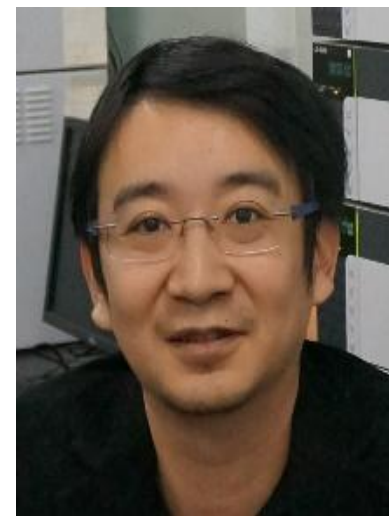


# Highly Enantioselective Aza-Michael Reaction between Alkyl Amines and $\beta$ -Trifluoromethyl $\beta$ -Aryl Nitroolefins

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Reporter: Lian-Jin Liu  
Checker: Wen-Xue Huang  
Date: 12/01/2016



Peking University,  
Shenzhen Graduate School

Huang, Y. *et al. Angew. Chem. Int. Ed.* **2015**, *54*, 15414–15418.

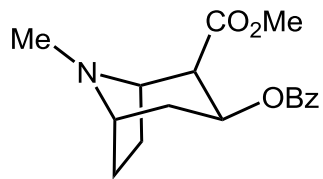
# Contents

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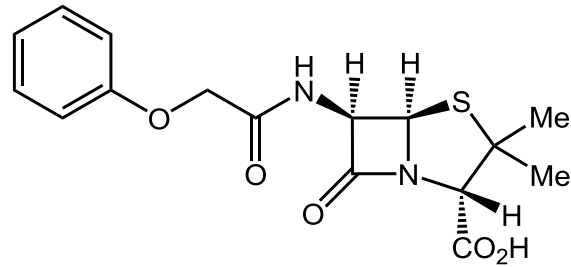
- ❖ **Introduction**
- ❖ **Enantioselective aza-Michael reaction with secondary-amine catalysis**
- ❖ **Enantioselective aza-Michael reaction with Brønsted acid catalysis**
- ❖ **Enantioselective aza-Michael reaction with noncovalent catalysis**
- ❖ **Summary**

## ➤ Introduction

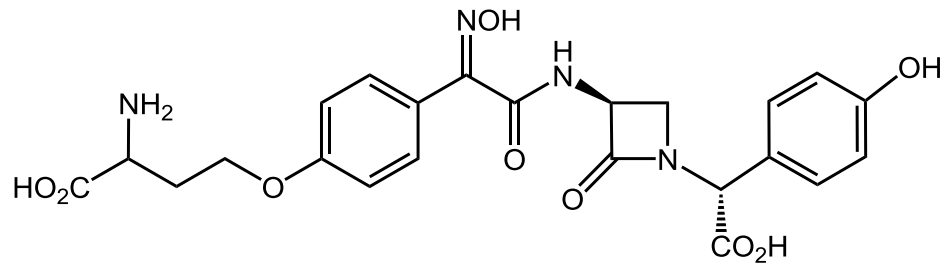
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**(-)-Cocaine**

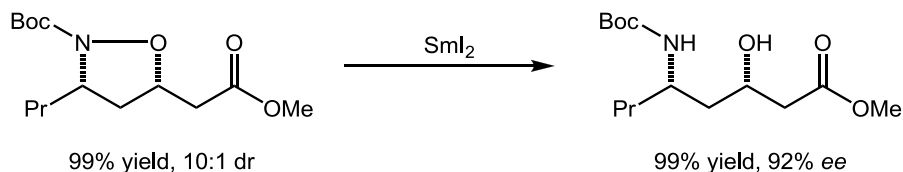
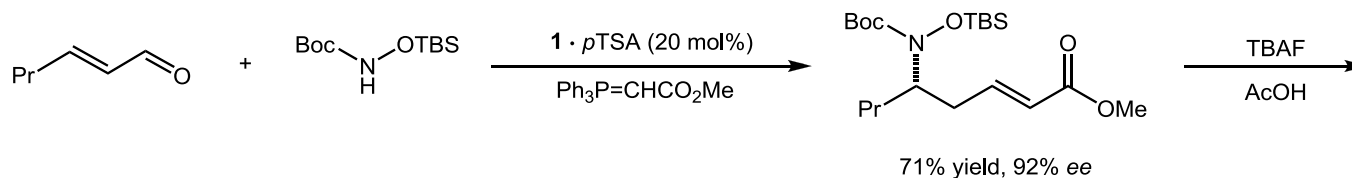
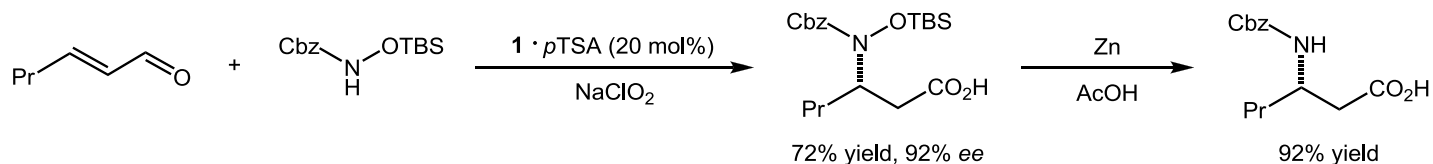
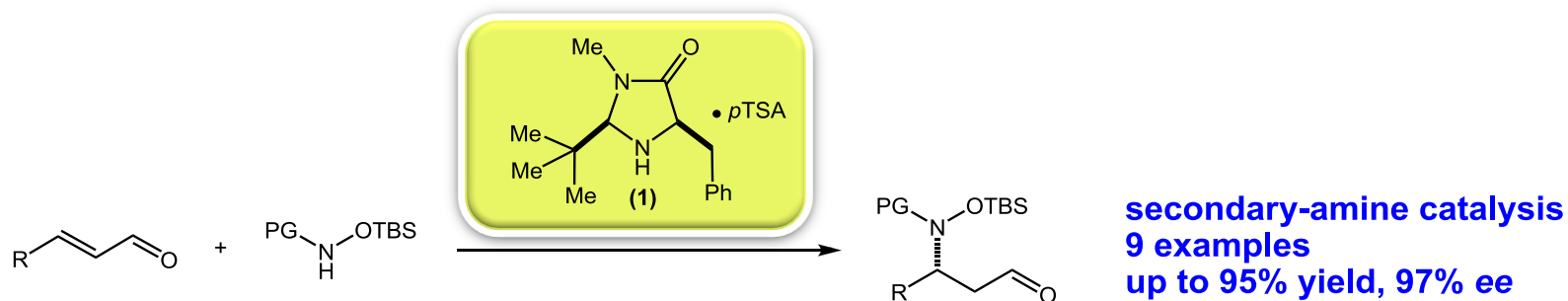


