

# Literature Report VI

## Palladium/L-Proline-Catalyzed Enantioselective $\alpha$ -Arylative Desymmetrization of Cyclohexanones

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Reporter : Shubo Hu  
Checker : Cong Liu  
Date : 2016-08-29

Jia, Y.-X. *et al.*  
*J. Am. Chem. Soc.* **2016**, 138, 5198.

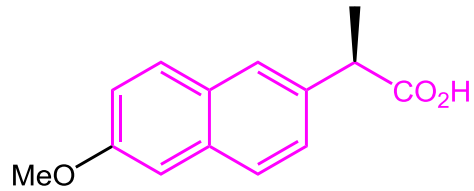
# Contents

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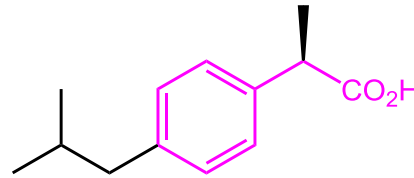
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  - 2** Pd-Mediated Racemic  $\alpha$ -Arylation of Cyclohexanones
  - 3** Pd/L-Proline-Catalyzed Asymmetric  $\alpha$ -Arylation of Cyclohexanones
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# Introduction

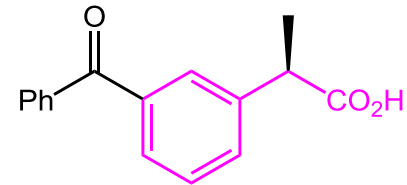
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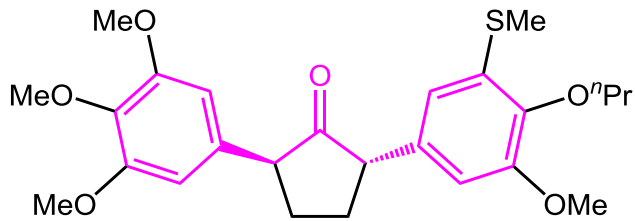
(R)-Naproxen



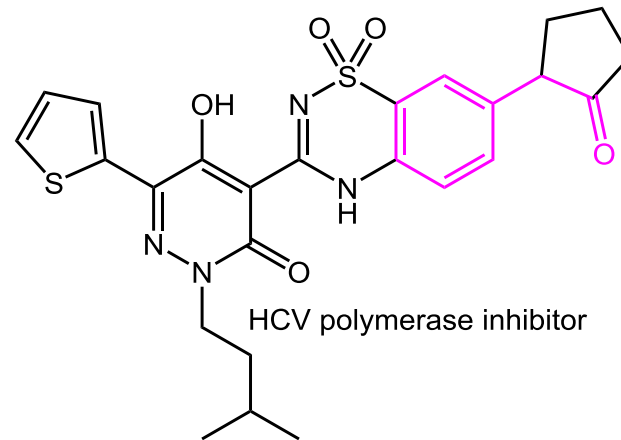
Ibuprofen



Ketoprofen



Platelet activating factor antagonist

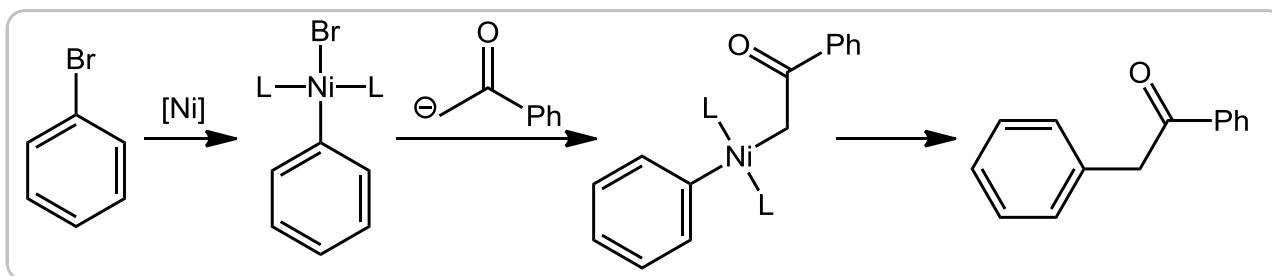


HCV polymerase inhibitor

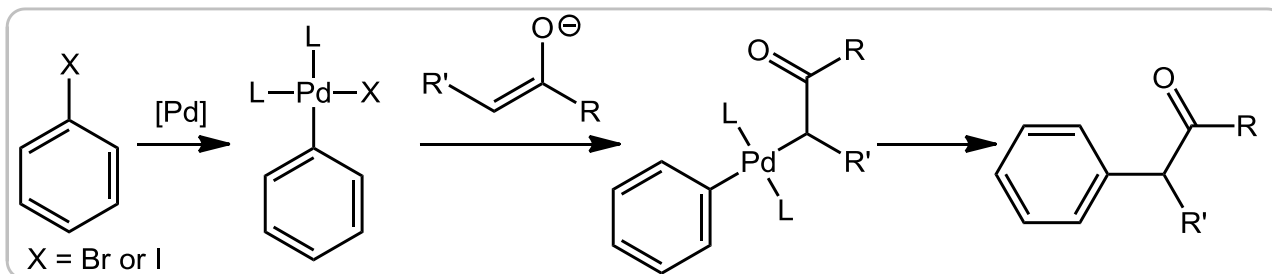
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# Introduction

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Semmelhack, M. F. *et al. Tetrahedron Lett.* **1973**, 14, 4519.



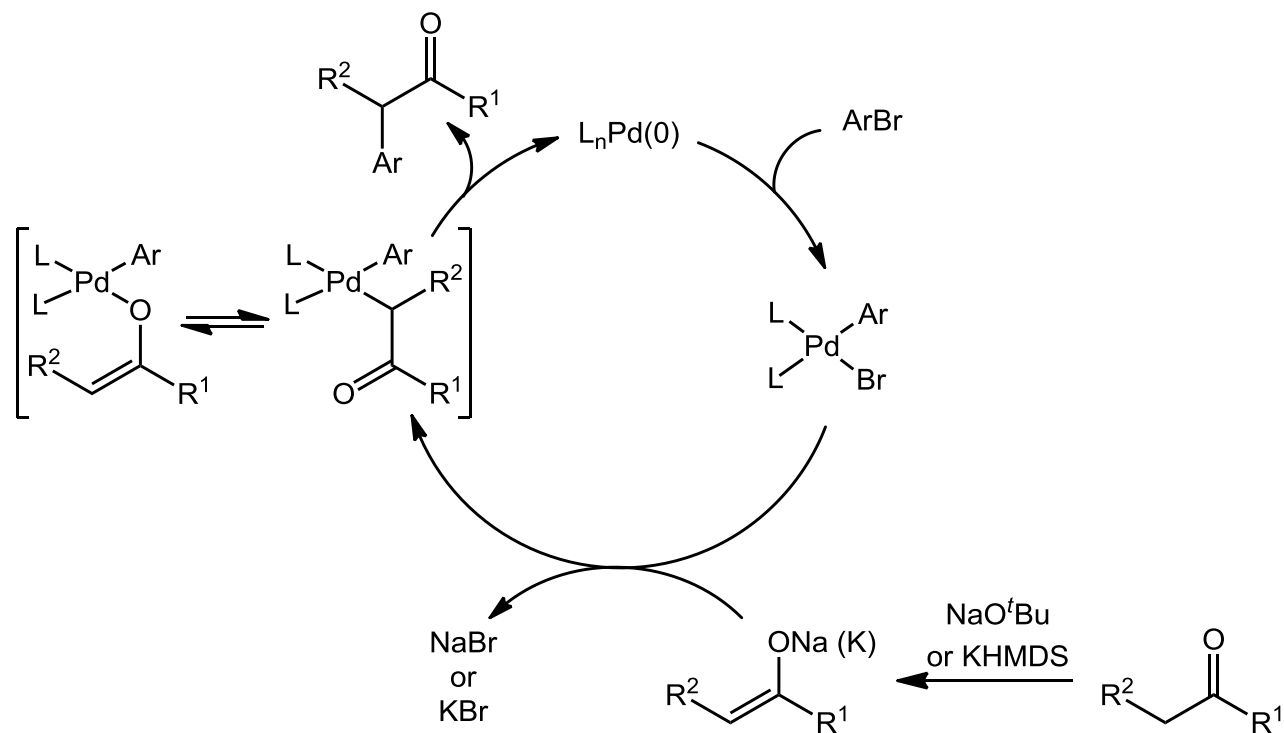
Buchwald, S. L. *et al. J. Am. Chem. Soc.* **1997**, 119, 11108.

