

Catalytic Enantioselective Reaction of Allenylnitriles with Imines Using Chiral Bis(imidazoline)s Palladium(II) Pincer Complexes

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Abstract: The first highly enantioselective reaction of allenyl nitriles with imines has been developed. Excellent yields and enantioselectivities were observed for the reaction with various imines using chiral Phebim-Pd^{II} complexes. This process offers a simple and efficient synthetic route for various functionalized α -vinylidene- β -aminonitriles and their derivatives.

Allene compounds and their derivatives have been proven to be useful building blocks for the preparation of pharmaceutical targets,^[1] and their unique reactivity renders them versatile intermediates in organic syntheses.^[2] Therefore, their synthetic importance has prompted large interest to develop the asymmetric synthesis of chiral compounds having an allenyl group.^[3] One of the most efficient methods for the preparation of chiral compounds having an allenyl group is the stereoselective addition of allenyl compounds carrying electron-withdrawing groups with electrophiles. There are several reports for the enantioselective addition reaction of allenylesters with imines,^[4,5] carbonyl compounds,^[6] and activated olefins,^[7] however, to the best of our knowledge, there are no reports on the catalytic enantioselective reaction of allenyl nitriles with imines (Figure 1).^[8] On the other hand, we recently developed the activation of nitrile compounds using chiral bis(imidazoline) palladium(II) pincer complexes.^[9,10] In these catalyst systems, palladium catalysts enhanced the acidity of the α -proton in alkynitriles to give palladium ketenimides, which react with imines to give products. We expected that the bis(imidazoline) palladium system could be applied to the reaction of allenyl nitriles with imines, because allenyl nitriles have reasonable acidity ($pK_a = 21.0$ in DMSO), similar to allylnitrile ($pK_a = 20.7$ in DMSO).^[11] Therefore, chiral palladium catalysts activate the acidity of allenyl nitriles, then the carbanion of allenyl nitriles

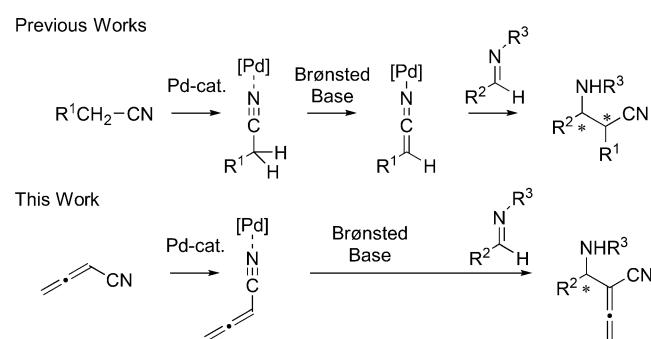


Figure 1. Palladium pincer complexes activate nitrile compounds.

attacks imines to give chiral α -vinylidene- β -aminonitriles. Herein, we report the first highly enantioselective reaction of allenyl nitriles with imines using palladium(II) pincer complexes with chiral bis(imidazoline)s as a chiral Lewis acid catalyst.

The enantioselective reaction of allenyl nitrile **2a** (1.5 equiv) with various imines **1** (1.0 equiv) was carried out by using palladium catalysts **4a-d** (2 mol %) and silver acetylacetone (2 mol %) in THF (Table 1). Although the reaction of *N*-diphenylphosphinoyl imine **1a** (R = DPP) and *N*-Boc imine **1b** with allenyl nitrile **2a** using catalyst **4a** and Ag(acac) did not afford any products (entries 1 and 2), the reaction of *N*-tosyl imine gave product **3c** in 73% yield with 54% ee (entry 3). Furthermore, the reaction with *N*-trimethylsilylethanesulfonyl imine **1d** (R = SES) gave product **3d** in moderate yield with 71% ee (entry 4). In order to improve enantioselectivity, we optimized the structure of bis(imidazoline) catalysts **4b-d** (entries 5–7). As a result, the reaction using catalyst **4d** having an acetyl group (R¹) and 2,4,6-trimethylphenyl group (Ar) emerged as the most suitable catalyst for this reaction giving product **3d** with high enantioselectivity but low yield due to the generation of a by-product by the over-reaction of product with imines (entry 7). To our delight, the reaction at a lower temperature suppressed the generation of by-product to give product **3d** in good yield with high enantioselectivity (entry 8). Finally, the addition of 20 mol % of trimethylsilylalcohol (TMSOH) improved the reactivity and yield of the product (entry 9).

