

Nickel-Catalyzed Enantioconvergent and Diastereoselective Allenylation of Alkyl Electrophiles: Simultaneous Control of Central and Axial Chirality

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ABSTRACT: In recent years, remarkable progress has been described in the development of methods that simultaneously control vicinal stereochemistry, wherein both stereochemical elements are central chirality; in contrast, methods that control central and axial chirality are comparatively rare. Herein we report that a chiral nickel catalyst achieves the enantioconvergent and diastereoselective coupling of racemic secondary alkyl electrophiles with prochiral 1,3-enynes (in the presence of a hydrosilane) to generate chiral tetrasubstituted allenes that bear an adjacent stereogenic center. A carbon–carbon and a carbon–hydrogen bond are formed in this process, which provides good stereoselectivity and is compatible with an array of functional groups.

Because the construction of carbon–carbon bonds and the control of stereochemistry are two central challenges in modern organic synthesis,^{1,2} the development of methods, particularly catalytic processes, that simultaneously achieve both objectives is an important goal.³ Metal-catalyzed coupling reactions of alkyl electrophiles with carbon (pro)nucleophiles are a powerful tool for carbon–carbon bond formation, and numerous enantioselective variants have been described that control a single stereochemical element at the new carbon–carbon bond (e.g., central or axial chirality).^{4,5} Controlling *two* stereochemical elements further elevates the challenge and the potential of such couplings, and initial progress has recently been described in addressing this objective, specifically, controlling *two stereogenic centers* (Figure 1A).^{6–9} In contrast, to our knowledge the challenge of controlling *two different* stereochemical elements (e.g., central and axial chirality) has not yet been achieved (Figure 1B).^{10–12}

Figure 1C provides an outline of part of a pathway through which this type of catalytic asymmetric coupling might proceed. Thus, a chiral transition-metal hydride undergoes a site-selective, regioselective, and stereoselective β -migratory insertion into the prochiral double bond of a 1,3-enyne to form a secondary propargylmetal complex (A). A stereospecific 1,3-migration of this intermediate generates an allenylmetal complex (B), wherein central chirality has been transformed into axial chirality. Intermediate B then couples with the racemic alkyl halide via an organic radical to furnish the desired allene, which includes a stereogenic axis and center.

Herein, we describe the development of a method that achieves the coupling of an alkyl electrophile to generate a carbon–carbon bond with control of both central and axial chirality (Figure 1B), specifically, a nickel-catalyzed reaction of racemic alkyl halides with prochiral 1,3-enynes to afford enantioenriched allenes (Figure 1D).¹³

RESULTS AND DISCUSSION

Chiral allenes are of substantial interest, as they are found in natural products and have utility in wide-ranging fields such as asymmetric catalysis,¹⁴ medicinal chemistry,^{15,16} and materials science;¹⁷ consequently, considerable attention has been focused on the development of methods for their synthesis.^{18–22} We envisioned a new approach to the synthesis of enantioenriched allenes, through a metal-catalyzed coupling of an alkyl halide with a 1,3-enyne in the presence of a hydride reagent (Figure 1C).²³

Upon surveying an array of parameters, we determined that NiBr₂·glyme and chiral pyridine–oxazoline ligand L* can catalyze the coupling of racemic alkyl electrophiles with 1,3-enynes in the presence of a hydrosilane to generate the desired allene not only in good yield but also with good stereoselectivity (Figure 2); it is noteworthy that, to control central chirality in this process, the catalyst must efficiently differentiate between two sp³-hybridized carbon substituents of the electrophile (alkyl vs CF₃). Compounds that bear a trifluoromethyl substituent are of significant interest in areas such as materials chemistry, agrochemistry, and particularly the pharmaceutical industry, since the trifluoromethyl group is an increasingly common functional group in marketed drugs.^{24–26}

This new nickel-catalyzed asymmetric allenylation is not highly sensitive to moisture or to air (Figure 2, entry 1: under otherwise identical conditions, in the presence of 1.0 equiv of water or 1.0 mL of air: $\geq 60\%$ yield, 92% ee, and $\geq 90:10$ dr; see Section VI of the Supporting Information), and it proceeds with

