Literature Report

Catalytic Enantioselective Minisci-type Addition to Heteroarenes

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Proctor, R. S. J.; Davis, H. J.; **Phipps, R. J.** *Science* **2018**, *360*, 419-422.

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Catalytic Minisci-type Additon to Heteroarenes by Fu

Catalytic Minisci-type Additon to Heteroarenes by Phipps



CV of Robert J Phipps



Education:

200x-2006 M.S., Imperial College London.

2006-2010 Ph. D., University of Cambridge.

2010-2013 Postdoctoral, University of California, Berkeley.

2013-Now Research Fellow, University of Cambridge.

Research:

- > Exploring new designs of multifunctional catalysts.
- Exploring the non-covalent interactions for catalysis.

Minisci-type Addition



Minisci, F. et al. Tetrahedron 1971, 27, 3575.



MacMillan, D. W. C. et al. Nature 2015, 525, 87.



Chen, G.; Liu, P. et al. Chem. Sci. 2016, 7, 6407.

 \mathbb{R}^1

 \dot{R}^2

up to 96% yield



Fu, Y.; Shang, R. et al. Chem. Eur. J. 2017, 23, 2537.





Ir-cat.: [Ir(dF(CF₃)ppy)₂(dtbbpy)]PF₆



Fu, Y.; Shang, R. et al. ACS Catal. 2017, 7, 907.

Proposed Mechanism



Optimization of the Reaction Parameters

N + 200 N N NPhth Blue LEDs		
1a 2a ^{3a} _{Ph}		
Entry Variations of conditions ^a Yield 3a (%	6) ^b	
1 None 85		
2 $Ru(bpy)_3(PF_6)_2$ instead of Ir-cat. <5	<5	
3 Without photocatalyst N.D.		
4 $AI(OTf)_3$ instead of PA-1 62		
5 PA-2 instead of PA-1 60		
6 PA-3 instead of PA-1 27	27	
7 Without PA-1 15		
8 Without light <5		
9 Using 1 mol% Ir-cat. 84		

^a Reaction conditions: **1a** (0.20 mmol), **2a** (0.30 mmol%), photocatalyst (2 mol%), acid catalyst (2 mol%), DMA (2 mL). ^b Isolated yields.







Enantioselective Minisci-type Addition



Phipps, R. J. et al. Science 2018, 360, 419.

Optimization of the Reaction Parameters



^a Reaction conditions: **4a** (0.10 mmol), **5a** (0.15 mmol), Photocatalyst (2 mol%), acid catalyst (5 mol%), Solvent (2 mL). ^b Isolated yields ^c Determined by HPLC.

Optimization of the Reaction Parameters



Entry	Variations of conditions ^a	Yield (%) ^b	Ee (%) ^c	
1	None	94	94	
2	Without Photocatalyst	0	0	
3	Without (R)-TRIP	0	0	
4	Without light	0	0	
5	1 mol% Photocatalyst	87	94	
6	1.1 equiv 5a	89	93	
7	4a (0.25 mmol), 5a (1.1 equiv)	93	96	

^a Reaction conditions: **4a** (0.10 mmol), **5a** (0.15 mmol), Photocatalyst (2 mol%), acid catalyst (5 mol%), Solvent (2 mL). ^b Isolated yields ^c Determined by HPLC.







Control Experiments

A: Investigation of a REA bearing no hydrogen bond donor



B: Kinetic effect experiment



Proposed Mechanism



Summary



Heteroarenes with basic nitrogen centers, of which pyridines and quinolines are the most common classes, are ubiquitous in pharmaceuticals, agrochemicals, and small molecules of medicinal interest. The nitrogen's basicity precludes traditional electrophilic aromatic substitution pathways, necessitating alternative strategies for elaboration. One widely used protocol is the addition of free radicals to protonated heterocycles, pioneered by Minisci in the 1960s and 1970s and often referred to as "Minisci-type" chemistry. Recent renewed interest has led to numerous advances in radical generation, allowing for milder conditions and convenient radical precursors.

Furthermore, photoredox catalysis has provided exciting avenues for radical generation in Minisci-type additions with precursors that include alcohols, ethers, boronic acids, carboxylic acids, and redoxactive esters (RAEs). However, control over absolute stereochemistry in the addition of prochiral radicals has proven elusive, even though many of the aforementioned protocols result in the formation of a stereocenter at the benzylic position.

Given the extraordinary pace of advancement of the field of photoredox catalysis and the range of molecules that are susceptible to Brønsted acid activation, we envisage that our successful merger of these strategies to address the challenge of enantioselective Minisci chemistry will have broad impact in both areas.

