

# Enantioselective Hydroacylation of Ketones

**Reporter: Yue Ji**  
**Checker: Shubo Hu**  
**Date: 2016/09/26**

Dong, V. M. *et al.*  
*J. Am. Chem. Soc.* **2016**, *138*, 12013.

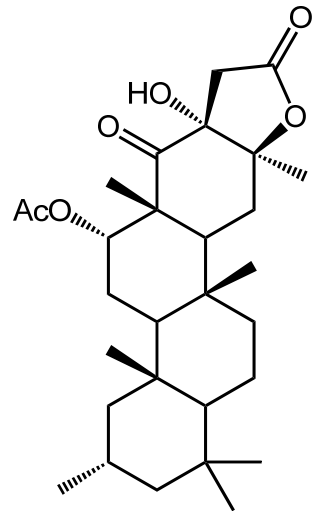
## Content

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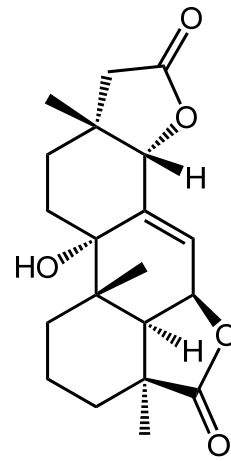
- **Introduction**
- **Rh-Catalyzed Enantioselective Hydroacylation of Ketones**
- **Co-Catalyzed Enantioselective Hydroacylation of Ketones**
- **Summary**

# Introduction

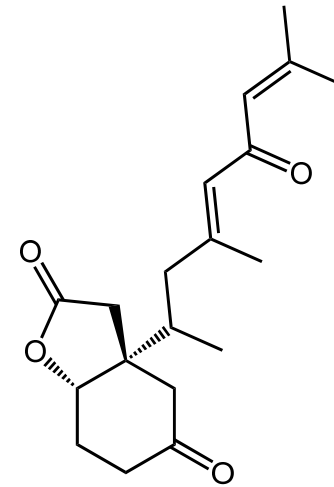
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**(+)-Hyatelone A**



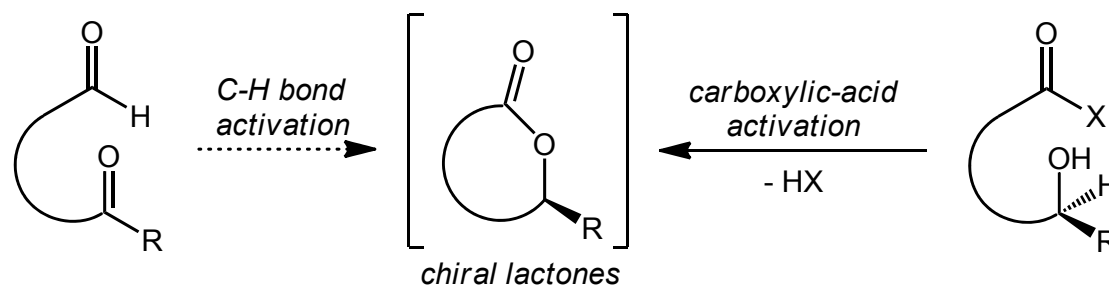
**(-)-Xylallantin B**



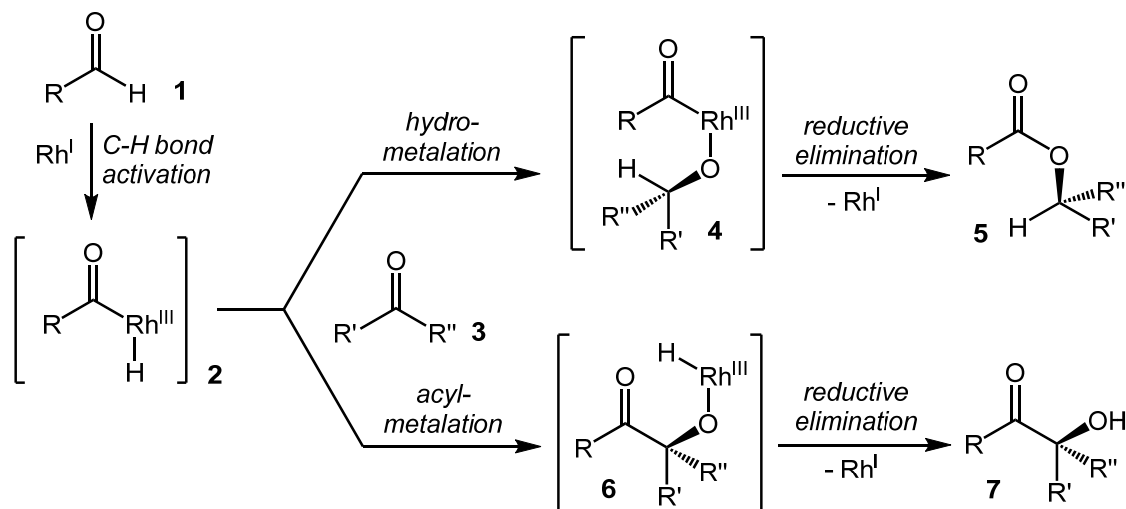
**(-)-Panamomon B**

# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

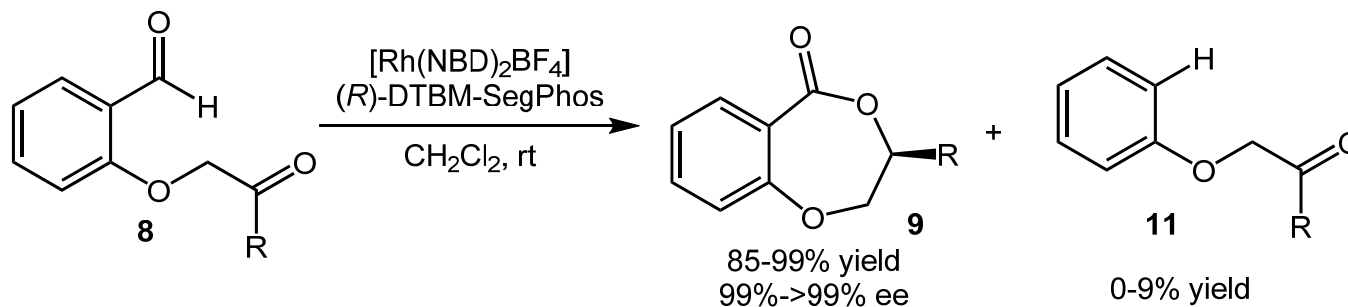
## An atom economical strategy for lactonization



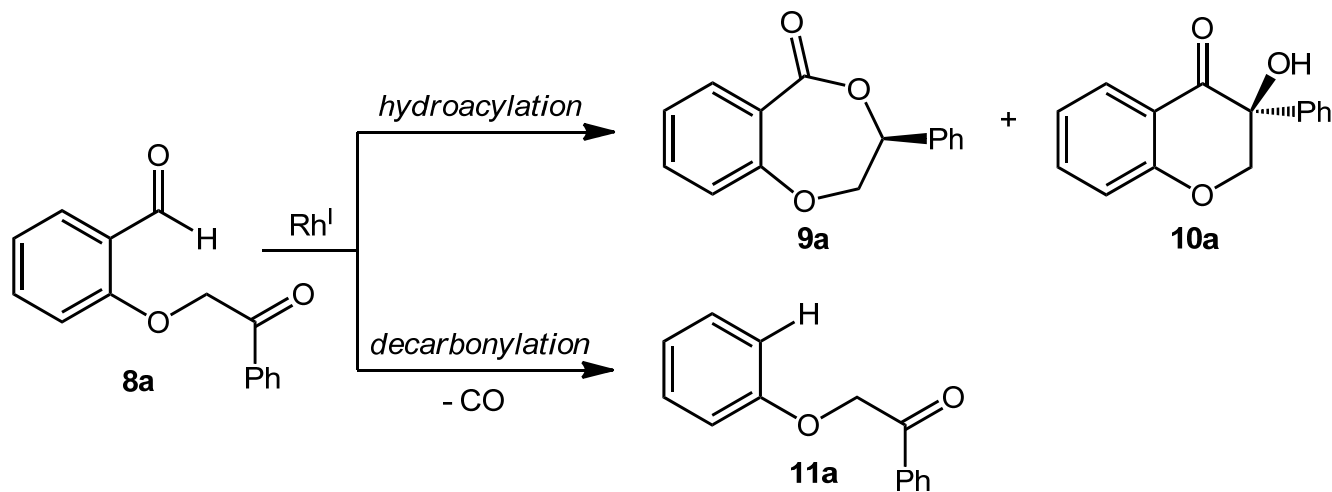
## Proposal for Tishchenko versus benzoin hydroacylation



