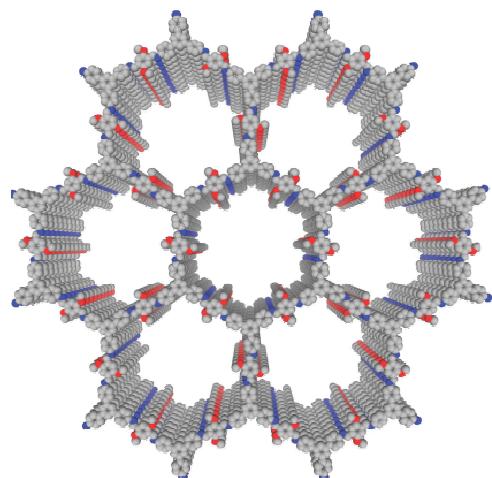




# 文献报告

## Covalent Organic Frameworks for *Heterogeneous* Catalysis



报告：时磊 检查：冯广收

手性合成研究组



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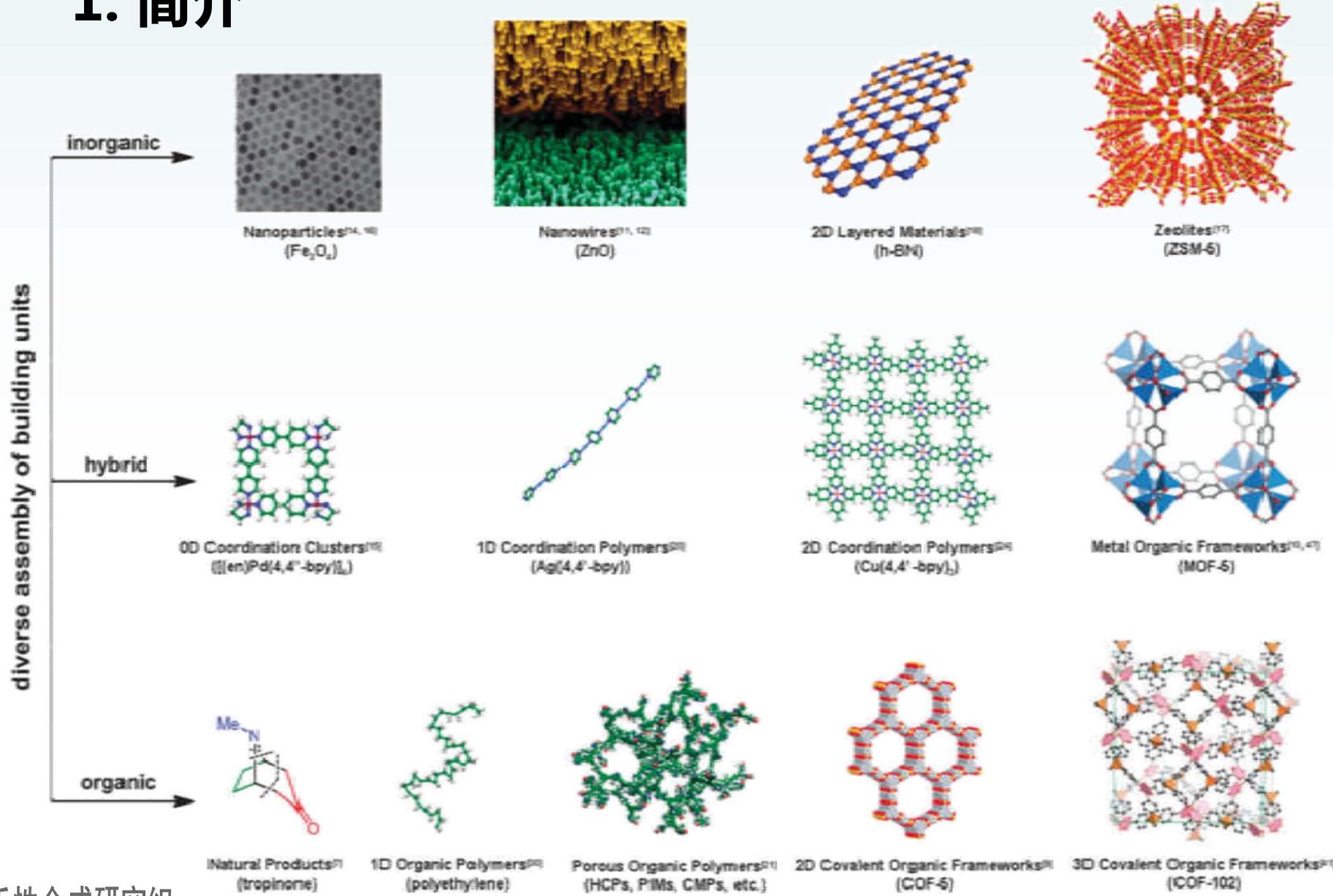
## COF材料的催化应用

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## 展 望



# 1. 简介





# ***(nano)porous organic polymer***

**Amorphous**

**HCPs**  
**Hypercrosslinked Polymers”**

**PIMs**  
**Polymers of Intrinsic Microporosity**

**CMPs**  
**Conjugated Microporous Polymers**

**Crystalline**

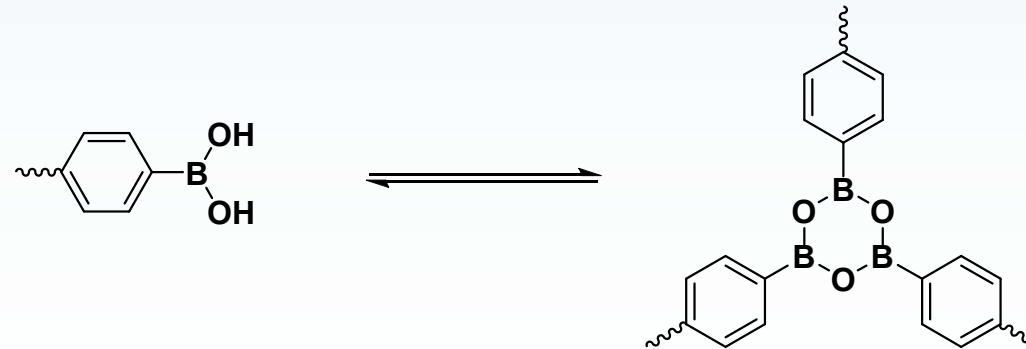
**COFs**  
**“Covalent Organic Frameworks”**