Next, we examined the reaction of **2a** with various imines **1d-p** using **4d**, Ag(acac), and TMSOH (Table 2). The reaction of various imines **1e-j** having an electron-withdrawing group, such as a fluoro, chloro, bromo, and trifluoromethyl group, in the *meta* or *para* position gave products **3e-j** in high yield with high enantioselectivity (63–88%, 99% ee, entries 1–7). The reaction with imines **1k** bearing an elec-

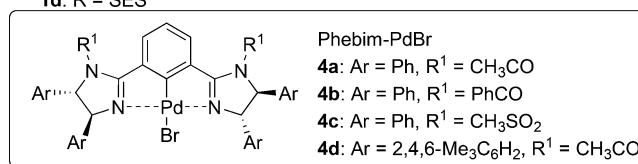
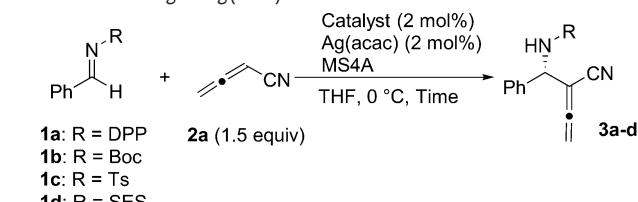
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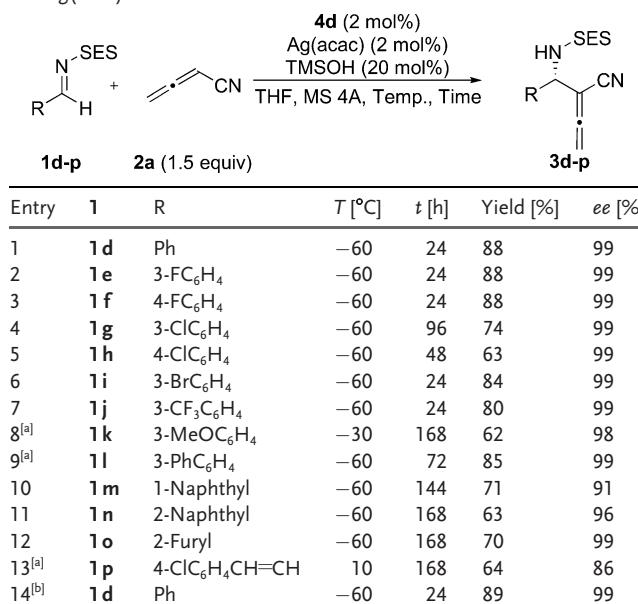
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Table 1: Optimization of the reaction of allenyl nitrile **2a** with various imines **1a–d** using **4d**–Ag(acac).

Entry	1	Catalyst	t [h]	Yield [%]	ee [%]
1	1a	4a	72	—	—
2	1b	4a	72	—	—
3	1c	4a	48	73	54
4	1d	4a	72	49	71
5	1d	4b	72	14	65
6	1d	4c	72	trace	—
7	1d	4d	72	47	91
8 ^[a]	1d	4d	72	82	99
9 ^[a,b]	1d	4d	24	88	99

[a] The reaction was carried out at –60 °C. [b] TMSOH (20 mol %) was added.

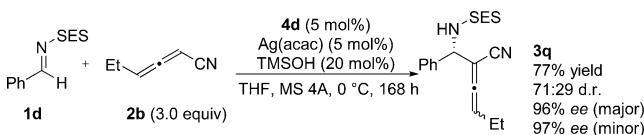
Table 2: The reaction of allenyl nitrile **2a** with various imines **1d–p** using **4d**–Ag(acac).

[a] The reaction was carried out using 5 mol % of **4d** and Ag (acac).