Hartwig, J. F. *et al. J. Am. Chem. Soc.* **1997**, 119, 12382.

Miura, M. *et al. Angew. Chem. Int. Ed.* **1997**, 36, 1740.

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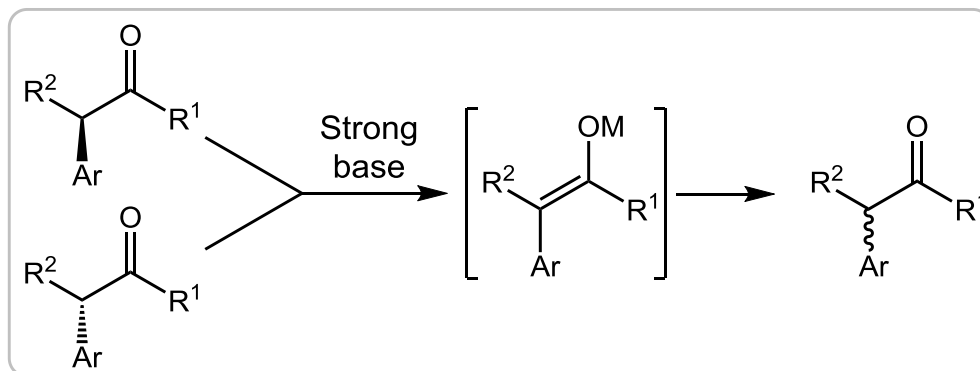
# Introduction



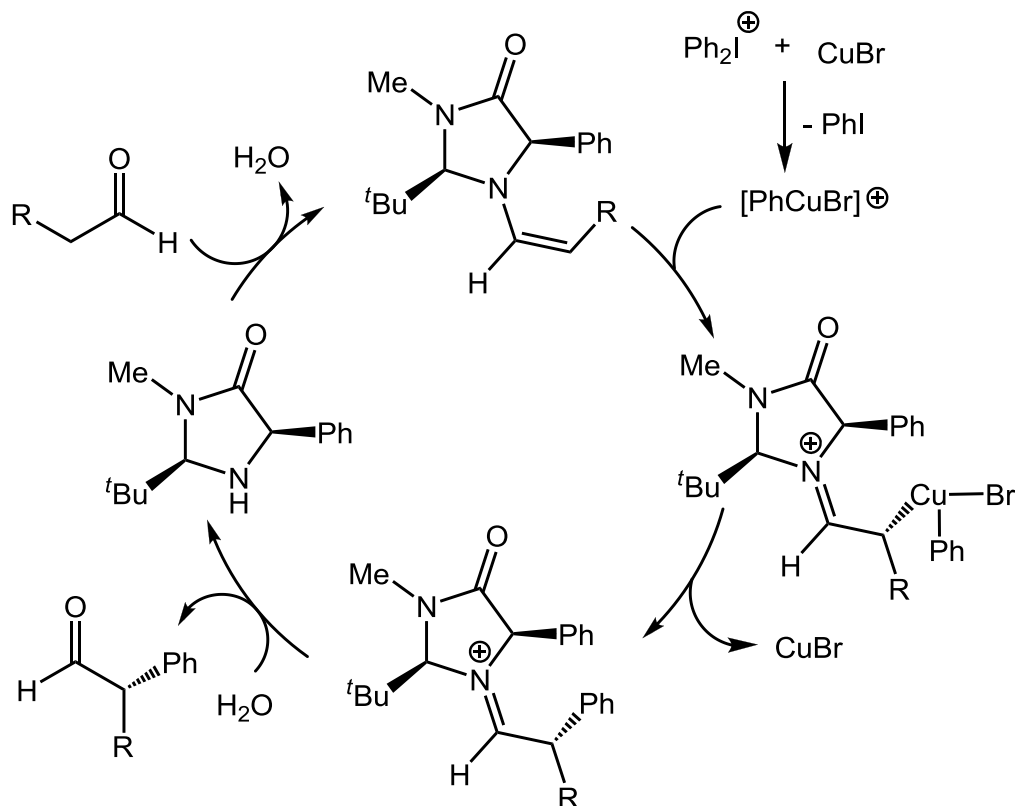
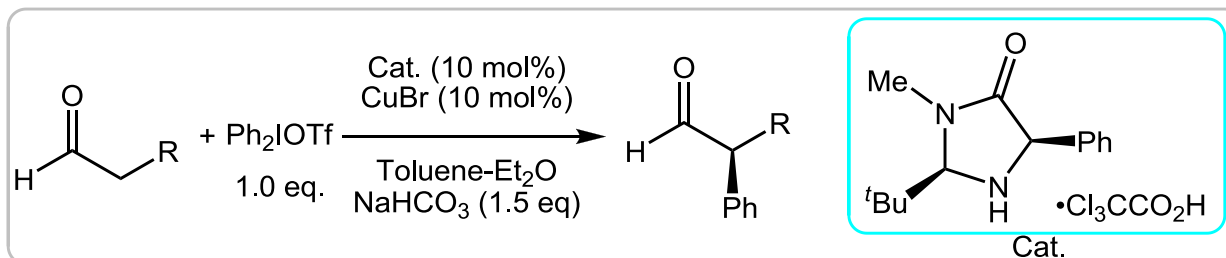
# Introduction

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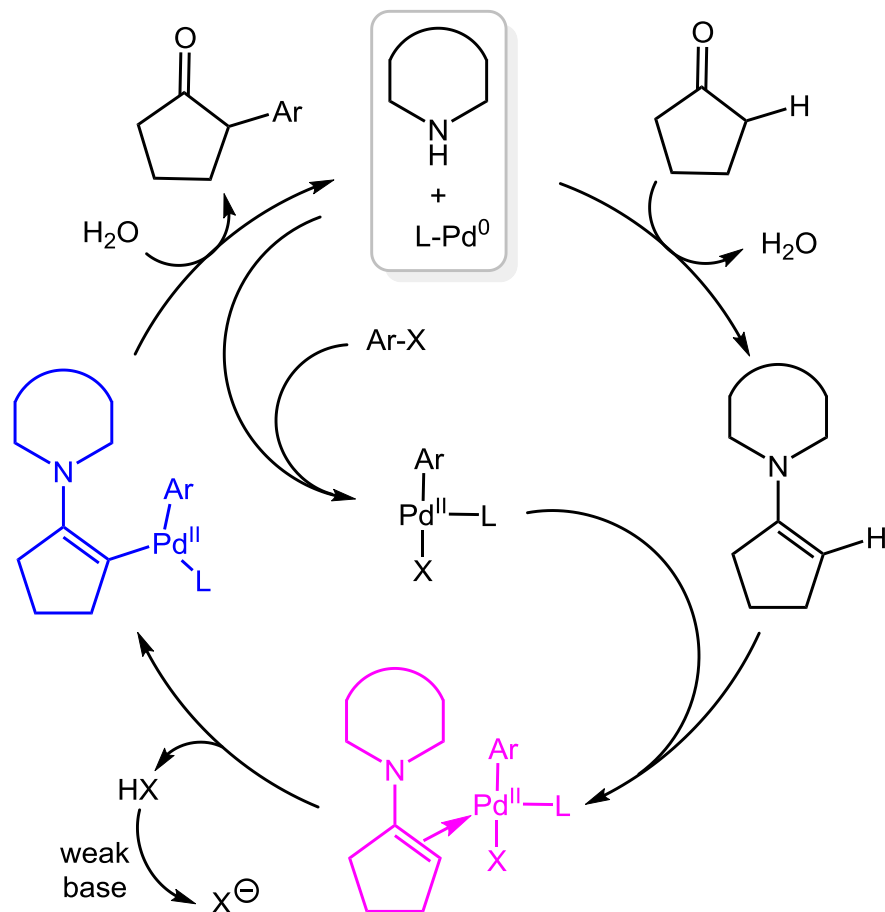
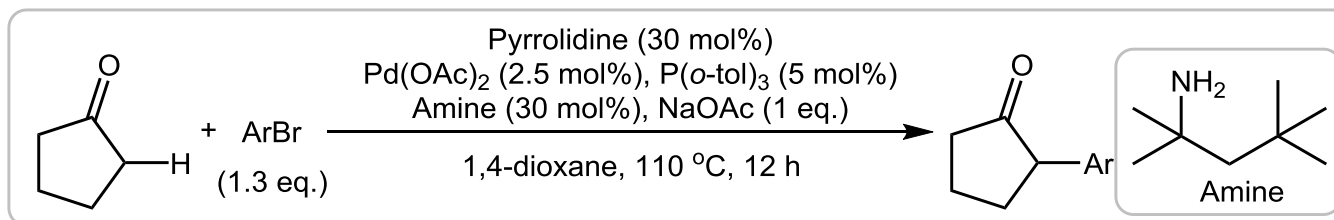
➤ The aforementioned approaches are very limited to the construction of quaternary carbon stereocenters:  $\alpha$ -H atom of the tertiary carbon stereocenter is more acidic and prone to racemization under basic conditions.