## Rh-Catalyzed Enantioselective Hydroacylation of Ketones

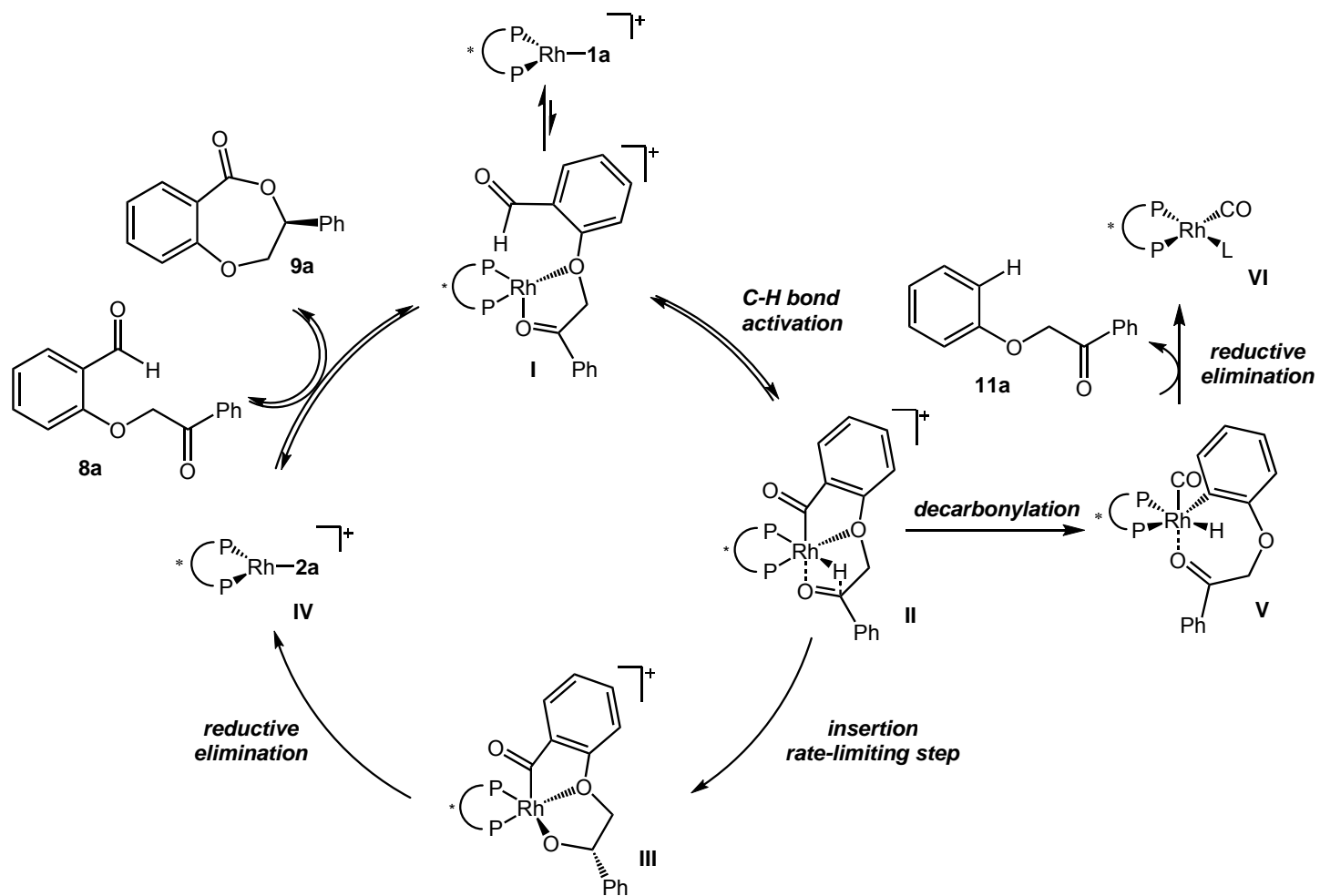


### Competing transformations for model substrate

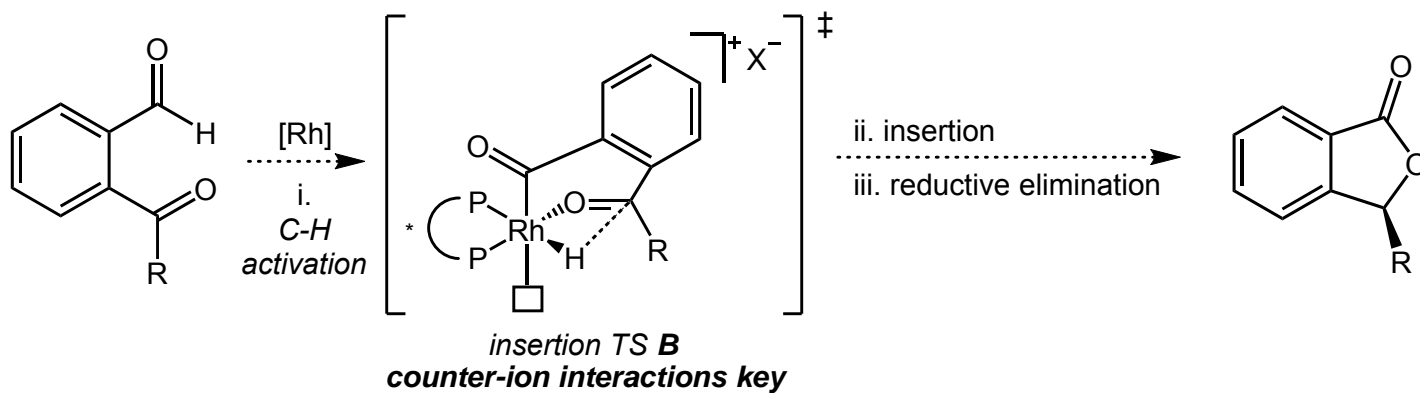
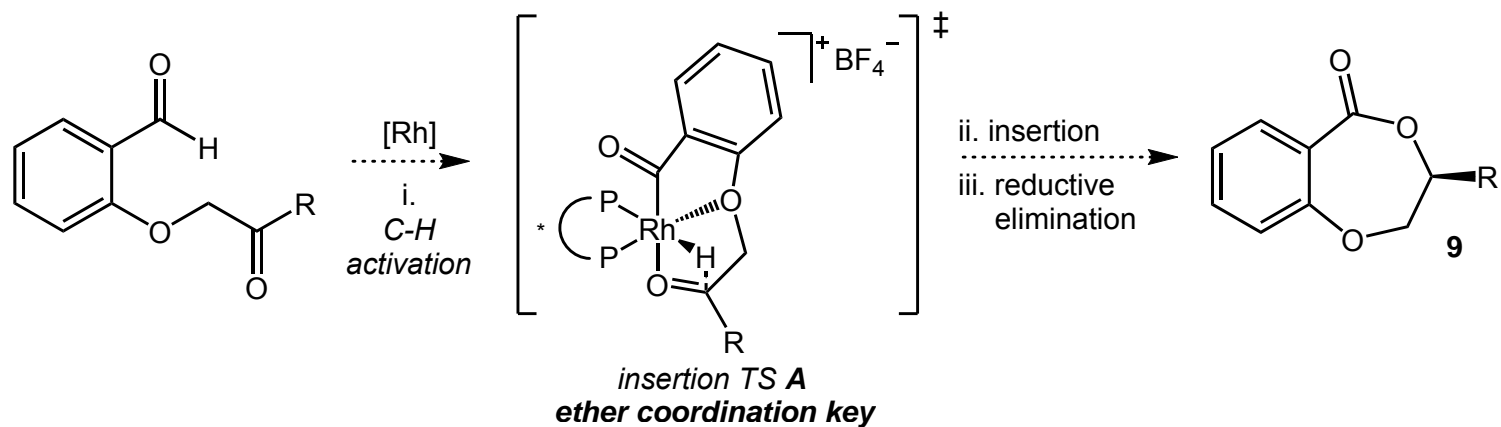


Dong, V. M. *et al.* *J. Am. Chem. Soc.* **2008**, *130*, 2916.  
Tsuji, J. *et al.* *Tetrahedron Lett.* **1965**, *44*, 3969.

# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

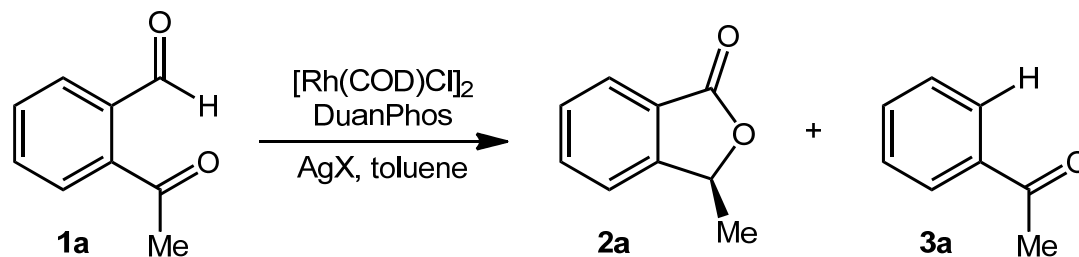


# Rh-Catalyzed Enantioselective Hydroacylation of Ketones



# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

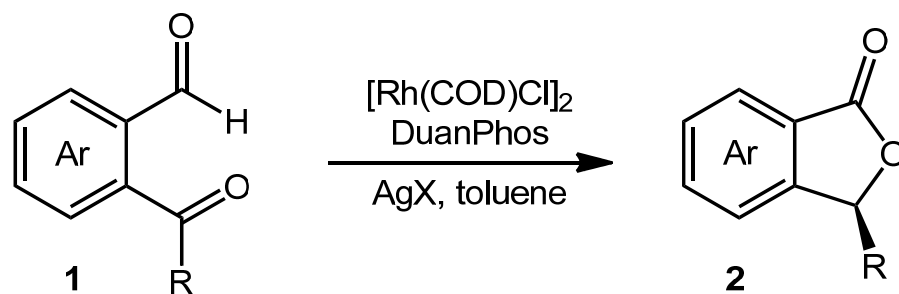
## Counterion effects on reactivity



entry	X	<b>2a</b> (%)	ee (%)	<b>3a</b> (%)	time
1	SbF <sub>6</sub>	30	40	52	1 d
2	BF <sub>4</sub>	76	29	24	1 d
3	OTf	> 95	81	< 5	0.5 h
4	OMs	> 95	91	trace	10 h
<b>5</b>	<b>NO<sub>3</sub></b>	<b>&gt; 95</b>	<b>97</b>	<b>trace</b>	<b>7 h</b>
6	Cl	92	97	trace	3 d



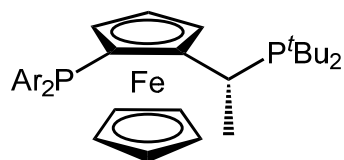
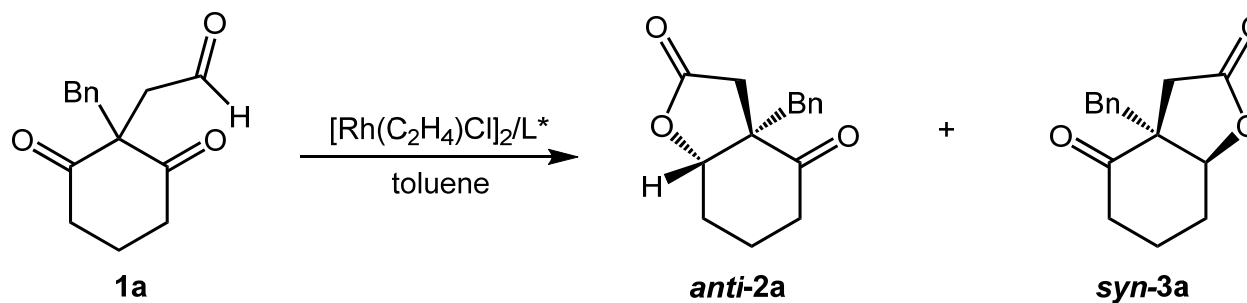
## Rh-Catalyzed Enantioselective Hydroacylation of Ketones



R	X	result
alkyl	$\text{NO}_3$	up to 97% yield, 98% ee
aryl	OMs or OTf	up to 93% yield, 96% ee

# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

## Ligand effect

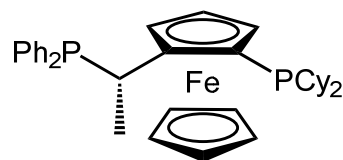


Ar = 4- $\text{CF}_3\text{C}_6\text{H}_4$

Josiphos **L1**

> 20:1 **2a:3a**

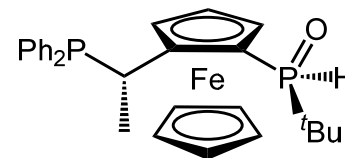
25%, 56% ee



Josiphos **L2**

> 20:1 **2a:3a**

32%, 87% ee



JoSPOphos **L3**

3:1 **2a:3a**

89%, 98% ee/99% ee

# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

## Parameters impacting diastereocontrol

a. Solvent effect:  $[\text{Rh}(\text{C}_2\text{H}_4)\text{Cl}]_2$ , 21 °C

	aprotic				protic		
<b>2a</b>	DME	toluene	DCE	THF	<i>t</i> BuOH	<i>t</i> AmOH	<b>3a</b>
<i>anti</i>	8:1	3:1	1.7:1	1.3:1	1:2	1:3	<i>Syn</i>

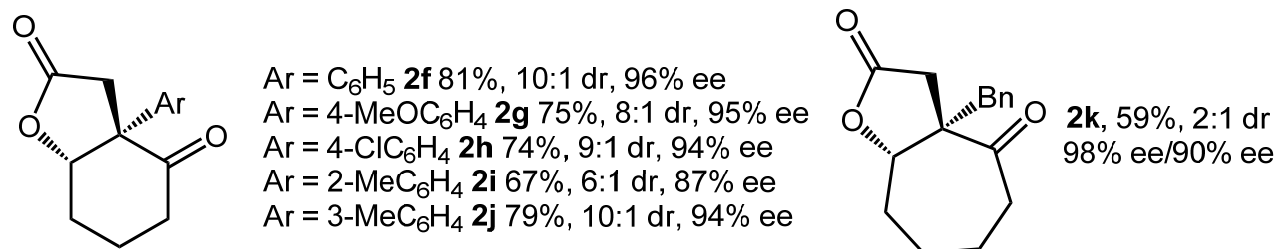
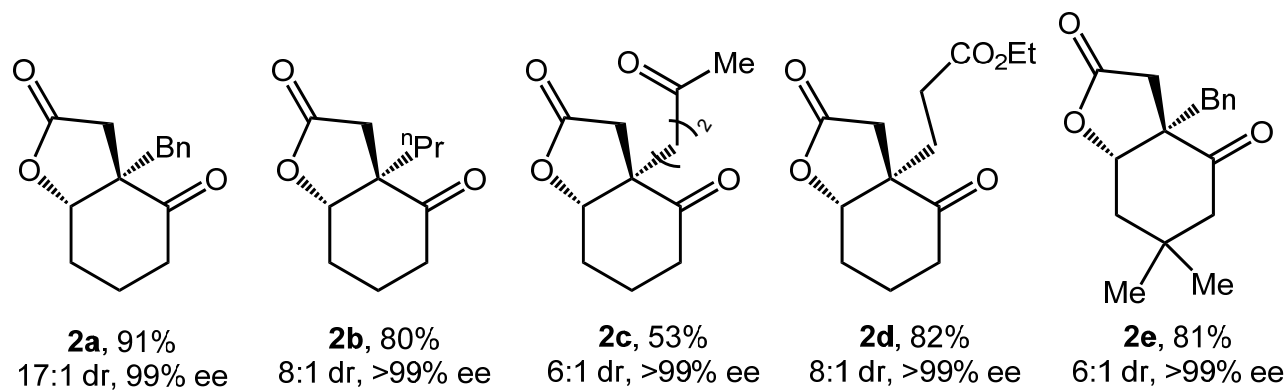
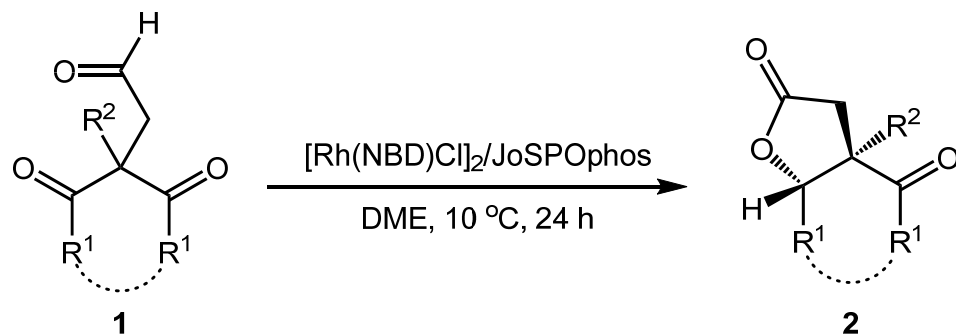
b. Temperature effect:  $[\text{Rh}(\text{C}_2\text{H}_4)_2\text{Cl}]_2$

	in DME			in <i>t</i> AmOH			
<b>2a</b>	10 °C	21 °C	80 °C	21 °C	50 °C	80 °C	<b>3a</b>
<i>anti</i>	←	←	←	→	→	→	<i>Syn</i>
	13:1	8:1	1.5:1	1:3	1:8	1:10	