**Penicillin V**

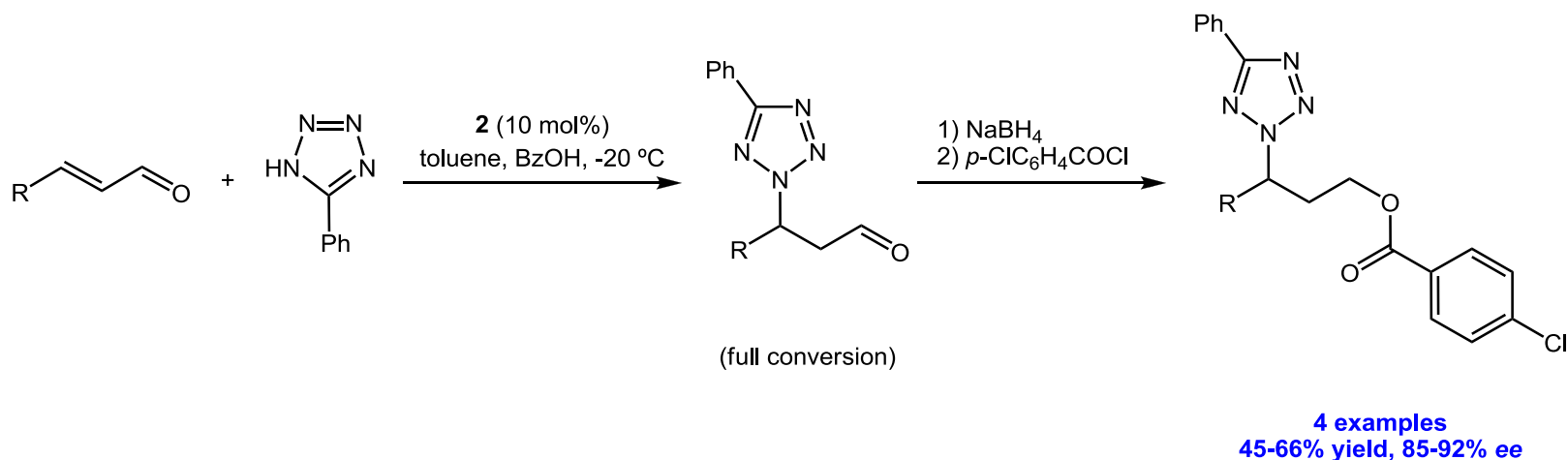
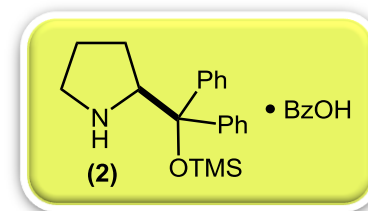
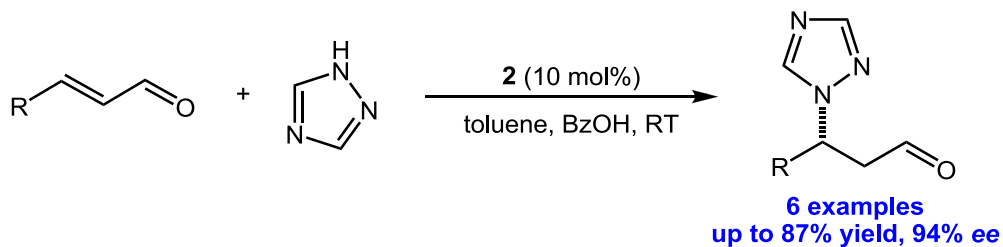


**(-)-Nocardicin A**

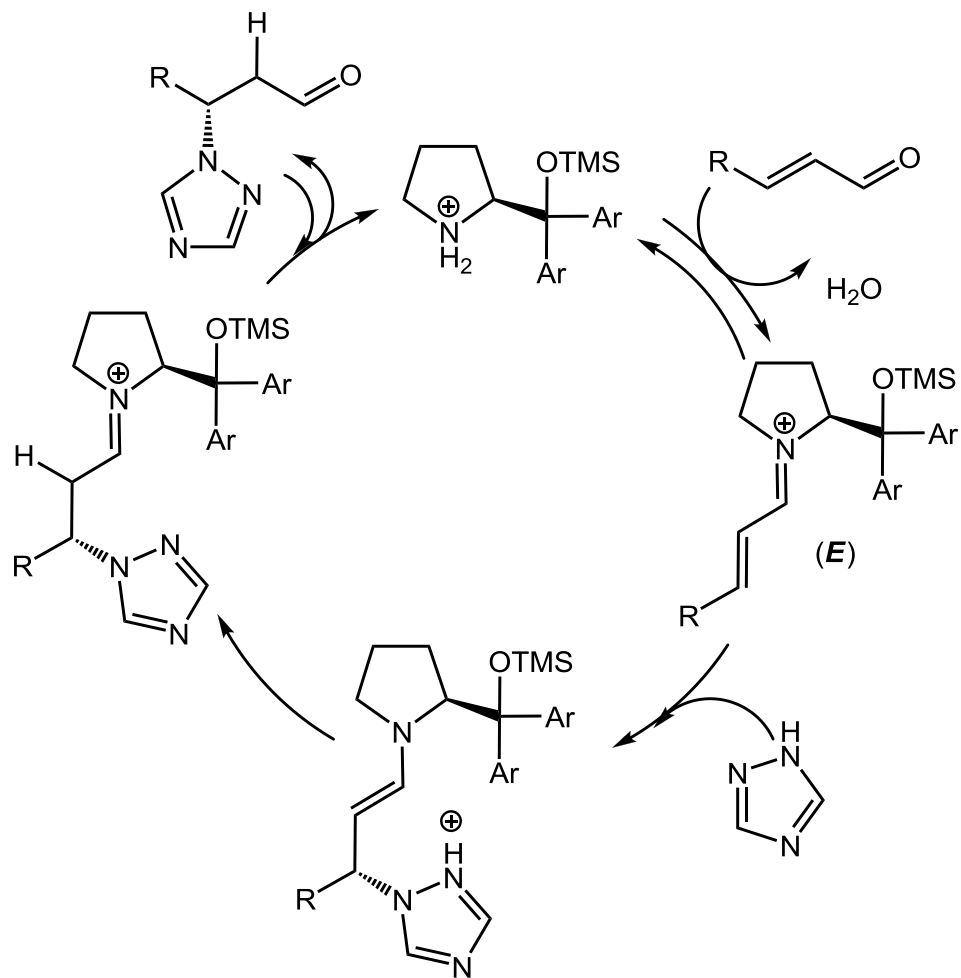
# Enantioselective aza-Michael reaction with secondary-amine catalysis