# Introduction



# Introduction

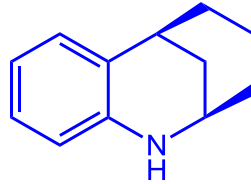


Dong, G. *et al.* *Angew. Chem. Int. Ed.* **2016**, *55*, 2559;  
Kurth, M. J. *et al.* *Angew. Chem. Int. Ed.* **2012**, *51*, 10588.

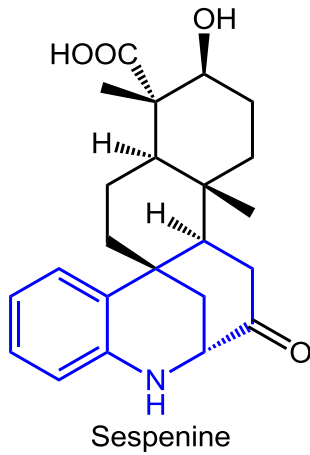


# Introduction

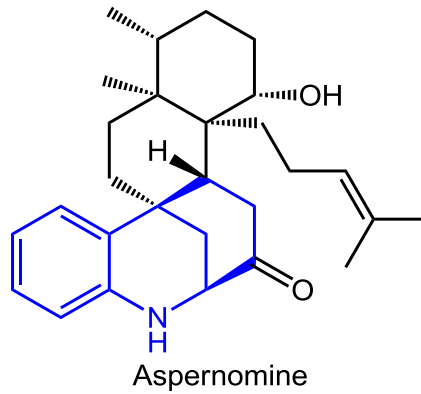
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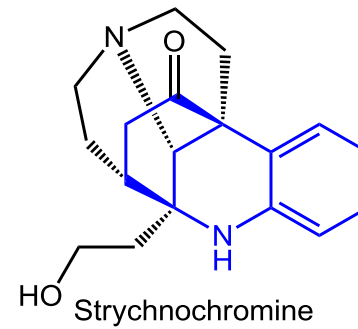
Hexahydro-2,6-methano-  
1-benzazocine



Sespenine



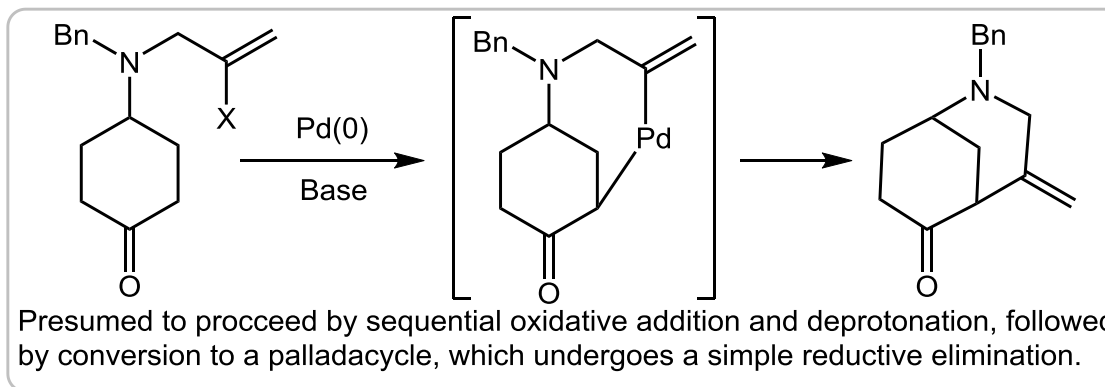
Aspernomine



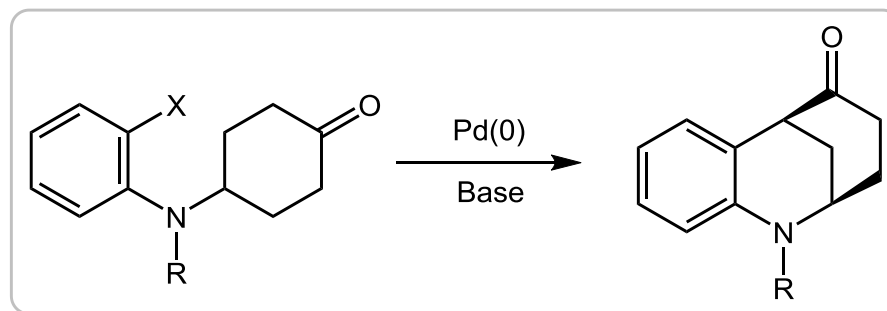
Strychnochromine

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# Racemic $\alpha$ -Arylation of Cyclohexanones

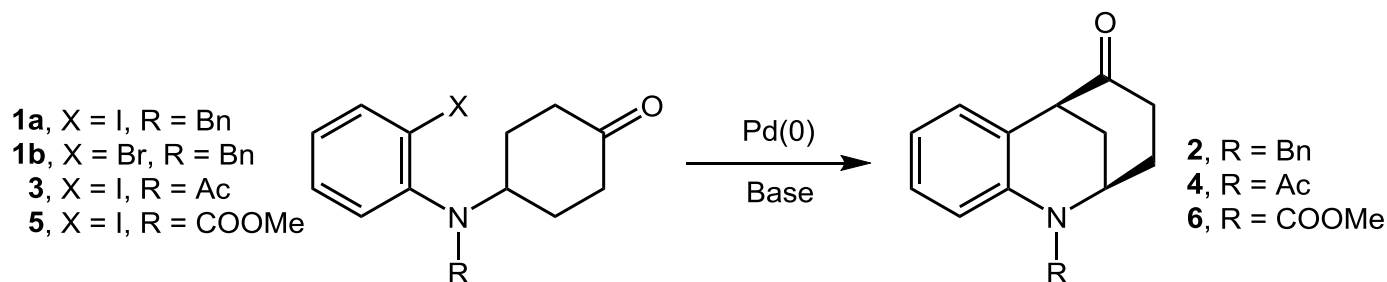


Solé, D.; Bonjoch, J. *et al. Org. Lett.* **2000**, 2, 2225.



Solé, D. *et al. J. Am. Chem. Soc.* **2003**, 125, 1587.