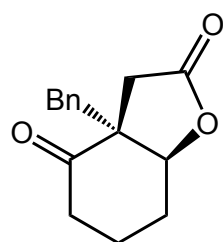
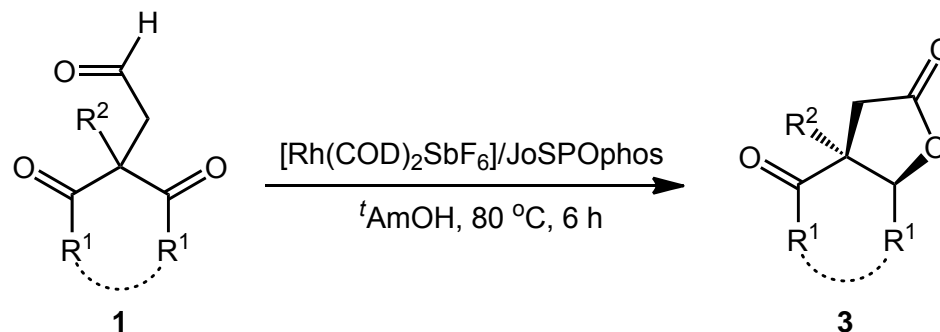
c. Counterion effect:  $[\text{Rh}(\text{COD})_2\text{X}]$  or  $[\text{Rh}(\text{COD})\text{X}]_2$

	in DME, at 10 °C				in <i>t</i> AmOH, at 80 °C				
<b>2a</b>	$\text{Cl}^-$	$\text{Br}^-$	$\text{I}^-$	$\text{SbF}_6^-$	$\text{Cl}^-$	$\text{OTf}^-$	$\text{BF}_4^-$	$\text{SbF}_6^-$	<b>3a</b>
<i>anti</i>	←	←	←	←	→	→	→	→	<i>Syn</i>
	13:1	6:1	3:1	2:1	1:11	1:>20	1:>20	1:>20	

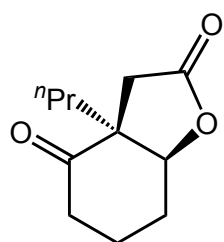
# Rh-Catalyzed Enantioselective Hydroacylation of Ketones



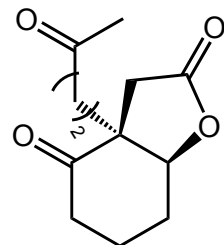
# Rh-Catalyzed Enantioselective Hydroacylation of Ketones



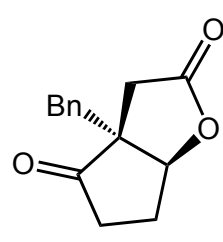
**3a**, 98%  
>20:1 dr, 98% ee



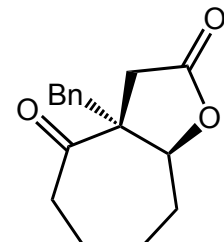
**3b**, 95%  
>20:1 dr, >99% ee



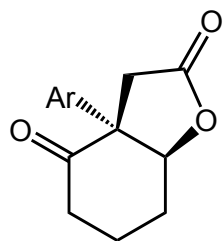
**3c**, 72%  
>20:1 dr, 83% ee



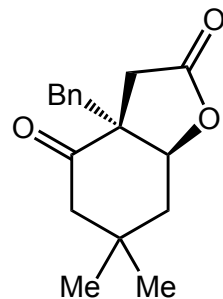
**3d**, 93%  
>20:1 dr, 95% ee



**3e**, 95%  
>20:1 dr, >99% ee



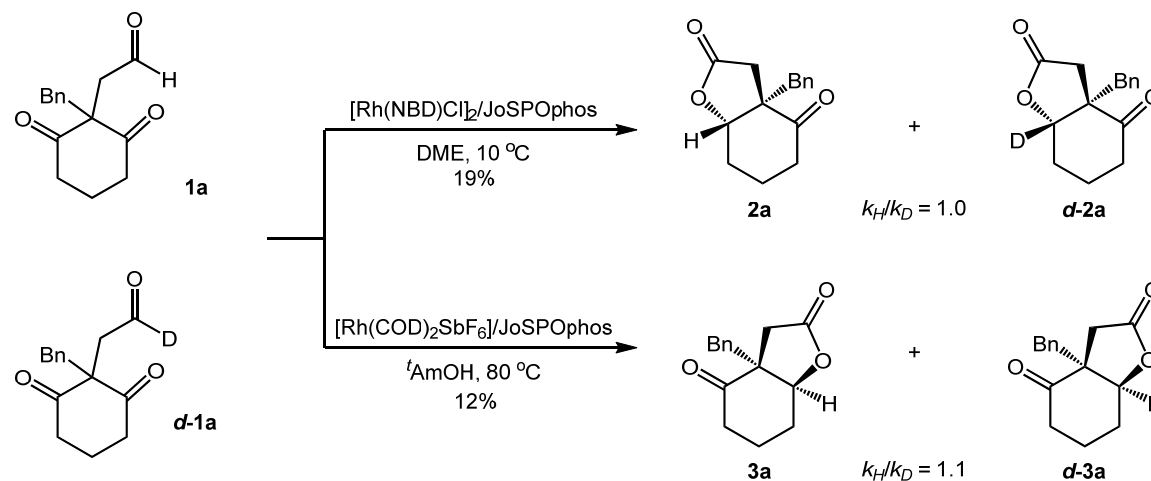
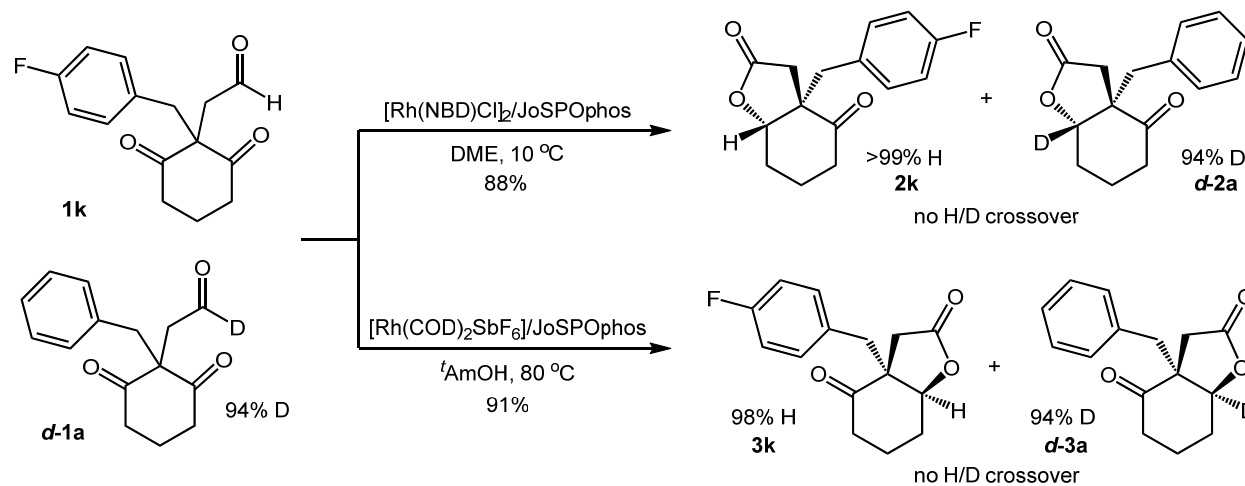
Ar = C<sub>6</sub>H<sub>5</sub> **3f** 95%, >20:1 dr, >99% ee  
Ar = 4-MeOC<sub>6</sub>H<sub>4</sub> **3g** 94%, >20:1 dr, >99% ee  
Ar = 4-ClC<sub>6</sub>H<sub>4</sub> **3h** 95%, >20:1 dr, 94% ee  
Ar = 3-MeC<sub>6</sub>H<sub>4</sub> **3i** 92%, >20:1 dr, 98% ee  
Ar = 2-MeC<sub>6</sub>H<sub>4</sub> **3j** 88%, >20:1 dr, >99% ee