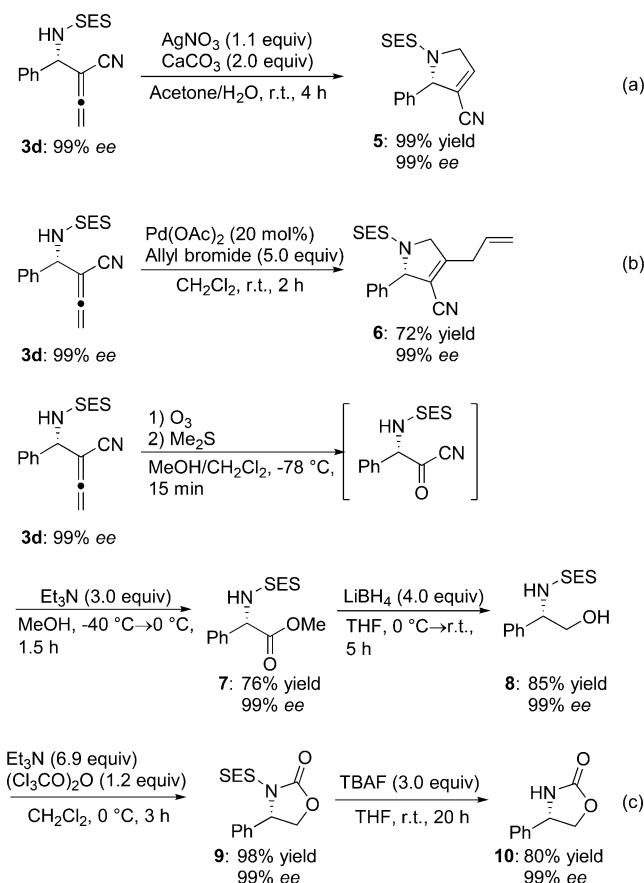
[b] 1.0 g of **1d** was used.

tron-donating methoxy group showed high enantioselectivity but moderate yield (entry 8). The reaction with biphenyl imine **1l** also gave product **3l** in 85% yield with 99% ee (entry 9). Naphthyl and a heteroatom containing imines **1m–o** also gave products **3m–o** with high enantioselectivity (entries 10–12). The reaction of conjugated imine **1p** was

tolerated in this reaction condition and gave product **3p** with good stereoselectivity (entry 13). To test the scalability, a gram-scale reaction was performed using 2 mol % of the catalyst **4d** (entry 14). The reaction of **1d** (1.0 g) with **2a** proceeded to give the product **3d** in 89% yield with 99% ee. Furthermore, the reaction of 3.0 equiv of racemic terminal ethyl-substituted allenyl nitrile **2b** with imine **1d** gave a diastereomer mixture of product **3q** in good yield with high enantioselectivity (Scheme 1).^[12] To the best of our knowledge, these results are first examples of the enantioselective reaction of allenyl nitriles with imines.

**Scheme 1.** Reaction of racemic terminal ethyl-substituted allenyl nitrile **2b** with imine **1d** using **4d**–Ag(acac) catalyst.

We next examined the transformation of α -vinylidene- β -aminonitrile **3d**. The reaction of **3d** using silver(I) nitrate and calcium carbonate in acetone/water gave the intramolecular cyclization product **5** in high yield without the loss of enantiopurity (Scheme 2a).^[13] Palladium acetate catalyzed

**Scheme 2.** Transformation of α -vinylidene- β -aminonitriles to chiral 2,5-dihydro-1*H*-pyrroles **5**, **6** and oxazolidinone **10**.

the coupling-cyclization of **3d** with allyl bromide, and afforded product **6** in moderate yield without racemization (Scheme 2b).^[14] Ozonolysis of **3d** gave acyl cyanide,^[15] which was directly converted into amino acid methyl ester **7** using MeOH and Et₃N.^[16] The reduction of **7** with LiBH₄ afforded amino alcohol **8**,^[17] which was cyclized to oxazolidinone **9**.^[18] Cleavage of the SES group from **9** using TBAF gave **10** without any loss of enantiopurity (Scheme 2c). The absolute configuration of **10** was determined as (*S*), based on the value of the specific rotation from a previous paper.^[19]

The assumed catalytic cycle for the reaction of allenynitrile with imines is shown in Figure 2. The addition of Ag(acac) to palladium complex **4** evokes the exchange reaction of bromide in **4** to acetylacetone (complex **A**). Coordination of the palladium in complex **A** to the cyano

