# **Literature Report 1**

Diastereo- and Atroposelective Synthesis of Bridged Biaryls Bearing an Eight-Membered Lactone through an Organocatalytic Cascade

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Zhao, Y. et al. J. Am. Chem. Soc. 2019, 141, 17062.

## CV of Prof. Zhao Yu



#### **Education:**

- > **1998-2002** B.S., Peking University
- > 2002-2008 Ph.D., Boston University
- 2008-2011 Postdoctoral associate, MIT
- 2011-2017 Assistant Professor, NUS
- > 2017-Present Associate Professor, NUS

#### **Research:**

- **D** Enantioselective Catalysis
- Organic and Organometallic Synthesis
- Mechanistic Studies

#### Introduction



### Introduction



Bringmann, G. et al. Angew. Chem. Int. Ed. 2005, 44, 5384.

## **Direct Arylation Strategy**





[Pd(NHC)(η<sup>3</sup>-cin)Cl] (0.5 mol%) KOH (2.0 equiv) <sup>t</sup>BuOH, 30 °C, 12 h



up to 99% yield

up to 99% ee

 $Y = B(OH)_2$ , Bneo Bpin, or  $BF_3K$ 

X = Br, Cl, or OTf



Shi, S.-L. et al. J. Am. Chem. Soc. 2019, 141, 14938.

#### **Dynamic Kinetic Resolution Strategy**



Miller, S. J. et al. Science 2010, 328, 1251.

#### **Construction of an Aromatic Ring**



Li, X. et al. J. Am. Chem. Soc. 2019, 141, 9527.

### **Construction of an Aromatic Ring**



Zhao, Y. et al. J. Am. Chem. Soc. 2019, 141, 17062.

#### **N-Heterocyclic Carbenes**

#### **General structures of N-heterocyclic carbens**



#### **Common reaction intermediates**



#### **Breslow Intermediates**



#### **Extended Breslow Intermediates**



Bode, J. W. et al. J. Am. Chem. Soc. 2004, 126, 14370.

#### α,β-Unsaturated Acylazolium Intermediates



## **Previous Work**



Zhao, Y. et al. Angew. Chem. Int. Ed. 2014, 53, 11041.



Zhao, Y. et al. Org. Lett. 2019, 21, 6169.

### **Proposed Mechanism**



Zhao, Y. et al. Org. Lett. 2019, 21, 6169.

#### **Reaction Assumption**



## **Reaction Assumption**



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	+ <i>n</i> -Pr	Cat. A (10 mol%) 2 equiv. DIPEA DCM, 24 °C, 36 h 4 Å MS A1 Ar = Ph A2 Ar = C <sub>6</sub> F <sub>5</sub> A3 Ar = 2,4,6-Me <sub>3</sub> C <sub>6</sub> H <sub>2</sub> $\Theta$ BF <sub>4</sub>	A4
aEntry	Cat.	Yield (%) <sup>b</sup>	Ee (%) <sup>c</sup>
1	A1	76	90
2	A2	72	80
3	A3	85	99
4	<b>A</b> 4	trace	n.d.

<sup>a</sup>Reaction were performed with **1a** (0.15 mmol,1.0 equiv.), **2b** (0.60 mmol,4.0 equiv.), Cat. **A** (10 mol%), DIPEA (0.30 mmol, 2.0 equiv) and 4 Å molecular sevie (100 mg) in DCM (2.0 mL) at 24 °C for 36 h. <sup>b</sup>Isolated yield. <sup>c</sup>Determined by HPLC analysis on a chiral stationary phase. n.d. = not determined.



<sup>a</sup> Entry	Base	Yield (%) <sup>b</sup>	Ee (%) <sup>c</sup>
1	DIPEA	85	99
2	NaOAc	trace	n.d.
3	$Cs_2CO_3$	trace	n.d.

<sup>a</sup>Reaction were performed with **1a** (0.15 mmol,1.0 equiv.), **2b** (0.60 mmol,4.0 equiv.), cat. **A3** (10 mol%), base (0.30 mmol, 2.0 equiv) and 4 Å molecular sevie (100 mg) in DCM (2.0 mL) at 24 °C for 36 h. <sup>b</sup>Isolated yield. <sup>c</sup> Determined by HPLC analysis on a chiral stationary phase. n.d. = not determined.



aEntry	Solvent	Yield (%) <sup>b</sup>	Ee (%) <sup>c</sup>
1	DCM	85	99
2	Toluene	70	97
3	CHCl <sub>3</sub>	70	98
4	THF	65	98

<sup>a</sup>Reaction were performed with **1a** (0.15 mmol,1.0 equiv.), **2b** (0.60 mmol,4.0 equiv.), Cat. **A3** (10 mol%), DIPEA (0.30 mmol, 2.0 equiv) and 4 Å molecular sevie (100 mg) in solvent (2.0 mL) at 24 °C for 36 h. <sup>b</sup>Isolated yield. <sup>c</sup>Determined by HPLC analysis on a chiral stationary phase.

#### **Substrate Scope**



#### **Reaction Assumption**





<sup>a</sup> Entry	Solvent	Yield (%) <sup>b</sup>	Ee (%) <sup>c</sup>
1	DCM	20	97
2	Toluene	17	96
3	THF	40	94
4	Dioxane	28	95
5	CH <sub>3</sub> CN	trace	n.d.

<sup>a</sup>Reaction were performed with **5** (0.15 mmol,1.0 equiv.), **2b** (0.60 mmol,4.0 equiv.), Cat. **A3** (10 mol%), DIPEA (0.30 mmol, 2.0 equiv) and 4 Å molecular sevie (100 mg) in solvent (2.0 mL) at 24 °C for 36 h. <sup>b</sup>Isolated yield. <sup>c</sup>Determined by HPLC analysis on a chiral stationary phase. n.d. = not determined.

#### **Substrate Scope**



#### **Proposed Mechanism**



#### **Proposed Mechanism**



#### **Product Transformation**



## Summary



> NHC-catalyzed propargylic substitution & cyclization;

- Bridged biaryls with an eight-membered lactons;
- Excellent control of both axial & central chirality.

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# Writing thought



The significance of axially chiral biaryls has been widely recognized in catalysis and drug delivery, due to their exceptionally compact architectures and promising pharmacological profiles. An important subclass is bridged biaryls, which bear an additional linkage between the two arens and often with stereocenters. These structures are of high natural abundance, as exemplified by the anti-leukemic lignan (-)-steganacin and the tubin inhibitor (-)-rhazinilam. Notably, these compounds are structurally related to helicenes, a significant class of helically chiral, ortho-condensed polycyclic compounds.





In summary, we have developed a catalytic, diastereo- and atropoenantioselective synthesis of bridged biaryls from readily available starting materials. This NHC-catalyzed transformation proceeds through a propargylic substitution-two-directional cyclization cascade and delivers the products with uniformly excellent control of axial and central chirality. The development of new catalytic procedures to access other challenging structures bearing different types of chirality is currently under investigation. Due to their significant utility, extensive efforts were devoted to the atropoenantioselective synthesis of axially chiral biaryls.

Previous investigation in our laboratory has led to the development of NHC-catalyzed enantioselective acylative kinetic resolution of alcohols/phenols as well as desymmetrization of meso-bisphenols.

In addition to desymmetrization of symmetrical triols, we were curious whether a chemoselective cascade cyclization could be achieved for unsymmetrical racemic triols as well.

To the best of our knowledge, the direct catalytic synthesis of bridged biaryls with simultaneous control of axial and central chirality remains elusive in the literature.

# Thanks for your attention