**3k**, 92%  
>20:1 dr, 90% ee

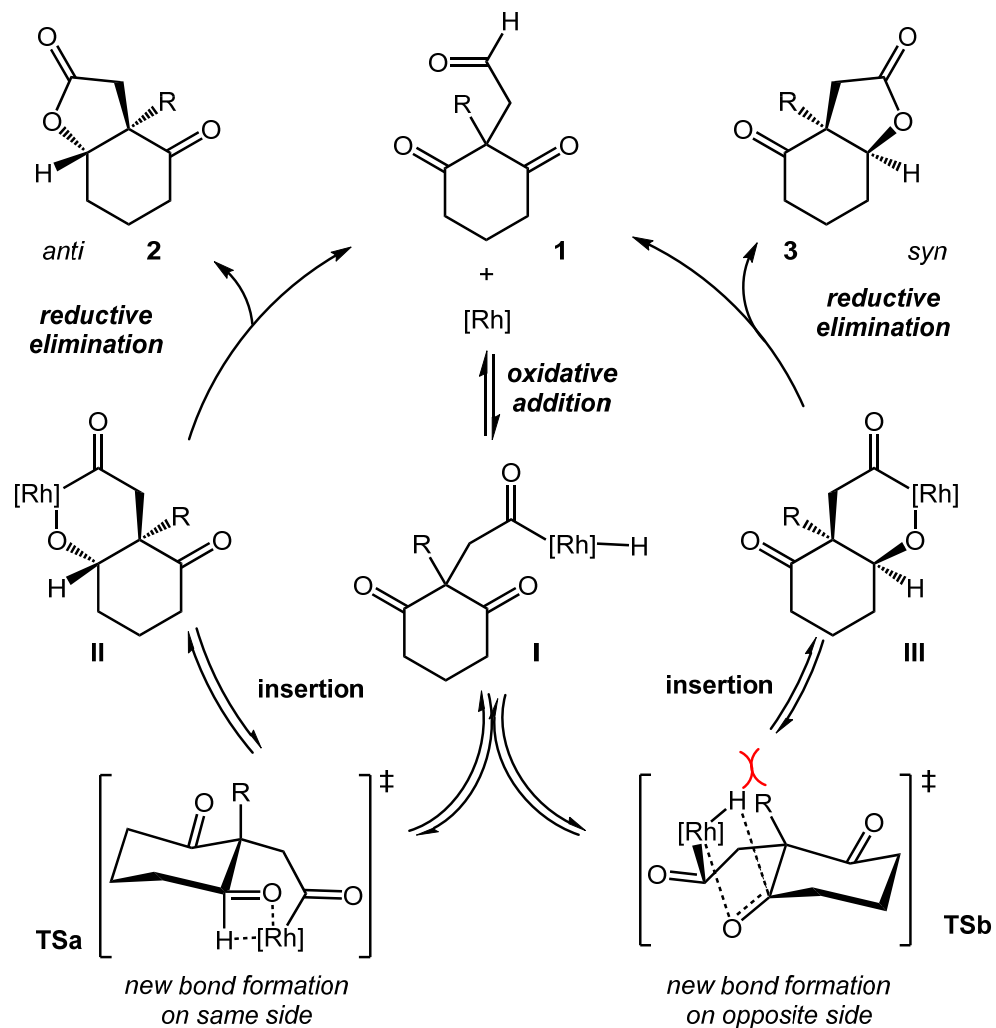
# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

## H/D crossover and kinetic isotope effect experiments



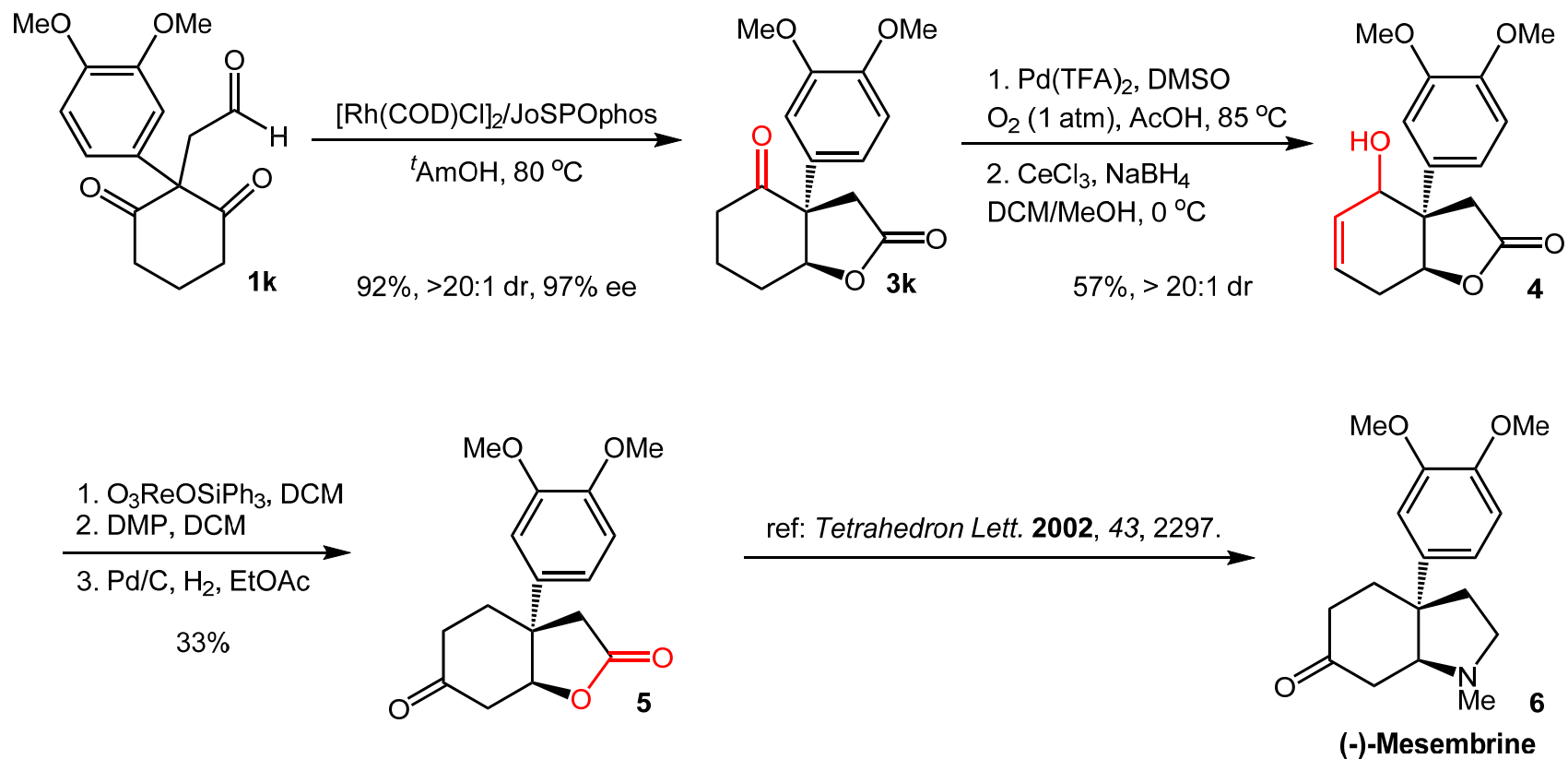
# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

## Mechanism



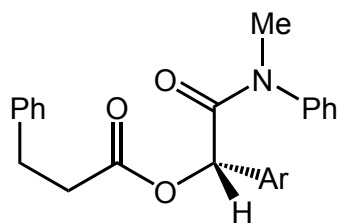
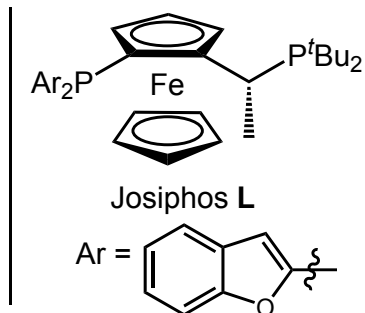
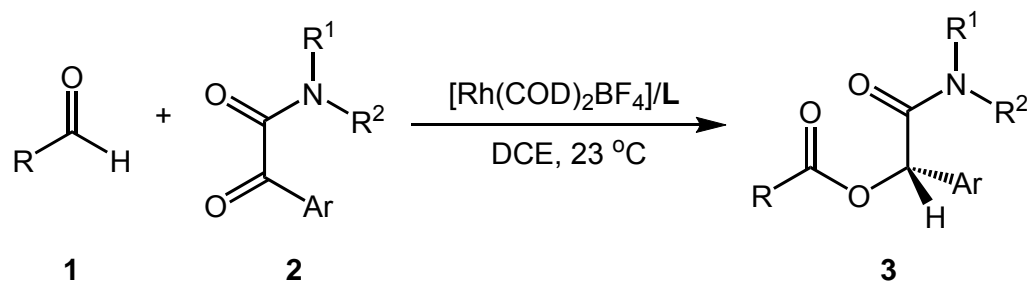
# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

## Transformations

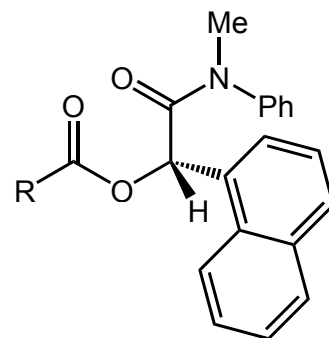




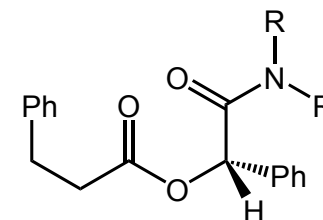
# Rh-Catalyzed Enantioselective Hydroacylation of Ketones



$\text{Ar} = \text{C}_6\text{H}_5$     **3a** 65%, 87% ee  
 $\text{Ar} = 2\text{-MeC}_6\text{H}_4$     **3b** 88%, 95% ee  
 $\text{Ar} = 3\text{-MeC}_6\text{H}_4$     **3c** 62%, 86% ee  
 $\text{Ar} = \text{Mes}$     **3d** 82%, 96% ee  
 $\text{Ar} = 1\text{-naphthyl}$     **3e** 86%, 95% ee