# Optimization of Reaction Conditions



Entry	Substrate	Method <sup>a</sup>	Product	Yield (%) <sup>b</sup>
1	<b>1a</b>	A	<b>2</b>	84
2	<b>1a</b>	B	<b>2</b>	68
3	<b>1a</b>	C <sup>c</sup>	<b>2</b>	76
4	<b>1b</b>	A	<b>2</b>	67
5	<b>1b</b>	B	<b>2</b>	60
6	<b>1b</b>	C <sup>d</sup>	<b>2</b>	78

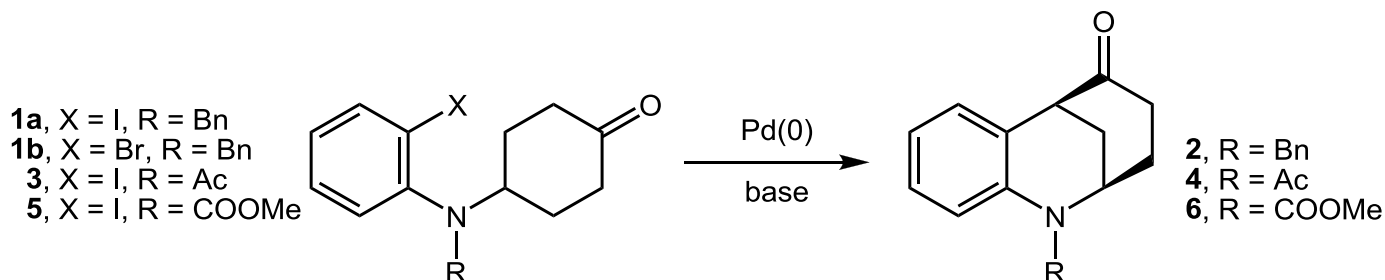
<sup>a</sup> Method A: Pd(PPh<sub>3</sub>)<sub>4</sub> (0.2 eq.), KO<sup>t</sup>Bu (3 eq.), THF, reflux, 3.5 h.

Method B: PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (0.2 eq.), Cs<sub>2</sub>CO<sub>3</sub> (3 eq.), THF, 100-110 °C, 24 h.

Method C: Pd(PPh<sub>3</sub>)<sub>4</sub> (0.2 eq.), K<sub>3</sub>PO<sub>4</sub> (3 eq.), THF, 100-110 °C, 24 h.

<sup>b</sup> The yield refers to pure isolated products. <sup>c</sup> Pd(PPh<sub>3</sub>)<sub>4</sub> (0.1 eq.). <sup>d</sup> 48 h.

# Optimization of Reaction Conditions



Entry	Substrate	Method <sup>a</sup>	Product	Yield (%) <sup>b</sup>
7	<b>3</b>	A	--	--
8	<b>3</b>	B	<b>4</b>	33
9	<b>3</b>	C	<b>4</b>	38
10	<b>5</b>	A	<b>6</b>	48
<b>11</b>	<b>5</b>	<b>B<sup>e</sup></b>	<b>6</b>	<b>92</b>
12	<b>5</b>	C	<b>6</b>	35

<sup>a</sup> Method A: Pd(PPh<sub>3</sub>)<sub>4</sub> (0.2 eq.), KO<sup>t</sup>Bu (3 eq.), THF, reflux, 3.5 h.

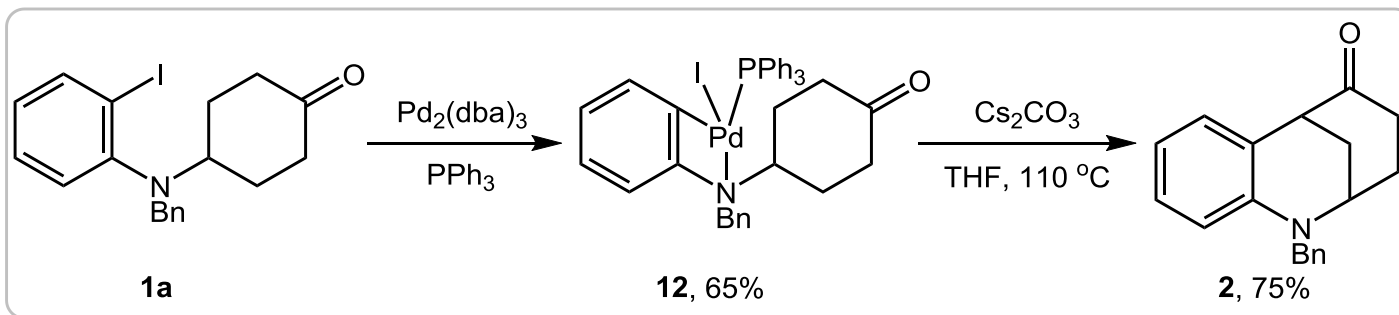
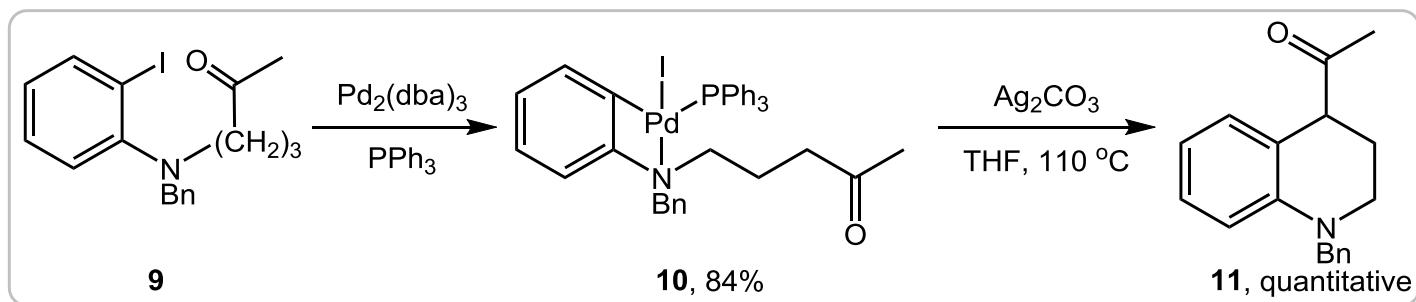
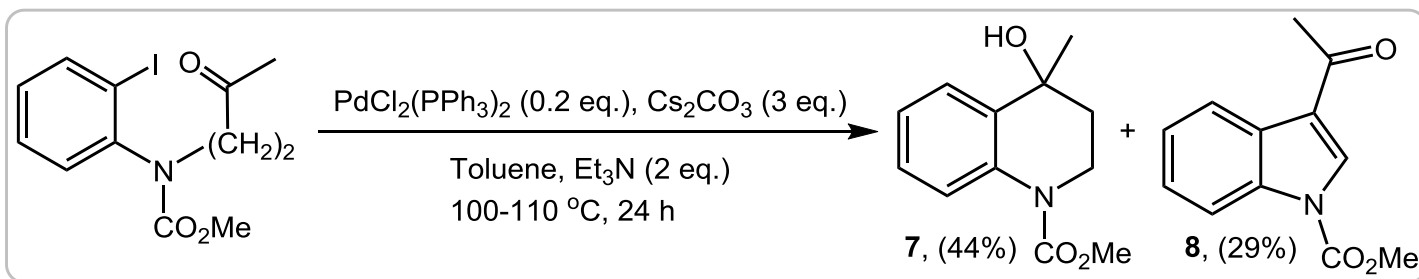
Method B: PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (0.2 eq.), Cs<sub>2</sub>CO<sub>3</sub> (3 eq.), THF, 100-110 °C, 24 h.

Method C: Pd(PPh<sub>3</sub>)<sub>4</sub> (0.2 eq.), K<sub>3</sub>PO<sub>4</sub> (3 eq.), THF, 100-110 °C, 24 h.