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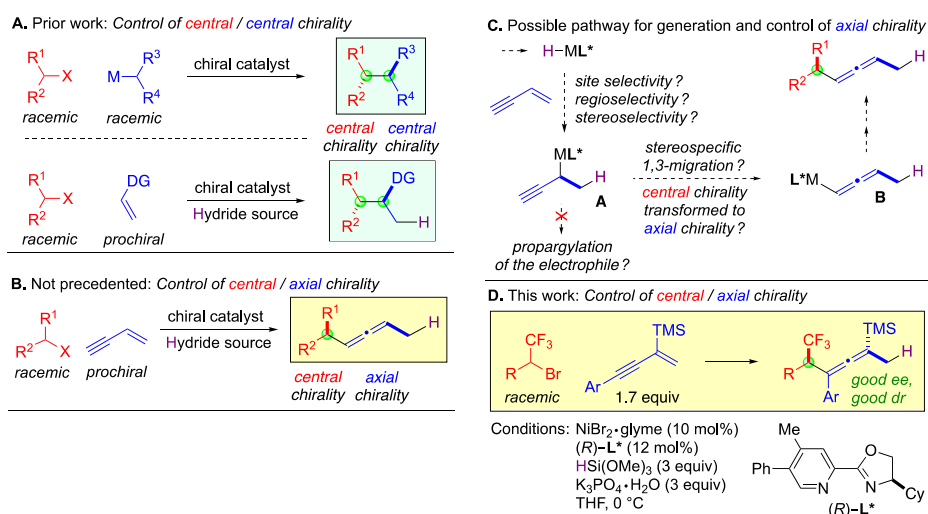


Figure 1. Metal-catalyzed couplings of racemic alkyl electrophiles with carbon (pro)nucleophiles: Simultaneous control of two stereochemical elements at the new carbon–carbon bond. (A) Prior work: Control of central/central chirality. (B) Not precedented: Control of central/axial chirality. (C) Outline of a possible pathway for generation and control of axial chirality. (D) This work: Control of central/axial chirality. DG = directing group, X = halide.