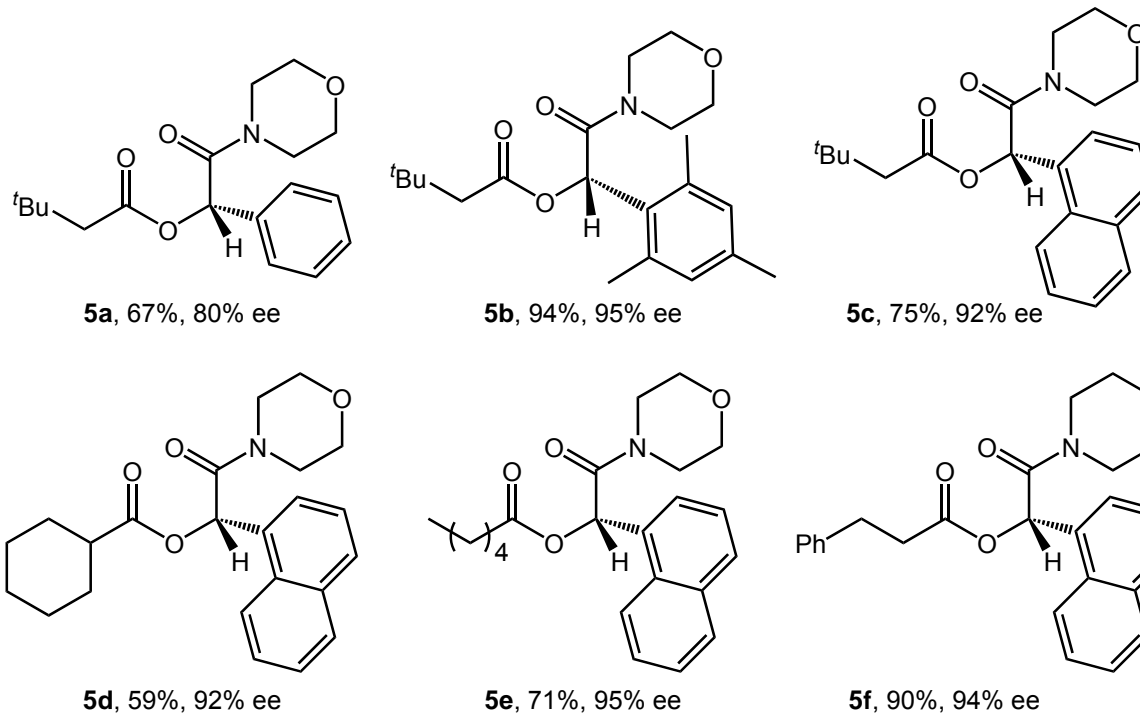
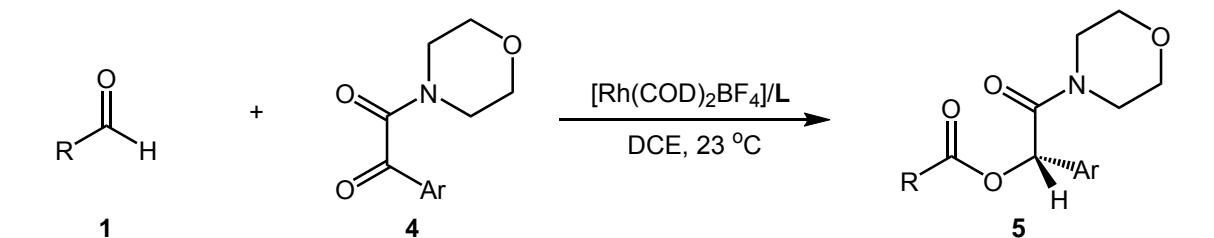


$\text{R} = \textit{i}\text{Bu}$     **3f** 91%, 94% ee  
 $\text{R} = \text{Neopentyl}$     **3g** 91%, 93% ee  
 $\text{R} = \text{Cy}$     **3h** 57%, 90% ee



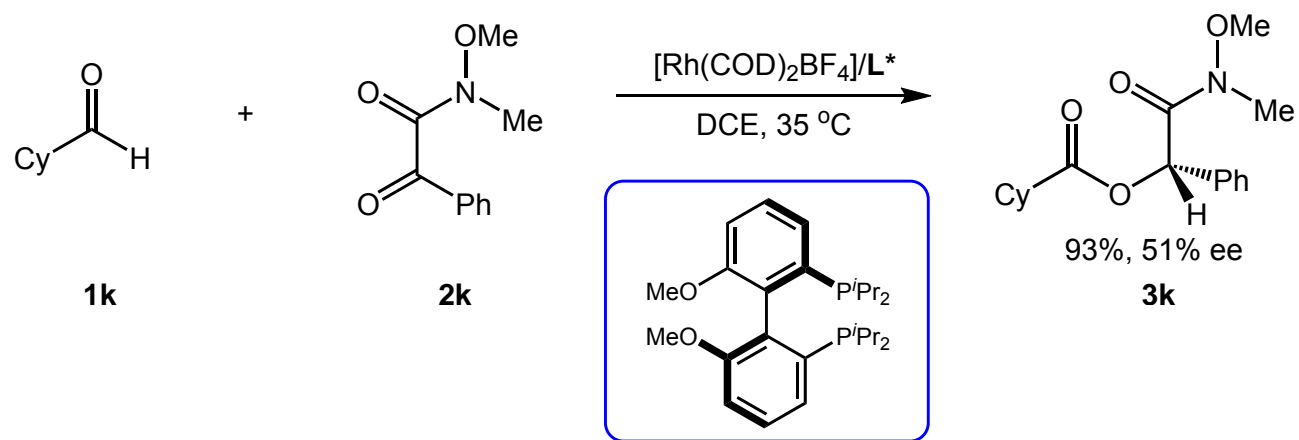
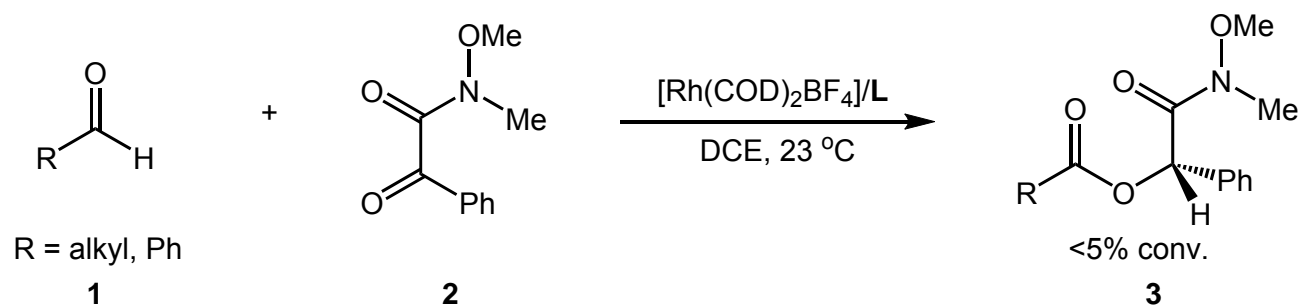
$\text{R} = \text{Ph}$     **3i** 47%, 78% ee  
 $\text{R} = \text{Bn}$     **3j** 46%, 81% ee

# Rh-Catalyzed Enantioselective Hydroacylation of Ketones



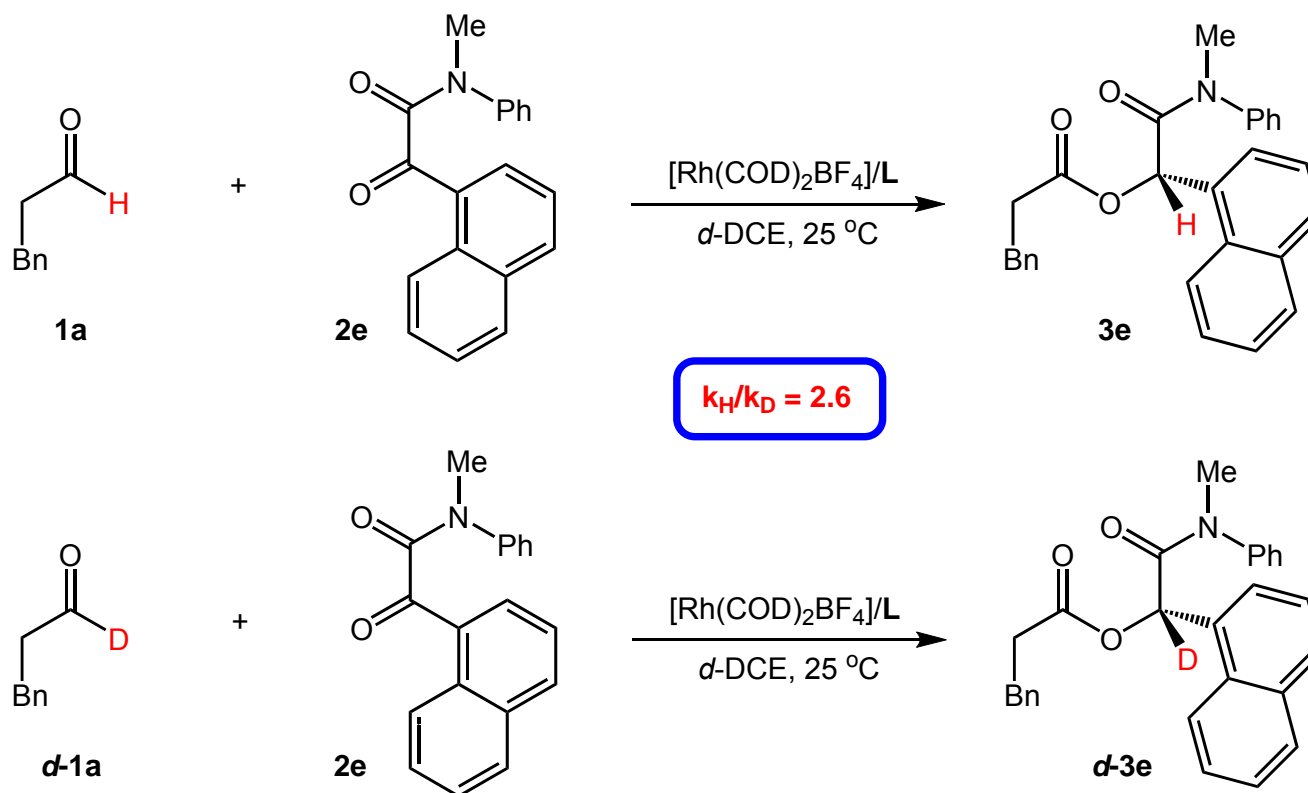
Dong, V. M. *et al.* *J. Am. Chem. Soc.* **2014**, *136*, 9471.