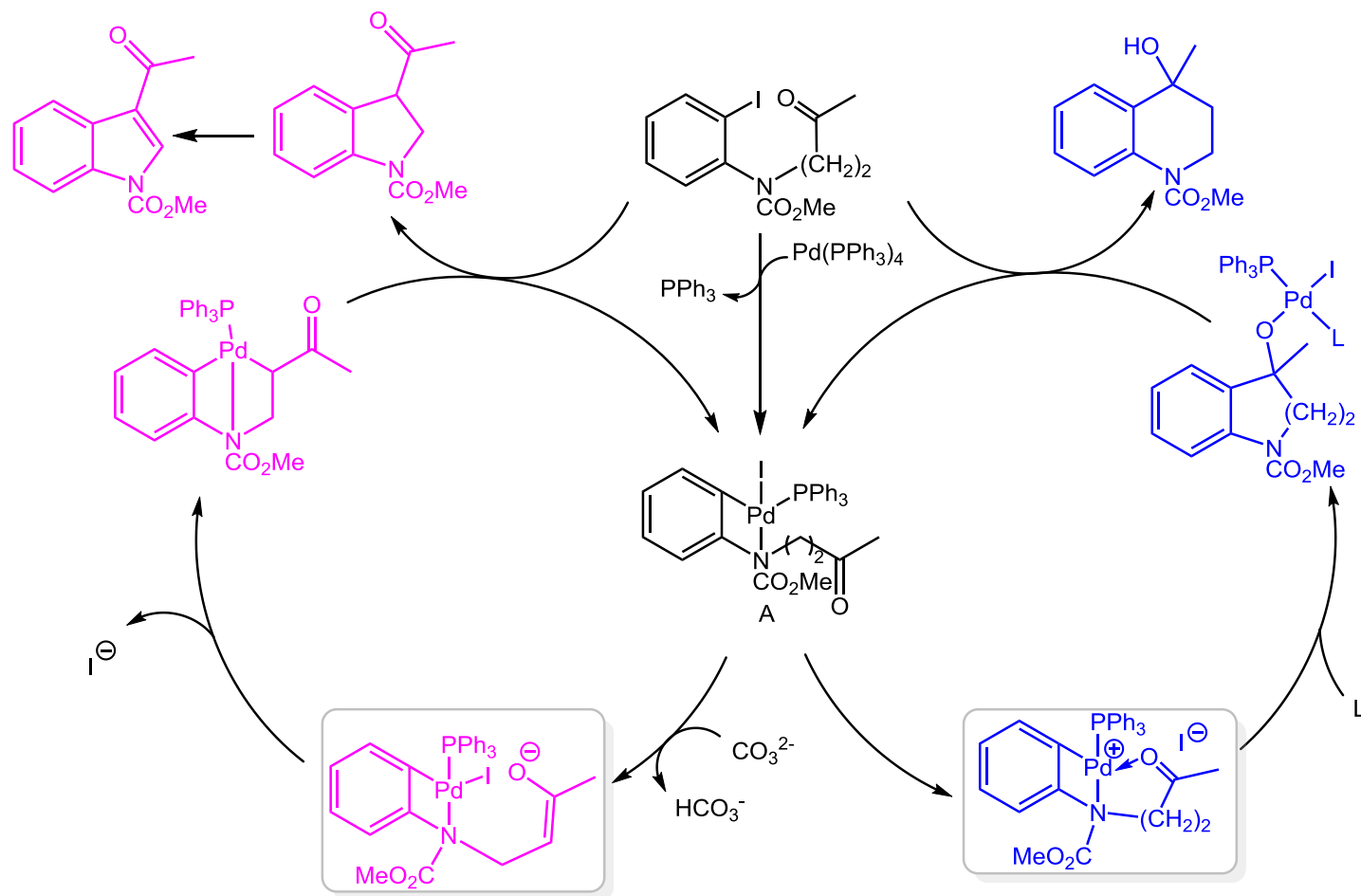
<sup>b</sup> The yield refers to pure isolated products. <sup>c</sup> Pd(PPh<sub>3</sub>)<sub>4</sub> (0.1 eq.). <sup>d</sup> 48 h.

<sup>e</sup> PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (0.3 eq.), 48 h.

# Substrate Scope

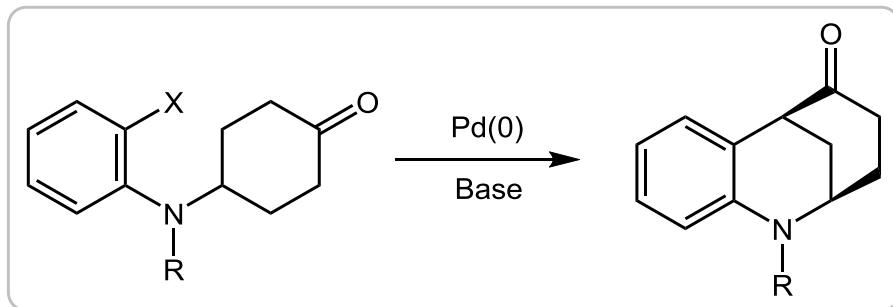


# Racemic $\alpha$ -Arylation of Cyclohexanones



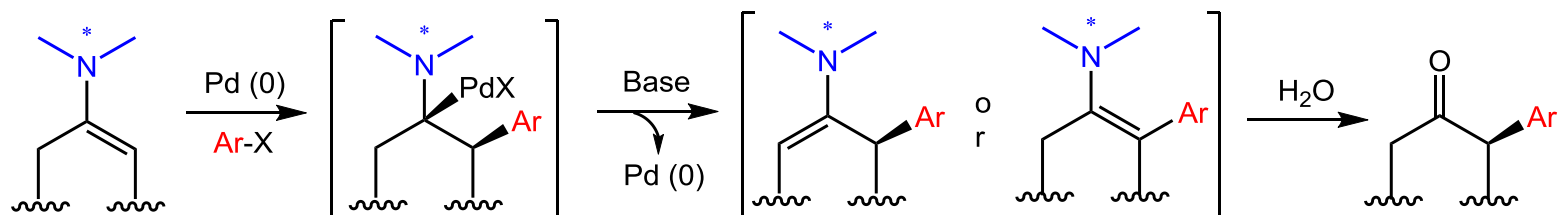
# Asymmetric $\alpha$ -Arylation of Cyclohexanones

## Racemic $\alpha$ -Arylation of Cyclohexanones

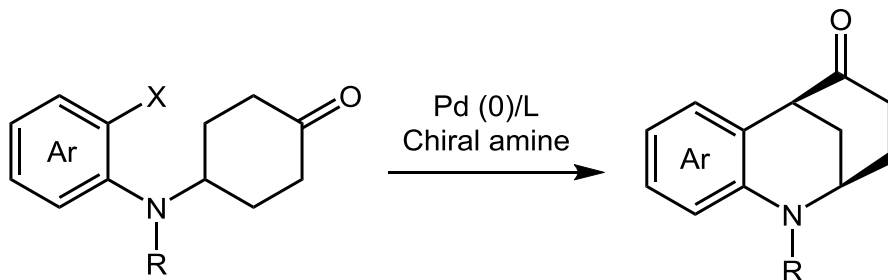


Solé, D. et al. *J. Am. Chem. Soc.* **2003**, 125, 1587.

## a) Proposed $\alpha$ -arylation of ketone via palladium/enamine catalysis

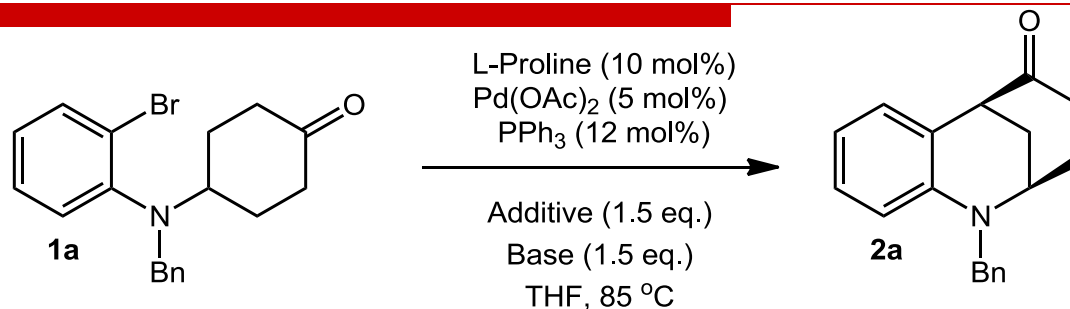


## b) $\alpha$ -Arylative desymmetrization of cyclohexanone



Jia, Y.-X. et al. *J. Am. Chem. Soc.* **2016**, 138, 5198.

