, 2 H). 7.95 (d, J = 8.2 Hz, 2 H).
- 12. The ratio of *trans / cis* was determined by ¹H-NMR and HPLC; the evalues of *trans* and *cis* compounds were determined by HPLC using chiral column (Chiralcel OD), we did not survey the specific rotation data and assign the absolute configuration because the products **4** cannot be separated by flash chromatography.

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