# Rh-Catalyzed Enantioselective Hydroacylation of Ketones



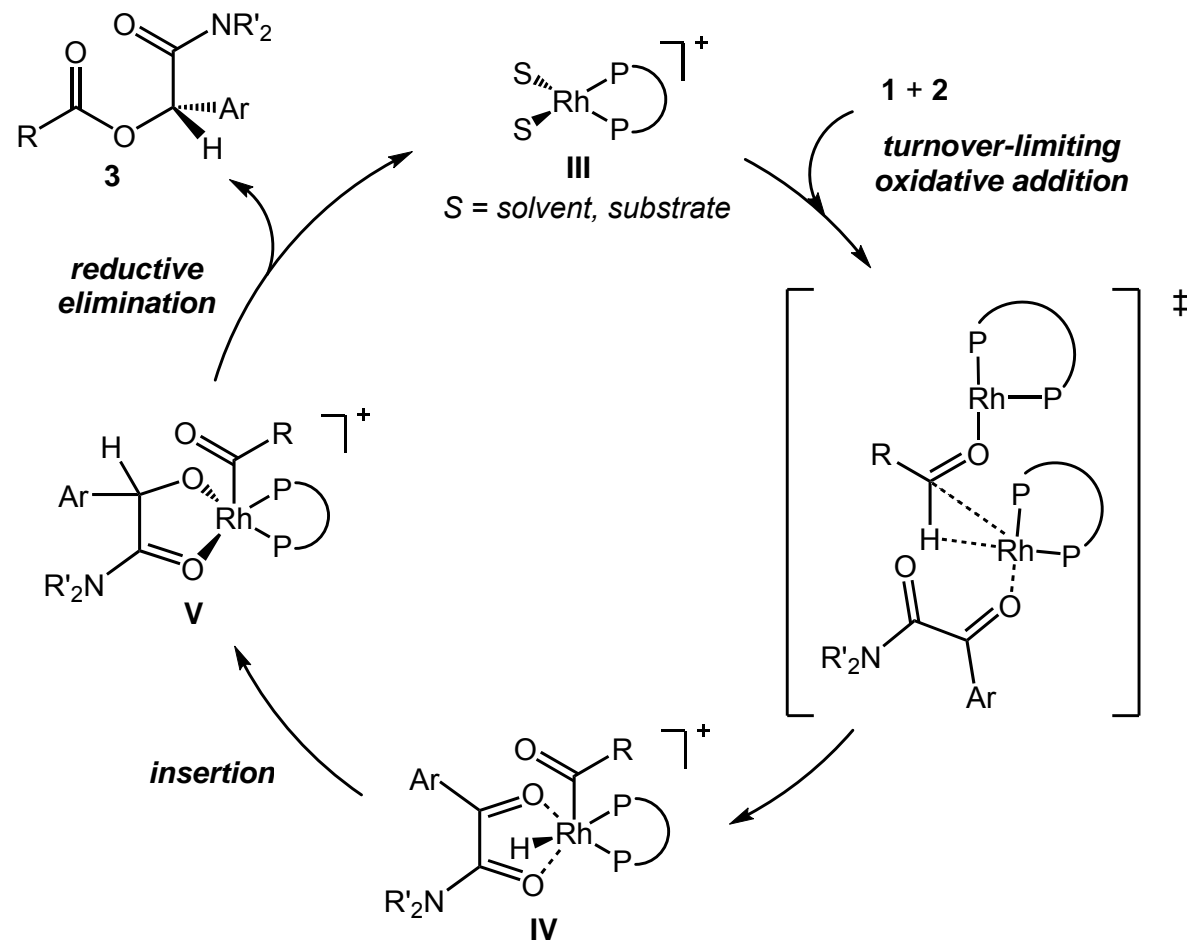
# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

## KIE experiments

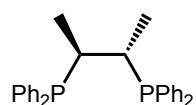
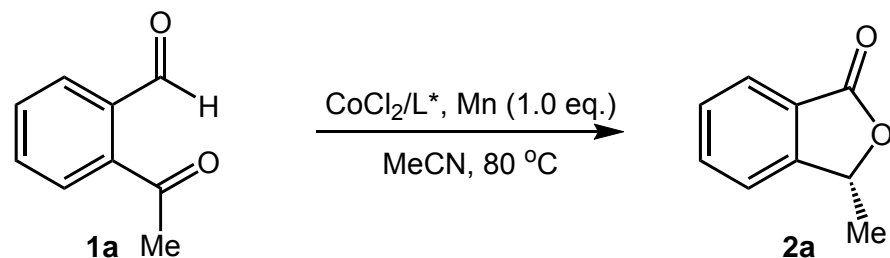


# Rh-Catalyzed Enantioselective Hydroacylation of Ketones

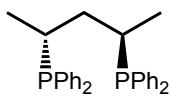
## Mechanism



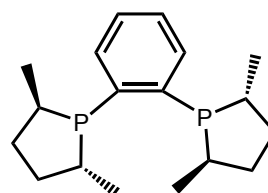
# Co-Catalyzed Enantioselective Hydroacylation of Ketones



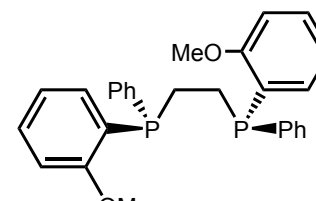
(*S,S*)-Chiraphos  
No reaction



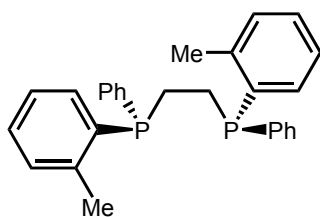
(*R,R*)-BDPP  
15%, --% ee



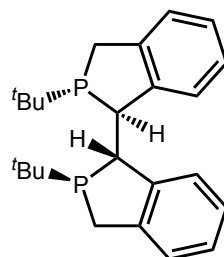
(*R,R*)-Me-DuPhos  
12%, --% ee



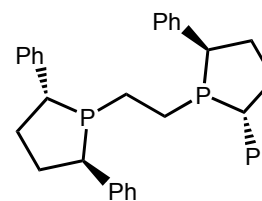
(*R,R*)-DIPAMP  
4%, --% ee



(*R,R*)-Tol-DIPAMP  
73%, 79% ee

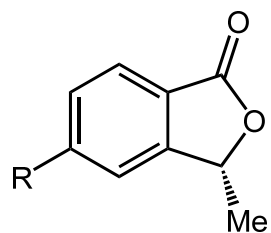
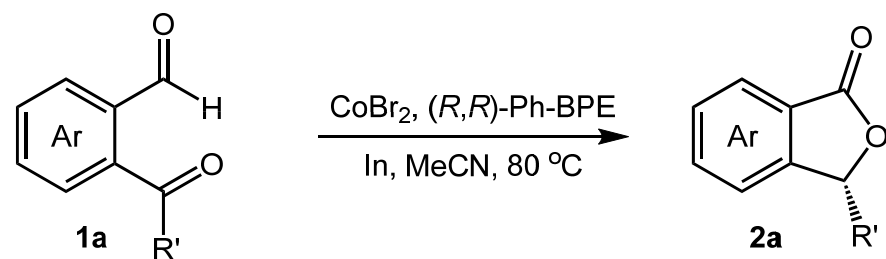