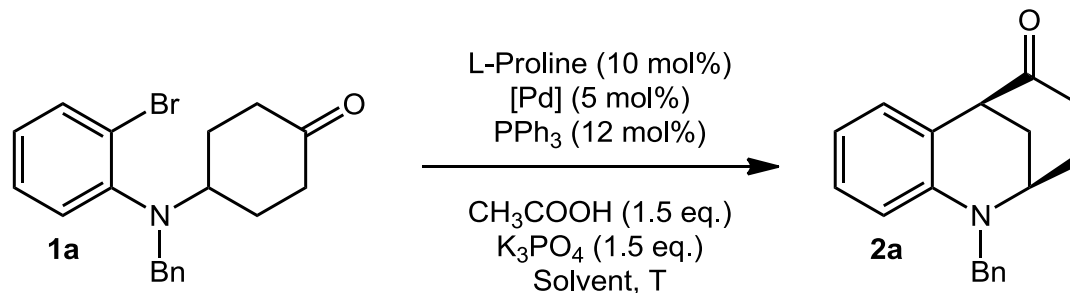
# Optimization of Reaction Conditions



Entry	Base	Additive	Yield (%)	Ee (%)
1	Cs <sub>2</sub> CO <sub>3</sub>	--	60	23
2	NaO <sup>t</sup> Bu	--	71	<10
3	K <sub>3</sub> PO <sub>4</sub>	--	60	60
4	K <sub>2</sub> CO <sub>3</sub>	--	NR	--
5	K <sub>3</sub> PO <sub>4</sub>	4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CO <sub>2</sub> H	60	73
6	K <sub>3</sub> PO <sub>4</sub>	2-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CO <sub>2</sub> H	20	70
7	K <sub>3</sub> PO <sub>4</sub>	2-MeOC <sub>6</sub> H <sub>4</sub> CO <sub>2</sub> H	40	72
8	K <sub>3</sub> PO <sub>4</sub>	3-ClC <sub>6</sub> H <sub>4</sub> CO <sub>2</sub> H	87	73
9	K <sub>3</sub> PO <sub>4</sub>	CF <sub>3</sub> COOH	79	81
10	K <sub>3</sub> PO <sub>4</sub>	CH <sub>3</sub> COOH	95	76



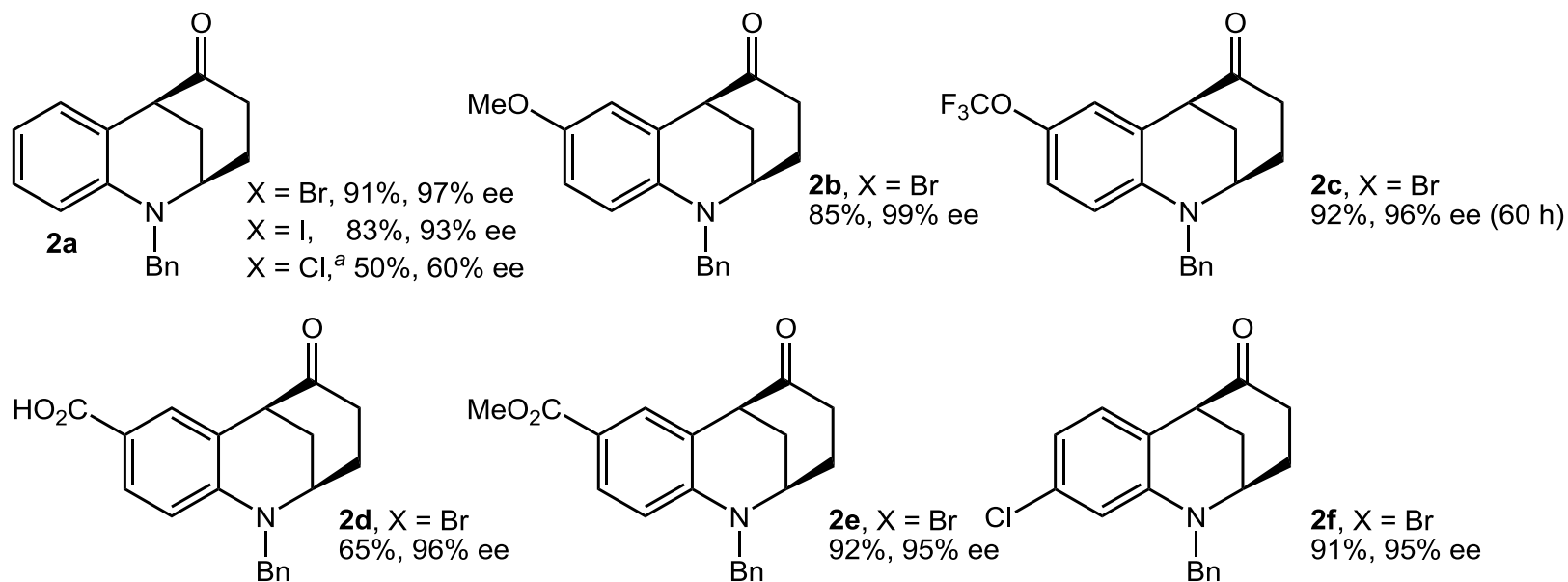
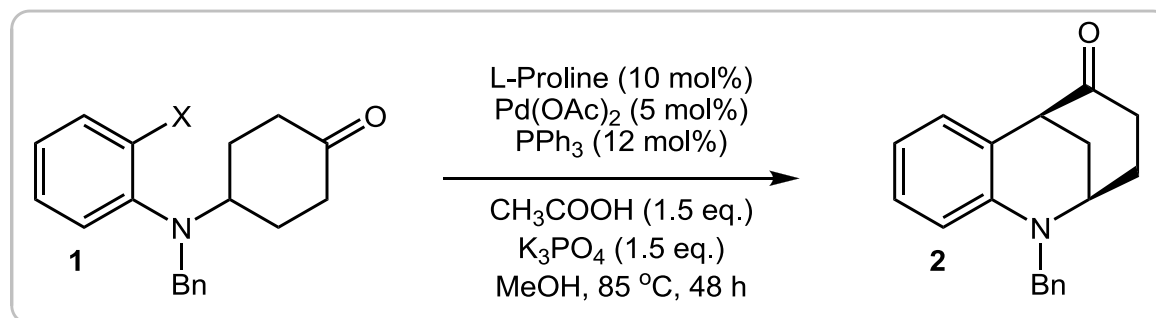
# Optimization of Reaction Conditions



Entry <sup>a</sup>	Solvent	[Pd]	T (°C)	Yield (%) <sup>b</sup>	Ee (%) <sup>c</sup>
11	DMSO	Pd(OAc) <sub>2</sub>	85	10	ND
12	Toluene	Pd(OAc) <sub>2</sub>	85	70	20
13	MeOH	Pd(OAc) <sub>2</sub>	85	91	97
14	MeOH	Pd(dba) <sub>2</sub>	85	70	63
15	MeOH	Pd(PPh <sub>3</sub> ) <sub>4</sub>	85	50	76
16 <sup>d</sup>	MeOH	Pd(OAc) <sub>2</sub>	70	31	ND