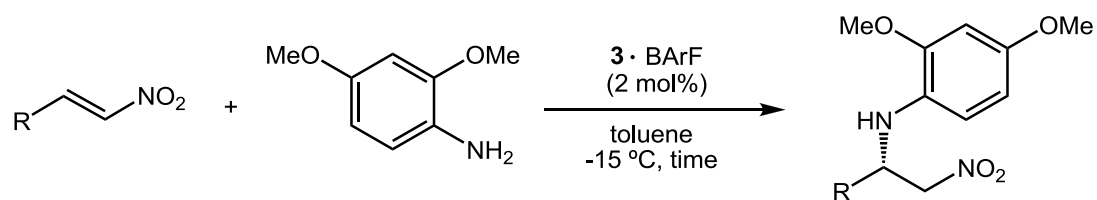
# Enantioselective aza-Michael reaction with secondary-amine catalysis



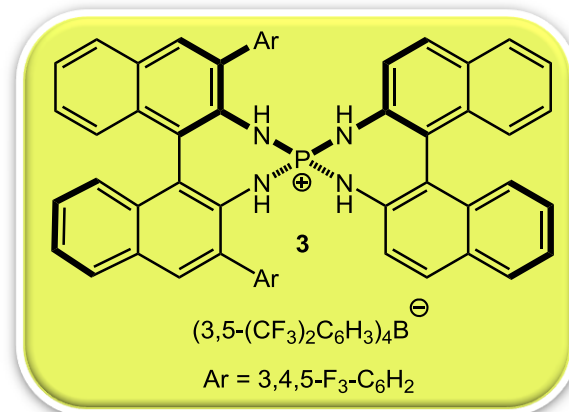
## ► Proposed catalytic cycle for aza-Michael reaction



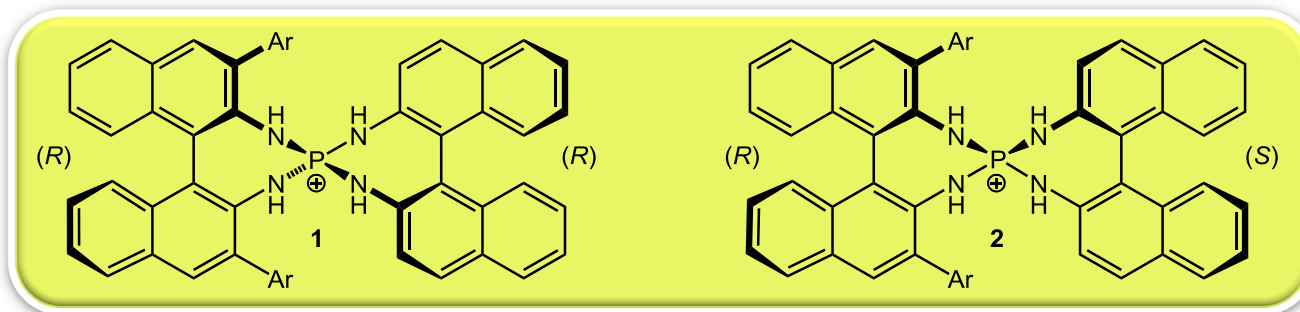
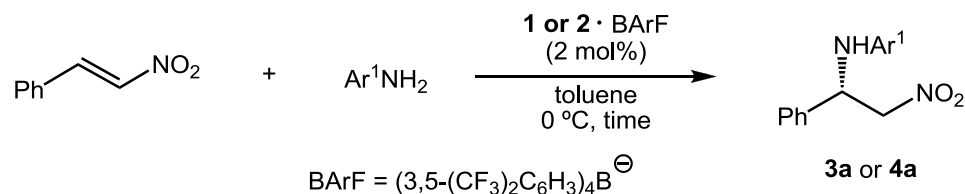
# Enantioselective aza-Michael reaction with Brønsted acid catalysis



**12 examples**  
up to 99% yield, 97% ee



## ➤ Effect of the structure of the arylamine and catalyst

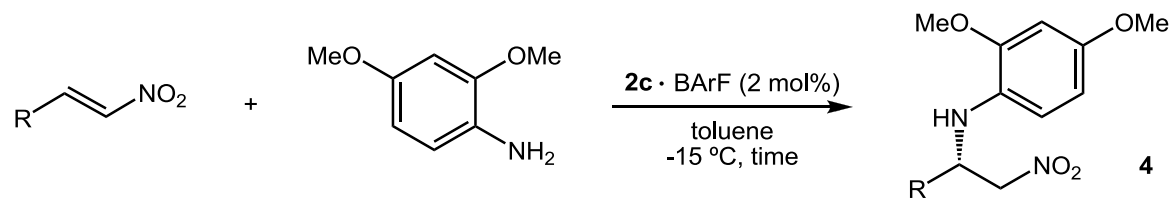


Entry	Catalyst (Ar)	Ar <sup>1</sup>	Time [h]	Yield [%]	Ee [%]	Prod.
1	<b>1a (H)</b>	4-MeO-C <sub>6</sub> H <sub>4</sub>	23	37	19	<b>3a</b>
2	<b>1a (H)</b>	2,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	23	42	41	<b>4a</b>
3	<b>1b (Ph)</b>	2,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	23	86	61	<b>4a</b>
4	<b>2b (Ph)</b>	2,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	27	87	83	<b>4a</b>
5	<b>2c (3,4,5-F<sub>3</sub>-C<sub>6</sub>H<sub>2</sub>)</b>	2,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub>	11	98	94	<b>4a</b>
<b>6<sup>a</sup></b>	<b>2c (3,4,5-F<sub>3</sub>-C<sub>6</sub>H<sub>2</sub>)</b>	<b>2,4-(MeO)<sub>2</sub>-C<sub>6</sub>H<sub>3</sub></b>	<b>12</b>	<b>98</b>	<b>95</b>	<b>4a</b>

<sup>a</sup> Reaction was conducted at -15 °C.



## ► Scope of nitroolefins



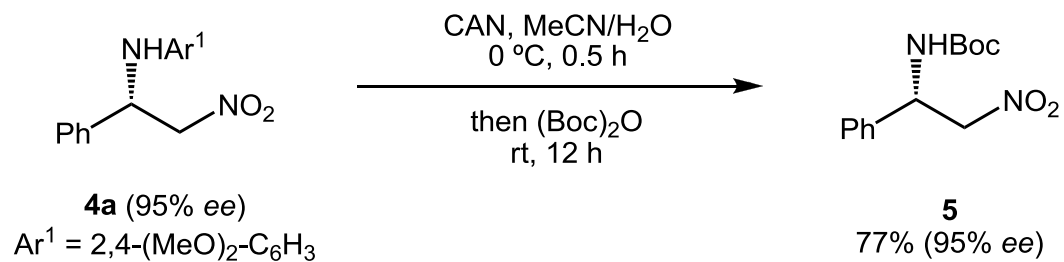
Entry	R	Time [h]	Yield [%]	Ee [%]	Prod.
1	4-F-C <sub>6</sub> H <sub>4</sub>	12	98	94	<b>4b</b>
2	4-Cl-C <sub>6</sub> H <sub>4</sub>	4	99	95	<b>4c</b>
3	4-Br-C <sub>6</sub> H <sub>4</sub>	4	99	94	<b>4d</b>
4	4-Me-C <sub>6</sub> H <sub>4</sub>	12	99	97	<b>4e</b>
5	3-MeO-C <sub>6</sub> H <sub>4</sub>	4	99	93	<b>4f</b>
6	3-Br-C <sub>6</sub> H <sub>4</sub>	9	99	93	<b>4g</b>
7	2-F-C <sub>6</sub> H <sub>4</sub>	12	99	92	<b>4h</b>
8	1-naphthyl	19	99	91	<b>4i</b>
9	2-naphthyl	19	99	95	<b>4j</b>
10	3-furyl	24	89	94	<b>4k</b>
11 <sup>a</sup>	Me <sub>2</sub> CHCH <sub>2</sub>	7	98	89	<b>4l</b>
12 <sup>a,b</sup>	Me(CH <sub>2</sub> ) <sub>4</sub>	0.5	93	87	<b>4m</b>