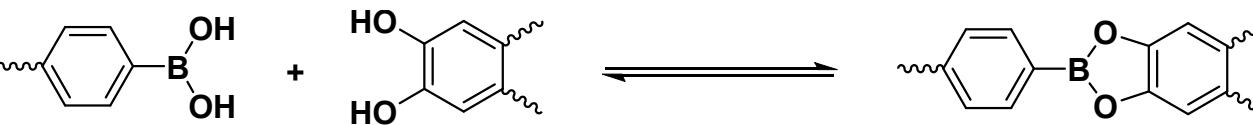
## Covalent Bond

✓ *reversibility*      ✓ *rigid planar*

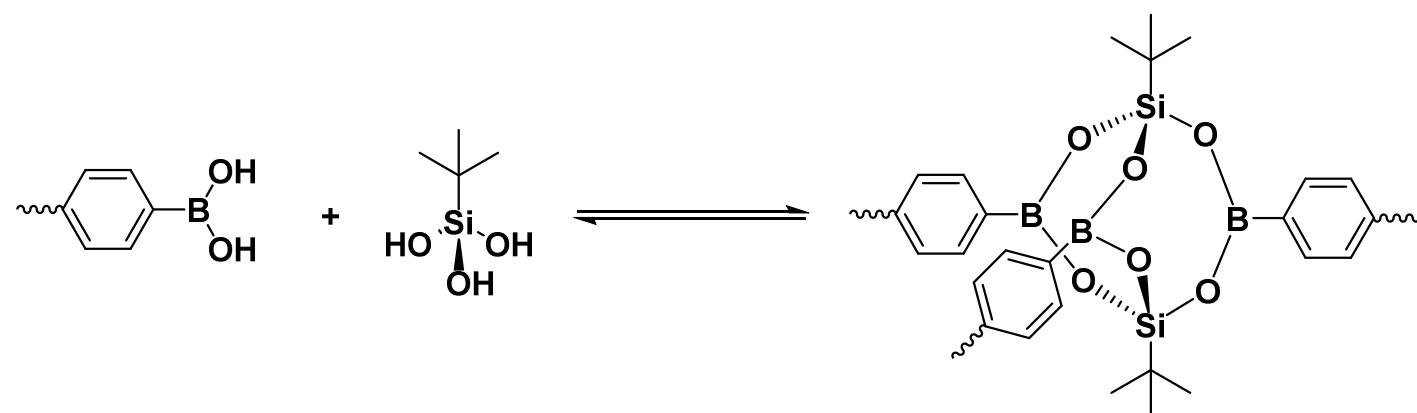
Boronate anhydride



Boronate Ester



Borosilicate



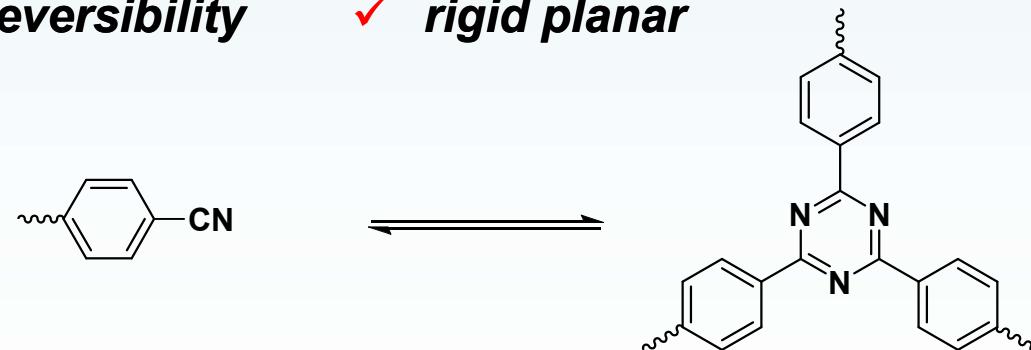


## Covalent Bond

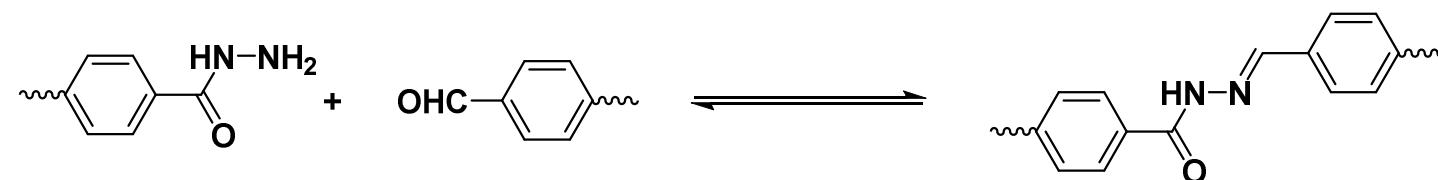
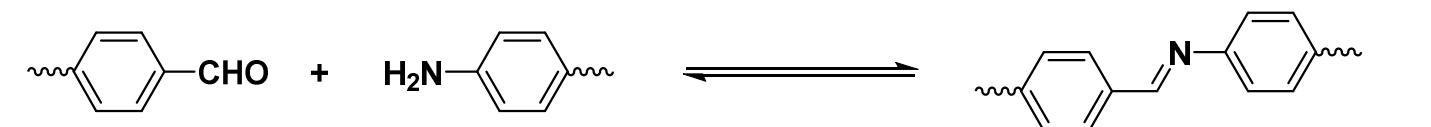
✓ *reversibility*

✓ *rigid planar*

### Nitrile Cyclotrimerization



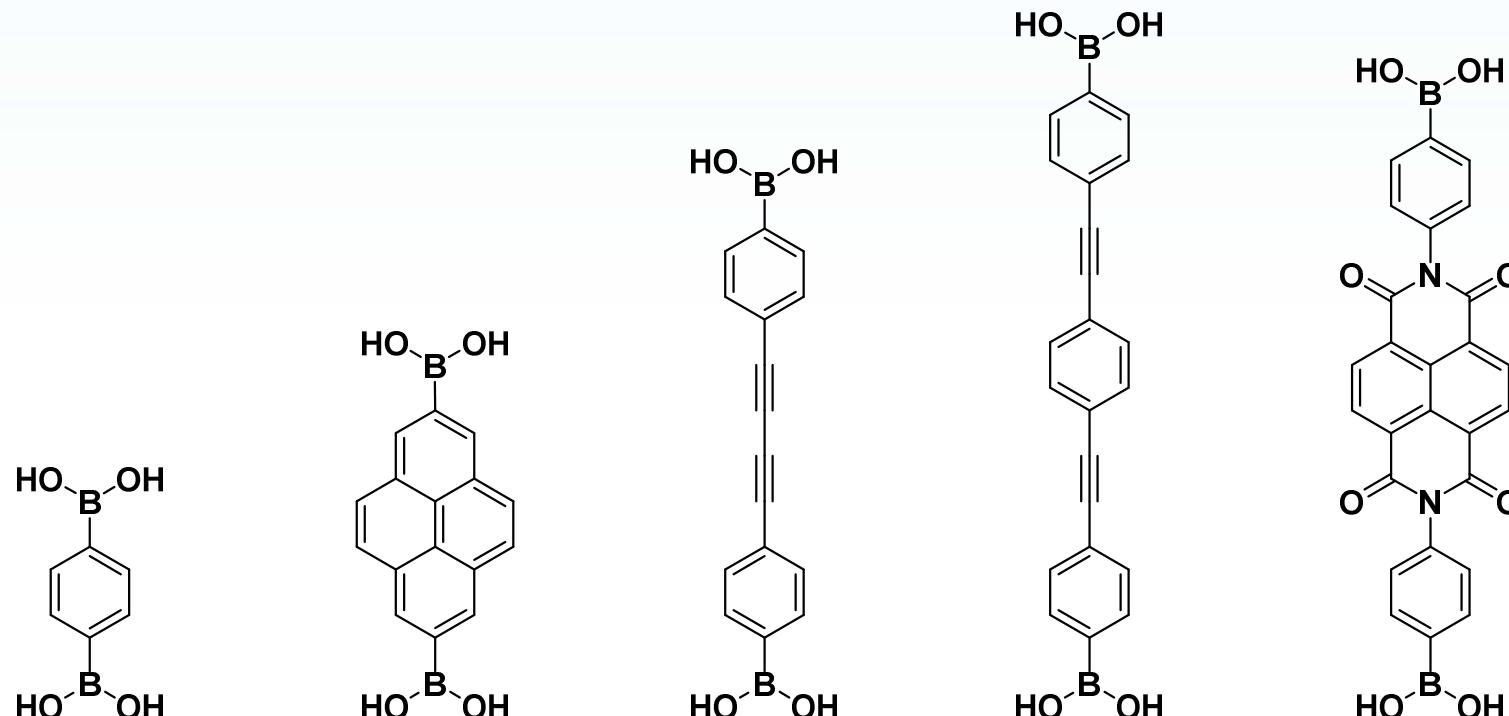
### Imine Formation





## Porous structure

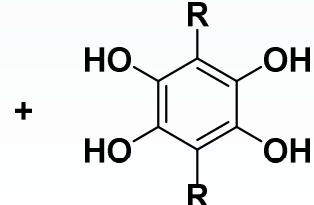
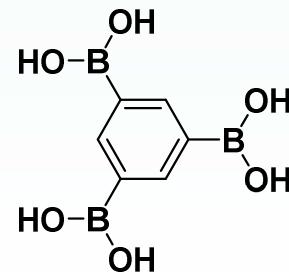
Length of units