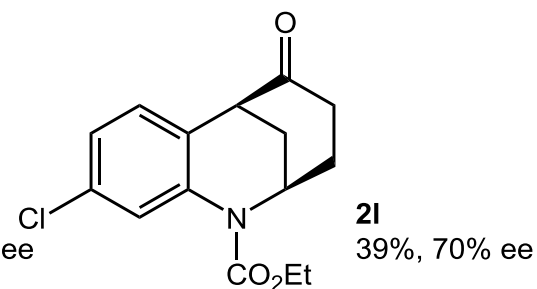
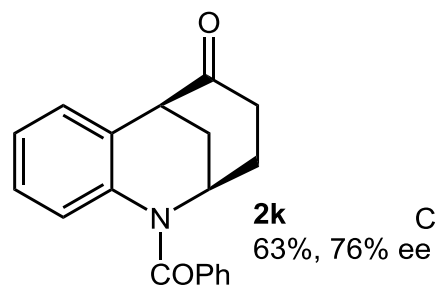
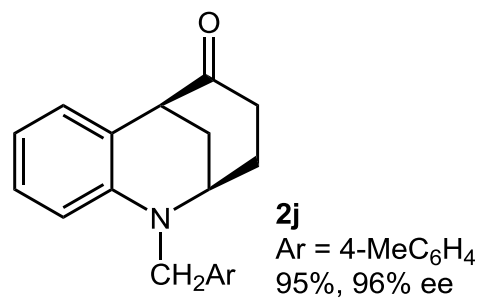
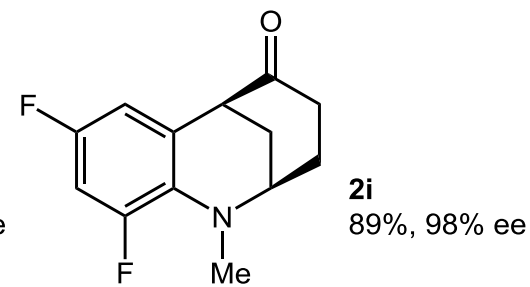
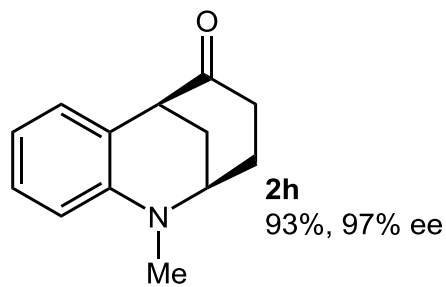
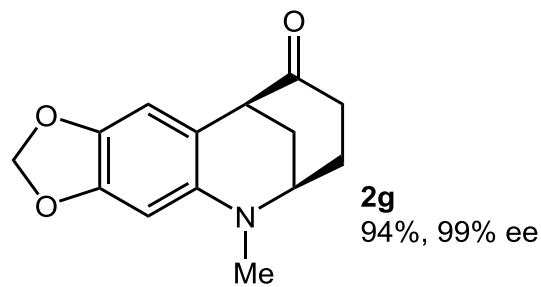
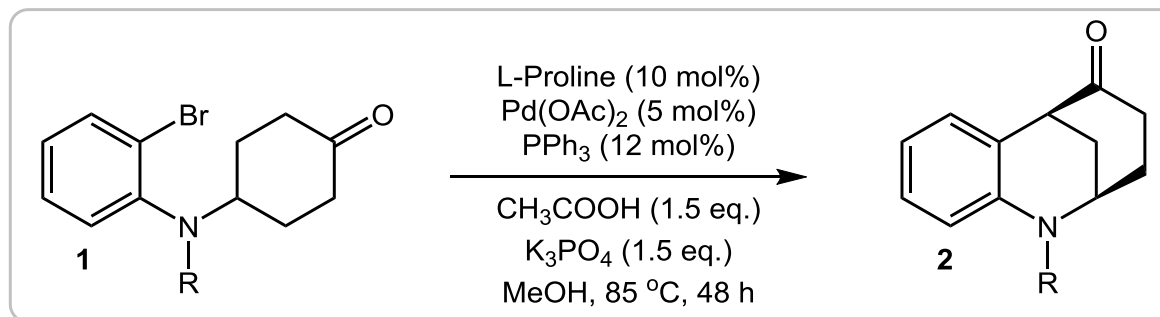
<sup>a</sup> Reaction conditions: **1a** (0.2 mmol), [Pd] (5 mol%), PPh<sub>3</sub> (12 mol%), L-proline (10 mol%), Base (0.3 mmol), Solvent (2.0 mL), 85 °C, 48 h. <sup>b</sup> Isolated yield. <sup>c</sup> Determined by HPLC. <sup>d</sup> 60 h.

# Substrate Scope

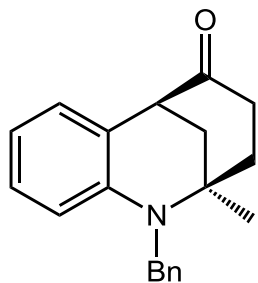
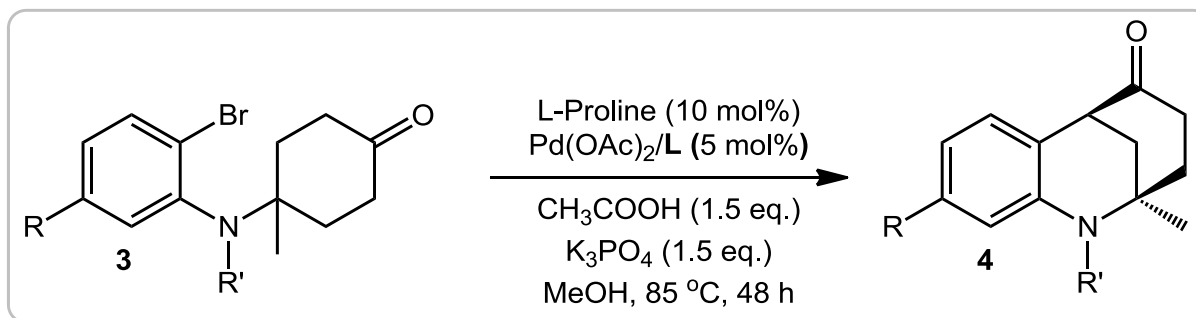


<sup>a</sup> With 12 mol% Cy<sub>3</sub>P·HBF<sub>4</sub> instead of PPh<sub>3</sub>.