<sup>a</sup> Diisopropyl ether was used as solvent instead of toluene .

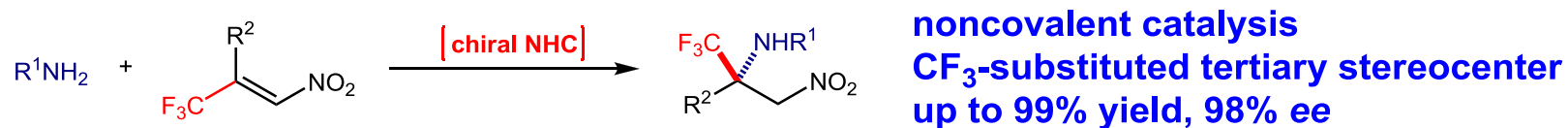
<sup>b</sup> Reaction was performed at room temperature.

## ► Deprotection-Reprotection process

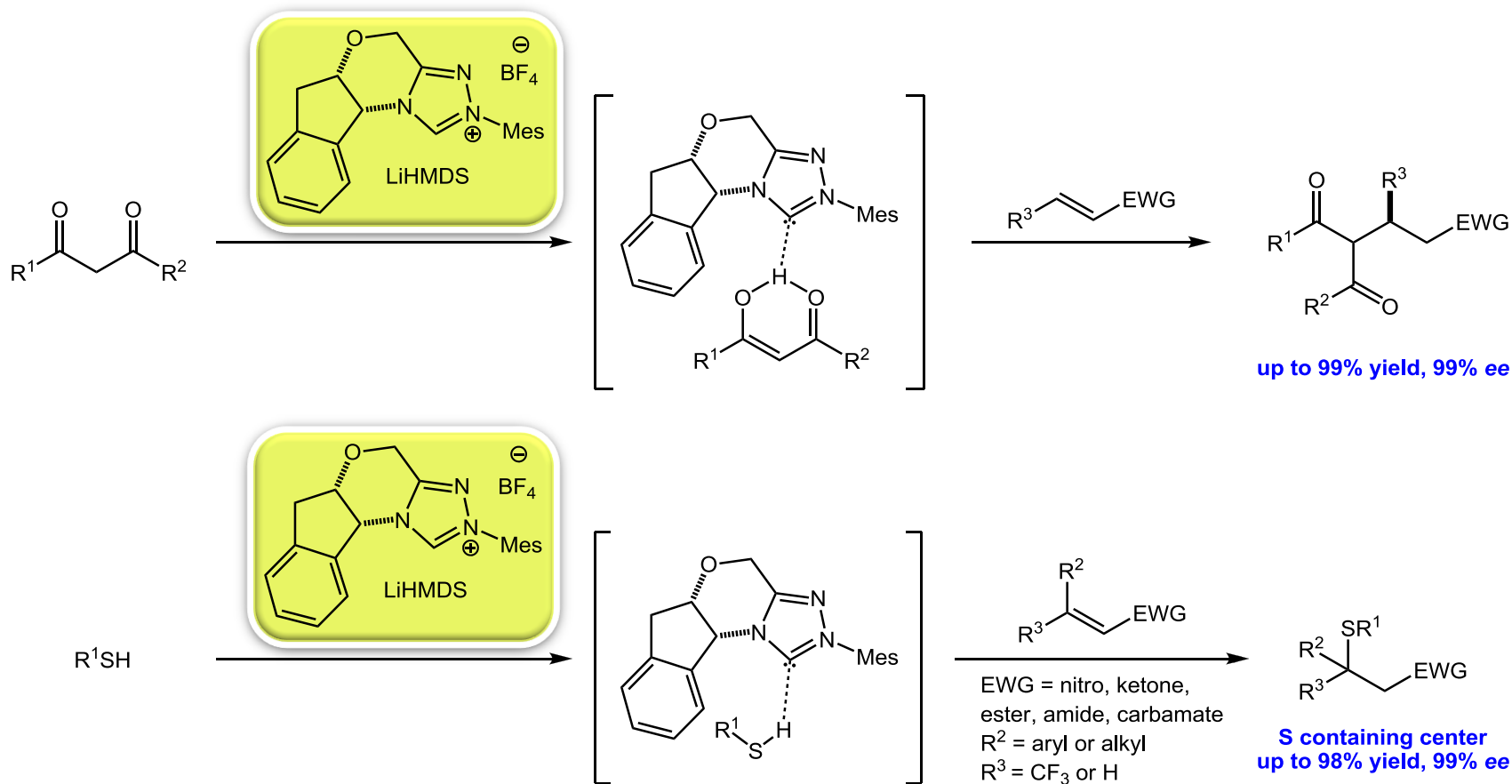
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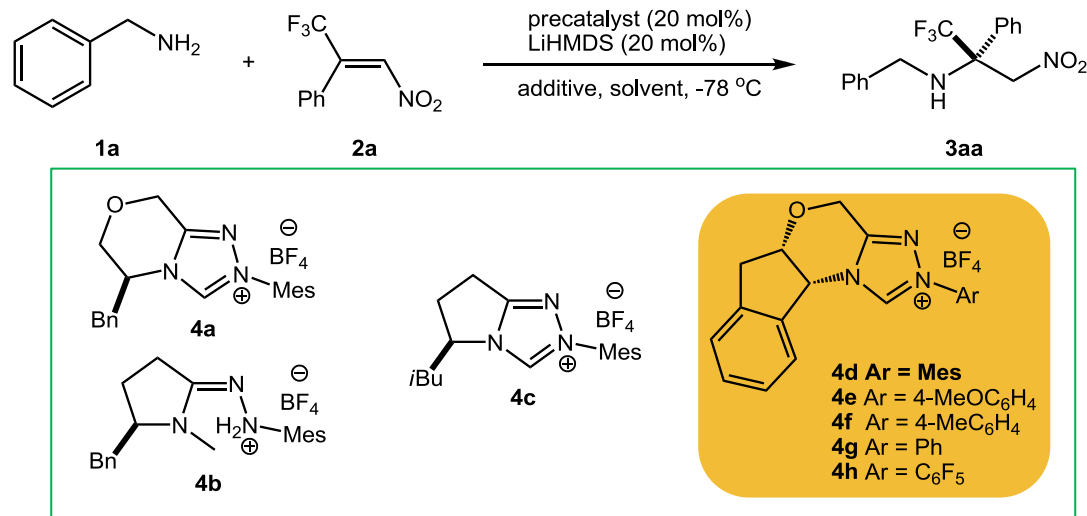
## ► Enantioselective aza-Michael reaction with noncovalent catalysis