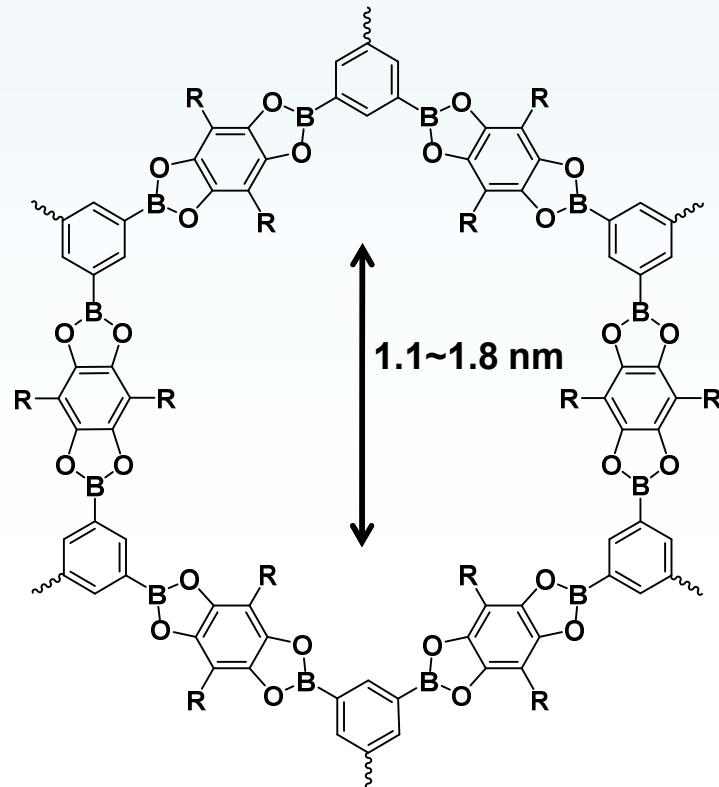
The pore size



## Length of substituent



+



- $R = H$   
 $R = CH_3$   
 $R = CH_2CH_3$   
 $R = CH_2CH_2CH_3$



# *Topology*

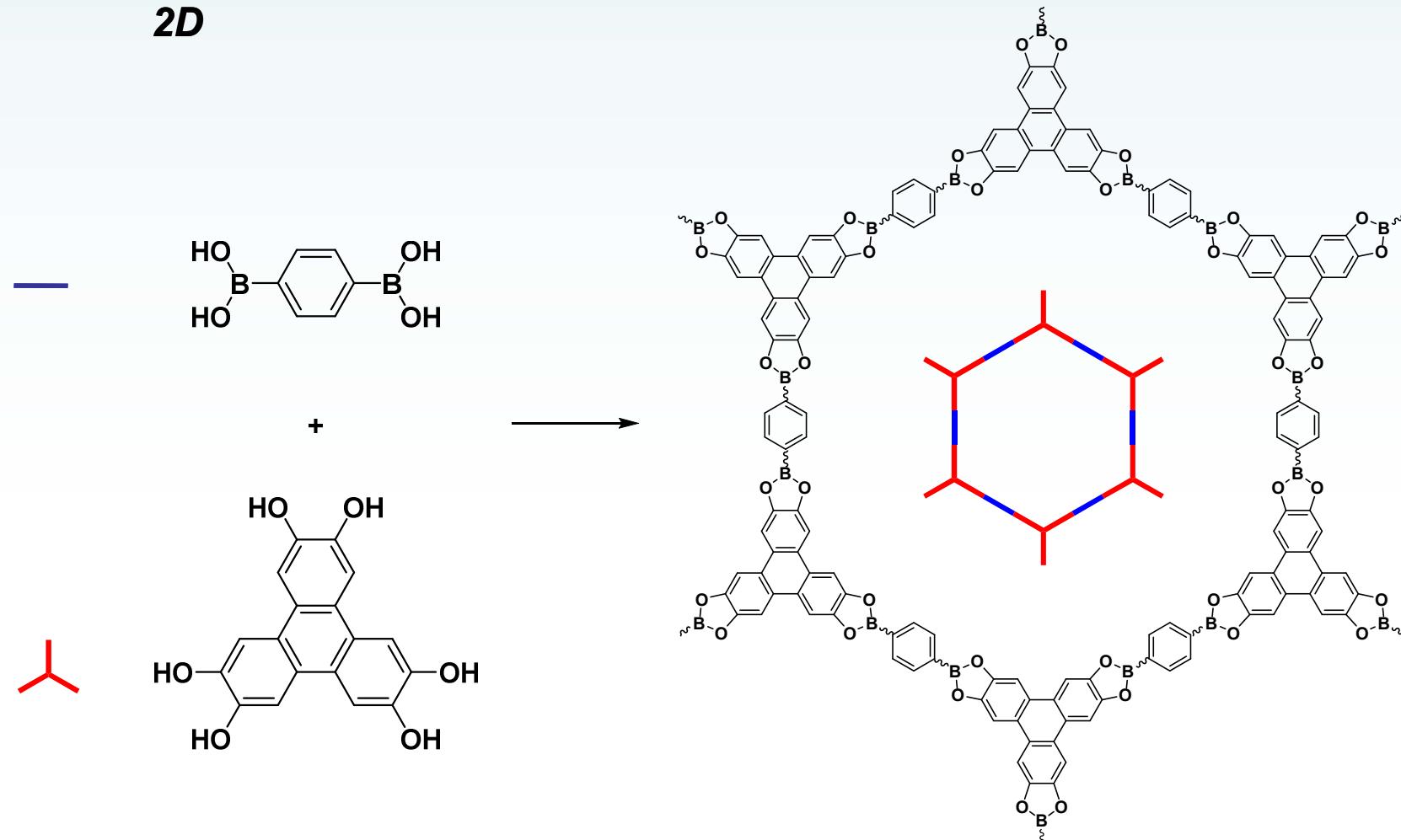
## Shape of units

*linear*



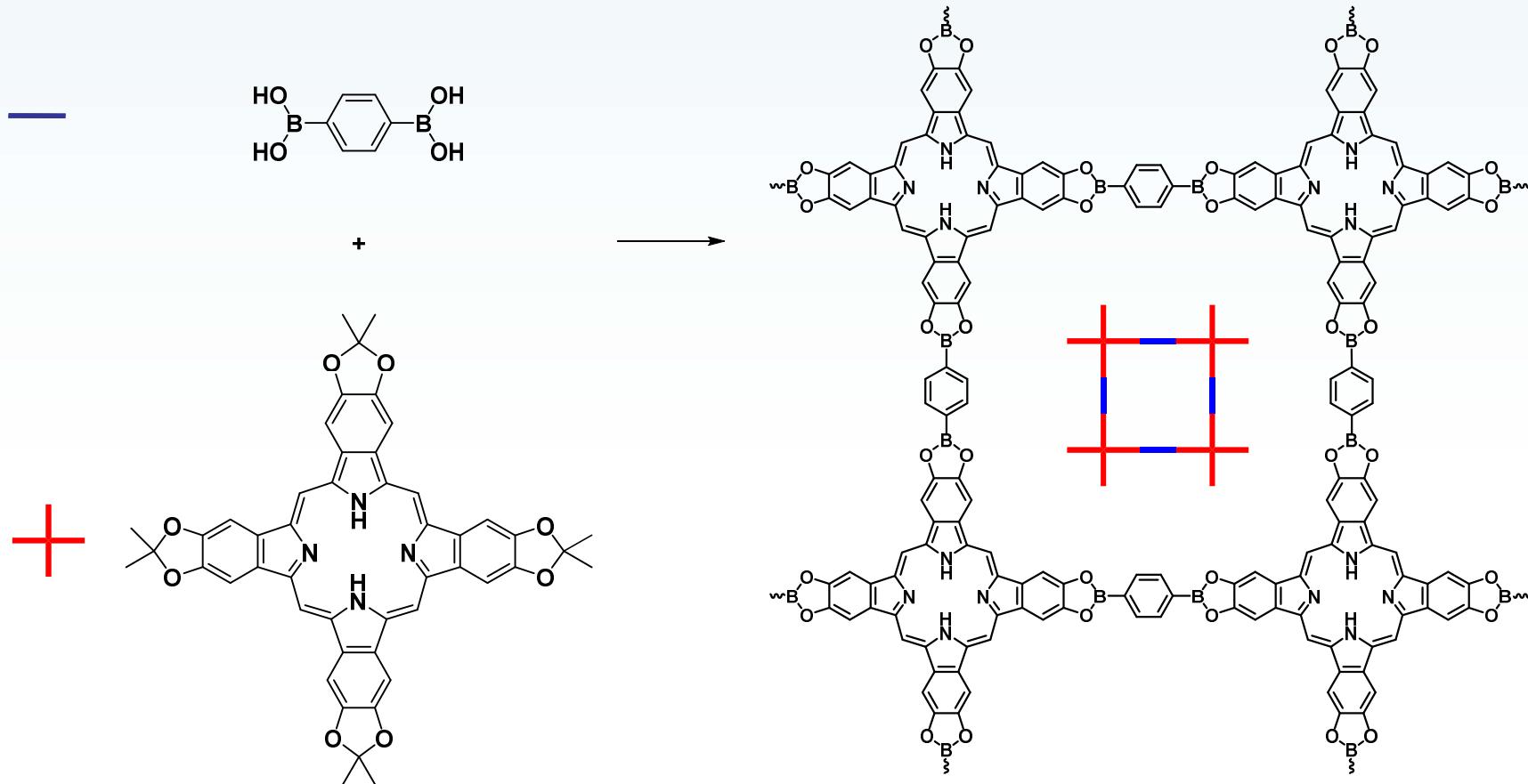


2D



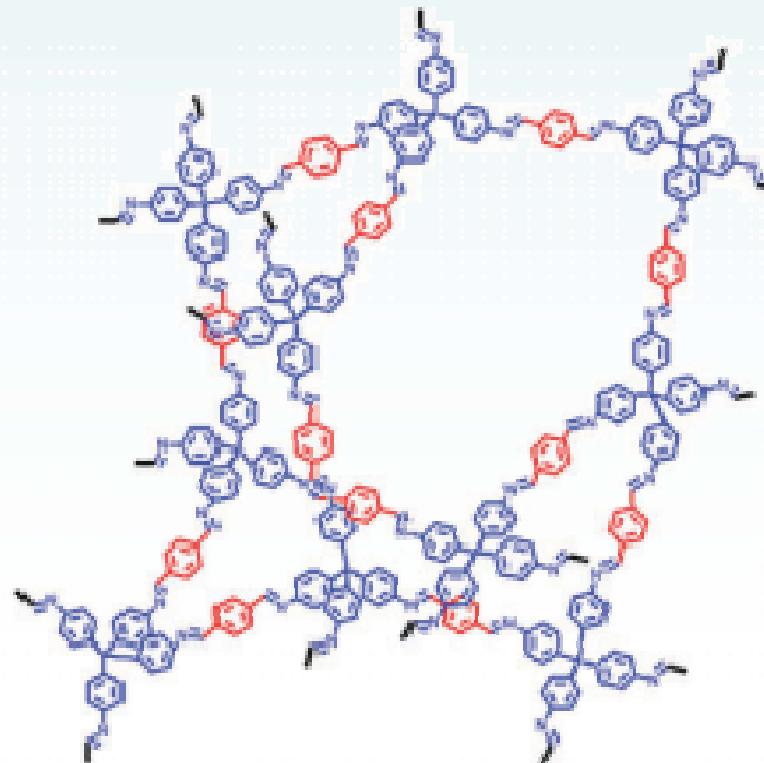
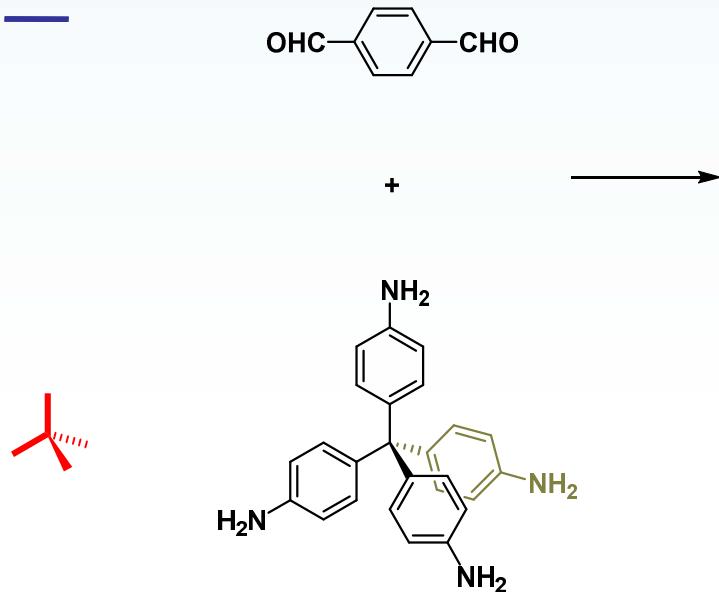