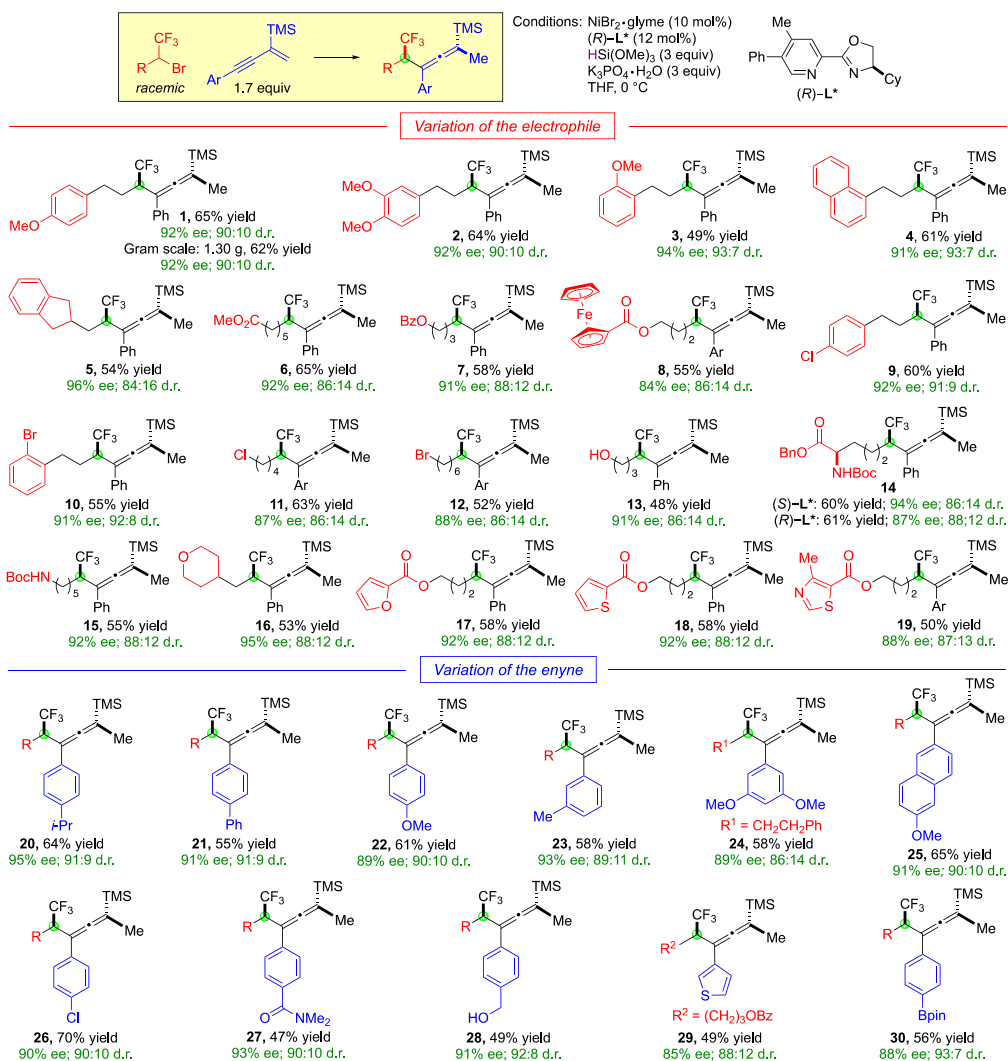


Figure 2. Asymmetric allenylations of racemic alkyl electrophiles: Scope. Reactions were conducted on a 0.8 mmol scale, unless otherwise noted. All data are the average of two experiments. All yields are of purified products. Ar = *p*-anisyl. R = CH₂CH₂(*p*-anisyl), unless otherwise noted.

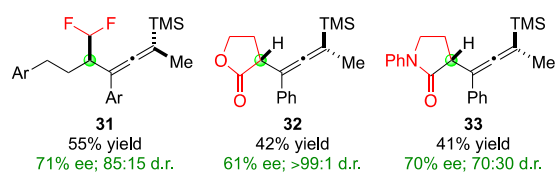


Figure 3. Asymmetric allenylations of racemic alkyl electrophiles: New families of electrophiles (preliminary results under the standard coupling conditions (Figure 2) without further optimization). Reactions were conducted on a 0.8 mmol scale. All data are the average of two experiments. All yields are of purified products. Ar = *p*-anisyl.

similar efficiency when conducted on a gram, rather than a 0.8 mmol, scale (entry 1). A variety of trifluoromethyl-substituted alkyl bromides serve as suitable coupling partners (entries 1–19), and the method is tolerant of an array of functional groups: in addition to an ether and ester (entries 1–3 and 6–8), it is compatible with potentially problematic functionalities such as an aryl chloride and bromide, primary alkyl chloride and bromide, hydroxyl, secondary carbamate, and a variety of heterocycles (tetrahydropyran, furan, thiophene, and thiazole) (entries 9–19; also, a secondary alkyl bromide,²⁷ quinoline,

indole, and epoxide: see the Supporting Information). The stereoselectivity is relatively consistent, with ee's typically >90% and d.r.'s generally $\geq 86:14$ (entries 1–19).

With respect to the 1,3-enyne coupling partner, an array of asymmetric allenylations proceed with good stereoselectivity (Figure 2, entries 20–30), including when the aromatic substituent is/bears an aryl chloride, tertiary amide, hydroxyl, thiophene, or boronate ester (entries 26–30).²⁸ While the coupling displays good functional-group tolerance, it is less tolerant of steric hindrance; for example, an α -branched electrophile and a 1,3-enyne that bears an ortho-substituted aromatic group are not suitable reaction partners under the present conditions (see Section V of the Supporting Information).

In preliminary studies, we have obtained promising lead results when applying this method directly to other families of electrophiles, indicating the potential breadth for this strategy for catalytic asymmetric allenylation (Figure 3). Not only an electrophile that is modestly different (CF_2H in place of CF_3) but also those that are entirely distinct (an α -bromolactone and an α -bromolactam) couple with encouraging stereoselectivity.