## ➤ NHCs as non-covalent chiral catalysts for asymmetric reaction

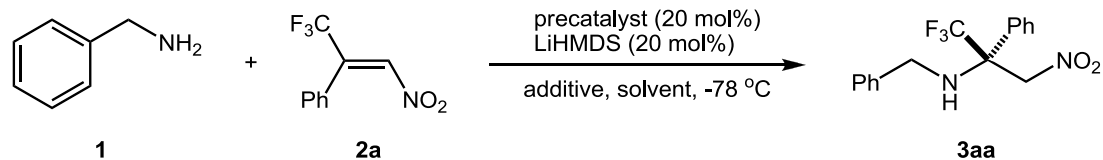


## ► Optimization of the reaction conditions



Entry	Catalyst	Additive	Solvent	Yield [%]	Ee [%]
1	<b>4a</b>	HFIP, 4 Å MS	toluene	88	-13
2	<b>4b</b>	HFIP, 4 Å MS	toluene	85	-16
3	<b>4c</b>	HFIP, 4 Å MS	toluene	90	2
<b>4</b>	<b>4d</b>	<b>HFIP, 4 Å MS</b>	<b>toluene</b>	<b>99</b>	<b>91</b>
5	<b>4e</b>	HFIP, 4 Å MS	toluene	80	51
6	<b>4f</b>	HFIP, 4 Å MS	toluene	90	28

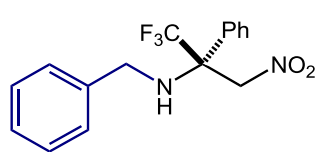
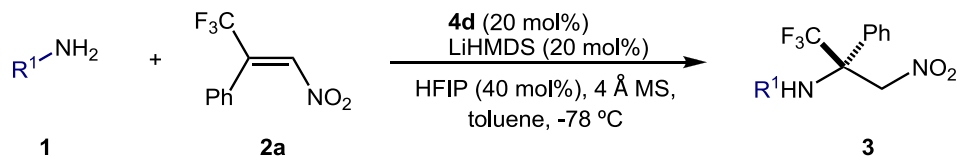
## ► Optimization of the reaction conditions



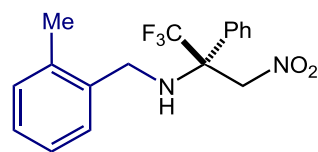
Entry	Catalyst	Additive	Solvent	Yield [%]	Ee [%]
7	<b>4g</b>	HFIP, 4 Å MS	toluene	70	7
8	<b>4h</b>	HFIP, 4 Å MS	toluene	50	0
9 <sup>a</sup>	<b>4d</b>	HFIP, 4 Å MS	toluene	83	15
10	<b>4d</b>	HFIP, 4 Å MS	toluene	57	52
11	<b>4d</b>	HFIP, 4 Å MS	CH <sub>2</sub> Cl <sub>2</sub>	70	3
12	<b>4d</b>	HFIP, 4 Å MS	THF	99	0
13	<b>4d</b>	HFIP, 4 Å MS	Et <sub>2</sub> O	70	76
14	<b>4d</b>	HFIP, 4 Å MS	MTBE	85	36
15	<b>4d</b>	4 Å MS	toluene	23	87
16	<b>4d</b>	HFIP	toluene	25	4

<sup>a</sup> The reaction was performed at -40 °C with 10 mol% of the catalyst.

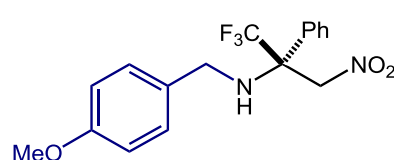
## ► Scope of the reaction with respect to the amine nucleophile



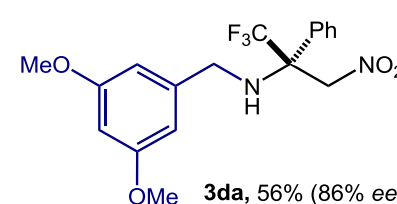
**3aa**, 90% (91% ee)



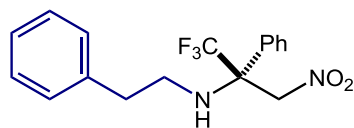
**3ba**, 67% (91% ee)



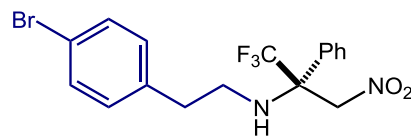
**3ca**, 89% (92% ee)



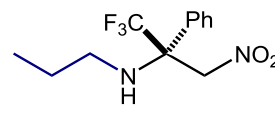
**3da**, 56% (86% ee)



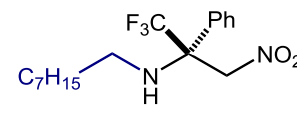
**3ea**, 85% (86% ee)



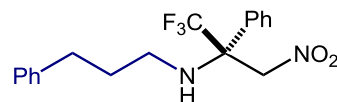
**3fa**, 68% (87% ee)



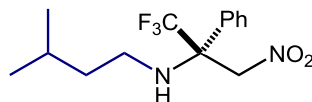
**3ga**, 80% (92% ee)



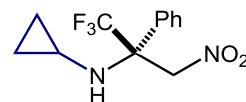
**3ha**, 87% (95% ee)



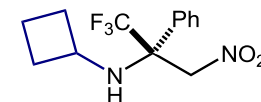
**3ia**, 82% (93% ee)



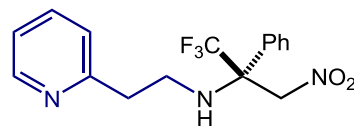
**3ja**, 79% (89% ee)



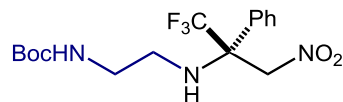
**3ka**, 79% (97% ee)



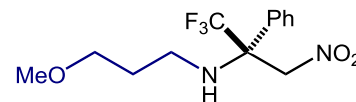
**3la**, 78% (94% ee)



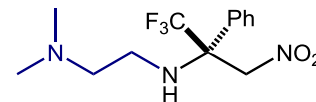
**3ma**, 87% (93% ee)