2D

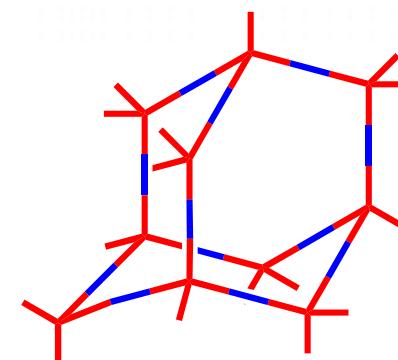




3D



COF-300



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## ***Synthesis of COFs***

- **Solvothermal synthesis**

*2~9 days 80~120 °C*

- **Ionothermal synthesis**

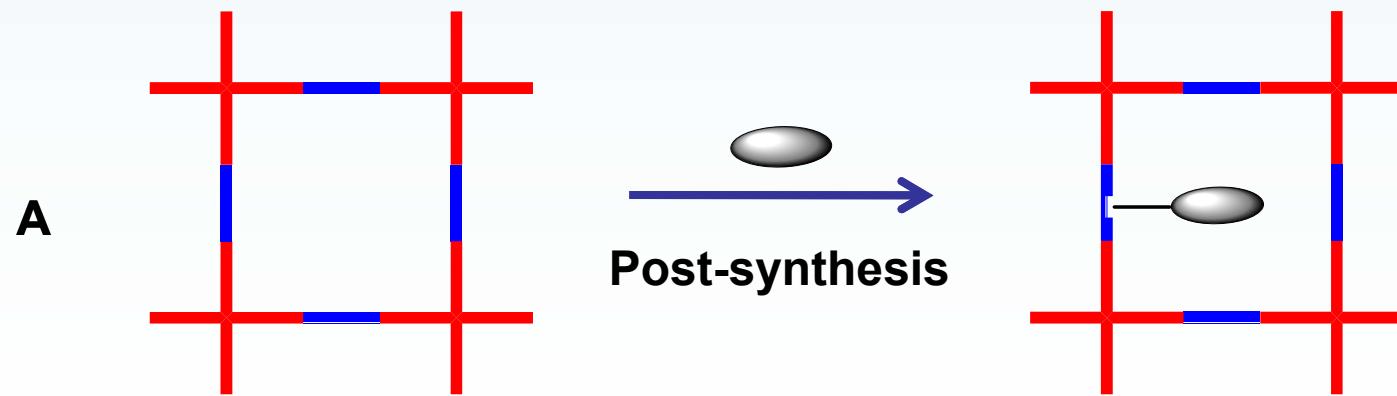
*Molten ZnCl<sub>2</sub> 400°C*

- **Microwave synthesis**

*200 times faster than solvothermal synthesis*



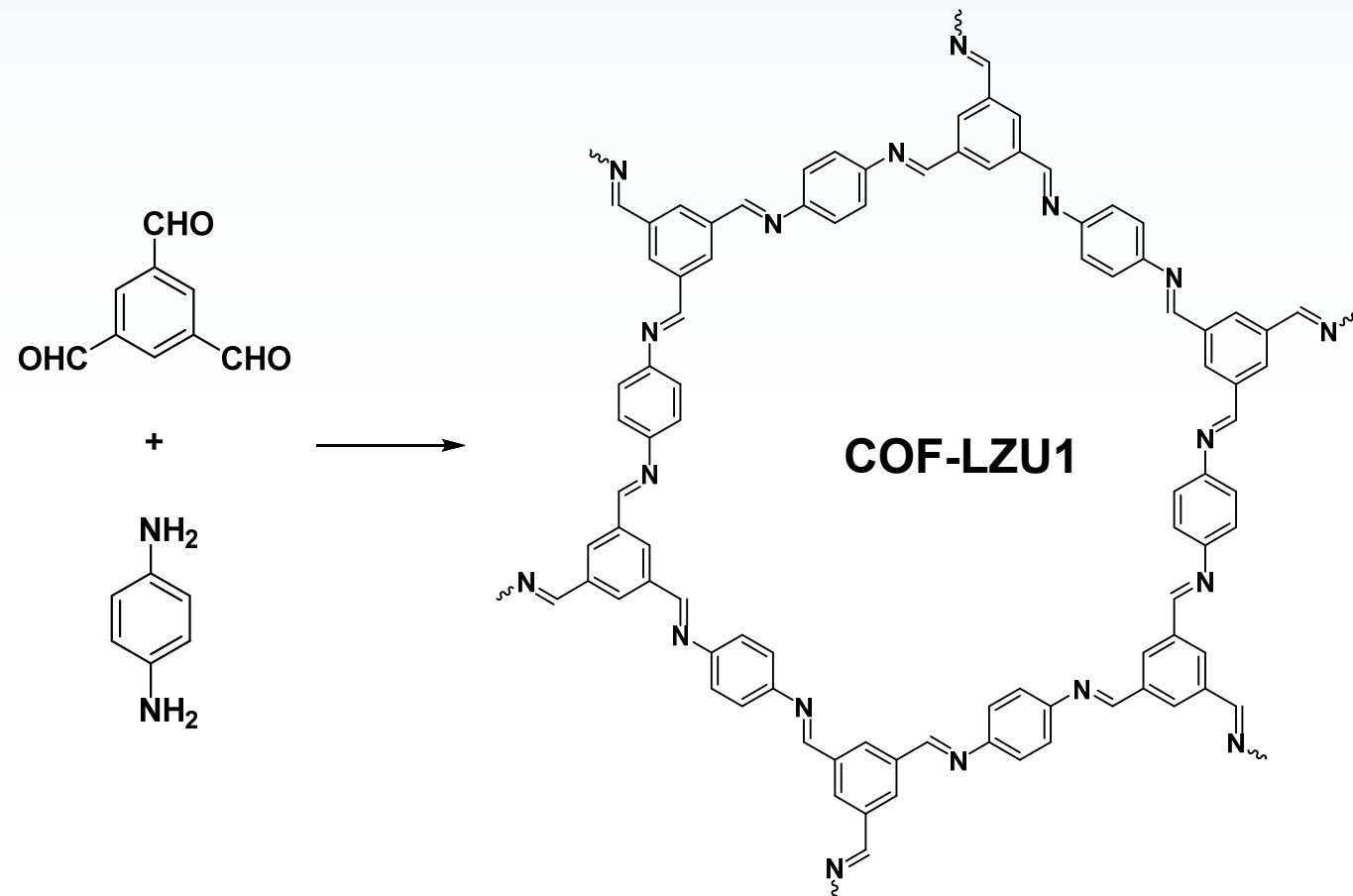
## *Functionality*



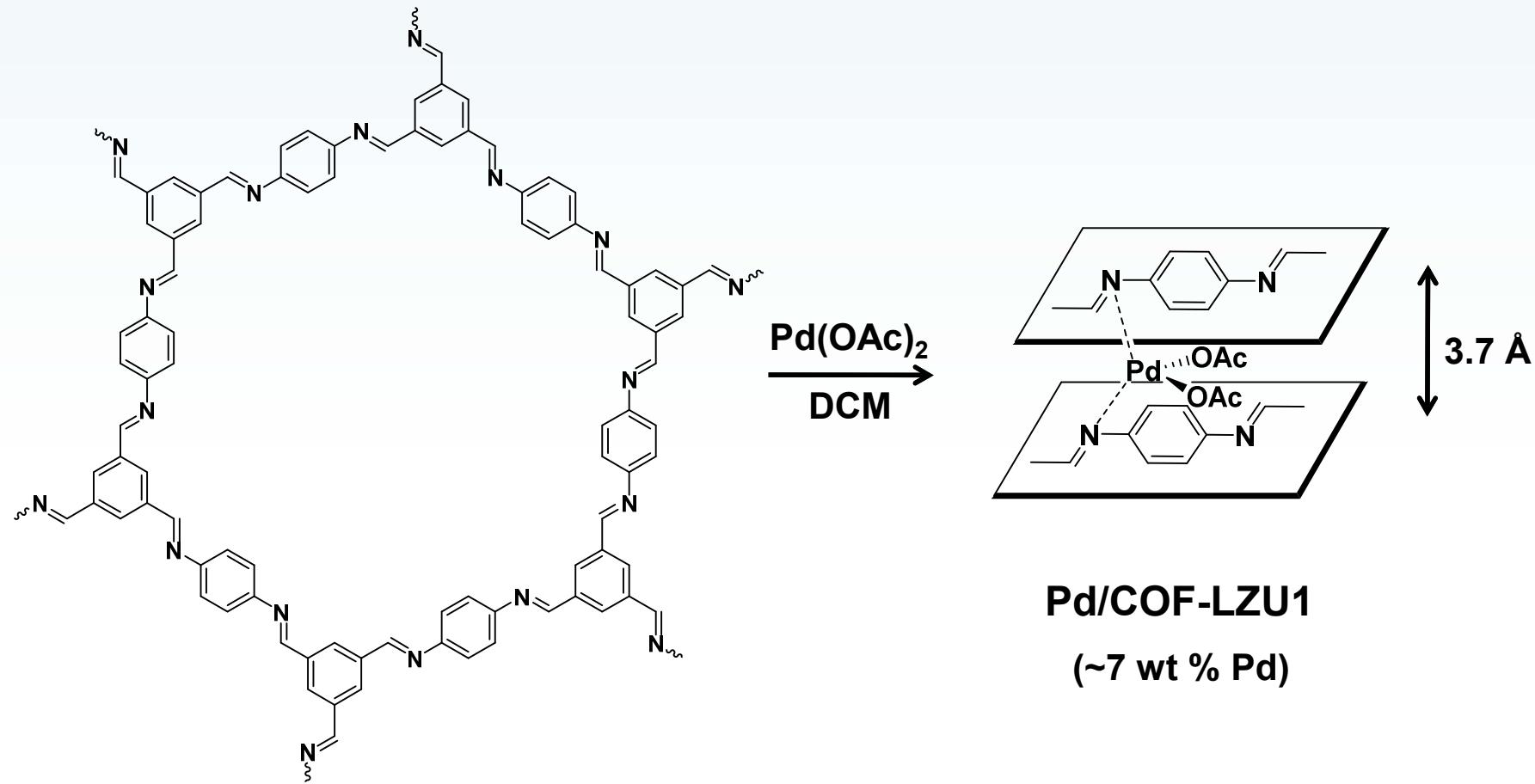


## 2. COF材料的催化应用

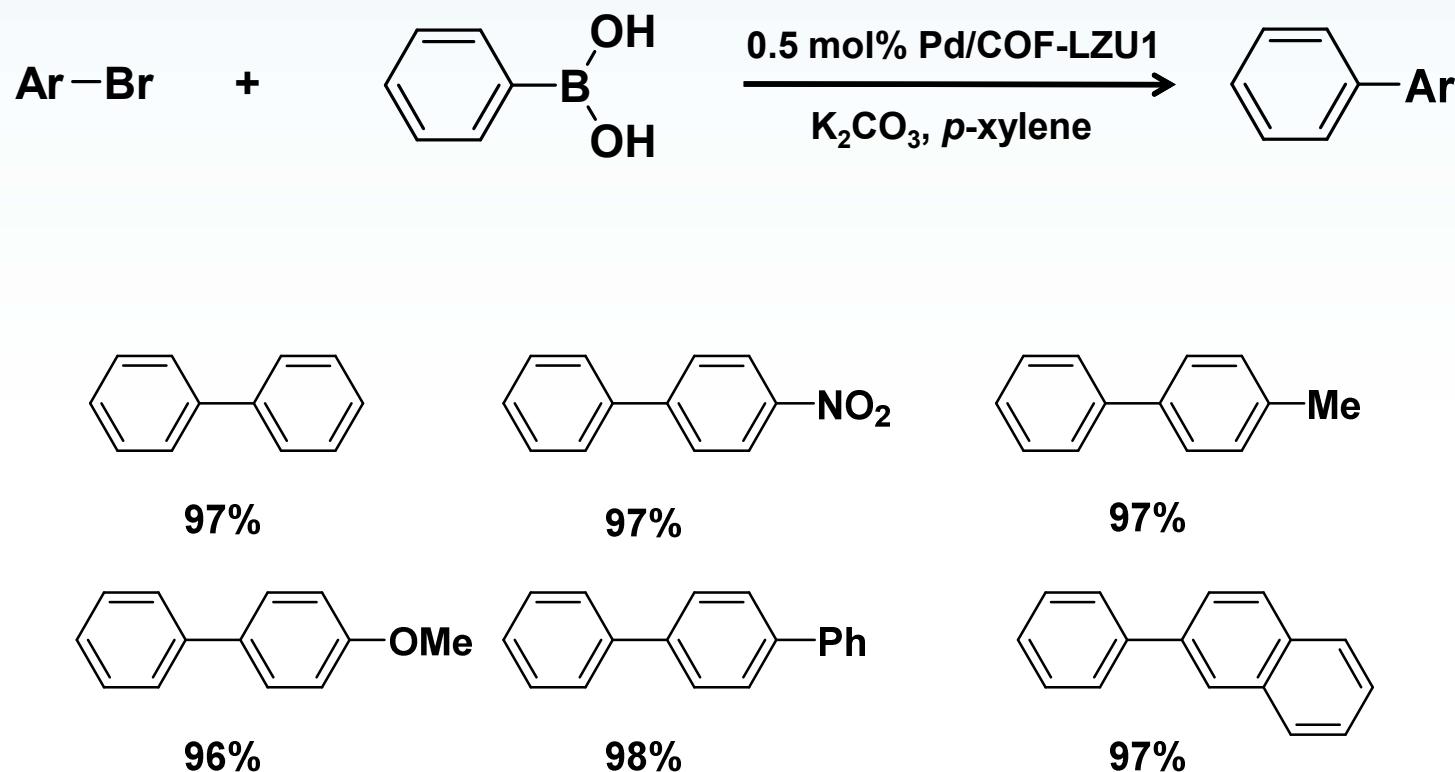
Suzuki coupling reaction



Wang, W. et al. *J. Am. Chem. Soc.* 2011, 133, 19816.