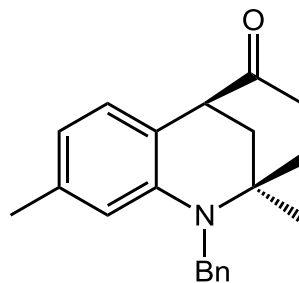
# Substrate Scope



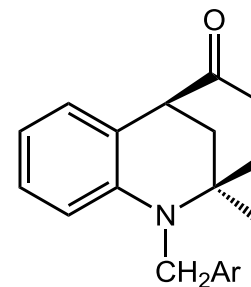
# Substrate Scope



**4a**  
L = PPh<sub>3</sub> (12 mol%)  
82%, 59% ee  
L = dppe (6 mol%)  
80%, 81% ee

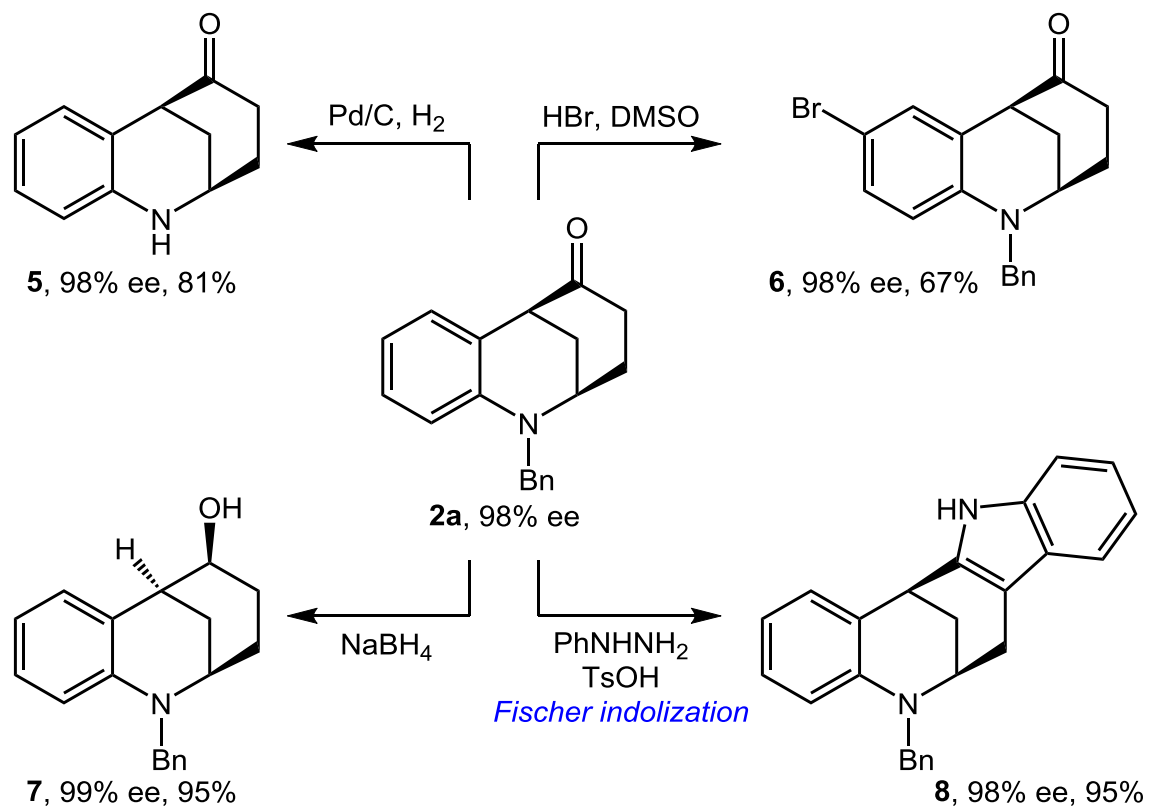


**4b**  
L = dppe (6 mol%)  
82%, 87% ee

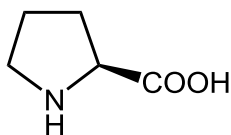
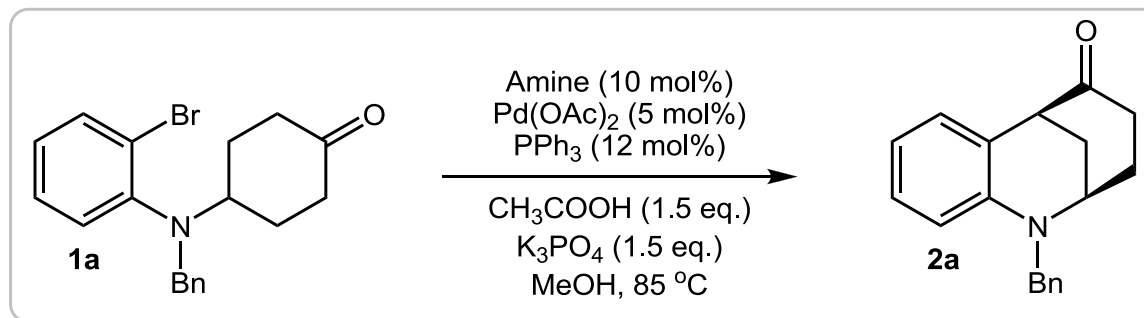


**4c**, Ar = 4-F-C<sub>6</sub>H<sub>4</sub>  
L = dppe (6 mol%)  
88%, 85% ee

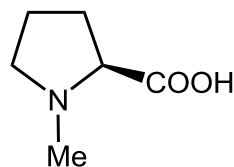
# Synthetic Transformations



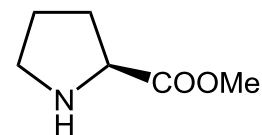
# Control Experiments



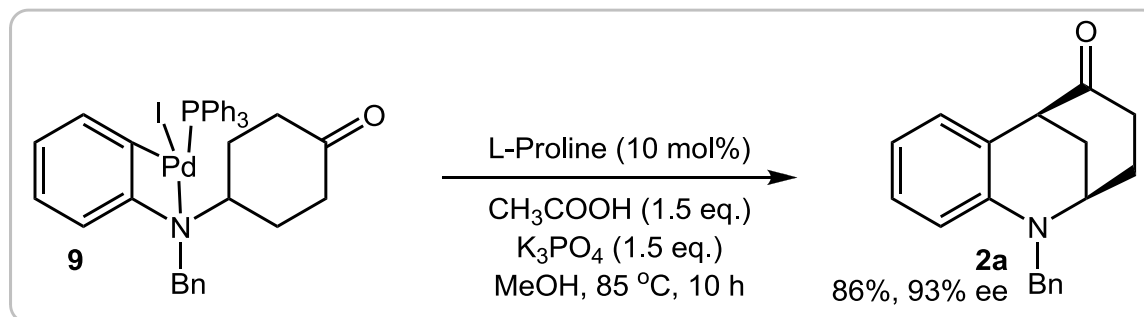
**A1**: 91%, 97% ee  
without **A1**: 92%, rac



**A2**: 89%, rac

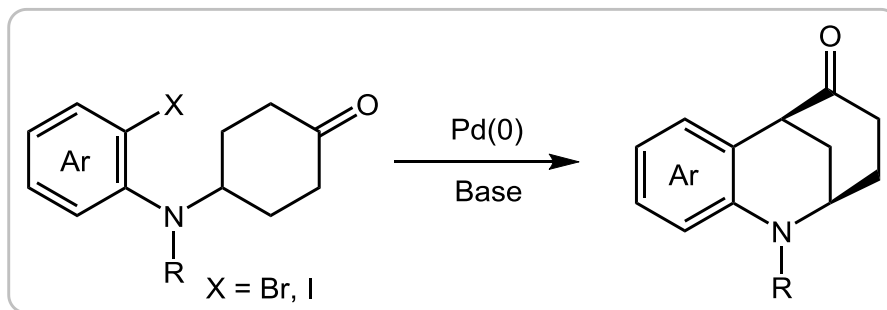


**A3**: 86%, 93% ee



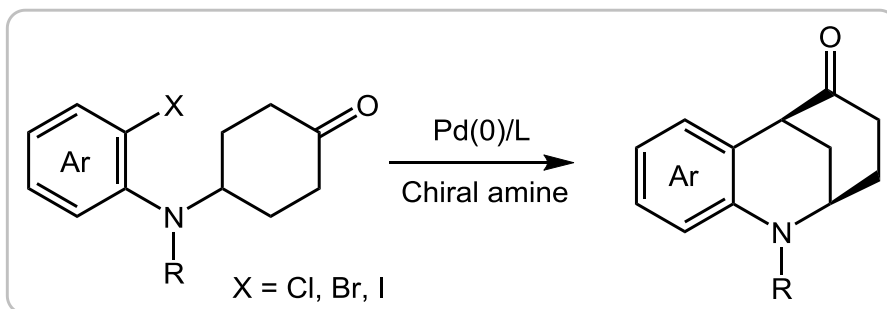
# Summary

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Transition-metal-catalyzed enantioselective  $\alpha$ -arylation of carbonyls represents one of the most important approaches to optically active  $\alpha$ -aryl carbonyl compounds. Since the seminal studies from the groups of Buchwald, Hartwig, and Miura, a range of carbonyl compounds have been investigated in enantioselective cross-coupling with aryl halides, including ketones, aldehydes, esters, and amides. However, the aforementioned approaches are very limited to the construction of quaternary carbon stereocenters because the benzylic  $\alpha$ -H atom of the tertiary carbon stereocenter is more acidic and prone to racemization under basic conditions.

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In addition, the demand of commercially unavailable chiral ligands still remains an issue for these transformations. It is noteworthy that a few other catalytic enantioselective arylation protocols to generate  $\alpha$ -carbonyl benzylic tertiary stereocenters have been developed, including (1) cross-coupling of  $\alpha$ -halo carbonyls with arylmetal reagents, (2) electrophilic coupling of diaryliodonium salts with silyl enolates, or aldehydes, and (3) cross-coupling of aryl triflates with silyl ketene acetals or tin enolates. In comparison, the direct coupling of readily available aryl halides with  $\alpha$ -C-H bonds of carbonyl moieties to form the tertiary carbon stereocenters is highly attractive from the viewpoint of atom economy and step efficiency.

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In conclusion, we have developed a new strategy for the enantioselective  $\alpha$ -arylation of cyclohexanones by using palladium/chiral amine catalysis. A series of optically active morphan derivatives bearing  $\alpha$ -carbonyl tertiary stereocenters were produced in good yields with excellent enantioselectivities. The present protocol offers new opportunities for the development of enantioselective  $\alpha$ -arylation of ketones.

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