**3na**, 79% (91% ee)

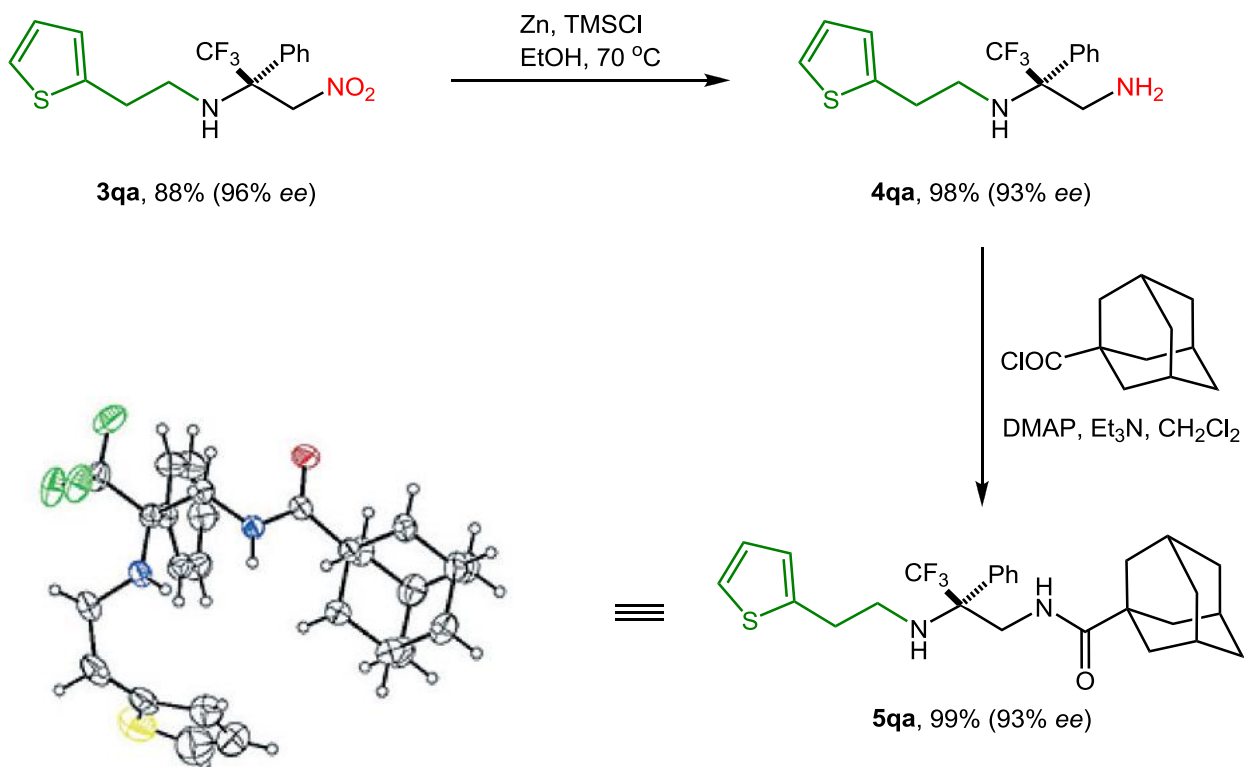


**3oa**, 87% (87% ee)



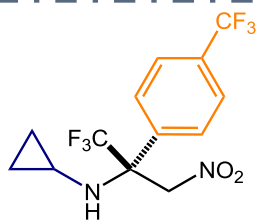
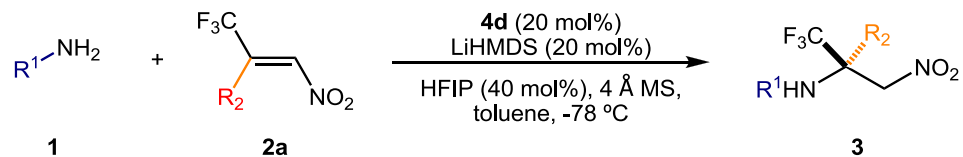
**3pa**, 99% (91% ee)

## ► The absolute configuration of product 3qa

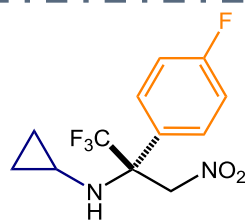




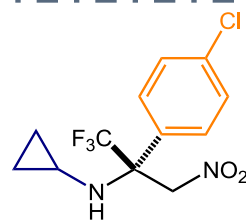
## ➤ Scope of the reaction with respect to the nitroalkene



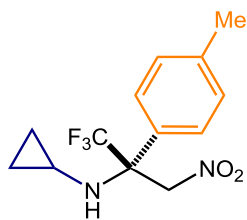
**3kb**, 64% (84% ee)



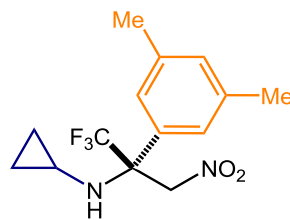
**3kc**, 77% (90% ee)



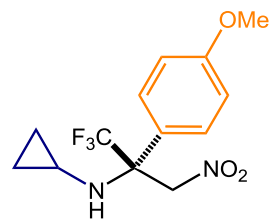
**3kd**, 92% (94% ee)



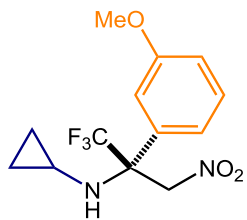
**3ke**, 78% (97% ee)



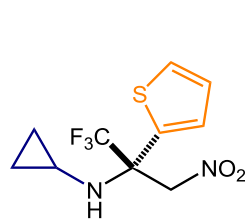
**3kf**, 57% (90% ee)



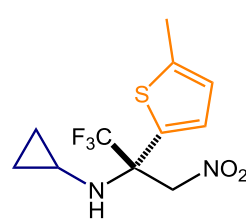
**3kg**, 99% (98% ee)