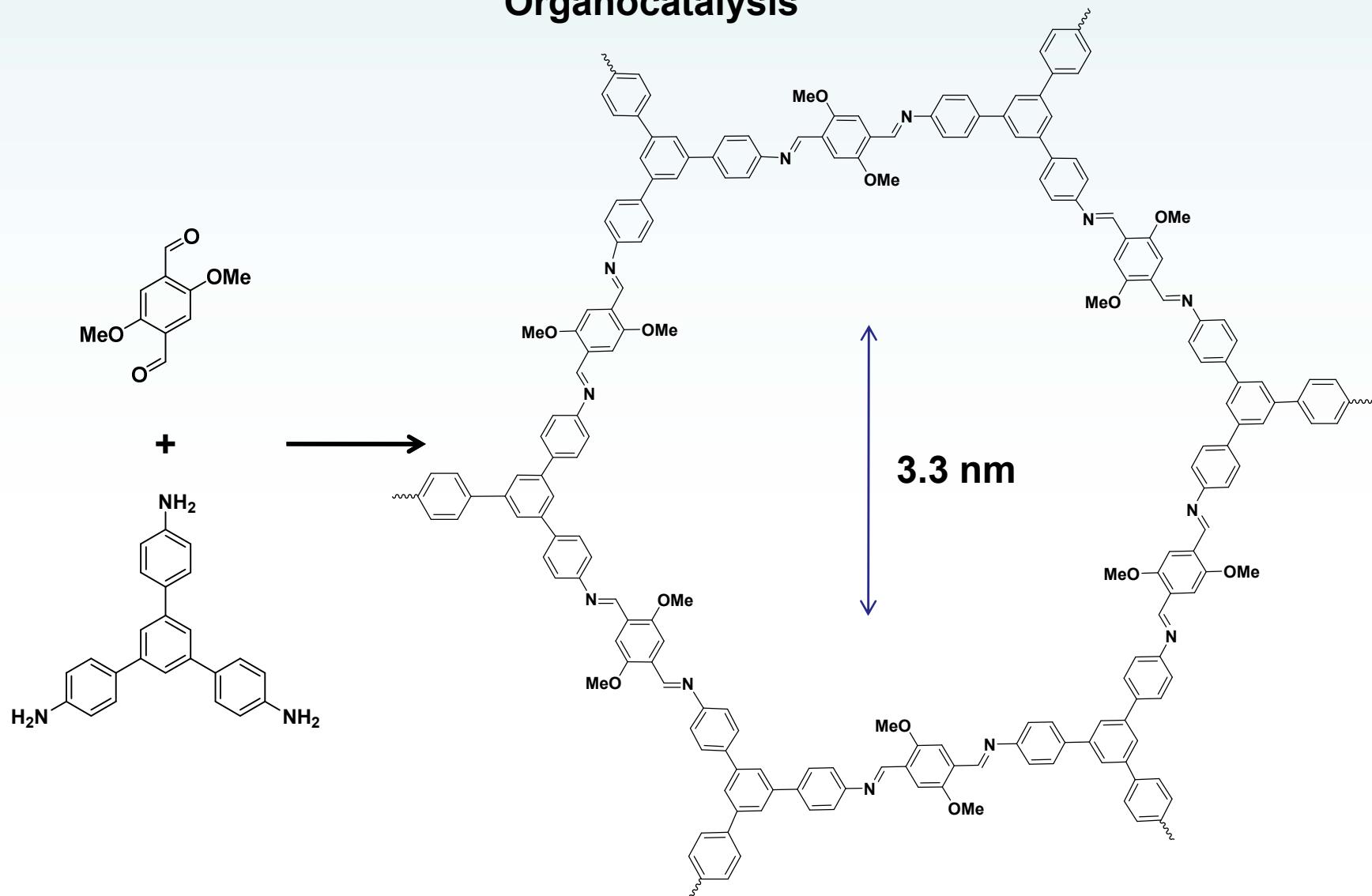
Wang, W. et al. *J. Am. Chem. Soc.* 2011, 133, 19816.



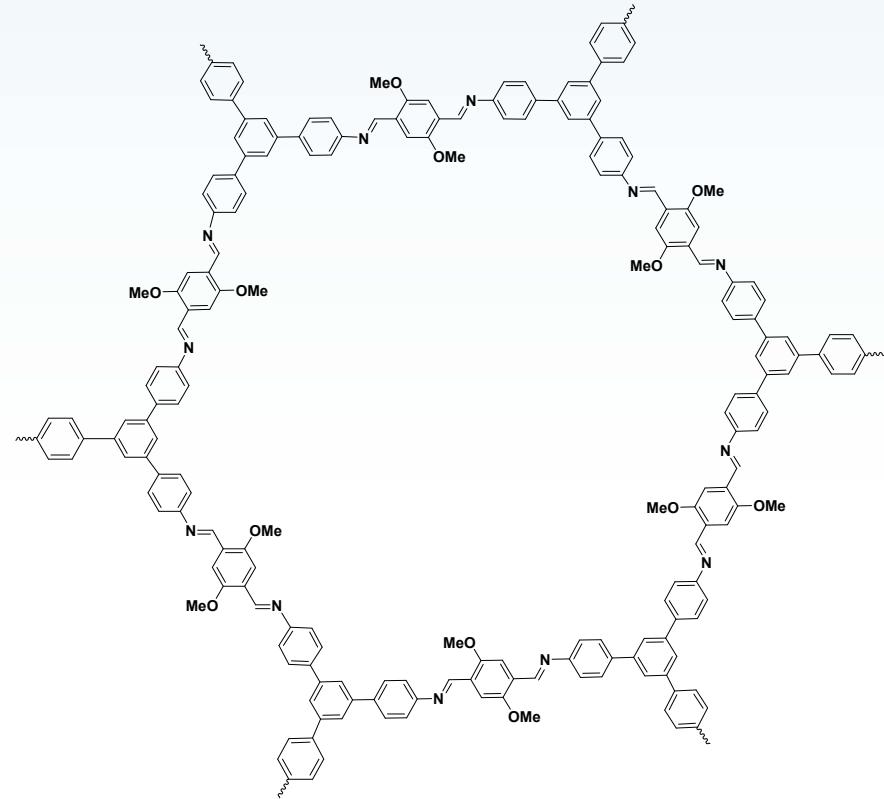
Wang, W. et al. *J. Am. Chem. Soc.* 2011, 133, 19816.