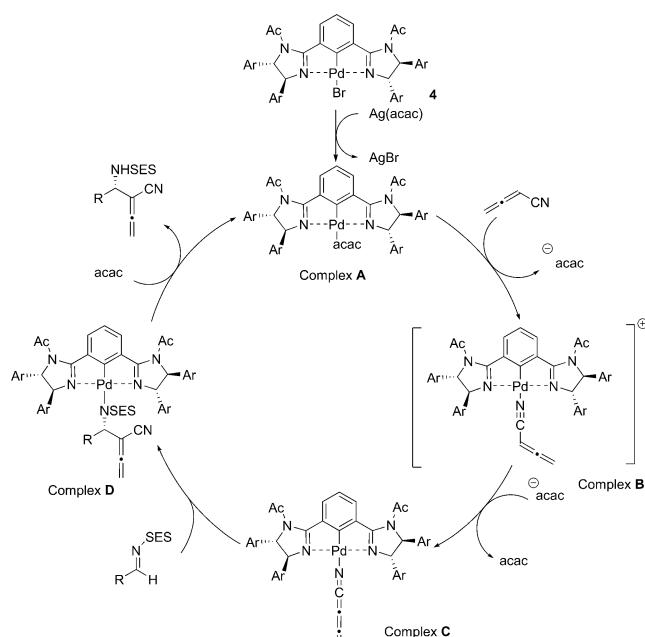
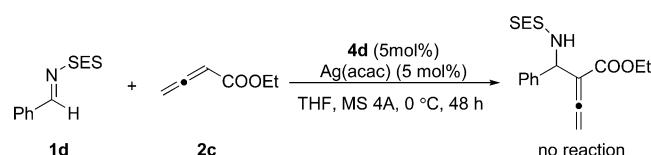


Figure 2. Proposed reaction cycle of the reaction of allenynitrile with imines using **4d**-Ag^I.

group in allenynitrile affords cationic complex **B**, which was observed by ESI-Mass analysis of the mixture of **4d**, Ag(acac) and allenynitrile in a 1:1:1 ratio (cation mode, calcd for C₅₆H₆₀N₅O₂Pd: 940.4, found: 940.3, see the Supporting Information). Complex **C** then reacts with *N*-SES imines to give complex **D**, which subsequently undergoes protonation and decomplexation, resulting in products **3** and regenerated complex **A**.^[20]

In order to clarify the activation of allenynitrile by palladium catalyst, the reaction of allenylester **2c** with **1d** using the **4d** was examined (Scheme 3). However, the reaction of allenylester **2c** with **1d** using **4d**-Ag(acac) catalyst did not give any products. This reactivity is in sharp contrast to the reaction of **2a** with **1d** using **4d**-Ag(acac). This result implied that the palladium pincer complex selectively coordinates and activates the cyano group in **2a**.



Scheme 3. Reaction of allenylester **2c** with imine **1d** using **4d**-Ag.

We examined the optimization of the structure of complex **4d** and allenynitrile **2a** using the Gaussian09^[21] B3LYP/LANL2DZ method (Figure 3). The optimization results

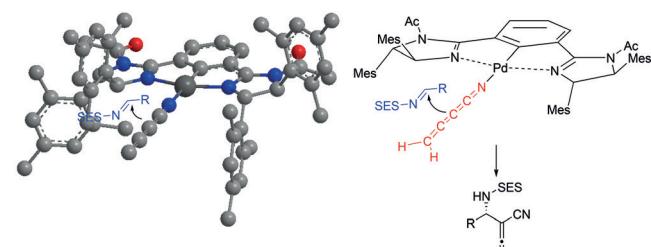


Figure 3. The optimized structure of the complex and the proposed transition state for the reaction of allenynitrile with an imine using **4d**.

showed that the cumulene-type intermediate and nitrogen in allenynitrile coordinate the palladium cation. Based on the calculation result and the absolute configuration of products, the assumed transition state for the reaction of palladium cumulene-type intermediate with imines using **4d** is depicted in Figure 3.

Avoiding steric repulsion between the substituent on imine and the mesityl group in **4d**, the imine approaches the cumulene-type intermediate of allenynitrile to give the (*S*)-isomer.

In conclusion, we developed the first highly enantioselective reaction of allenynitrile with imines using chiral phebim-Pd^{II} catalyst. Various imines can be applicable in the process. This process offers a simple and efficient synthetic route for α -vinylidene- β -aminonitriles and their derivatives. Further studies are in progress to determine the potential of these catalytic systems to other processes.

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Conflict of interest

The authors declare no conflict of interest.

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