(*S,S,R,R*)-DuanPhos  
37%, 85% ee

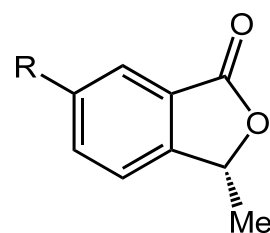


(*R,R*)-Ph-BPE  
Mn (1.0 eq.), 70%, 95% ee  
In (0.2 eq.), 92%, 95% ee (CoBr<sub>2</sub>)

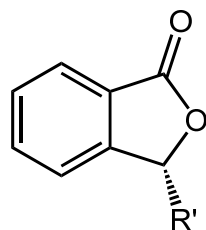
# Co-Catalyzed Enantioselective Hydroacylation of Ketones



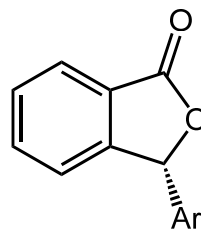
**2a** R = H, 88%, 95% ee  
**2b** R = MeO, 97%, 93% ee  
**2c** R = Me, 82%, 94% ee



**2d** R = Me, 84%, 97% ee  
**2e** R = Cl, 75%, 92% ee  
**2f** R = Br, 61%, 90% ee  
**2g** R = CO<sub>2</sub>Et, 91%, 93% ee



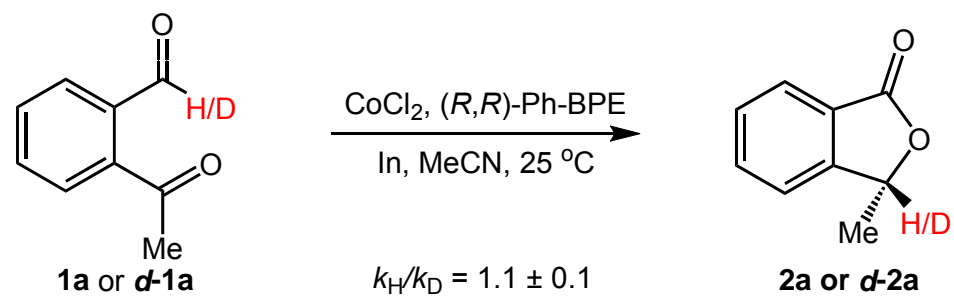
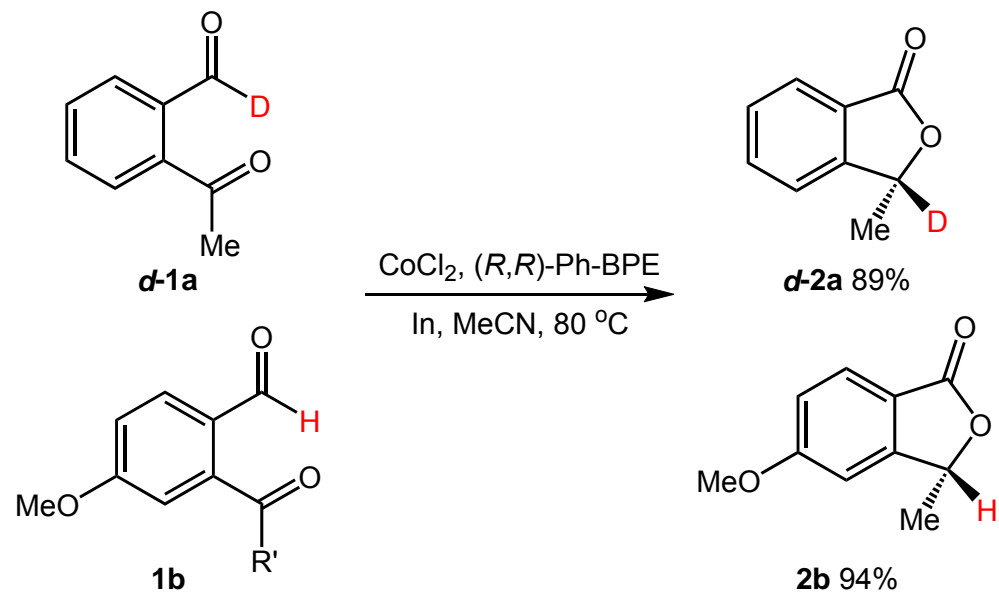
**2h** R' = Et, 93%, 91% ee  
**2i** R' = *i*Pr, 91%, 95% ee



**2j** Ar = C<sub>6</sub>H<sub>5</sub>, 92%, 91% ee  
**2k** Ar = 4-MeOC<sub>6</sub>H<sub>4</sub>, 90%, 98% ee  
**2l** Ar = 4-MeC<sub>6</sub>H<sub>4</sub>, 86%, 92% ee  
**2m** Ar = 4-ClC<sub>6</sub>H<sub>4</sub>, 83%, 88% ee  
**2n** Ar = 3-MeC<sub>6</sub>H<sub>4</sub>, 86%, 90% ee  
**2o** Ar = 2-MeC<sub>6</sub>H<sub>4</sub>, 54%, 82% ee

# Co-Catalyzed Enantioselective Hydroacylation of Ketones

Deuterium-labeling experiments



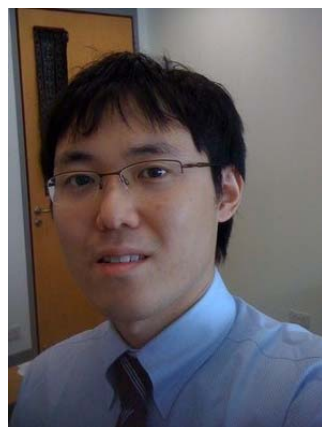
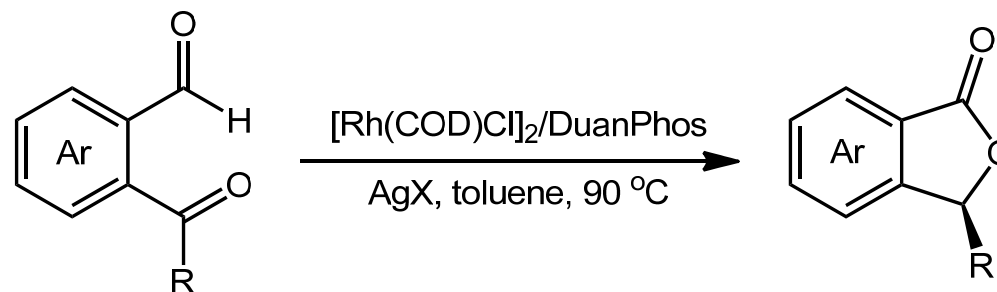


## Summary



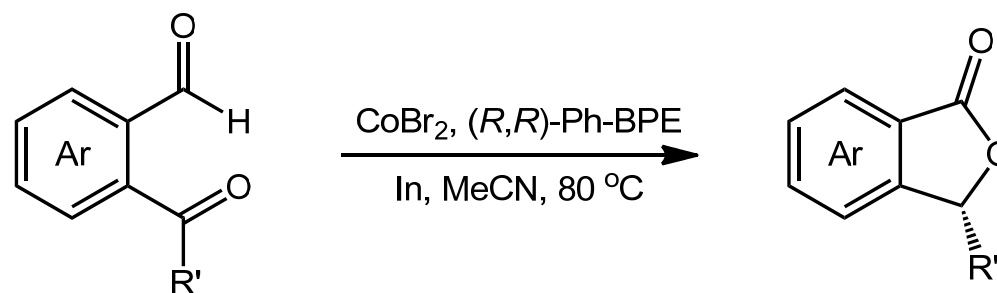
V. M. Dong

### *Rh-Catalyzed Enantioselective Hydroacylation of Ketones*



N. Yoshikai

### *Co-Catalyzed Enantioselective Hydroacylation of Ketones*



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Cyclic architectures comprise a large number of natural products with diverse biological activity. Nature uses enzymes to access both stereoisomers of any bicycle through kinetic control. The use of metal catalysis to construct bicyclic motifs with high enantio- and diastereocontrol thus represents a modern challenge for organic synthesis. Inspired by the occurrence of bicyclic  $\gamma$ -lactones in natural products, we sought an atom-economical strategy to access both the *syn* and *anti* diastereoisomers by ketone hydroacylation. Toward this goal, we herein report the construction of bicyclic  $\gamma$ -lactones featuring the rare activation of aliphatic aldehydes, without competitive decarbonylation.

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By computational studies, we find that the *syn* bicycle **3a** is thermodynamically more stable than the *anti* isomer **2a**. We recognize that the *syn* isomer can undergo a chair flip and thus has more conformational degrees of freedom than its *anti* counterpart. A survey of literature reveals that bond formation to generate related fused bicycles typically occurs to the carbonyl *via* the same side of the reactive tether, suggesting that such additions are rapid and irreversible. In contrast, our hydroacylation strategy enables access to both stereoisomers *via* kinetic control. Under our standard conditions, the *anti* and *syn* products do not interconvert, further supporting the idea that reductive elimination is irreversible. Further kinetic and computational studies are underway to better understand these effects to guide development of future stereodivergent strategies.

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