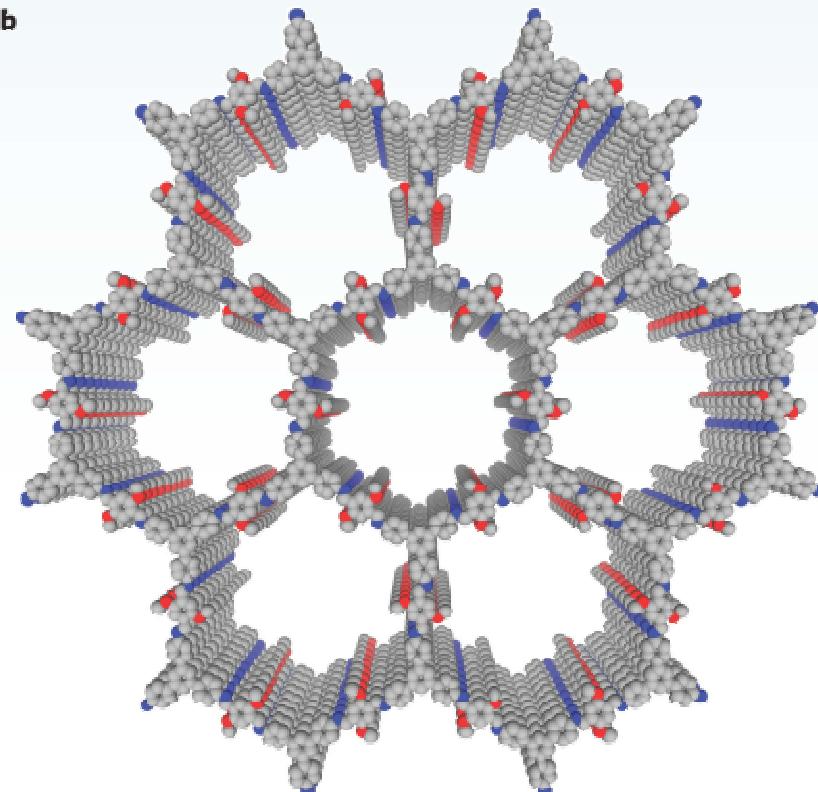
## Organocatalysis



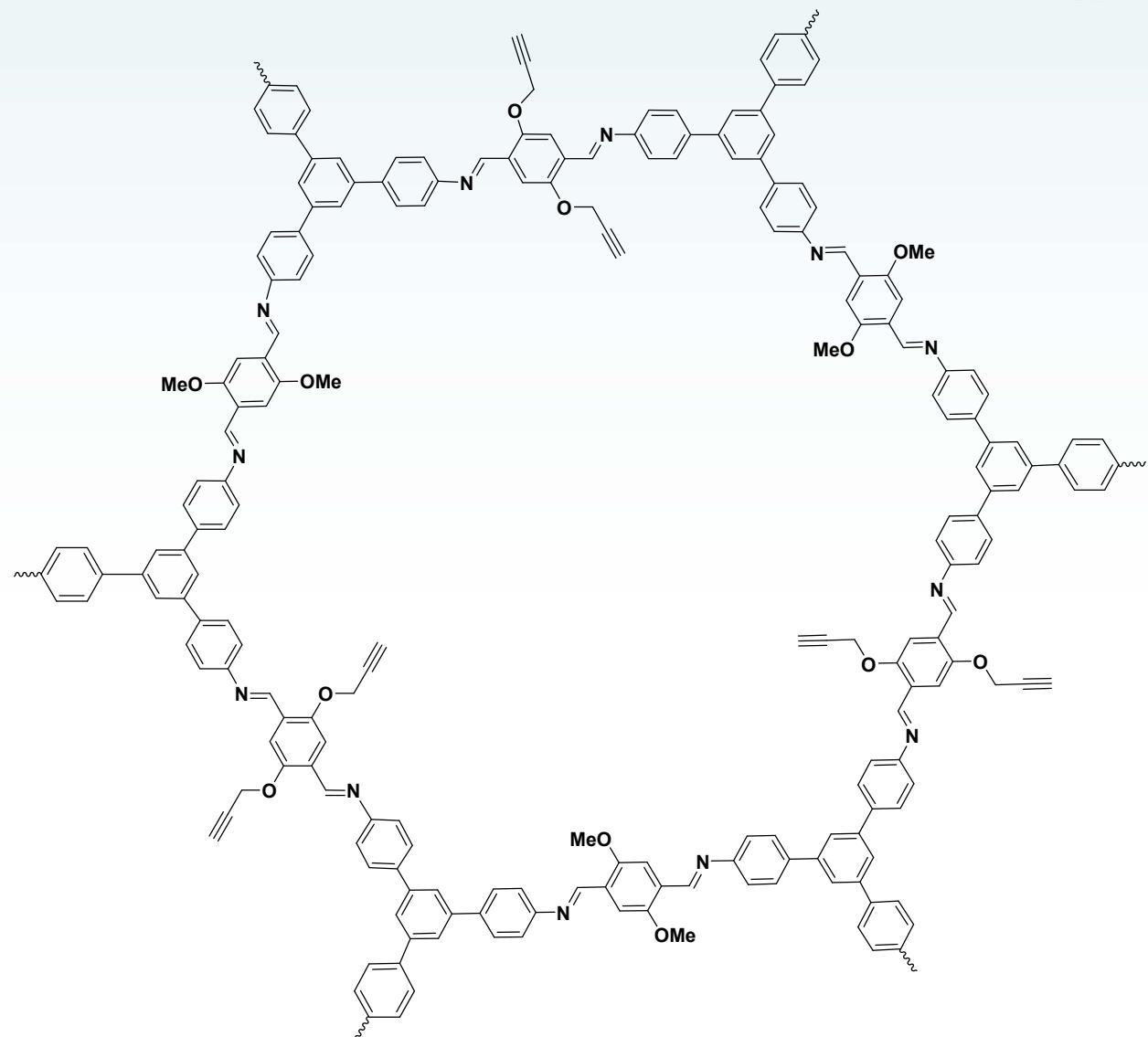
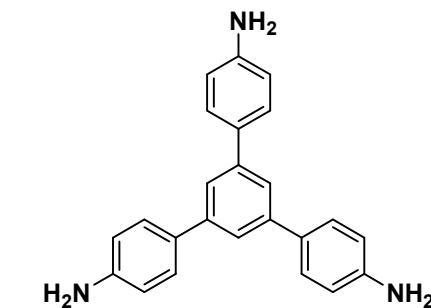
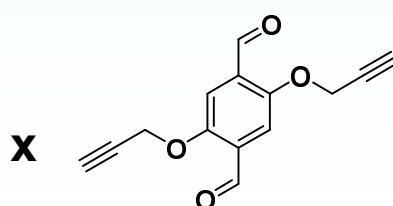
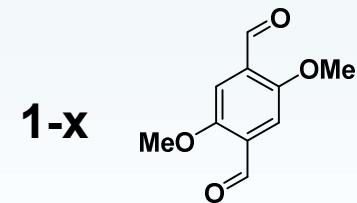