**3kh**, 83% (95% ee)

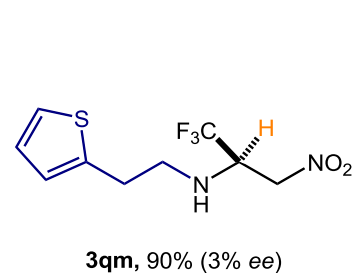
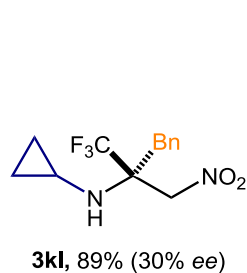
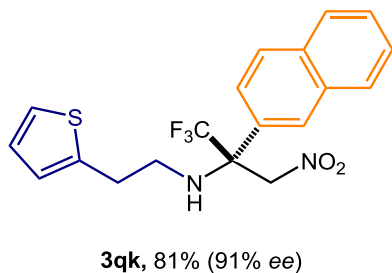
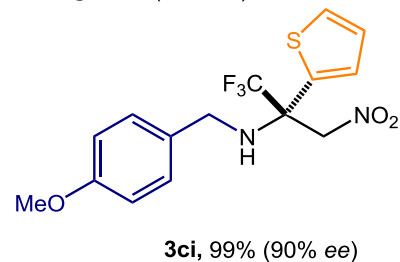
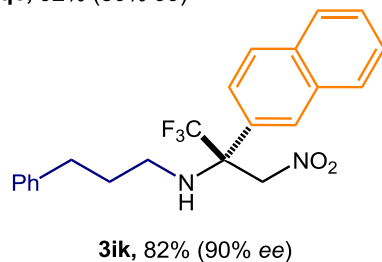
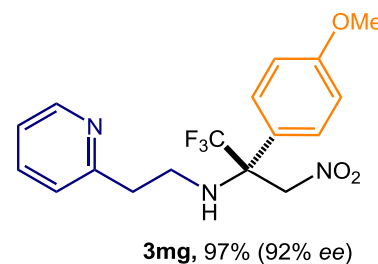
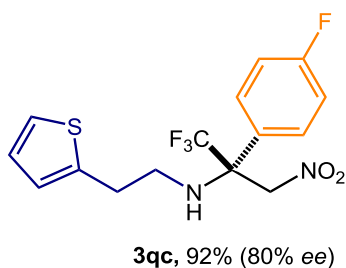
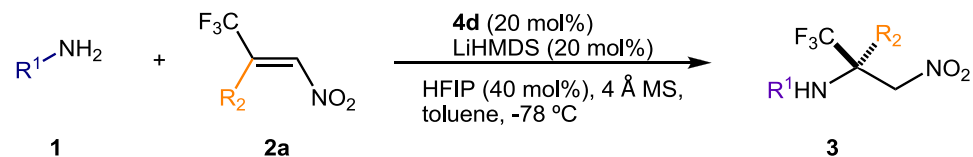


**3ki**, 98% (96% ee)

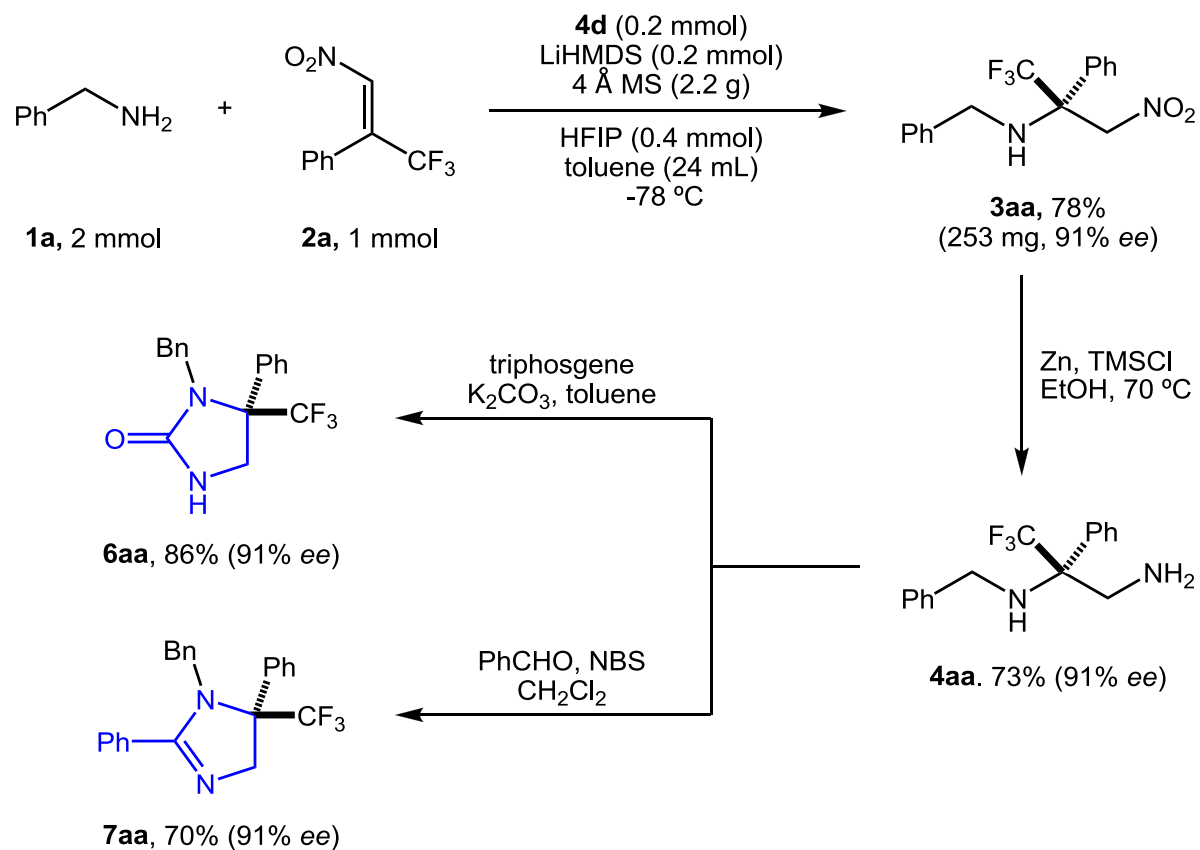


**3kj**, 99% (96% ee)

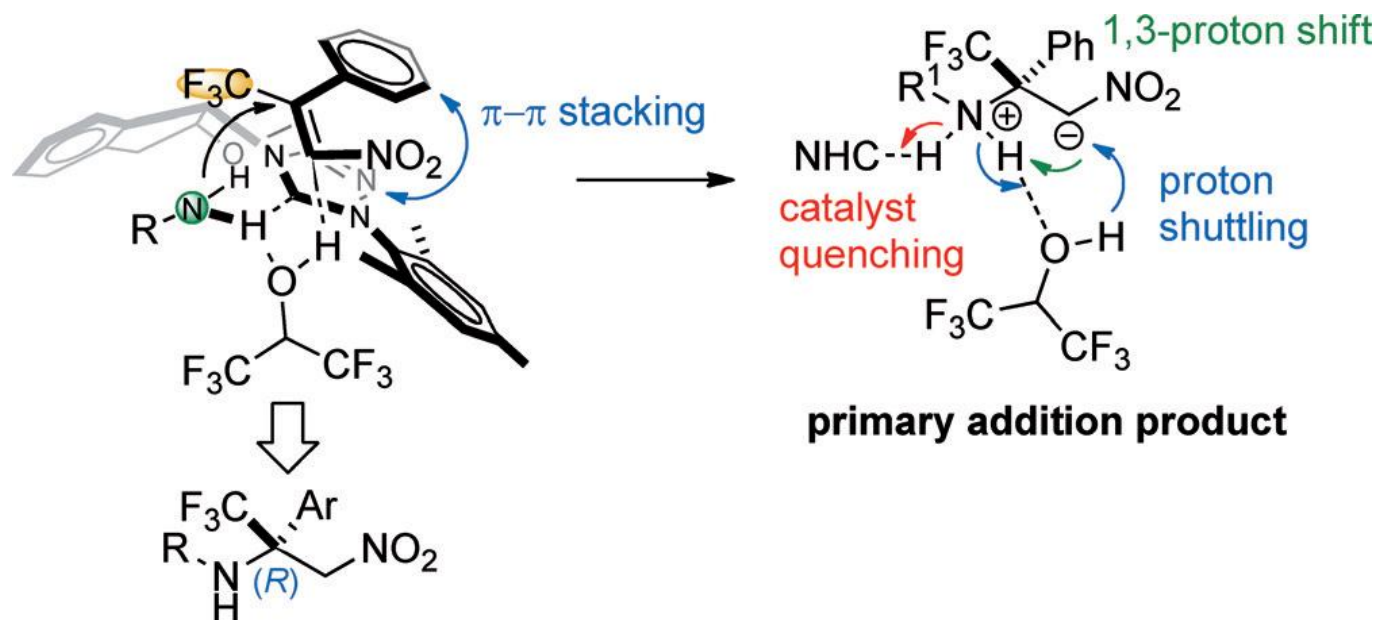
## ➤ Scope of the reaction with respect to the nitroalkene