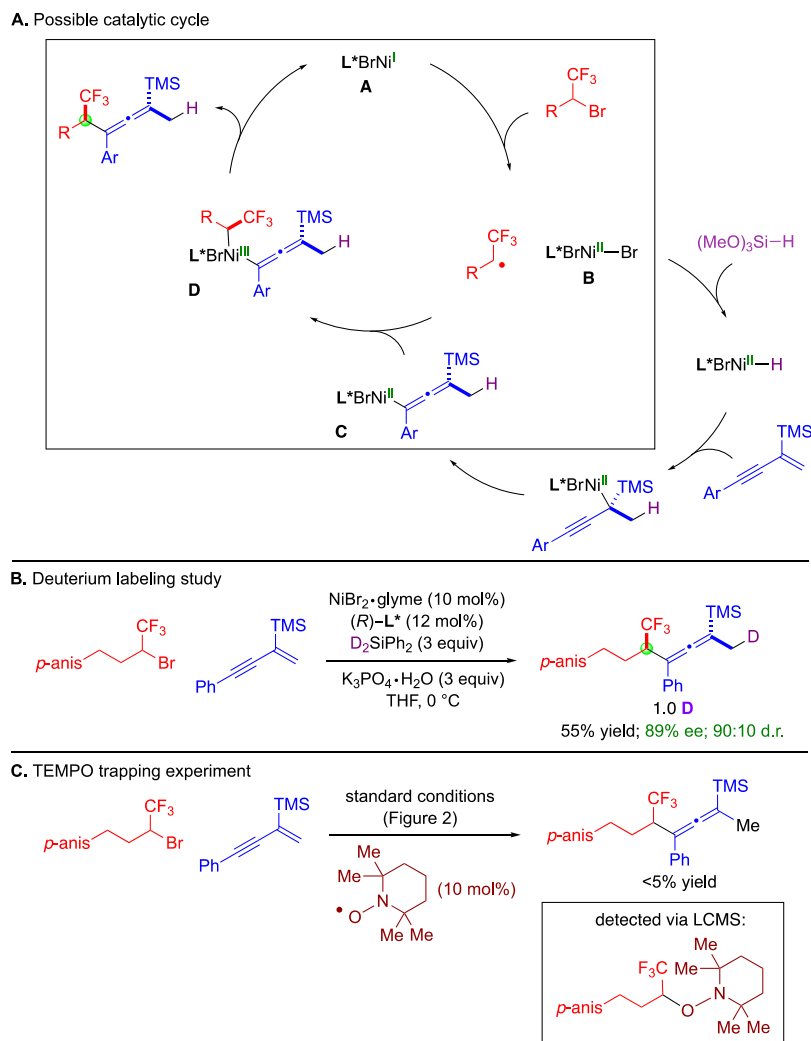


Figure 4. Asymmetric allenylations of racemic alkyl electrophiles: Mechanistic studies. (A) Possible catalytic cycle. Not every intermediate is illustrated, and for the sake of simplicity the elementary steps are depicted as irreversible. (B) Deuterium labeling study. (C) TEMPO trapping experiment.

With regard to mechanism, we hypothesize that these asymmetric allenylations may be proceeding in analogy to a pathway that we recently suggested for nickel-catalyzed enantioconvergent couplings of racemic alkyl halides with terminal olefins in the presence of a hydrosilane (Figure 4A).^{29,30} We carried out preliminary mechanistic studies to gain insight into our new process. To examine the reversibility of oxidative addition/olefin binding/ β -migratory insertion, we employed a deuterated silane in the asymmetric allenylation (commercially available D_2SiPh_2 ; H_2SiPh_2 provides comparable results to $HSi(OMe)_3$ under our standard coupling conditions); our observation that 1.0 deuterium is incorporated into the allene product suggests that oxidative addition/olefin binding/ β -migratory insertion are not highly reversible (Figure 4B). Furthermore, when an asymmetric allenylation is conducted in the presence of 2,2,6,6-tetramethylpiperidin-1-oxyl (TEMPO), a radical trap, coupling is inhibited and a TEMPO adduct of the electrophile is observed (Figure 4C), consistent with the generation of an organic radical from the alkyl bromide under the reaction conditions.³¹

CONCLUSIONS

We developed a nickel-catalyzed enantioconvergent and diastereoselective cross-coupling between racemic alkyl halides and prochiral 1,3-enynes to form chiral tetrasubstituted allenes. Whereas tremendous advances have been reported in achieving stereoselective coupling reactions of alkyl electrophiles, to the best of our knowledge this is the first such process wherein elements of central and axial chirality are controlled in the new carbon–carbon bond. Further efforts are underway to exploit earth-abundant metals in the catalytic asymmetric synthesis of motifs of interest in medicinal chemistry.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/jacs.4c00593>.

Experimental details: general information, preparation of coupling partners, preparation of L^* , procedure for enantioconvergent and diastereoselective cross-coupling, effect of reaction parameters, studies of functional-group compatibility, assignment of absolute configuration, mechanistic studies, NMR spectra, SFC data, and references (PDF)

Accession Codes

CCDC 2311642 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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- (28) Obtaining a 70% yield of coupling product with good stereoselectivity (entry 26 of Figure 2) when the racemic electrophile is the limiting reagent indicates that both enantiomers of the electrophile are being converted into the product; that is, this is an enantioconvergent reaction of the electrophile, not a simple kinetic resolution. At partial conversion, the unreacted electrophile is racemic.
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