Jiang, D. et al. *Nature Chem.*, 2015, 7, 905.



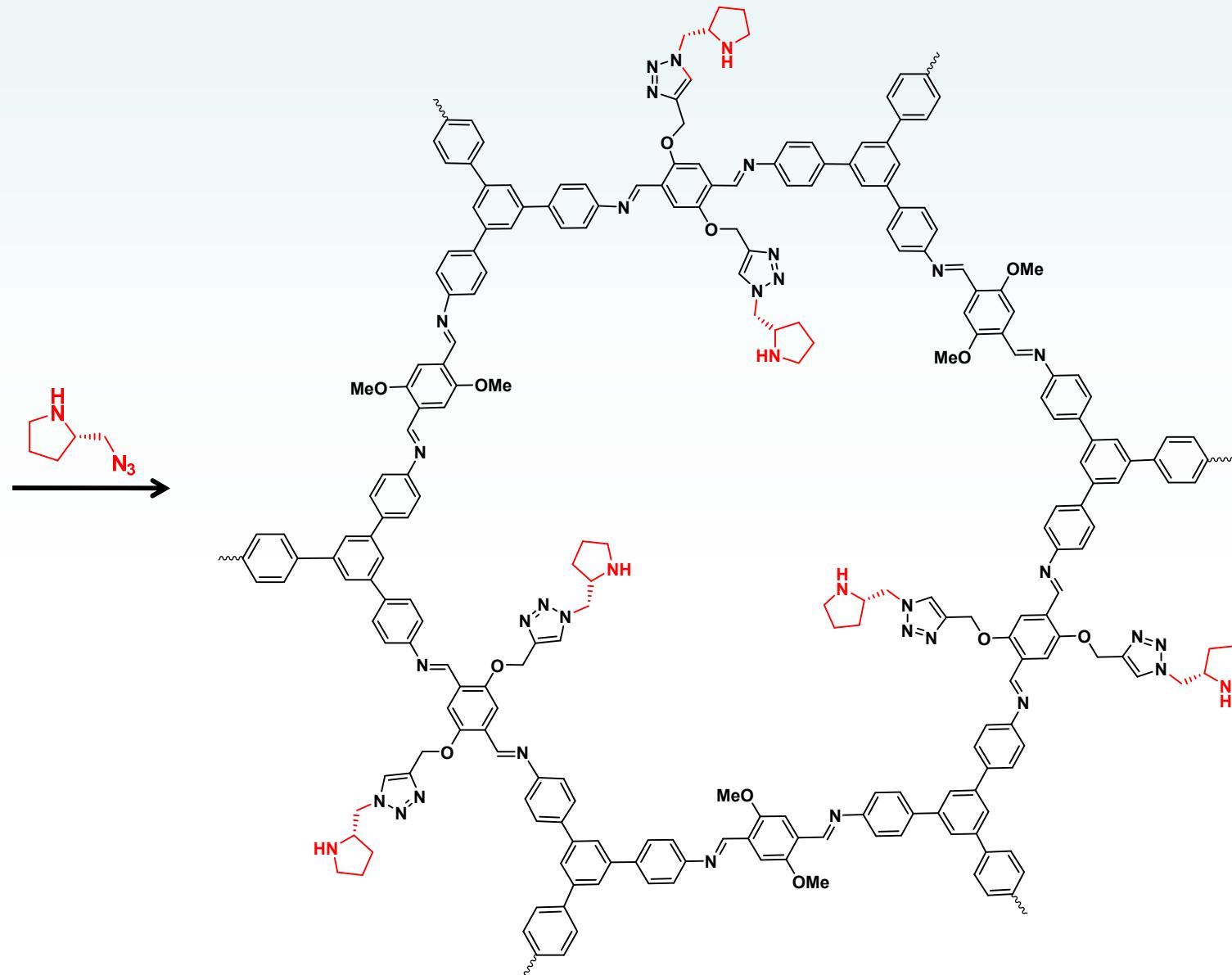
b



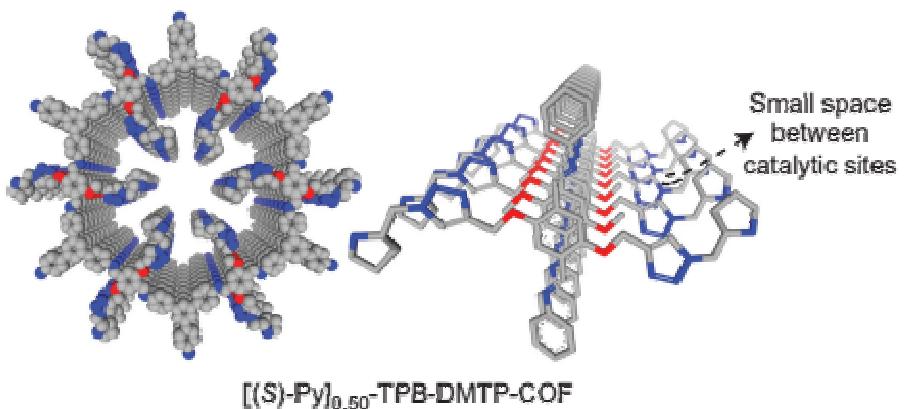
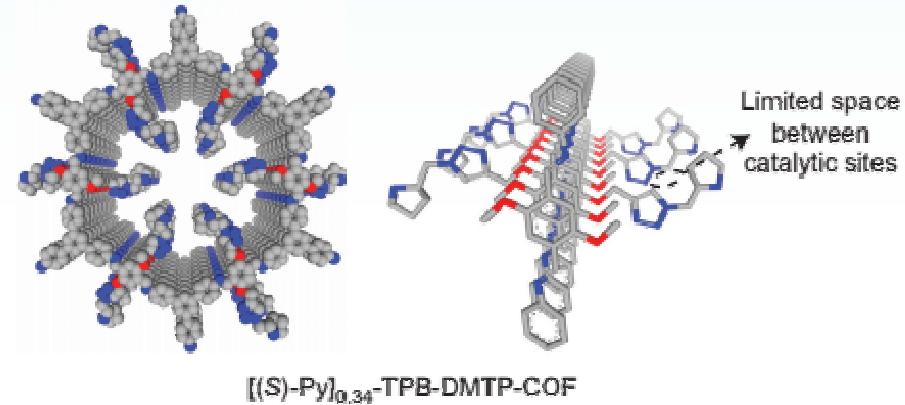
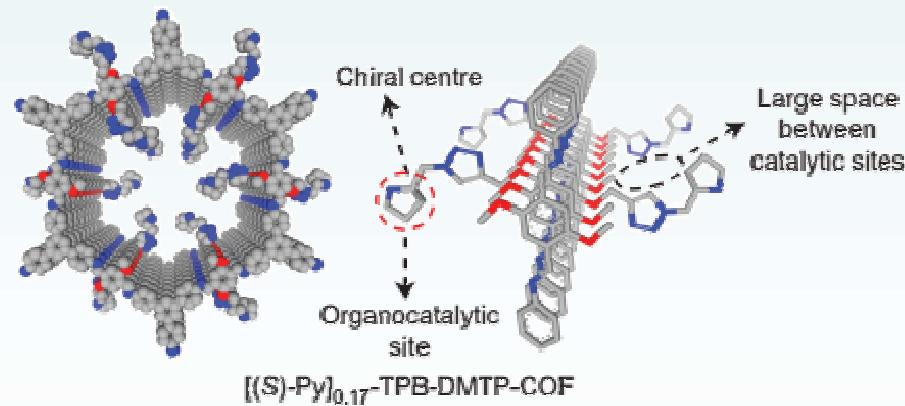
Jiang, D. et al. *Nature Chem.*, 2015, 7, 905.