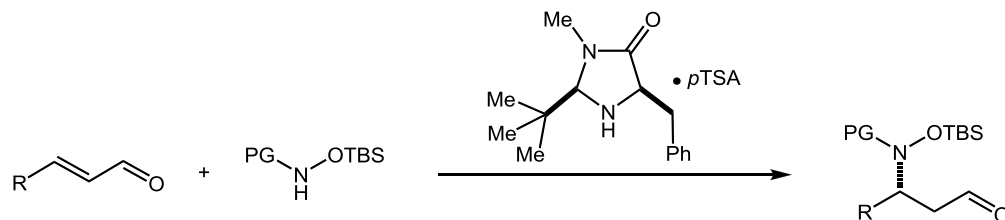
## ➤ Synthesis of CF<sub>3</sub>-containing chiral heterocycles



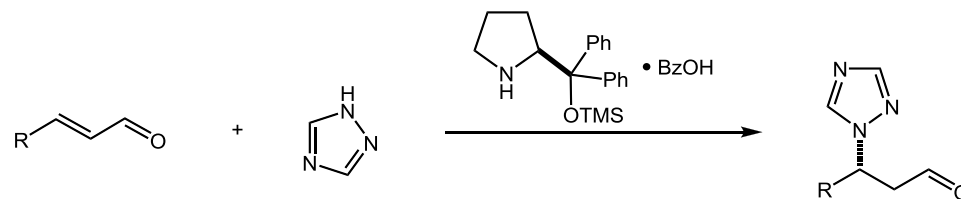
## ➤ Proposed transition state and proton-shuttling mechanism



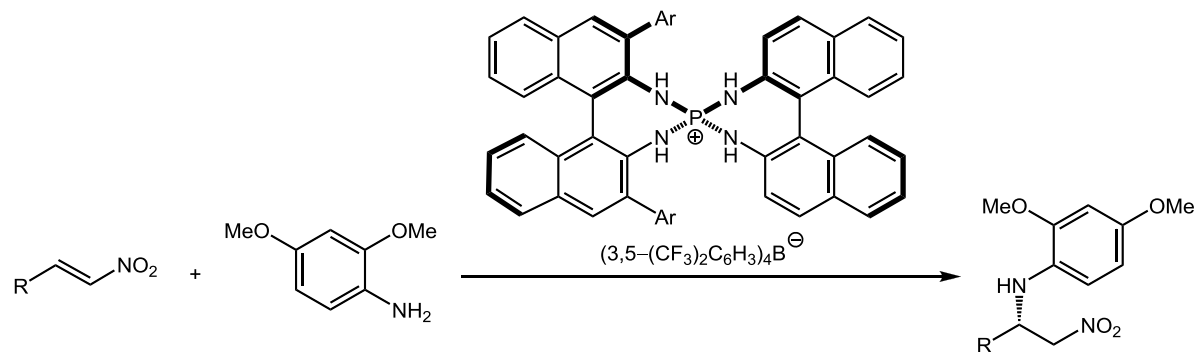
## ► Summary



**secondary-amine catalysis**  
MacMillan and co-workers  
up to 95% yield, 97% ee



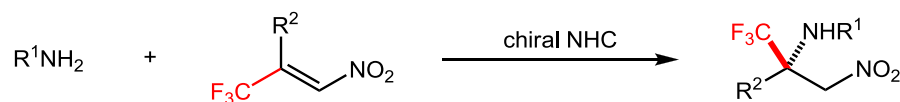
**secondary-amine catalysis**  
Jørgensen and co-workers  
up to 87% yield, 94% ee



**Brønsted acid catalysis**  
Ooi and co-workers  
up to 99% yield, 97% ee

## ➤ Summary

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**$R^1$  = alkyl**  
**HOMO-raising activation**  
**noncovalent catalysis**  
**Huang and co-workers**  
 **$CF_3$ -substituted tertiary stereocenter**  
**up to 99% yield, 98% ee**

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The asymmetric aza-Michael reaction is arguably one of the most straightforward methods for the synthesis of important 1,2-difunctionalized chiral building blocks, such as 1,2-diamines. Many successful combinations of nitrogen nucleophiles and electron-deficient olefins have been reported, with either Lewis acid catalysts or organocatalysts. However, two key problems remain largely unsolved: the use of alkyl amines as the nitrogen source, and the construction of tertiary stereocenters with a nitrogen substituent. Simple alkyl amines tend to form stable complexes with Lewis acids, which not only deactivates the catalyst, but also renders the aza-Michael reaction reversible.

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In the case of organocatalysis, primary and secondary amines interfere with both LUMO-lowering iminium activation (competing imine/iminium formation) and proton catalysis (acid–base neutralization). As a result, research efforts have been mostly concentrated on modulating the basicity of the donor nitrogen atom.



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In summary, we have developed the first highly enantioselective aza-Michael reaction of simple aliphatic amines. 1,2-Diamine analogues with a unique CF<sub>3</sub>-bearing tertiary stereocenter can be prepared in high yield with high enantioselectivity. The noncovalent, HOMO-raising activation of the NHC overcomes the intrinsic problem of using a basic amine in LUMO-lowering catalysis. Furthermore, an acidic proton shuttle is used to prevent catalyst quenching and retain turnover. A dual role of the NHC is proposed: activation by hydrogen bonding and π-π stacking. We expect that this generic HOMO-raising noncovalent activation mode by NHCs will find wide application in enantioselective catalysis and cascade reactions.