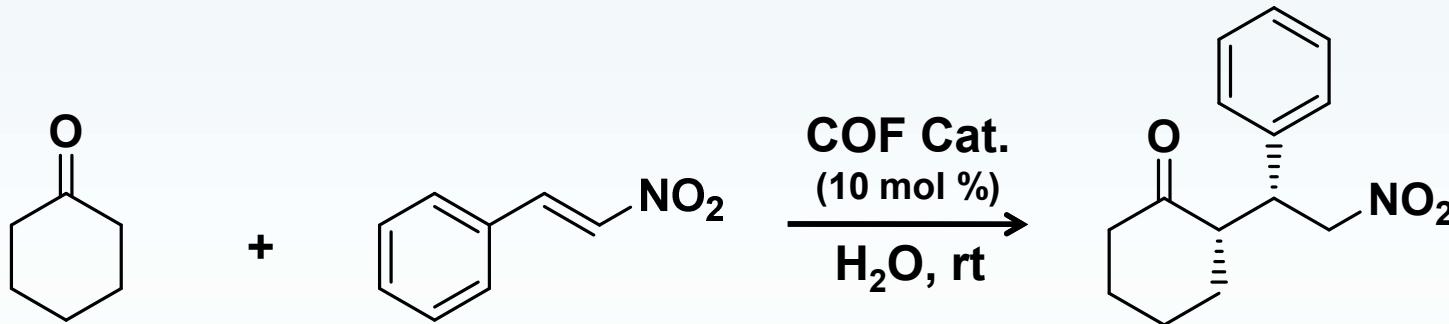


**X= 0.17, 0.34, 0.50**

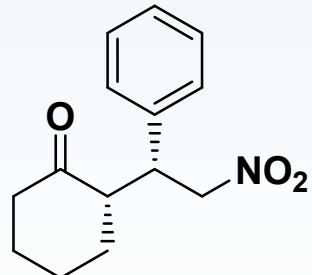


$[(S)\text{-Py}]_x\text{-TPB-DMTP-COFs } X = 0.17, 0.34, 0.50$

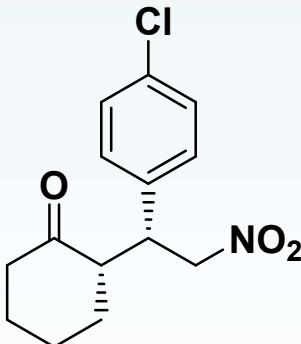




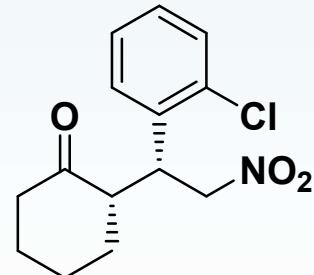
Entry	Catalyst	Time (h)	Conv. (%)	Yield (%)	Ee (%)	dr
1	$[(S)\text{-Py}]_{0.17}\text{-TPB-DMTP-COFs}$	12	100	95	92	90:10
2	$[(S)\text{-Py}]_{0.34}\text{-TPB-DMTP-COFs}$	17	100	93	91	90:10
3	$[(S)\text{-Py}]_{0.50}\text{-TPB-DMTP-COFs}$	34	100	95	89	88:12
4	$(S)\text{-1-(pyrrolidin-2-yl)methyl)-4-phenoxyethyl-triazole}$	22	100	96	92	91:9
5	$[(S)\text{-Py}]_{0.25}\text{-H}_2\text{P-COF}$	36	-	-	-	-



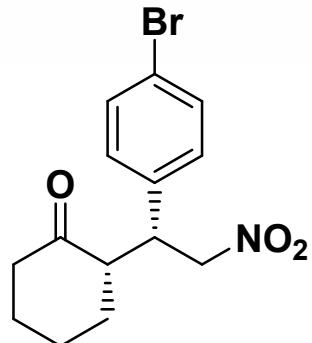
12 h, dr = 90/10  
ee = 92%



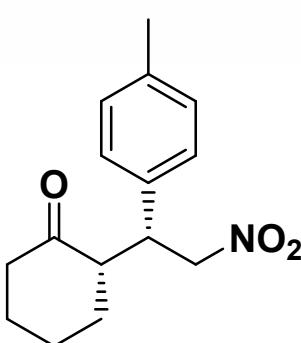
10 h, dr = 90/10  
ee = 90%



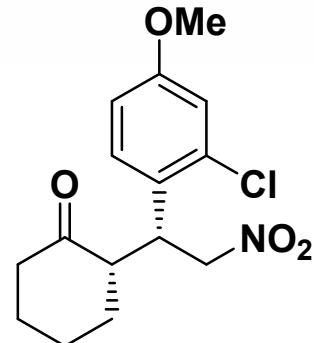
6 h, dr = 97/3  
ee = 94%



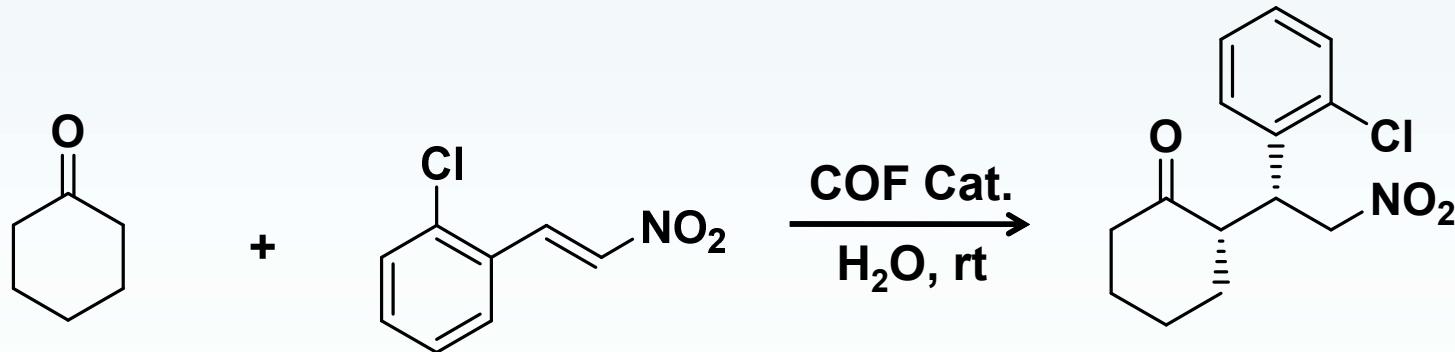
12 h, dr = 93/7  
ee = 95%



16 h, dr = 92/8  
ee = 93%



26 h, dr = 94/6  
ee = 96%



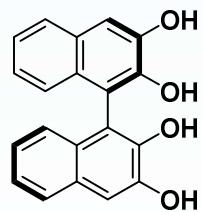
Cycle	Time (h)	Conv. (%)	Yield (%)	Ee (%)	dr	Ry(wt %)
1	6	100	95	94	97:3	>99
2	7	100	93	94	97:3	>99
3	8	100	94	94	97:3	>99
4	11	100	92	94	97:3	>99
5	13	100	92	93	97:3	>99

**COF Cat. = [(S)-Py]<sub>0.17</sub>-TPB-DMTP-COFs**



# 展望

■ 手性COF材料 → 手性催化 手性分离



■ 模板法合成COF材料

MOF → COF      破坏配位键

COF → COF      B-O键水解

2D → 3D      环加成

